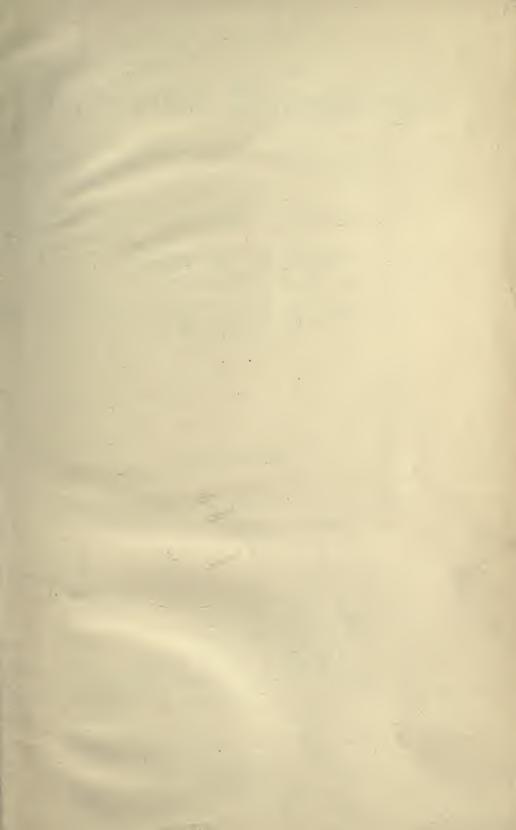




Digitized by the Internet Archive in 2007 with funding from Microsoft Corporation





INTERIOR BALLISTICS

RY

JAMES M. INGALLS Colonel United States Army, retired

Formerly Instructor of Ballistics at the U. S. Artillery School; Author of Treatises on Exterior and Interior Ballistics, Ballistic Machines, Ballistic Tables, Etc.

THIRD EDITION

NEW YORK **IOHN WILEY & SONS** LONDON: CHAPMAN & HALL, LIMITED 1912

VF820

COPVRIGHT 1912 BY JAMES M. INGALLS

PREFACE TO THE EDITION OF 1894

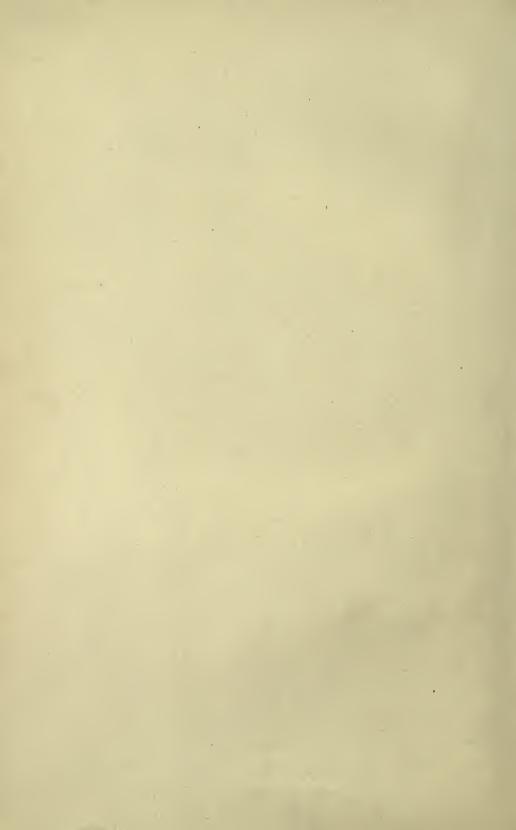
(SECOND EDITION)

When, in the summer of 1889, it was decided by the Staff of the Artillery School to add to the curriculum a course of interior ballistics, the instructor of ballistics, knowing of no text-book on the subject in the English language entirely suited to the needs of the school, employed the time at his disposal before the arrival of the next class of student officers in studying up and arranging a course of instruction upon this subject, so important to the artillery officer. The text-book then planned was partially completed and printed on the Artillery School press, and has been tested by two classes of student officers.

In the summer of 1893 the author again had leisure to work on the unfinished text-book, but in the meantime he had found so much of it which admitted of improvement that, with the encouragement of Lieutenant-Colonel Frank, Second Artillery, the Commandant of the School, it was decided to rewrite nearly the entire work as well as to complete it according to the original plan by the addition of the last two chapters.

With the exception of portions of Chapters IV and V, the author claims no originality. He has simply culled from various sources what seemed to him desirable in an elementary textbook, arranged it all systematically from the same point of view and with a uniform notation.

ARTILLERY SCHOOL, February 15, 1894.



PREFACE TO THE THIRD EDITION

The second edition of this work was used as a text-book at the Artillery School until the School suspended operations at the outbreak of war with Spain, in April, 1898. This edition, having become exhausted, the author has been induced by the request of officers for whose wishes he has great respect, to prepare a new edition embodying the results-of some investigations which were published in Volumes 24, 25 and 26 of the *Journal of the United States Artillery*, and which have been favorably received by Artillery Officers, both at home and abroad. The Journal articles have been rewritten, and many improvements attempted, suggested by friendly criticisms for which the author wishes to express his thanks.

As most of the formulas of interior ballistics in the present state of our knowledge of the subject are more or less empirical in their nature, many applications of the formulas deduced in Chapter IV are given in the following chapter to show their agreement with the results of actual firing with guns of widely different calibers. In this connection it is gratifying to be able to quote from an article published in the *Journal of the Royal Artillery*, vol. 36, No. 9, by Captain J. H. Hardcastle, R.A., who states, with reference to the formulas of Chapter IV as applied to firing-practice with English guns loaded with cordite, that "After many dozens of calculations I can find no serious disagreement between the results of calculation and experiment."

vi PREFACE

In this paper Captain Hardcastle has very ingeniously adapted the formulas of Chapter IV to slide-rule operations, thereby lessening the labor of calculation somewhat, though at the expense of accuracy in some cases. For the benefit of those who are accustomed to use the slide rule in making logarithmic computations a supplemental table of the X functions has been added to Table I, omitting the function X_5 , which is not used in Captain Hardcastle's method.

This work was prepared primarily for the officers of our Coast Artillery Corps; but it is hoped that gun-designers and powder-manufacturers may find in it something useful to them.

The author desires to express his indebtedness to Lieut.-Colonel Ormand M. Lissak and Major Edward P. O'Hern of the Ordnance Department for valuable suggestions and for data employed in the "applications." Also to Captains Ennis and Bryant, for assistance in computing Table I.

PROVIDENCE, R. I., September 20, 1911.

TABLE OF CONTENTS

CHAPTER I

PAGE

Definition and object.—Early history of gunpowder. Robins' experiments and deductions. Hutton's experiments. D'Arcy's method. Rumford's experiments with fired gunpowder. Rodman's inventions and experiments. Modern explosives. Density of powder. Inflammation and combustion of a grain of powder. Inflammation and combustion of a charge of powder.

CHAPTER II

Properties of perfect gases.—Marriotte's law. Specific volume. Specific weight. Law of Gay-Lussac. Characteristic equation of the gaseous state. Thermal units. Mechanical equivalent of heat. Specific heat. Specific heat of a gas under constant pressure. Specific heat under constant volume. Numerical value of R for atmospheric air. Law of Dulong and Petit. Determination of specific heats. Ratio of specific heats. Relations between heat and work in the expansion of a perfect gas. Isothermal expansion. Adiabatic expansion. Law of temperatures. Law of pressures and volumes. Examples. Theoretical work of an adiabatic expansion in the bore of a gun. Noble and Abel's researches on fired gunpowder in close vessels. Description of apparatus employed. Summary of results. Pressure in close vessels deduced from theoretical considerations. Value of the ratio of the non-gaseous products to the volume of the charge. Determination of the force of the powder, and its interpretation. Theoretical determination of the temperature of explosion of gunpowder. Mean specific heat of the products of combustion. Pressure in the bore of a gun derived from theoretical considerations. Table of pressures. Theoretical work effected by gunpowder. Factor of effect. Actual work realized as expressed by muzzle energy.

CHAPTER III

Combustion of a grain of powder under constant atmospheric pressure.—

Notation. Definition of the vanishing surface. General expression for the burning surface of a grain of powder. Expression for the

PAGE

volume consumed in terms of the thickness burned. Definition of the form characteristics. Fraction of grain burned. Applications. Spheres. Cubes. Strips. Solid cylinders. Pierced cylinders. Multiperforated grains. General expression for surface of combustion of multiperforated grains. Maximum surface of combustion. Slivers. Expression for volume of slivers. Proposed ratio of dimensions of multiperforated grains to web thickness. Expression for weight of charge burned at any instant. Expressions for initial volume and surface of combustion of a charge of powder. Expression for specific gravity of grain. Initial surface of unit weight of powder. Volume of charge. Gravimetric density. Density of loading. Reduced length of initial air space. Working formulas for English and metric units. Examples.

CHAPTER IV

Combustion and work of a charge of powder in a gun.-Introductory remarks. Sarrau's law of burning under a variable pressure and reason for adopting it. Expression connecting the velocity of burning of grain with velocity of projectile in bore. Expression for fraction of charge burned in terms of volumes of expansion of the gases generated. Expression for velocity of projectile while powder is burning. Velocity of projectile after powder is all burned. Pressure on base of projectile while powder is burning. Pressure after powder is burned. Expression for the initial pressure upon the supposition that the powder was all burned before the projectile had moved from its seat, and the relation of this pressure to the force of the powder. Method of computing the X functions. Special formulas. Expressions for maximum pressure. Formula for velocity of combustion under atmospheric pressure. Working formulas. English units. Metric units. Characteristics of a powder. Expressions for constants in terms of the characteristics for English and metric units. Expressions for force of powder when weights of charge and projectile vary. 79 to 97

CHAPTER V

Applications.—Formulas which apply only while powder is burning. Formulas which apply only after powder is all burned. Formulas which apply at instant of complete combustion. Discontinuity of pressure curve for certain forms of grain. Monomial formulas for velocity and pressure. Typical pressure and velocity curves. Example of

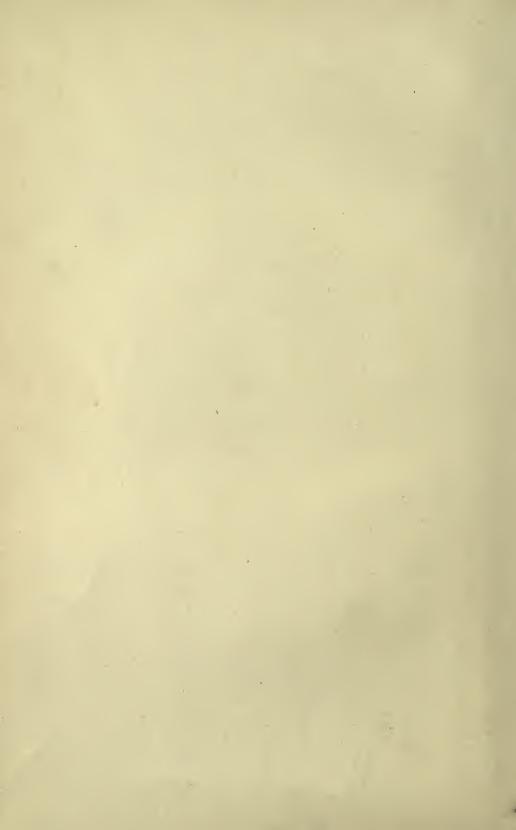
PAGE

monomial formulas, as applied to the 8-inch B. L. R. Comparison of computed velocities and maximum pressures with observed values. Determination of travel of projectile at point of maximum pressure and also when powder is all burned. Expression for fraction of charge burned for any travel of projectile. Examples, Greatest efficiency when charge is all consumed at muzzle. Application to hypothetical 7-inch gun. Binomial formulas for velocity and pressure. Forms of grain for which binomial formulas must be employed Methods for determining the constants from experimental firing. Applications to Sir Andrew Noble's experiments with a 6-inch gun. Description of the experiments. Discussion of the data for cordite. 0.4", 0.35", and 0.3" diameter. Remarks on the so-called "force of the powder" as deduced from the calculations. Examples. Application to the Hotchkiss 57-mm, rapid-firing gun. Data obtained by D'Arcy's method. New method for determining the form characteristics of the grains. Application to the magazine rifle, model of 1003. Powder characteristics. Formulas for designing guns for cordite, with application to a hypothetical 7-inch gun. Trinomial formulas. Grains for which these formulas are necessary. Spherical and cubical grains. Formulas for computing the constants. Application to Noble's experiments with ballistite in a 6-inch gun. Table of computed velocities and pressures. Remarks on the velocity and pressure curves. Examples. Multiperforated grains. Special formulas required for these grains. Discussion of the data obtained by the Ordnance Board with the 6-inch Brown wire gun. Remarks on the discontinuity of the pressure curves. Examples. Superiority of uniperforated to multiperforated grains. Application to the 14-inch rifle. Effect of increasing the volume of the chamber upon the maximum pressure. Better results can be obtained by lengthening the powder grains. Table of pressures. . . . 98 to 169

CHAPTER VI

On the rifling of cannon.—Advantages of rifling. The developed groove.
Uniform twist. Increasing twist. General expression for pressure
on the lands. Angular acceleration. Pressure for uniform twist.
Increasing twist. Semi-cubical parabola. Common parabola. Rel-
ative width of grooves and lands. Application to the 10-inch
B. L. R., model of 1888. Application to the 14-inch gun. Retarding
effect of a uniform twist of one turn in twenty-five calibers 170 to 186

Tables





INTERIOR BALLISTICS

CHAPTER I

INTRODUCTION

Definition and Object.—Interior ballistics treats of the formation, temperature and volume of the gases into which the powder charge, in the chamber of a gun, is converted by combustion, and the work performed by the expansion of these gases upon the gun, carriage and projectile. Its object is the deduction and discussion of rules and formulas for calculating the velocity, both of translation and of rotation, which the gases of a given weight of powder of known composition and quality are able to impart to a projectile and their reaction upon the gun and carriage. The discussion of the formulas deduced will bring out many important questions, such as the proper relation of weight of charge to weight of projectile and length of bore, the best size and shape of the powder grains for different guns and their effect upon the maximum and muzzle pressures, the velocity of recoil, etc. The most approved formulas for calculating the pressures upon the surface of the bore will be given; but the methods which have been devised for building up the gun, so as best to resist these pressures, will not be entered upon here as their consideration belongs to another branch of the subject.

Early History of Interior Ballistics.—For more than five hundred years gunpowder—an intimate mixture of nitre, sulphur and charcoal,—was used almost exclusively as the pro-

. .

pelling agent in firearms; and though it has been entirely superseded within the last quarter of a century by gun-cotton. nitro-glycerine, and their various compounds, vet it possessed many admirable qualities which the modern powders do not as yet so fully enjoy. It ignited easily without deflagration: its effects were regular and sure; its manufacture was economical. rapid and comparatively safe; it produced but little erosion in the bore. Finally, it kept well in transportation, and indefinitely in properly ventilated magazines. It is on record that experiments made with gunpowder, manufactured more than two centuries before, showed that it had lost none of its ballistic qualities. The principal objection to gunpowder, as compared to nitrocellulose powders, are the dense volumes of smoke accompanying its explosion, the fouling of the bore. and the comparatively large charges required to give the desired muzzle velocity, necessitating an abnormal enlargement of the powder chamber or an impracticable lengthening of the gun.

Robins' Experiments and Deductions.—The celebrated Benjamin Robins seems to have been the first investigator who had a tolerably correct idea of the circumstances relating to the action and force of fired gunpowder. In a paper which was read before the Royal Society in 1743 entitled, "New principles of gunnery." Robins described among other things some experiments he had made for determining the velocities of musket balls when fired with given charges of powder. These velocities were measured by means of the ballistic pendulum invented by Robins, "the idea of which is simply that the ball is discharged into a very large but movable block of wood, whose small velocity, in consequence of that blow, can be easily observed and accurately measured. Then, from this small velocity thus obtained, the large one of the ball is immediately derived from this simple proportion, viz., as the weight of the ball is to the sum of the weights of the ball and the block, so is the observed velocity of the last to a fourth proportional, which is the velocity of the ball sought."*

The deductions which Robins makes from these experiments, so far as they relate to interior ballistics, may be summarized as follows:

- (1) Gunpowder fired either in a vacuum or in air produces, by its combustion, a permanent elastic fluid or air.
- (2) The pressure exerted by this fluid is, cæteris paribus, directly as its density.
- (3) The elasticity of the fluid is increased by the heat it has at the time of explosion.
- (4) The temperature of the fluid at the moment of combustion is at least equal to that of red-hot iron.
- (5) The maximum pressure exerted by the fluid is equal to about 1,000 atmospheres.
- (6) The weight of the permanent elastic fluid disengaged by the combustion is about three-tenths that of the powder, and its volume at ordinary atmospheric temperature and pressure is about 240 times that occupied by the charge.

These deductions, considering the extremely erroneous and often absurd opinions that were entertained by those who thought upon the subject at all in Robins' time—and even down to the close of the century—show that Robins is well entitled to be called the "father of modern gunnery."

Hutton's Experiments.—Dr. Charles Hutton, professor of mathematics in the Royal Military Academy, Woolwich, continued Robins' experiments at intervals from 1773 to 1791. He improved and greatly enlarged the ballistic pendulum so that it could receive the impact of 1-pound balls, whereas that used by Robins was adapted—for musket balls only. Hutton's experiments are given in detail in his thirty-fourth, thirty-fifth, thirty-sixth, and thirty-seventh tracts. They verify most of

^{*} Hutton's "Mathematical Tracts," vol. 3, p. 210 (Tract 37), London, 1812.

Robins' deductions, but with regard to Robins' estimate of the temperature of combustion and the maximum pressure Hutton says: "This was merely guessing at the degree of heat in the inflamed fluid, and, consequently, of its first strength, both which in fact are found to be much greater."* His own estimate of the temperature is double that of Robins, and he places the maximum pressure of fired gunpowder at 2,000 atmospheres. Hutton gives a formula for the velocity of a spherical projectile at any point of the bore, upon the assumption that the combustion of the charge is instantaneous and that the expansion of the gas follows Mariotte's law—no account being taken of the loss of heat due to work performed—a principle which at that time was unknown.

D'Arcy's Method.—In 1760 the chevalier D'Arcy sought to determine the law of pressure of the gas in the bore of a musket by measuring the velocity of the projectile at different points of the bore. This he accomplished by successively shortening the length of the barrel and measuring for each length the velocity of the bullet by means of a ballistic pendulum. Having obtained from these experiments the velocities of the bullets for several different lengths of travel, the corresponding accelerations could be calculated, and then the pressures, by multiplying the accelerations by the mass. This was the first attempt to determine the law of pressures dynamically.

Rumford's Experiments with Fired Gunpowder.—The first attempt to measure directly the pressure of fired gunpowder was made, in 1792, by our countryman, the celebrated Count Rumford. A most interesting account of his experiments is given in his memoir entitled "Experiments to determine the force of fired gunpowder," † which must be regarded as the most important contribution to interior ballistics which had been

^{*} Tracts, vol. 3, p. 211.

[†] Philosophical Transactions, London, 1797, p. 222; also "The Complete Works of Count Rumford," Boston, 1870, vol. 1, p. 98.

made up to that time. The apparatus used by Rumford consisted of a small and very strong wrought-iron mortar (or eprouvette), which rested with its axis vertical upon a solid stone foundation. This mortar (or barrel, as Rumford calls it), was 2.78 inches long and 2.82 inches in diameter at its lower extremity and tapered slightly toward the muzzle. The bore (or chamber) was cylindrical, one-fourth of an inch in diameter and 2.13 inches deep. At the centre of the bottom of the barrel there was a projection 0.45 inch in diameter and 1.3 inches long, having an axial bore 0.07 inch in diameter connecting with the chamber above, but closed below, forming a sort of vent, but having no opening outside.

By this arrangement the charge could be fired without any loss of gas through the vent by the application of a red-hot ball provided with a hole, into which the projecting vent-tube could be inserted, which latter would thus become in a short time sufficiently heated to ignite the powder. The upper part of the bore or muzzle was closed by a stopper made of compact. well-greased sole leather, which was forced into the bore, until its upper surface was flush with the face of the mortar, and upon this was placed the plane surface of a solid hemisphere of hardened steel, whose diameter was 1.16 inches. "Upon this hemisphere the weight made use of for confining the elastic fluid generated from the powder in its combustion reposed. This weight in all the experiments, except those which were made with very small charges of powder, was a piece of ordnance of greater or less dimensions or greater or less weight, according to the force of the charge, placed vertically upon its cascabel upon the steel hemisphere which closed the end of the barrel; and the same piece of ordnance, by having its bore filled by a greater or smaller number of bullets, as the occasion required, was made to serve for several experiments." *

^{*} Rumford's Works, vol. 1, p. 121.

As one of the objects of Rumford's experiments was to determine the relation between the pressure of the powder gases and their density, he varied the charge, beginning with I grain. and for each charge placed a weight, which he judged was about equivalent to the resulting pressure, upon the hemisphere. on firing, the weight was lifted sufficiently to allow the gases to escape, it was increased for another equal charge; and this was repeated until a weight was found just sufficient to retain the gaseous products—that is, so that the leathern stopper would not be thrown out of the bore, but only slightly lifted. density of the powder gases could easily be determined by comparing the weight of the charge with the weight of powder required to completely fill the chamber and vent, which latter was about 25½ grains troy. Rumford increased the charges a grain at a time from 1 grain to 18 grains, and from a mean of all the observed pressures he deduced the empirical formula,

$$p = 1.841x^{1 + .0004x}$$

in which p is the pressure in atmospheres and x the density of loading to a scale of 1000—that is, for a full chamber x = 1000; for one-half full x = 500, and so on. This formula gives 29,178 atmospheres for the maximum pressure—that is, when the powder entirely fills the space in which it is fired. In this case the value of x is 1000, and Rumford's pressure formula becomes

$$p = 1.841 \times 1000^{1.4} = 29178$$

Nearly a century later Noble and Abel (see Chapter II) found by their experiments, which are entirely similar in character to those of Rumford, that the maximum pressure of fired gunpowder is but 6,554 atmospheres, or 43 tons per square inch; and this result has been accepted by all writers on interior

ballistics as being very near the truth. Their formula for the pressure in terms of Rumford's x is

$$p = \frac{2.818x}{1 - 0.00057x}$$

in which p and x are defined as before. If in this formula we make x = 1000, we have, as already stated,

$$p = \frac{2.818 \times 1000}{1 - 0.57} = 6554$$

For small densities of loading, Noble and Abel's formula gives greater pressures than Rumford's principally because the powder used by the later investigators was the stronger; but as the densities increase this is reversed. With a charge of 18 grains, for which x = 702, Noble and Abel's formula gives a pressure of 3,208 atmospheres, while Rumford's gives 8,140 atmospheres. To enable us to understand the cause of this great difference in the results obtained by these eminent savants (which is very instructive), we will go a little into detail. experiments were made by Rumford with a charge of 18 grains of powder. In the first of these a 24-pounder gun, weighing 8,081 pounds, was placed vertically on its cascabel upon the steel hemisphere closing the muzzle of the barrel. When the charge was fired "the weight was raised with a very sharp report, louder than that of a well-loaded musket." The barrel was again loaded with 18 grains as before, and enough shot were placed in the bore of the 24-pounder gun to increase its weight to 8,700 pounds. Upon firing the powder by applying the red-hot ball "the vent-tube of the barrel was burst, the explosion being attended with a very loud report." These experiments were the eighty-fourth and eighty-fifth, and closed the series.

In the eighty-fourth experiment a weight of 8,081 pounds was actually raised by the explosion of 18 grains of powder (about one-fourth the service charge of the Springfield rifle),

acting upon a circular area one-quarter of an inch in diameter. To raise this weight under the circumstances would require a pressure of more than 11.200 atmospheres, while, as we have seen, the actual pressure due to this density of loading, according to Noble and Abel's formula, is but 3,208 atmospheres. Evidently then the weight in this experiment was not raised by mere pressure: but we must attribute a great part of the observed effect (in consequence of the position of the charge at the bottom of the bore) to the energy with which the products of combustion impinged against the leathern stopper, which had only to be raised 0.13 inch (the thickness of the leather) to allow the gases to escape. In Noble and Abel's experiments there was no such blow from the products of combustion because the apparatus for determining the pressure (crusher gauge) was placed within the charge. Had the leathern stopper in Rumford's experiments been a little longer, it is probable that his conclusions would have been entirely different.

Rodman's Inventions and Experiments.—We have space only to mention the names of Gay-Lussac, Chevreul, Graham, Piobert, Cavalli, Mayevski, Otto, Neumann, and others, who did original work, of more or less value, for the science of interior ballistics prior to the year 1860. We will, however, dwell a few moments on the important work done by Captain (afterwards General) T. J. Rodman, of our own Ordnance Department, between the years 1857 and 1861.* The objects of Rodman's experiments were: First, to ascertain the pressure exerted upon different points of the bore of a 42-pounder gun in firing under various circumstances. Second, to determine the pressures in the 7-inch, 9-inch, and 11-inch guns when the charges of powder and the weight of projectiles were so proportioned that there should be the same weight of powder behind, and

^{* &}quot;Experiments on Metal and Cannon and Qualities of Cannon Powder," by Captain T. J. Rodman, Boston, 1861.

the same weight of metal in front of each square inch of area of cross-section of the bore. Third, to determine the differences in pressure and muzzle velocity due to the variations in the size of the powder grains; and, fourth, to determine the pressures exerted by gunpowder burned in a close vessel for different densities of loading.

For the purpose of carrying out these experiments Rodman, instead of using the system of varying weights employed by Rumford, invented what he called the "indenting apparatus," which has since been extensively used, not only in this country but in all foreign countries, under the name of Rodman's pressure (or cutter) gauge; and which is too well known to require a description.

The maximum pressure of gunpowder when exploded in its own space, as determined by Rodman by the bursting of shells filled with powder, ranged from 4,000 to 12,600 atmospheres: the mean of all the experiments giving 8,070 atmospheres, or 53 tons per square inch. These results, though much nearer the truth than those deduced by Rumford, are still about 25 per cent. greater than Noble and Abel's deductions; and this is undoubtedly due to the position of the pressure gauge, which was placed near the exterior surface of the shell, so that when the products of combustion had reached the gauge they had acquired a considerable energy which probably exaggerated the real pressure. The same causes, it will be remembered, vitiated Rumford's experiments. In both cases it was as if a charge of small shot had been fired with great velocity against the leathern stopper in the one case, or the end of the piston of the indenting tool in the other.

General Rodman was the first person to suggest the proper shape for powder grains, in order to diminish the initial velocity of emission of gas and to more nearly equalize the pressure in the bore of the gun. For this purpose he employed what he termed a "perforated cake cartridge" composed of disks of

compressed powder from 1 to 2 inches thick and of a diameter to fit the hore. Rodman demonstrated that such a form of cartridge would relieve the initial strain by exposing a minimum surface at the beginning of combustion, while a greater volume of gas would be evolved from the increasing surfaces of the cylindrical perforations as the space behind the projectile became greater; and this would tend to distribute the pressure more uniformly along the bore. Rodman's experiments with this powder in the 15-inch cast-iron gun which he had recently constructed for the government—and which is without doubt the most effective and the best smooth-bore gun ever made—fully confirmed his theory; but for many reasons he found it more convenient and equally satisfactory to build up the charge by layers of pierced hexagonal prisms about an inch in diameter fitting closely to one another, instead of having them of the diameter of the bore.

The war of the rebellion which was inaugurated while General Rodman was in the midst of his discoveries and inventions, put an end forever to his investigations, but his ideas were speedily adopted in Europe, and his "prismatic powders," but slightly modified, are extensively used.

Modern Explosives.—Gun-cotton, made by immersing cleaned and dried cotton waste in a mixture of strong nitric and sulphuric acids, was discovered by Schönbein of Basel, in 1846, who immediately proposed to employ it as a substitute for gunpowder. General von Lenk made many experiments with gun-cotton by compressing it into cubes or cylinders, with the idea of employing it for artillery use. But all his efforts failed from the fact that, no matter how much it was compressed, it was still mechanically porous; and when ignited in a gun the flame and hot gases speedily penetrated the mass causing it to detonate, or, at least, to approach dangerously near to detonation. It was not until the discovery in the early eighties that gun-cotton could be dissolved or made into a paste, or colloid,

3

by acetone and other so-called solvents, that it was possible to employ it as a propellant. In this condition, when moulded into grains and thoroughly dried, it loses its mechanical porosity and burns from the surface in parallel layers, the grain retaining its original form until completely consumed.

Gun-cotton is mixed in certain proportions with nitroglycerine to form nearly all the powders employed for war purposes. For example, the powder used in the British army and navy (called cordite), consists of 65 per cent. of gun-cotton, 30 per cent. of nitro-glycerine and 5 per cent. of mineral jelly or vaseline, this latter being used as a preservative. This is also very nearly the composition of the powder used in the United States army and navy. For a full account of the properties, manufacture and uses of gun-cotton and nitroglycerine, the reader is referred to General Weaver's "Notes on Explosives."

Density of Powder.—By density of a powder is meant its specific gravity, or the ratio of the weight of a given volume of the powder to the weight of an equal volume of water at the standard temperature. It is sometimes referred to as mercurial density, since it may be determined by an apparatus which utilizes the property of mercury of filling the interstices between the grains without penetrating into the pores or uniting chemically with the powder. The density varies somewhat according to the pressure to which the grains were subjected during the manufacture and ranges from about 1.56 to 1.65.

Inflammation and Combustion of a Grain of Powder.—Inflammation is the spreading of the flame over the free surface of the grain from the point of ignition. Combustion is the propagation of the burning into the interior of the grain. Ignition is produced by the sudden elevation of the temperature of a small portion of the grain to about 180° C. (in the case of cordite) either by contact with an ignited body, by mechanical shock or friction, or by detonation of a fulminate. The velocity

of inflammation depends upon the nature of the source of heat which ignites it, upon the state of the surface of the grain and upon its density and dryness. The combustion of a grain takes place in successive concentric layers, and in free air equal thicknesses are burned in equal times. As the mass of gas disengaged in any given time is proportional to the quantity of powder burned during the same time, and, therefore, proportional to the surface of inflammation, it follows that the emission of gas is largely influenced by the form of the grain. For example, if the grain is spherical the surface of inflammation decreases rapidly up to the end of its burning where it is zero. On the other hand, the surface of inflammation (or of combustion) of a multi-perforated grain increases until it is nearly consumed.

Inflammation and Combustion of a Charge of Powder.— The inflammation of a charge of powder involves the transmission of the flame from one grain to another. Its velocity depends not only upon the inflammability of the grain but also upon the facility with which the gases first generated are able to penetrate the charge. This is assisted by a proper arrangement of the grains composing the charge and also by placing an igniter of fine rifle powder at each end of the cartridge. The combustion of a charge composed of grains of the same form and dimensions should, from what has been said, practically terminate at the same time with each or any grain; and, therefore, the time of combustion of a charge increases with the size of the grains, and is in all cases with service powders much longer than the time of inflammation.

If a charge of powder be confined in a close vessel and ignited, its combustion takes place silently, and permanent gases and a certain amount of solid matter are produced which can be collected for analysis by opening the vessel, as in the experiments of Noble and Abel described in Chapter II. In this case no work is performed by the gases, and the accompanying phe-

nomena are comparatively simple. But if the combustion takes place in a chamber of which one of the walls is capable of moving under the tension of the gases, which condition is realized in cannon, the resulting phenomena are much more complicated, as a little consideration will show.

When the charge of powder in the chamber of a loaded gun is ignited at both ends of the cartridge, all the grains will be inflamed practically simultaneously. The first gaseous products formed will expand into the air-spaces of the chamber and almost immediately acquire a tension sufficient to start and overcome the forcing of the projectile. This latter once in motion will encounter no resistances in the bore comparable with those which opposed its start, and its velocity will rapidly increase under the continued action of the pressure of the gases. This pressure will also increase at first: for, though the displacement of the projectile gives a greater space for the expanding gases, this is more than compensated for by a more abundant disengagement of gas. But the pressure soon reaches its maximum: for if, on the one hand, the disengagement of gas is accelerated by the increase of pressure, on the other hand the increasing velocity of the projectile offers more and more space for the gases to expand in. The velocity itself would soon reach a maximum if the bore were sufficiently prolonged; for in addition to the friction and the resistance of the air, both of which retard the motion of the projectile, the propulsive force decreases by the expansion and cooling of the gases. Therefore the retarding forces will in time predominate and the projectile be brought to rest. Its velocity starting from zero passes to its maximum and if the gun terminated at this point the projectile would leave the bore with the greatest velocity the charge was capable of communicating to it.

So far only charges in general have been considered. Take, now, a charge composed of small grains of slight density. The initial surface of inflammation will be very great and the emission

of gas correspondingly abundant. The pressure will increase rapidly, and, in consequence, the velocity of combustion. It results from this that the grains will be consumed nearly as soon as inflamed, and this before the projectile has had time to be displaced by a sensible amount. Hence all the gases of the charge, disengaged almost instantaneously, will be confined an instant within the chamber; their tension will be very great, and they will exert upon the walls of the gun a sudden and violent force which may rupture the metal, and which in all cases will produce upon the gun and carriage shocks which are destructive to the system and prejudicial to accuracy of fire. On the other hand the projectile will be thrown quickly forward, as by a blow from a hammer.

If, on the contrary, the charge is made up of large grains of great density, the total weight of gas emitted will be the same as before; but the mode of emission will be different. The initial surface of inflammation will be less, and the initial tension of the gas not so great. The combustion will take place more slowly, and will be only partially completed when the projectile shall have begun to move. The pressure of the gases will attain a maximum less than in the preceding case, but the pressure will decrease more slowly. Under the continued action of this pressure, the velocity of the projectile will be rapidly accelerated and at the muzzle will differ but little from that obtained by the fine powder, without producing upon the gun and carriage the destructive effects mentioned above.

CHAPTER II

PROPERTIES OF PERFECT GASES

Mariotte's Law.—When a mass of gas is subjected to pressure the volume diminishes until the increased tension just balances the pressure; and it was found by experiment that if the temperature of the gas remains constant, the tension, or pressure, is inversely proportional to the volume. Thus, if v_1 and v_2 represent different volumes of the same mass of gas and p_1 and p_2 the corresponding tensions, or pressures, then if the temperature is the same for both volumes we have the proportion:

$$v_1:v_2=rac{\mathrm{I}}{p_1}:rac{\mathrm{I}}{p_2}$$

Hence

$$v_1 p_1 = v_2 p_2 = \text{constant.}$$

That is, for every mass of gas at invariable temperature the product of the volume and tension is constant. This law is generally called Mariotte's law, though it was first discovered by the English chemist Robert Boyle, in 1662, and verified by Mariotte in 1679.

Specific Volume.—The specific volume of a gas is the volume of unit weight at zero temperature and under the normal atmospheric pressure. Designate the specific volume by v_o and the normal atmospheric pressure by p_o . Then we have by Mariotte's law

$$v p = v_o p_o$$
.

Specific Weight.—The specific weight of a gas is the weight of unit volume at zero temperature and under the pressure p_o . It is therefore the reciprocal of the specific volume v_o .

Law of Gay-Lussac.—The coefficient of expansion of a gas is the same for all gases, and is sensibly constant for all temperatures and pressures. Let, as before, v_o be the specific volume, v_t the volume at temperature t and α the coefficient of expansion. Then the variation of volume by Gay-Lussac's law will be expressed by the equation

$$v_t - v_o = \alpha t v_o;$$

whence

$$v_t = v_o (\mathbf{1} + \alpha t)$$

The value of the coefficient α is approximately $\frac{1}{273}$ for each degree centigrade. The last equation may, therefore, be written

$$v_t = v_o \left(\mathbf{1} + \frac{t}{273} \right)$$

Characteristic Equation of the Gaseous State.—The last equation, which expresses Gay-Lussac's law, may be combined with Mariotte's law, introducing the pressure p. The problem may be enunciated as follows: Having given the specific volume of a gas v_o to determine its volume v_t at a temperature t under the corresponding pressure p_t .

Let x be the volume v_t would become at \circ° C., under the pressure p_t . Then by Gay-Lussac's law

$$v_t = x (1 + \alpha t)$$

and by Mariotte's law

$$p_t x = p_o v_o;$$

whence eliminating x,

$$p_t v_t = p_o v_o (1 + \alpha t) = \frac{p_o v_o}{273} (273 + t)$$

Since $\frac{p_o v_o}{273}$ is constant, put

$$R = \frac{p_o v_o}{273};$$

whence

$$p_t v_t = R(273 + t);$$

or, dropping the subscripts as no longer necessary,

$$p v = R (273 + t)$$

The temperature (273 + t) is called the absolute temperature, and is reckoned from a zero placed 273 degrees below the zero of the centigrade scale. Calling the absolute temperature T there results finally

$$p v = R T \qquad . \qquad . \qquad . \qquad . \qquad (1)$$

which is called the characteristic equation of the gaseous state. It is simply another expression of Mariotte's law in which the temperature of the gas is introduced.

Equation (1) expresses the relation existing between the pressure, volume and absolute temperature of a unit weight of gas. For any number y units of weight occupying the same volume v the relation evidently becomes

A gas supposed to obey exactly the law expressed in equation (1) is called a perfect gas, or is said to be theoretically in the perfectly gaseous state. This condition represents an ideal toward which gases approach more nearly as their state of rarefaction increases. Of all gases, hydrogen approximates most closely to such an hypothetical substance, though at ordinary temperatures the simple gases, nitrogen, oxygen and atmospheric air, may for most practical purposes be considered perfect gases.

Thermal Unit.—The heat required to raise the temperature of unit weight of water at the freezing point one degree of the thermometer is called a thermal unit. There are two thermal units in general use, namely: the British thermal unit (B. T. U.), which is the heat required to raise the temperature of one pound

2

of water from 32° F. to 33° F.; and the French thermal unit (called a calorie), which is the heat required to raise the temperature of one kilogram of water from o° C. to 1° C. There is still another thermal unit of frequent use, namely: the heat required to raise the temperature of one pound of water from o° C. to 1° C., and which may be designated as the pound-centigrade (P. C.) unit.

Mechanical Equivalent of Heat.—The mechanical equivalent of heat is the work equivalent of a thermal unit, and will be designated by E. According to Rowland the value of E is 778 foot-pounds for a B. T. U. Since a degree of the centigrade scale is $\frac{9}{5}$ of a degree of the Fahrenheit scale, we have

for a P. C. thermal unit, $E = \frac{9}{5} \times 778 = 1400.4$ foot-pounds. Also since there are 3.280869 feet in a metre, the value of E for a calorie is

$$\frac{1400.4}{3.280869}$$
 = 426.84 kilogram-metres.

Specific Heat.—The quantity of heat, expressed in thermal units, which must be imparted to a unit weight of any substance to increase its temperature one degree of the thermometer, or the quantity of heat given up by the substance while its temperature is lowered one degree, is called its specific heat. The specific heat of different substances varies greatly. Thus, if a pound of mercury and a pound of water receive the same quantity of heat the temperature of the former will be much greater that the latter. Indeed, it requires about 32 times as much heat to raise the temperature of water 1° as it does to raise the temperature of mercury by the same amount.

The heat imparted to a substance is expended in three different ways: 1. Increasing the temperature, which may be called vibration work; 2. In doing internal or disgregation work; 3. In doing external work by expansion. If it were

possible to eliminate the two latter, we should get the true specific heat, or the heat necessary to increase the temperature simply. For a perfect gas, however, the disgregation work is zero, and for all substances the disgregation work is small in comparison with the vibration work. The specific heat of a gas may be determined in two different ways, giving results which are of fundamental importance in thermodynamics, namely: Specific heat under constant pressure, and specific heat under constant volume.

Specific Heat of a Gas Under Constant Pressure.—To fix the ideas suppose a unit weight of gas to be confined in a spherical envelope capable of expanding without the expenditure of work and which allows no heat the gas may have to escape, and to be in equilibrium with the constant pressure of the atmosphere. Under these conditions let a certain quantity of heat be applied to the gas just sufficient to raise its temperature one degree of the thermometer after it has expanded until equilibrium is again restored. This quantity of heat, in thermal units (designated by C_p), is called specific heat under constant pressure.

Specific Heat Under Constant Volume.—Next repeat the experiment just described, but replacing the elastic envelope, which by hypothesis permitted the gas to expand freely, by a rigid envelope, thus keeping the volume of the gas constant while heat is applied. It will now be found that there will less heat be required to raise the temperature of the gas one degree. The quantity of heat required in this case is called the specific heat under constant volume, and in terms of the thermal unit employed, is designated by C_v .

The number of molecules of gas being the same in both experiments and the temperatures being equal, it follows that the quantity of heat absorbed by the gas, or the vibration work, is the same in both experiments. But in the experiment made under constant volume the heat absorbed is necessarily equal

to the total heat supplied, namely, C_v thermal units, since the envelope is considered impermeable to heat. Therefore in the first experiment there is a loss of heat equal to $C_p - C_v$ thermal units. This last heat must then have been expended in overcoming the atmospheric pressure in expanding; and the work done will be found by multiplying $C_p - C_v$ by the mechanical equivalent of heat. That is, for an increase of one degree of temperature,

Work of expansion =
$$(C_p - C_v) E$$
.

The work of overcoming a constant resistance is measured by the product of the resistance into the path described. In the case of the expanding gas just considered the constant resistance is the atmospheric resistance p_o ; and the path described is measured by the increase of volume of the gas. To determine this latter Gay-Lussac's law gives for the centigrade scale

$$v_t - v_o = \frac{t \ v_o}{273}$$

and therefore for an increase of temperature of one degree there is an increase of volume equal to $v_o/273$. The work of expansion for one degree is, therefore,

$$\frac{p_o \ v_o}{273} = R.$$

The quantity R is, then, the external work of expansion performed under atmospheric pressure by unit weight of gas when its temperature is raised one degree centigrade. But this work of expansion has already been found equal to $(C_p - C_v)$ E. There results, therefore, the important equation

$$(C_p - C_v) E = \frac{p_o v_o}{273} = R$$
 . . . (3)

for the centigrade scale of temperature. For the Fahrenheit scale the equation becomes

$$(C_p - C_v) E = \frac{p_o v_o}{491.4} (31)$$

Numerical Value of R.—The numerical value of R for any particular gas depends upon the units of length and weight adopted, the atmospheric pressure, the specific weight of the gas and the scale of temperature. Throughout this chapter the foot and pound will be employed for the units of length and weight, respectively; and generally the centigrade scale of temperature will be used. The adopted value of the atmospheric pressure is

$$p_o = 10333$$
 kgs. per m.² log = 4.01423.
 $p_o = 2116.3$ lbs. per ft.² log = 3.32558.
 $p_o = 14.6967$ lbs. per in.² log = 1.16722.

As an example, find the numerical value of R for atmospheric air. The specific weight of this gas, according to the best authorities, is 0.080704 lbs. The specific volume is the reciprocal of this; or $V_o = 12.3909$ c. ft. Therefore,

$$R = \frac{2116.3 \times 12.3909}{273} = 96.056$$
 foot-pounds.

Therefore, for one pound of this gas,

$$pv = 96.056 T;$$

and for y pounds

$$p v = 96.056 y T.$$

Law of Dulong and Petit.—The product of the specific heat of a perfect gas under constant volume, by its density, is a constant number.

By the density of a gas is meant its specific weight expressed in terms of the specific weight of atmospheric air taken as unity. If C_{va} is the specific heat of air at constant volume and C_v and d the specific heat at constant volume, and density, respectively, of any other gas, then in accordance with this law,

$$C_v d = C_{va}$$
.

Determination of Specific Heats.—The specific volume and the specific heat at constant pressure of a gas can both be determined with great accuracy by experiment; but the specific heat under constant volume is almost impossible to measure directly on account of the dissipation of heat through the sides of the vessel containing the gas. It can, however, be computed by equation (3) which gives

$$C_v = C_p - \frac{R}{E} \quad . \quad . \quad . \quad . \quad (4)$$

By means of this equation and the direct determination of specific heats under constant pressure, Regnault has deduced the following law for perfect gases:

The specific heats under constant pressure and constant volume are independent of the pressure and volume.

The following table gives the specific weights, volumes and heats of those gases which approximate most nearly to the theoretically perfect gas. The values of R were computed by (1) and those of C_v by (4). The temperature is supposed to be 0° C., and the barometer to stand at 760 mm. = 29.922 in.:

Gas	Specific Weight	Specific Volume	R	C_p	C_{v}
Atmospheric air	Pounds 0.080704	Cubic Feet 12.3909	96.056	0.23751	0.16892
Nitrogen	0.078394	12.7569	98.887	0.24380	0.17319
Oxygen	0.089230	11.2070	86.878	0.21751	0.15547
Hydrogen	0.005590	178.8910	1386.8	3.40900	2.41873

Ratio of Specific Heats.—In the study of interior ballistics the values of C_p and C_v for the gases given off by the explosion of the charge are of little importance. It suffices generally to know their ratio which is constant for perfect gases and approximately so for all gases at the high temperature of explosion. That this ratio is constant for perfect gases may be shown as follows: Since

$$R = \frac{p_o v_o}{273} = \frac{p_o}{273 \ w_o} = \frac{p_o}{273 \ d \ w_a}$$

in which w_a is the specific weight of atmospheric air, we shall have for two gases distinguished by accents, the relation

$$\frac{R'}{R''} = \frac{d''}{d'} \; ;$$

that is, the values of R for two perfect gases are inversely as their densities. But by the law of Dulong and Petit we have

$$\frac{C'_v}{C''_v} = \frac{d''}{d'} = \frac{R'}{R''}$$
 (as shown above).

Therefore

$$\frac{R'}{C'_v} = \frac{R''}{C''_v} = \text{constant.}$$

Therefore from equation (4),

$$\frac{C_p}{C_n} = r + \frac{R}{C_n E} = \text{constant} = n \text{ (say)}.$$

If we compute n by means of atmospheric air, we shall have

$$n = 1 + \frac{96.056}{0.16892 \times 1400.4} = 1.406.$$

Relations Between Heat and Work in the Expansion of Perfect Gases.—The relations which exist between the variations of the volume and pressure of a given weight of gas and the heat necessary to produce them, may now be determined from equation (1) as follows: This equation is

$$pv = RT$$

and contains three arbitrary variables p, v and T. If we suppose an element of heat, dq, to be applied to the gas, the temperature will generally be augmented by an elementary amount dT, and this may be accomplished in three different ways:

1. The volume may increase by the element dv without altering the pressure. 2. The pressure may increase by dp

while the volume remains constant. 3. The volume and pressure may both vary at the same time. We will consider each of these cases separately.

1. Differentiating (1), supposing p constant, we have

$$d\ T = \frac{p\,dv}{R} \ ;$$

and therefore the quantity of heat communicated to the gas will be, in thermal units, from the definition of specific heat,

$$dq = C_p dT = \frac{C_p p dv}{R}$$

2. If, the volume v remaining constant, the pressure is varied by d p, we shall have, proceeding as before,

$$dq = C_v dT = \frac{C_v v dp}{R}$$

3. If the volume and pressure vary together, the corresponding element of heat will be the sum of the partial variations given above. That is

$$d q = \frac{\mathbf{I}}{R} \left(C_p \, p \, d \, v + C_v \, v \, d \, p \right) \qquad . \qquad . \qquad (5)$$

The differential of (1) is

$$R d T = p d v + v d p; \qquad (6)$$

whence, eliminating v d p between (5) and (6), there results

Whence, since C_p , C_v and R are constants for the same gas,

$$q = C_v \int dT + \frac{C_p - C_v}{R} \int p \, dv.$$

The first integral represents the change of temperature and the second the external work of expansion. Denoting by T_1 and T the initial and final temperatures of the expanding gas and by W the external work, we have

$$q = (T_1 - T) C_v + \frac{C_p - C_v}{R} W (8)$$

Isothermal Expansion.—If we suppose the initial temperature T_1 to remain constant, that is, that just sufficient heat is imparted to the gas while it expands to maintain its initial temperature, equation (8) becomes

$$q = \frac{C_p - C_v}{R} W.$$

We see in this case that the quantity of heat absorbed by the gas is proportional to the external work done. The quantity $\frac{R}{C_p - C_v}$ is, therefore, the ratio of the effective work of a unit weight of gas to the quantity of heat absorbed, or the mechanical equivalent of heat, E. Therefore

$$E = \frac{R}{C_n - C_n}$$

a result already established by another method.

The work performed, therefore, by the isothermal expansion of unit weight of gas is given by the equation

$$W = E q = 1400.4 q \text{ foot-pounds}, . . . (9)$$

where q is expressed in P. C. thermal units.

The work of an isothermal expansion may also be expressed in terms of the initial and final volumes or pressures. Thus, substituting in the general equation of the work of expansion,

$$W = \int p \, dv,$$

the value of p from (1) and integrating between the limits v_1 and v, we have

$$W = R T_1 \log_e \frac{v}{v_1} = p_1 v_1 \log_e \frac{v}{v_1}$$
 . . (10)

where v is the greater volume and v_1 the less.

Since from (1)

$$\frac{v}{v_1} = \frac{p_1}{p}$$

we also have

in which p_1 is the greater tension and p the less.

The reciprocal of E may be called the heat equivalent of work, that is, the quantity of heat equivalent to a unit of work. Therefore from (9), (10) and (11), we have

$$q = \frac{W}{E} = \frac{p_1 v_1}{E} \log_e \frac{v}{v_1}$$

$$= \frac{p_1 v_1}{E} \log_e \frac{p_1}{p}$$
(12)

Equations (10) and (11), by inverting the ratios of volumes or pressures, evidently hold good when the initial volume v_1 and initial tension p_1 are changed by compression under constant temperature into the less volume v and greater tension p.

Adiabatic Expansion.—If a gas expands and performs work in an envelope impermeable to heat, so that it neither receives nor gives up heat during the expansion, the transformation is said to be adiabatic. In such an expansion the temperature and tension of the gas both diminish and the work performed must be less than for an isothermal expansion, other things being equal. For an adiabatic expansion, q is zero in (8) and, therefore, since the temperature diminishes,

$$W = \frac{R C_v}{C_p - C_v} (T_1 - T)$$

$$= \frac{R}{n - 1} (T_1 - T)$$

$$= C_v E (T_1 - T)$$
(13)

Therefore in an adiabatic expansion the work done is proportional to the fall of temperature.

Next consider equation (7), where, if we make dq zero, it becomes

$$o = C_v d T + \frac{C_p - C_v}{R} p d v;$$

which, by dividing by C_v and replacing p by its value $\frac{RT}{v}$, reduces to

$$o = \frac{dT}{T} + (n - 1) \frac{dv}{v}$$

Integrating between limits, we have

$$\frac{T}{T_1} = \left(\frac{v_1}{v}\right)^{n-1}. \qquad (14)$$

Again, making d q zero in (5), we have

$$o = C_p \not p d v + C_v v d \not p,$$

which may be written (dividing by $C_v \not p v$)

$$o = n \, \frac{d \, v}{v} + \frac{d \, p}{p}$$

Integrating between limits, we have

Combining (14) and (15) gives the important relations

$$\left(\frac{v_1}{v}\right)^{n-1} = \frac{T}{T_1} = \left(\frac{p}{p_1}\right)^{\frac{n-1}{n}} . \qquad (16)$$

By means of (16) the work of an adiabatic expansion given by (13) may be expressed either in terms of the initial and terminal volumes, or of the initial and terminal pressures. Thus, since

$$T_1 - T = T_1 \left(1 - \frac{T}{T_1} \right)$$

the last of equations (13) may be written,

$$W = C_v E T_1 \left\{ \mathbf{I} - \left(\frac{v_1}{v}\right)^{n-1} \right\}$$

$$= C_v E T_1 \left\{ \mathbf{I} - \left(\frac{p}{p_1}\right)^{\frac{n-1}{n}} \right\}$$

$$(17)$$

EXAMPLES.

1. Determine the volume of 5 pounds of oxygen at a pressure of 50 pounds per square inch by the gauge, and at a temperature of 60° C.

The real pressure is gauge pressure plus the atmospheric pressure = 50 + 14.6967 = 64.6967 lbs. per in.² Therefore, $p = 144 \times 64.6967$ lbs. per ft.² $T = 273 + 60 = 333^{\circ}$. R = 86.878. Therefore from (2),

$$v = \frac{5 \times 86.878 \times 333}{144 \times 64.6967} = 15.527 \text{ ft.}^3$$

2. One pound of atmospheric air occupying a volume of one cubic foot has a tension of 50,000 lbs. per ft.² What is its temperature by the Fahrenheit scale?

For the centigrade scale we have R = 96.056, v = 1, p = 50,000 and y = 1.

Therefore
$$T = \frac{50000}{96.056} = 520.^{\circ}54 \text{ C.} = 968.^{\circ}97 \text{ F.}$$

$$\therefore t = 968.97 - 491.4 = 477.^{\circ}57 \text{ F.}$$

3. A gas-receiver having a volume of 3 cubic feet contains half a pound of oxygen at 70° F. What is the pressure by the gauge?

Here
$$y = \frac{1}{2}$$
, $v = 3$, $R = 86.876$, $t = 21^{\circ} \frac{1}{9}$ C., and $T = 294\frac{1}{9}$.

Therefore,

$$p = \frac{86.876 \times 294\frac{1}{9}}{2 \times 3 \times 144} - 14.697 = 14.876 \text{ lbs. per in.}^2$$

4. A spherical balloon 20 feet in diameter is to be inflated with hydrogen at 60° F., when the barometer stands at 30.2 in., so that gas may not be lost on account of expansion when the

balloon has risen till the barometer stands at 19.6 in., and the temperature falls to 40° F. How many pounds and how many cubic feet of gas are to be run in?

Here
$$v = \frac{4}{3}\pi \times 10^3 = 4188.8 \text{ ft.}^3$$

 $p = \frac{19.6 \times 2116.3}{29.9215} = 1386.8 \text{ lbs. per ft.}^2$
 $T = 277\frac{4}{9} \text{ C.}$
 $R = 1386.8.$
 $\therefore y = \frac{pv}{RT} = 15.092 \text{ lbs.}$

To determine the number of cubic feet of gas run in, we have

$$v = \frac{y R T}{p} = 2827.4 \text{ ft.}^3,$$

where

$$p = \frac{30.2 \times 2116.3}{29.9215} = 2136.0.$$

$$T = \frac{5}{9} (60 - 32) + 273 = 288\frac{5}{9}.$$

5. "The balloon in which Wellman intends to seek the North Pole has a capacity of 224,244 cubic feet, and weighs, with its car and machinery, 6,600 lbs. What will be its lifting capacity when filled with hydrogen at 10° C. and 760 mm. of the barometer?" (Lissak's "Ordnance and Gunnery," p. 61.)

The balloon, when inflated, will hold at 10° C., 17,458 lbs. of air and 1,209 lbs. of hydrogen. Its lifting capacity will, therefore, be 17,458 - (1,209 + 6,600) = 9,649 lbs.

6. Two pounds of air expand adiabatically from an initial temperature of 60° F., and a pressure of 65.3 lbs. per in.² to a pressure of 50 lbs. per in.² Determine the initial and terminal volumes, the terminal temperature and the external work done.

Here $p_1 = 144 \times 65.3 = 9403.2$; $p = 50 \times 144 = 7200$; $T_1 = 288\frac{5}{9}$ C.; R = 96.056; y = 2. Take n = 1.4

$$v_{1} = \frac{y R T_{1}}{p_{1}} = 5.8954 \text{ ft.}^{3}$$

$$v = v_{1} \left(\frac{p_{1}}{p}\right)^{\frac{5}{7}} = 7.1338 \text{ ft.}^{3}$$

$$T = T_{1} \left(\frac{p}{p_{1}}\right)^{\frac{2}{7}} = 267.37 \text{ C.} = 481.266 \text{ F.}$$

$$\therefore t = 21.87 \text{ F.}$$

$$W = \frac{y R}{n-1} (T_{1} - T) = 10177 \text{ ft.-lbs.}$$

7. Compute the work of expansion of 2 pounds of air at temperature 100° C., which expands adiabatically until it doubles its volume. Also determine the temperature after expansion and the ratio of the initial and terminal pressures.

Answers:
$$W = 43378$$
 ft.-lbs.
 $t = 19^{\circ}.68$ C.
 $p = 0.3789$ p_1 .

8. A mass of air occupying a volume of 3 ft. expands adiabatically from an initial temperature of 70° F., and pressure of 85 lbs. per in. until external work of 8,000 ft.-lbs. has been done. Compute the terminal volume, pressure, temperature, and weight of air.

Answers:
$$v = 3.768 \text{ ft.}^3$$

 $p = 61.78 \text{ lbs. per in.}^2$
 $t = 23^{\circ}.86 \text{ F.}$
 $y = 1.3 \text{ lbs.}$

Theoretical Work of an Adiabatic Expansion in the Bore of a Gun.—If, in the first of equations (17), we replace $C_v E T_1$, by its equal $\frac{R T_1}{n-1}$ it becomes for y pounds of gas

$$W = \frac{y R T_1}{n-1} \left\{ 1 - \left(\frac{v_1}{v}\right)^{n-1} \right\}$$

This equation gives the work of y pounds of gas at the initial temperature T_1 , expanding from the initial volume v_1 to volume v_2 . Suppose the mass of gas to occupy the chamber of a gun with the projectile at its firing seat; and to expand by forcing the projectile along the bore. In this case v_1 will be the volume of the chamber, which is an enlargement of the bore, and is measured by what is called the reduced length of the chamber; that is, by the length of a cylinder whose cross-section is the same as the bore and whose volume is that of the chamber. If u_0 is the reduced length required, V_c the volume of the chamber and d the diameter corresponding to the area of cross-section of bore, and which on account of the rifling is slightly greater than the caliber, we evidently have

$$u_o = \frac{4 V_c}{\pi d^2}$$

The variable volume v is the volume of the chamber plus the volume of the bore in rear of the projectile after it has moved any distance u; and is, therefore, measured by $u_o + u$. Therefore the above expression for the work of expansion becomes

$$W = \frac{y R T_1}{n-1} \left\{ 1 - \left(\frac{u_0}{u_0 + u} \right)^{n-1} \right\}$$

There is some uncertainty as to the proper value of n for the gases of fired powder. As we have seen, the value of this ratio for perfect gases is approximately 1.4; and it has been generally assumed that at the high temperature of combustion of powder the gases formed may be regarded as possessing all the properties of perfect gases; and therefore most of the earlier writers on interior ballistics employed this value of n in their deductions. But more recent experiments have shown that this value is too great, but have not fixed its true value. The experiments of Noble and Abel with the gases of fired gunpowder, at or near the temperature of combustion, made $n = 1\frac{1}{3}$ nearly;

and this is the value which, for want of a better, we will adopt in what follows. Introducing this value of n into the above expression for the work of expansion; and making $R T_1 = f$ and the ratio $u/u_0 = x$, we have

$$W = 3fy \left\{ 1 - \frac{1}{(1+x)^{\frac{1}{3}}} \right\} \qquad . \qquad . \tag{18}$$

The work of expansion in the bore of a gun is expended in many ways, but chiefly in the energy of translation imparted to the projectile. If we assume that the entire work is thus expended, we shall have

$$v^{2} = \frac{2 g}{w} W = \frac{6 gfy}{w} \left\{ \mathbf{I} - \frac{\mathbf{I}}{(\mathbf{I} + x)^{\frac{1}{6}}} \right\} \qquad . \tag{19}$$

It is evident from (2) that f is the pressure per unit of surface of unit weight of gas at temperature T_1 . The ratio x is the number of volumes of expansion of y pounds of gas due to the travel u.

The assumption that the work of expansion is measured by the energy of translation of the projectile does not change the form of the second member of (19); and it is evident that by giving to f a suitable value determined by experiment, the equality expressed in (19) may be strictly true. But in this case f ceases to have the value RT_1 and becomes simply an experimental coefficient.

In English units (pound and foot), f would be theoretically the pressure in pounds per square foot of one pound of gas at temperature T_1 confined in a volume of one cubic foot.

In metric units (kilogramme and decimetre), f would be defined as above, making the proper change of units.

We may deduce a second approximation to the velocity impressed upon the projectile by the expansion of the gas by taking into account the work performed upon the gun and carriage, as well as upon the projectile. We will suppose the gun mounted upon a free-recoil carriage. Let M be the mass

of the gun and carriage, V their velocity at any period of motion and m the mass of the projectile. The expression for the work of expansion will now be

$$2 W = m v^2 + M V^2$$
. (20)

A second equation between the velocities v and V can be deduced by equating the momenta of the system projected upon the axis of the gun. We thus obtain

Eliminating V from (20) and (21) there results

This expression for v^2 is the same as that given by (19) with the exception of the small fraction m/M which can be safely neglected in comparison with unity. Similarly it may be shown that the work expended upon the projectile in giving it rotation about its axis is small in comparison with the work of translation.

Noble and Abel's Researches on Fired Gunpowder.—Noble and Abel's experiments on the explosion of gunpowder in close vessels were given to the world in two memoirs which were read before the Royal Society in 1874 and 1879, respectively. These experiments have an important bearing upon the subject of interior ballistics, since they furnish the most reliable values we possess of the temperature of combustion of fired gunpowder, the mean specific heat of the products of combustion (solid as well as gaseous), the ratio of solid to gaseous products, and, lastly, what is known as the *force of the powder*,—all of which are important factors in computing the work done by the gases of a charge of gunpowder exploded in the chamber of a gun.

The vessels in which the explosions were produced were of two sizes, the smaller one for moderate charges and for experi-

ments connected with the measurement or analysis of the gases, while in the larger one Captain Noble states that he has succeeded in absolutely retaining the products of combustion of a charge of 23 pounds of gunpowder.* These vessels consisted of a steel barrel open at both ends, the two open ends being closed by carefully fitted screw plugs (firing plug and crusher plug), furnished with gas checks to prevent any escape of gas past the screw. In the firing plug was a conical hole closed from within by a steel cone which was ground into its place with great exactness, and which, when the cylinder was prepared for firing, was covered with very fine tissue paper to give it electrical insulation from the rest of the apparatus. The two wires from a Leclanché battery were attached, the one to the insulated cone and the other to the firing plug, and were connected within the powder chamber by a fine platinum wire passing through a glass tube filled with mealed powder. This platinum wire became heated when the electric current passed through it, and the charge was thus fired. At the opposite end of the cylinder from the firing plug was another plug fitted with a crusher gauge for determining the pressure of the gases. The vessel was also provided with an arrangement for collecting the gases after an explosion for analysis, measurement of quantity, or for other purposes.

Results of the Experiments.—It was found that about 57 per cent. by weight of the products of combustion were nongaseous, consisting principally of potassium carbonate, potassium sulphate, and potassium sulphide, the first named greatly preponderating. The remaining 43 per cent. were permanent gases, principally CO₂, CO and N. These gases, when brought to a temperature of o° C., and under the normal atmospheric pressure of 760 millimetres, occupied about 280 times the volume of the unexploded powder.

^{*} Lecture on Internal Ballistics, by Captain Noble, London, 1892, p. 12.

Pressure in Close Vessels, Deduced from Theoretical Considerations.—The expression for the pressure of the gases developed by the combustion of gunpowder in a close vessel is deduced upon the following suppositions:

1st. That a portion of the products of combustion is in a liquid state.

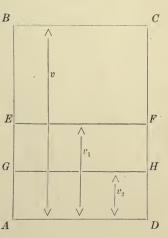
2d. That the pressure due to the permanent gases can only be calculated by deducting the volume of the liquid products from the volume of the vessel.

Upon these hypotheses the expression for the pressure may be deduced as follows:

Let A B C D be a section of a close vessel of volume v in which a given charge of powder is exploded.

Let A E F D represent the space (v_1) occupied by the charge, and A G H D the space (v_2) occupied by the non-gaseous products. Let Δ_1 be the so-called density of the products of combustion,—that is $\Delta_1 = \frac{v_1}{v}$; and α the ratio of the non-gaseous products to the volume of the charge, or

 $\alpha = \frac{v_2}{v_1} = \frac{v_2}{\Delta_1 v}$. The gases after ex- A



plosion will occupy the space $v - v_2 = v - \alpha \ \Delta_1 v = v \ (\mathbf{1} - \alpha \ \Delta_1)$. Let p_1 be the pressure that would be developed if the volume of the vessel were $A \ E \ F \ D$ (or v_1). In this case the density of the products of combustion (Δ_1) (the charge remaining the same) would be unity; and the space occupied by the gases would be $v_1 - v_2 = v_1 \ (\mathbf{1} - \alpha) = \Delta_1 \ v \ (\mathbf{1} - \alpha)$. Now if p is the pressure when the volume of the vessel is v, we have by Mariotte's law (assuming that the temperature is the same for all densities of the products of combustion),

$$p = p_1 \frac{\Delta_1 v(\mathbf{I} - \alpha)}{v(\mathbf{I} - \alpha \Delta_1)} = (\mathbf{I} - \alpha) p_1 \frac{\Delta_1}{\mathbf{I} - \alpha \Delta_1};$$

or, making

$$f = (\mathbf{I} - \alpha)_{\cdot} p_{\mathbf{I}},$$

we have

$$p = f \frac{\Delta_1}{1 - \alpha \Delta_1} . \qquad (23)$$

The factor f is called the force of the powder.

Value of the Ratio α .—Let p_2 and p_3 be the pressures in the same vessel produced by two different charges, and Δ_2 and Δ_3 the corresponding densities of the products of combustion. Then from equation (23) (assuming f to be the same for all values of Δ_1),

$$p_2 = f \frac{\Delta_2}{1 - \alpha \, \Delta_2},$$

and

$$p_3 = f \frac{\Delta_3}{1 - \alpha \Delta_3};$$

whence by division,

$$\frac{\mathbf{I} - \alpha \, \Delta_3}{\mathbf{I} - \alpha \, \Delta_2} = \frac{p_2 \, \Delta_3}{p_3 \, \Delta_2}.$$

Therefore

$$\alpha \, = \frac{\mathrm{I}}{\Delta_2 \, \Delta_2} \bigg\{ \frac{p_3 \, \Delta_2 - \, p_2 \, \Delta_3}{p_3 - \, p_2} \bigg\} \, , \label{eq:alpha_parameters}$$

by means of which the mean value of α can be determined when a sufficient number of pressures, corresponding to different values of Δ_1 , have been found by experiment. The value of α finally adopted by Noble and Abel is 0.57.

Determination of the Force of the Powder.—To determine f we have from equation (23),

$$f = p \frac{1 - .57 \Delta_1}{\Delta_1};$$

from which f may be found by means of a single measured pressure corresponding to a given density of the products of

combustion. When $\Delta_1 = r$, that is, when the vessel is completely filled by the charge, p was found to be 43 tons per square inch, and therefore f = 43 (r - .57) = 18.49 tons or 41417.6 pounds per square inch. Therefore Noble and Abel's formula for the pressure in a close vessel is, for different densities of the products of combustion,

$$p = 18.49 \frac{\Delta_1}{1 - .57 \Delta_1} \text{ tons per sq. in.}$$

$$= 41417.6 \frac{\Delta_1}{1 - .57 \Delta_1} \text{ lbs. per sq. in.}$$
(24)

To transform this equation so that it shall express the pressure in kilos per dm.² we may employ a simple rule which, as it is of frequent use, is here inserted for convenience:

Rule:—To reduce a pressure expressed in tons per square inch to the same pressure expressed in kilos per dm.², add to the logarithm of the former the constant logarithm 4.1972544 and the sum is the logarithm of the pressure required.

If the pressure to be reduced is in pounds per in.² then the constant logarithm to be added is 0.8470064.

Applying this rule the expression for the pressure of the products of combustion of a charge of gunpowder fired in a close vessel is found to be

$$p = 291200 \frac{\Delta_1}{1 - .57 \Delta_1}$$

 $\therefore f = 291200 \text{ kilos per dm.}^2$

It will be seen from the definition given to Δ_1 that it is the density of loading as defined in Chapter III when the gravimetric density of the powder is unity,—that is, when a kilo of the powder fills a volume of a dm.³; or, what is the same thing, when a pound occupies a volume of 27.68 cubic inches; and in this case, when Δ_1 is unity the charge just fills the receptacle. Noble and Abel were careful to keep the gravimetric density of the powder they experimented with as near unity as possible.

Interpretation of f.—It will be seen from Equation (23) that the quantity designated by f is the pressure of the gases when

$$\frac{\Delta_1 v}{v (1-\alpha \Delta_1)} = 1,$$

that is, when the space occupied by the gases is equal to the volume of the charge, which requires that the vessel should have $1 + \alpha$ units of volume. Thus if the kilogramme and litre are the units of weight and volume, respectively, the volume of the vessel must be 1.57 litres in order that the gases may occupy a volume of one litre, and have a tension equal to f. From this f may be defined to be the pressure of the gases of unit weight of powder occupying unit volume at the temperature of combustion T_1 .

If ϵ is the weight of gas furnished by the combustion of unit weight of powder we have from Equation (2),

$$p_1 v_1 = \epsilon R T_1;$$

and if v_1 is the unit of volume, there results

$$p_1 = f = \epsilon R T_1 \quad . \quad . \quad (25)$$

If the pound is the unit of weight the unit of volume is 27.68 cubic inches. In this case the definition of f requires that the volume of the vessel should be $1.57 \times 27.68 = 43.459$ cubic inches.

The value of ϵ , according to Noble and Abel, is 0.43; and therefore the pressure of unit weight of the gases of fired gunpowder at temperature T_1 is

$$\frac{f}{0.43}$$
.

From this it follows that the pressure of one pound of the gases of fired gunpowder at temperature of combustion, confined in a volume of 27.68 cubic inches, is

$$\frac{41417.6}{0.43}$$
 = 96320 lbs. per square inch.

Also, the pressure of one pound of the gases of the paragraph immediately preceding, confined in a volume of one cubic foot, is, in pounds per square foot,

$$\frac{96320 \times 27.68}{12} = 222180 \text{ lbs.}$$

If the gravimetric density of the powder be unity, and y and v be taken in pounds and cubic inches, respectively, then Equation (23) becomes

$$p = f \frac{27.68 \, y}{v - 27.68 \, \alpha \, y} \qquad . \qquad . \qquad . \tag{26}$$

Solving with reference to y and to v gives

$$y = \frac{p v}{27.68 (\alpha p + f)}$$
 . . . (27)

and

$$v = \frac{27.68 \, y \, (\alpha \, p + f)}{p} \quad . \quad . \quad (28)$$

These equations are useful in questions involving the bursting of shells, etc.

Theoretical Determination of the Temperature of Explosion of Gunpowder.—Having determined the value of f from the experiments, we can deduce the temperature of explosion by means of the formula

$$T_1 = \frac{273}{\epsilon \, p_o \, v_o} f$$

According to Noble and Abel's experiments, if the gravimetric density of the powder is such that a kilogramme occupies one litre, the gases furnished by its combustion will fill a volume of 280 litres at 0° C. under the normal atmospheric pressure of 103.33 kgs. per square decimetre. We therefore have

$$v_o = \frac{280}{y \epsilon}$$

and

$$p_o = 103.33$$

whence

$$T_1 = \frac{273 \times 291200}{103.33 \times 280} = 2748^{\circ} \text{ C}.$$

This is the absolute temperature of combustion of gunpowder according to Noble and Abel's latest deductions from their experiments. Subtracting 273° from this temperature we have temperature of explosion = 2475° C. $(4487^{\circ}$ F.).

Mean Specific Heat of the Products of Combustion.—From equation (8), we have when W = 0, that is, when no external work is performed,

$$Q = C_v \left(T_1 - 273 \right)$$

in which Q is the heat of combustion; that is, the quantity of heat that unit of weight of the explosive substance evolves, under constant volume, when the final temperature of the products of combustion is \circ° C. From this equation we find

$$C_v = \frac{Q}{T_1 - 273}$$

The heat of combustion was determined by Noble and Abel in the following manner:

"A charge of powder was weighed and placed in one of the smaller cylinders, which was kept for some hours in a room of very uniform temperature. When the apparatus was throughout of the same temperature, the thermometer was read, the cylinder closed, and the charge exploded.

"Immediately after explosion the cylinder was placed in a calorimeter containing a given weight of water at a measured temperature, the vessel being carefully protected from radiation, and its calorific value in water having been previously determined.

"The uniform transmission of heat through the entire volume of water was maintained by agitation of the liquid, and the thermometer was read every five minutes until the maximum was reached. The observations were then continued for an equal time to determine the loss of heat in the calorimeter due to radiation, etc.; the amount so determined was added to the maximum temperature."

In this way the heat of combustion of R. L. G. and F. G. powders was found to be 705 heat-units; that is, the combustion of a unit weight of the powder liberated sufficient heat to raise the temperature of 705 unit-weights of water 1° C. We therefore have

$$C_v = \frac{705}{2475} = 0.285$$

This result is accepted by Noble and Abel, and also by Sarrau, as a very close approximation to the mean specific heat of the entire products of combustion. If we assume that the mean specific heat of gunpowder of different compositions is constant, we can compute the temperatures of combustion when the heat of combustion has been determined by the calorimeter, by the formula

$$T = \frac{Q}{0.285}$$

in which T is given by the centigrade scale.

Pressure in the Bores of Guns Derived from Theoretical Considerations.—"At an early stage in our researches, when we found, contrary to our expectation, that the elastic pressure deduced from experiments in close vessels did not differ greatly (where the powder might be considered entirely consumed, or nearly so) from those deduced from experiments in the bores of guns themselves, we came to the conclusion that this departure from our expectation was probably due to the heat stored up in the liquid residue. In fact, instead of the expansion of the permanent gases taking place without addition of heat, the residue, in the finely divided state in which it must be on the ignition of the charge, may be considered a source of heat of the most per-

fect character, and available for compensating the cooling effect due to the expansion of the gases on production of work.

"The question, then, that we now have to consider is—What will be the conditions of expansion of the permanent gases when dilating in the bore of a gun and drawing heat, during their expansion, from the non-gaseous portions in a very finely divided state?" *

Let c_i be the specific heat of the non-gaseous portion of the charge, which we can assume, without material error, to be constant. We shall then have c_i d T for the elementary quantity of heat yielded to the gases per unit of weight of liquid residue. If there are w_i units of weight of liquid residue it will yield to the gases w_1 c_i d T units of heat; and if there are w_2 units of weight of gas we shall have in heat-units,

$$d q = -\frac{w_1}{w_2} c_i d T = -\beta c_i d T,$$

in which

$$\beta = \frac{w_1}{w_2};$$

that is, β is the ratio between the weights of the non-gaseous and gaseous portions of the charge. The negative sign is given to the second member because T decreases while q increases.

Substituting the above value of dq in Equation (7), it becomes

$$- (C_v + \beta c_i) dT = \frac{C_p - C_v}{R} p dv . . (29)$$

and this combined with Equation (6), gives, by a slight reduction,

$$- (\beta c_{i} + C_{p}) \frac{d v}{v} = (\beta c_{i} + C_{v}) \frac{d p}{p} . . . (30)$$

Since C_p , C_v , c_1 and β are, by hypothesis, constant during the expansion, the integration of Equation (30) between the limits v_2

^{*} Noble and Abel, Researches, etc., page 98.

and v_3 —the former being the initial volume occupied by the permanent gases and the latter their volume after the projectile has been displaced by a distance u, gives

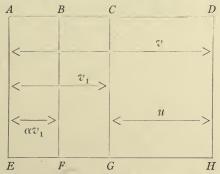
$$p = p_1 \left(\frac{v_2}{v_3}\right)^r \qquad . \qquad . \qquad . \qquad (31)$$

in which

$$r = \frac{\beta c_i + C_p}{\beta c_i + C_n}$$

Equation (31), it will be seen, becomes identical with Equation (15), when $\beta = 0$; that is, when there is no liquid residue.

To introduce v_1 and v, that is the volumes occupied by the charge and the entire volume in the rear of the projectile, into Equation (30) in place of v_2 and v_3 , proceed as follows: Let



A C E G represent the chamber of the gun, which we will suppose filled with powder without compression, and further that one pound of the powder fills a space of 27.68 cubic inches. The gravimetric density and density of loading are each unity; and if v_1 is the volume of the chamber, it follows that

$$v_1 = 27.68 \,\tilde{\omega}.$$

 $\tilde{\omega}$ being the weight of charge.

Suppose the powder to be entirely consumed before the projectile moves any perceptible distance; and that the non-gaseous products fill the space A B E F, whose volume is α v_1 . The gases,

therefore, which by their expansion give motion to the projectile will occupy the space B C F G before perceptible motion begins. The volume of the space B C F G is evidently $v_2 = v_1 - \alpha v_1 = v_1 (\mathbf{1} - \alpha)$. Let D H be the base of the projectile after it has moved a distance u; and, designating the volume A D E H by v, we evidently have $v_3 = v - \alpha v_1$. Substituting these values of v_2 and v_3 in Equation (31) gives

$$p = p_1 \left(\frac{v_1 (1 - \alpha)}{v - \alpha v_1} \right)^{r} . \qquad (32)$$

In this equation p_1 is the pressure produced by the combustion of a charge of powder in a close vessel when the density of loading is unity. The values of the constants are given by Noble and Abel as follows:*

 $p_1 = 43$ tons per square inch $\alpha = 0.57$ $\beta = 1.2957$ $C_p = 0.2324$ $C_v = 0.1762$ $c_t = 0.45$ $v_1 = 27.68 \tilde{\omega}$

from which we find r = 1.074. Substituting these values in the expression for p it becomes

$$p = 43 \left(\frac{0.43 \, v_1}{v - 0.57 \, v_1} \right)^{1.074} \, . \qquad (33)$$

which gives the pressure in tons per square inch.

If, as in a close vessel, we let

$$\Delta_1 = \frac{v_1}{v},$$

then

$$p = (0.43)^{1.074} \times 43 \left\{ \frac{\Delta_1}{1 - .57 \Delta_1} \right\}^{1.074}$$

^{*} Researches, etc., page 167,

$$= 17.375 \left\{ \frac{\Delta_1}{1 - .57 \Delta_1} \right\}^{1.074} . . . (34)$$

REMARKS:—The value of $\beta = 1.2957$, adopted by Noble and Abel, gives for unit weight, $w_1 = 0.5644$ and $w_2 = 0.4356$, while the values of these quantities adopted in our equations are 0.57 and 0.43, respectively. These last-named values would make $\beta = 1.3256$.

Noble and Abel's values of the specific heats of the permanent gases of combustion, namely, $C_p = 0.2324$ and $C_v = 0.1762$, make n = 1.32; while for perfect gases, as has been shown, n = 1.4 very nearly.

Table of Pressures.—In the following table of pressures the third column gives the pressures in the bore of a gun corresponding to the values of Δ_1 in the first column. They were computed by Equation (34) upon the assumption that the permanent gases in expanding, and thereby doing work, borrow heat from the nongaseous residue; and also that the combustion is complete before the projectile has moved perceptibly; and finally that there is no conduction of heat to the walls of the gun. The tensions in the fifth column were computed by Equation (24) and agree with Noble and Abel's experiments.*

Temperatures of Products of Combustion in Bores of Guns.— The temperature in the bore of a gun during the expansion of the products of combustion, may be determined from Equation (29), which replacing R by its value from Equation (1), becomes

$$\frac{dT}{T} = -\frac{C_p - C_v}{\beta c_i + C_v} \frac{dv}{v};$$

whence integrating between the same limits as before, and observing that

$$\frac{C_p - C_v}{\beta c_i + C_v} = r - 1,$$

^{*} For table of pressures see page 46.

TABLE OF PRESSURES.

$\begin{array}{c c} \text{Mean density} \\ \text{of products} \\ \text{of products} \\ \text{of combustion.} \\ \hline & & & \\ \hline & & \\ \hline & & \\ \hline & & & \\ \hline & & & \\ \hline & & & \\ \hline \ & & \\$		Tensions ca Equation	lculated by on (33).	Tensions in close cylinders, or where gases expand without doing work.	
		Tons per square Differences.		Tons per square inch.	Differences.
1.00	1.000	43.00	5.01	43.00	4.69
.95	1.053	37.99	4.40	38.31	4.14
.90	I.III	33.59	3.88	34.17	3.68
.85	1.176	29.72	3.44	30.49	3.30
.80	1.250	26.28	3.06	27.19	2.97
.75	1.333	23.22	2.76	24.22	2.68
.70	1.429	20.46	2.48	21.54	2.45
.65	1.539	17.98	2.25	19.09	2.23
.60	1.667	15.73	2.04	16.86	2.05
- 55	1.818	13.69	1.86	14.81	1.88
.50	2.000	11.83	1.70	12.93	1.74
· 45	2.222	10.13	1.56	11.19	1.61
. 40	2.500	8.57	1.43	9.58	1.50
.35	2.857	7.14	1.31	8.08	1.39
. 30	3 · 333	5.83	I.2I	6.69	1.30
.25	4.000	4.62	1.11	5.39	I.22
. 20	5.000	3.51	I.02	4.17	1.14
. 15	6.667	2.49	.93	3.03	1.07
. 10	10.000	1.56	.84	1.96	1.01
.05	20.000	.72		.95	

we have

$$T = T_1 \left(\frac{v_2}{v_3}\right)^{r-1}.$$

Replacing v_2 by v_1 $(1 - \alpha)$, and v_3 by $v - \alpha v_1$, for reasons already given, we have for the absolute temperature of the gases during expansion, the equation

$$T = T_1 \left(\frac{v_1 (1 - \alpha)}{v - \alpha v_1} \right)^{r - 1}.$$

Introducing the density of the products of combustion (Δ_1) , and the numerical values of α and r into this last equation, it becomes

$$T = (.43)^{.074} \times T_1 \left\{ \frac{\Delta_1}{1 - .57 \Delta_1} \right\}^{.074}$$
$$= 0.93946 T_1 \left\{ \frac{\Delta_1}{1 - .57 \Delta_1} \right\}^{.074} . \qquad (35)$$

The value of T for any given density of the products of combustion (represented by Δ_1) will depend upon their initial temperature (or absolute temperature of combustion), T_1 . Its theoretical value, based upon Noble and Abel's latest deductions from their experiments, as published in their second memoir, has already been found to be 2748° C. But there are very great difficulties in the way of verifying by experiment the theoretical value of T_1 , and Captain Noble in his Greenock lecture (February 12th, 1892) takes the absolute temperature of combustion at 2505° C., as deduced in their first memoir. Therefore making

$$T_1 = 2505^{\circ} \text{ C.},$$

the expression for T becomes

$$T = 2353.3 \left\{ \frac{\Delta_1}{1 - .57 \Delta_1} \right\}^{.074} . . . (36)$$

The temperatures in degrees Centigrade and Fahrenheit, calculated from Equation (36), are given in the following table. "It is hardly necessary to point out that the values given in this table are only strictly accurate when the charge is ignited before the projectile is sensibly moved; but in practice the correction due to this cause will not be great."*

Theoretical Work Effected by Gunpowder.—The theoretical work which a charge of gunpowder is capable of effecting during the expansion of its volume from v_1 to any volume v is expressed by the definite integral

$$W = \int_{v_1}^{v} p \, dv;$$

^{*} For table of temperatures see next page.

TABLE OF TEMPERATURES.

Mean density of prod- ucts of combustion.	Number of volumes of expansion.	TEMPERATURES.		
Δ_1	$\frac{1}{\Delta_1}$	Centigrade.	Fahrenheit.	
1.00	1.0000	2231	4048	
.95	1.0526	2210	4010	
.90	1.1111	2189	3972	
.85	1.1765	2168	3934	
.80	1.2500	2147	3897	
.75	1.3333	2126	3859	
.70	1.4286	2106	3823	
.65	1.5385	2085	3785	
.60	1.6667	2063	3745	
∙55	1.8182	2041	3706	
. 50	2.0000	2018	3664 *	
·45	2.2222	1994	3621	
.40	2.5000	1968	3574	
∙35	2.8571	1940	3524	
.30	3 · 3333	1909	3468	
.25	4.0000	1874	3405	
.20	5.0000	1834	3333	
.15	6.6667	1785	3245	
.10	10.0000	1719	3126	
.05	20.0000	1615	2939	
.00	∞	0	0	

or, substituting for p its value from (32),

$$W = \int_{v_1}^{v} p_1 \left(\frac{v_1(\mathbf{I} - \alpha)}{v - \alpha v_1} \right)^r dv,$$

= $p_1 v_1^{\mathsf{r}} (\mathbf{I} - \alpha)^{\mathsf{r}} \int_{v_1}^{v} \frac{dv}{(v - \alpha v_1)^r};$

whence, integrating, we have

$$W = \frac{p_1 v_{,1}^{\mathsf{r}} (1-\alpha)^{\mathsf{r}}}{r-1} \left\{ \frac{1}{[v_1 (1-\alpha)]^{\mathsf{r}-1}} - \frac{1}{(v-\alpha v_1)^{\mathsf{r}-1}} \right\}.$$

Multiplying and dividing the second member by $[v_1 (1-\alpha)]^{r-1}$, we have

$$W = \frac{p_1 v_1(\mathbf{I} - \alpha)}{(r - \mathbf{I})} \left\{ \mathbf{I} - \left(\frac{v_1(\mathbf{I} - \alpha)}{v - \alpha v_1} \right)^{r - l} \right\}.$$

If, in this last equation, p_1 be expressed in kilogrammes per square decimetre, and v_1 be made unity (one litre), the work will be expressed in decimetre-kilogrammes per kilogramme of powder burned. To express the work in foot-tons per pound of powder burned, we must make $v_1 = 27.68$ cubic inches; and then, since p_1 is given in tons per square inch, divide the result by 12, the number of inches in a foot. Making these substitutions and replacing α and r by their values already given, we have, in foot-tons,

$$W = 576.369 \left\{ r - \left(\frac{.43 v_1}{v - .57 v_1} \right)^{r-1} \right\};$$

or, in terms of Δ_1 ,

$$W = 576.369 \left\{ 1 - 0.93946 \left(\frac{I_1}{1 - .57 \Delta_1} \right)^{.074} \right\} \quad . \quad (37)$$

Substituting in Equation (37) from Equation (35) we have

$$W = \frac{576.369}{T_1} (T_1 - T)$$

or, since, according to Noble and Abel,

$$T_1 = 2505^{\circ}$$

we have

$$W = 0.23008 (T_1 - T)$$
 (38)

which gives the work in terms of the loss of temperature of the products of combustion.

Table III gives the work of expansion of the gases of one pound of gunpowder of the normal type and free from moisture, computed by Equation (37). By means of the work given in this table, and by the use of a proper factor of effect determined by experiment, Noble and Abel consider that the actual work of a given charge of powder upon a projectile may be computed with considerable accuracy. Their method of using this table will be clearly seen by the following extract:

"If we wish to know the maximum work of a given charge,

fired in a gun with such capacity of bore that the charge suffered five expansions ($\Delta_1 = 0.2$) during the motion of the projectile in the gun, the density of loading being unity, the table shows us that for every pound in the charge, an energy of 91.4 foot-tons will as a maximum be generated.

"If the factor of effect for the powder and gun be known, the above values, multiplied by that factor, will give the energy per pound that may be expected to be realized in the projectile.

"But it rarely happens, especially with the very large charges used in the most recent guns that densities of loading so high as unity are employed; and in such cases, from the total energy realizable must be deducted the energy which the powder would have generated, had it expanded from a density of unity to that actually occupied by the charge. Thus in the example above given, if we suppose the charge instead of a density of loading of unity to have a density of 0.8, we see from Table 3, that from the 91.4 foot-tons above given, there must be subtracted 19.23 foot-tons; leaving 72.17 foot-tons as the maximum energy realizable under the given conditions, per pound of the charge." *

To apply these principles practically for *muzzle velocities*, let, as before,

 v_1 be the volume occupied by the charge, in cubic inches. v the total volume of bore and chamber, in cubic inches.

 V_b the volume of the bore.

 V_c the volume of the chamber, in cubic inches.

Then

$$v = V_b + V_c$$
;

and, if the gravimetric density of the powder be unity,

$$v_1 = 27.68 \, \tilde{\omega},$$

where $\tilde{\omega}$ is the weight of the charge in pounds. Therefore the

^{*} Noble and Abel, Researches, page 176.

number of volumes of expansion of the products of combustion will be, at the muzzle,

$$\frac{v}{v_1} = \frac{1}{\Delta_1} = \frac{V_b}{27.68 \, \bar{\omega}} + \frac{V_c}{v_1};$$

which may be written, if the gravimetric density of the powder be unity,

$$\frac{1}{\Delta_1} = 0.0361263 \frac{V_b}{\tilde{\omega}} + \frac{1}{\Delta}$$

in which Δ is the density of loading as defined in Chapter III.

If the gravimetric density of the powder be not unity, let v_2 be the volume in cubic inches of one pound of powder not pressed together except by its own weight; and let

$$\frac{27.68}{v_2}=m;$$

then we have in all cases,

$$\frac{1}{\Delta_1} = m \left\{ 0.0361263 \frac{V_b}{\tilde{\omega}} + \frac{1}{\Delta} \right\},\,$$

in which $\frac{1}{\Delta_1}$ is the number of volumes of expansion of the products of combustion.

Let W_2 be the work taken from Noble and Abel's table (Table III) of the gases of one pound of powder for a given value of $\frac{1}{\Delta_1}$, and W_1 the work due to the expansion $\frac{m}{\Delta}$. Also, let F be the factor of effect. Then if we assume that the work of expansion is all expressed in the energy of translation of the projectile, we shall have approximately,

$$\frac{w v^2}{4480 g} = F W \tilde{\omega} \qquad (39)$$

in which w is the weight of the projectile and

$$W = W_2 - W_1$$

From (39) the muzzle velocity v may be computed when the

factor of effect is known; or, we may determine the factor of effect when the muzzle velocity has been measured by a chronograph. These two equations reduced to practical forms are the following:

$$v = 379.57 \sqrt{FW \frac{\tilde{\omega}}{w}} \quad . \quad . \quad . \quad (40)$$

and

$$F = 0.000006941 \frac{v^2 w}{W \tilde{\omega}}$$
 . . . (41)

As an illustrative application of these formulas to interior ballistics take the following data from Noble and Abel's second memoir, relative to the English 8-inch gun: It was found by firing a charge of 70 pounds of a certain brand of pebble powder, with a projectile weighing 180 pounds, that a muzzle velocity of 1694 foot-seconds was obtained. What was the factor of effect (F) pertaining to this gun and brand of powder? For this particular gun and charge we have $\tilde{\omega} = 70$ pounds, w = 180 pounds, $\Delta_1 = 0.1634$, $\Delta = 0.605$ and m = 1. In Noble and Abel's table of work (Table III) the first column gives values of $\frac{1}{\Delta_1}$, increasing by a common difference, while the second column contains the corresponding values of Δ_1 . By a simple interpolation we find for the values of Δ_1 and Δ given above, $W_2 = 99.4$ and $W_1 = 37.6$; whence W = 61.8 foot-tons. Substituting these values in Equation (41) we have

$$F = 0.000006941 \frac{180 \times (1694)^2}{70 \times 61.08} = 0.8287.$$

That is, the actual work realized, as expressed and measured by the projectile's energy of translation, as it emerges from the bore, is nearly 83 per cent. of the theoretical maximum work which the powder gases are capable of performing, leaving but 17 per cent. for the other work done by the gases, namely, the work expended upon the charge, the gun and carriage, and in

giving rotation to the projectile; the work expended in overcoming passive resistances, such as forcing the rotating band into the groove, the subsequent friction as the projectile moves along the bore, and the resistance of the air in front of the projectile; and lastly, the heat communicated to the walls of the gun. It is very difficult to evaluate these non-useful energies, but it is probable that they do not consume more than 17 per cent. of the maximum work of the gases. Longridge finds by an elaborate calculation that this lost work in a 10-inch B. L. Woolwich gun amounts to 30 per cent. of the maximum work; * but it is believed that he has greatly overestimated the work required to give motion to the products of combustion. Colonel Pashkievitsch makes the lost work rather less than 17 per cent. of that expressed in the energy of translation of the projectile.†

To test the correctness of Equation (40) for determining muzzle velocities we will apply it to the same gun by means of which the factor of effect was determined, increasing the charge from 70 to 90 pounds, and again to 100 pounds, and compare the computed velocities with those measured with a chronograph. For a charge of 90 pounds of powder we have $\Delta_1 = 0.210$ and $\Delta = 0.780$; whence $W_2 = 89.3$, $W_1 = 20.86$, and W = 68.44

$$v = 379.57 \sqrt{\frac{.8287 \times 90 \times 68.44}{180}} = 2021 \text{ foot-sec.}$$

The measured velocity with this charge was 2027 foot-seconds. In a similar way we find by the formula that for a charge of 100 pounds v=2174 foot-seconds, while the measured velocity was 2182 foot-seconds. The differences between the computed and observed velocities in these examples are about one-third of one

^{*&}quot;Internal Ballistics." By Atkinson Longridge. London, 1889. Chapter V.

^{† &}quot;Interior Ballistics." By Colonel Pashkievitsch. Translated from the Russian by Captain Tasker H. Bliss, U. S. Army. Washington, 1892.

per cent., and are well within the limits of probable error in measuring them.

The factor of effect increases with the caliber of the gun, as is shown by experiment. Thus with the English 10-inch gun fired with charges of 130 and 140 pounds of the pebble powder we have been considering, the factor of effect is 0.855; while with the 11-inch gun, and charge of 235 pounds, the factor of effect is 0.89.

CHAPTER III

COMBUSTION UNDER CONSTANT PRESSURE

Combustion of a Grain of Powder Under Constant Atmospheric Pressure.—In what follows it is assumed that the powder grain is of some regular geometrical form to which the elementary rules of mensuration can be applied. It will also be assumed as the result of observation, that the combustion of the grain takes place simultaneously on all sides and that, under the constant pressure of the atmosphere, parallel layers of equal thickness are burned away in equal successive intervals of time—that is, that the velocity of combustion under constant pressure is uniform.

The form and dimension of each grain of powder constituting the charge are of the utmost importance, as upon them depends the proper distribution of the mean effective pressure within the bore. If the initial surface of combustion of the charge be large and the web thickness of the grains small, then the maximum pressure will be excessive and the muzzle velocity inadequate. On the other hand, if the web thickness be too great the chase pressure may prove destructive to the gun. More than one of our heavy guns it is believed have been wrecked during the past ten years simply from excessive web thickness.

Many forms of grain have been adopted by different manufacturers in this and foreign countries, but they may all be divided into two general groups, viz.: those burning with a continuously decreasing surface, and those in which the surface of combustion may increase (or decrease) to a certain stage, the grain then breaking up into other forms entirely dissimilar to the original and which are then consumed with a rapidly decreasing surface. To the first group belong spherical, cubical, ribbon-shaped, and indeed all solid grains of whatever form,

and cylindrical grains with an axial perforation. To the latter group belong pierced prismatic and the so-called multiperforated grains employed by both our army and navy.

Notation.-Let

l = thickness of layer burned in time t.

 l_o = one-half the least dimension of the grain. Since combustion takes place on all sides of a grain at once, it may be assumed that when $l=l_o$ all grains of the first group are totally consumed. This, of course, is not the case with m. p. grains.

 S_o = the total initial surface of combustion of the grain.

S = surface of combustion at time t, corresponding to l.

S' = the total burning surface when $l = l_o$; that is, when the grain, as a grain, is about to disappear. This surface may be called the vanishing surface of combustion.

 V_o = the initial volume of a grain.

V =volume of grain burned at time t. That is, the volume comprised between the surfaces S_o and S.

k =fraction of grain burned in time t. That is, $k = \frac{V}{V_o}$.

The general expression for the burning surface of a grain of powder moulded into any one of the simple geometrical forms adopted by powder manufacturers may take the form,

$$S = S_o + a l + b l^2$$
 . . . (1)

where l is the thickness of layer burned from instant of ignition. At that instant l is zero and S the initial surface of combustion S_o . In the course of burning when l is about to become l_o , S is about to become S'. Therefore

$$S' = S_o + a l_o + b l_o^2$$
 (2)

In these two equations a and b are constants for the same form of grain, whose values will be deduced later.

The general expression for the volume consumed while a thickness l is burned away, is

$$V = \int_{o}^{l} S \, d \, l \,;$$

whence substituting for S its general value from (τ) and integrating,

$$V = S_0 l + \frac{a}{2} l^2 + \frac{b}{3} l^3 \qquad . \qquad . \qquad . \qquad . \qquad . \qquad . \qquad (3)$$

The initial volume V_o is evidently what V becomes when the grain is completely consumed, that is, when $l = l_o$. Therefore

This, of course, gives the entire original volume only for those grains which are completely consumed when $l=l_o$, or, in other words, when the web thickness is burned. It need hardly be said that it does not apply to m. p. grains. In this latter case, it gives the original volume minus the "slivers," so called.

$$S_a = S' - a l_0 - b l_0^2$$

If, in (4), we substitute for S_o its value from (2), namely,

it becomes

$$V_o = S' l_o - \frac{a}{2} l_o^2 - \frac{2b}{3} l_o^3 \quad . \quad . \quad . \quad . \quad (5)$$

From (4) and (5) we readily find

$$a = -\frac{2}{l_o}(2 S_o + S') + \frac{6 V_o}{l_o^2} \quad . \quad . \quad . \quad (6)$$

and

$$b = \frac{3}{l_o^2} (S_o + S') - \frac{6 V_o}{l_o^3} \quad . \quad . \quad . \quad (7)$$

These equations give a and b when S_o , S' and V_o can be computed by the rules of mensuration. It will be observed that a is a linear quantity while b is of zero order of magnitude. These properties afford tests, as far as they go, as to whether the work of deducing a and b in any particular case has been correctly performed.

Fraction of Grain Burned for any Value of 1.—We have by definition

$$k = \frac{V}{V_o} = \frac{S_o l + \frac{a}{2} l^2 + \frac{b}{3} l^3}{V_o}.$$

This may be transformed into

$$k = \frac{S_o \, l_o}{V_o} \cdot \frac{l}{l_o} \left\{ \ \mathbf{I} \ + \frac{a \, l_o}{2 \, S_o} \cdot \frac{l}{l_o} + \frac{b \, l_o^2}{3 \, S_o} \cdot \frac{l^2}{l_o^2} \right\}.$$

Put for convenience,

$$\frac{S_o l_o}{V_o} = \alpha; \ \frac{a l_o}{2 S_o} = \lambda; \ \frac{b l_o^2}{3 S_o} = \mu \ . \tag{8}$$

Then

$$k = \alpha \frac{l}{l_o} \left\{ 1 + \lambda \frac{l}{l_o} + \mu \frac{l^2}{l_o^2} \right\} \qquad (9)$$

For all grains of the first group k becomes unity when $l = l_o$, that is, when the grain is all burned; in this case (9) reduces to

$$I = \alpha (I + \lambda + \mu) (10)$$

This relation always subsists between these numerical constants and serves to test the correctness of their derivation in any case.

The following relations which are easily established will be useful:

$$S' = (1 + 2 \lambda + 3 \mu) S_o = S_o + \frac{\alpha V_o}{l_g} (2 \lambda + 3 \mu);$$
 (11)

or, more generally,

$$S = (1 + 2 \lambda \frac{l}{l_o} + 3 \mu \frac{l^2}{l_o^2}) S_o.$$

We also have

$$\alpha (\lambda + 2 \mu) = \frac{S' l_o}{V_o} - 1.$$

Therefore for all grains whose vanishing surface (S') is zero, we have

$$1 + 2\lambda + 3\mu = 0.$$

and

$$\alpha (\lambda + 2 \mu) = -1.$$
 . . . (12)

Applications.—We will now apply these formulas to a discussion of various forms of grain now in use or which may come into use.

1. Sphere.—For a spherical grain l_o is evidently the radius. Then by mensuration

$$S_o = 4 \pi l_o^2$$

$$S' = 0$$

$$V_o = \frac{4 \pi l_o^3}{3}$$

Substituting these in (6) and (7) we readily find

$$a = -8\pi l_0$$
 and $b = 4\pi$.

Therefore from (1)

$$S = 4 \pi l_o^2 - 8 \pi l_o l + 4 \pi l^2 = 4 \pi (l_o - l)^2;$$

and, therefore, S is a decreasing function of l.

From (8) we find

$$\alpha = 3, \lambda = -1 \text{ and } \mu = \frac{1}{3};$$

and these substituted in (9) give

$$k = 3\frac{l}{l_o} \left\{ \mathbf{1} - \frac{l}{l_o} + \frac{\mathbf{1}}{3} \frac{l^2}{l_o^2} \right\} = \mathbf{1} - \left(\mathbf{1} - \frac{l}{l_o} \right)^3$$

which is the fraction of grain burned in terms of the thickness of the layer l.

If we divide the thickness of web (radius of grain) into five equal parts the following table may be computed, which will be useful for comparing this form of grain with others to be given:

$\frac{l}{l_o}$.	k.	First Differences.
0.0	0.000	
0.2	0.488	0.488
0.4	0.784	0.296
0.6	0.936	0.152
0.8	0.992	0.056
1.0	1.000	0.008

The second column gives the entire fraction of grain burned and the third column the fraction of grain burned for each layer. It will be observed that nearly one-half the initial volume of the grain is in the first layer.

2. Parallelopipedon.—Let $2 l_o$ be the least dimension of the parallelopipedon and m and n the other two dimensions. Then, by the rules of mensuration,

$$S_o = 4 l_o m + 4 l_o n + 2 m n$$

 $S' = 2 (m - 2 l_o) (n - 2 l_o) = 2 m n - 4 l_o m - 4 l_o n + 8 l_o^2$
 $V_o = 2 l_o m n$

Substituting these values of S_o , S' and V_o in (6) and (7), gives

$$a = -8 (2 l_0 + m + n)$$
 and $b = 24$.

Making the following substitutions, viz.:

$$\frac{2 l_o}{m} = x$$
 and $\frac{2 l_o}{n} = y$

in which x and y are generally less than unity, we have, finally,

$$\alpha = \mathbf{1} + x + y; \ \lambda = -\frac{x + y + xy}{\alpha}; \ \mu = \frac{xy}{\alpha}.$$

It may be noted that these values of α , λ , μ satisfy equation (10).

There are three special parallelopipedons worthy of separate notice:

(a) Cube.—The cubical form has been used for ballistite and for some other powders. For this form we evidently have

$$x = y = 1$$
.

Therefore

$$\alpha = 3$$
; $\lambda = -1$; $\mu = \frac{1}{3}$.

These are the same as were found for spherical grains, as might have been inferred. They also apply approximately to sphero-hexagonal, mammoth and rifle powders (old style).

(b) Square Flat Grains.—For these grains (still used with certain rapid-firing guns), m and n are equal and greater than $2 l_o$. Therefore x and y are equal and less than unity. Therefore,

$$\alpha = 1 + 2x$$
; $\lambda = -\frac{x(2+x)}{\alpha}$; $\mu = \frac{x^2}{\alpha}$

If these grains are very thin, x becomes a very small fraction and may be omitted in comparison with unity. In this case λ and μ are approximately zero and α unity. This gives

$$k = \frac{l}{l_o}$$

or, a constant emission of gas during the burning; but the grain would be consumed in a very short interval of time.

(c) Grains Made into Long Slender Strips (or "Ribbons"), with Rectangular Cross-Section.—These grains are approximately those of the new English powder called "axite." Also of the French "B N" powders, and others. If we suppose the width of the strip to be five times, and the length one-hundred and fifty times, its thickness (which corresponds nearly with the

"B N" powders), we shall have
$$x = \frac{1}{5}$$
 and $y = \frac{1}{150}$. Therefore

$$\alpha = 1.207$$
; $\lambda = -0.172$; $\mu = 0.001$;

and the expression for k becomes

$$k = 1.207 \frac{l}{l_o} \left\{ 1 - 0.172 \frac{l}{l_o} + 0.001 \frac{l^2}{l_o^2} \right\}$$

The following table illustrates the progressiveness of this particular grain:

$\frac{l}{l_o}$.	<i>k</i> .	First Differences.		
0.0	0.000			
0.2	0.233	0.233		
0.4	0.450	0.217		
0.6	0.650	0.200		
0.8	0.833	0.183		
		0.167		
1.0	1.000	1.000		

These strips, made up into compact bundles or fagots to form the charge, seem well adapted for rapid-firing guns of moderate caliber. In the application of the expression for k for computing velocities and pressures in the gun, μ may be regarded as zero, and thus greatly shorten the calculations without impairing their accuracy.

If the cross-section of the strip is square, we shall have

$$m = 2 l_o, x = 1 \text{ and } y = \frac{2 l_o}{n},$$

n being the length of the strip. Therefore, in this case,

$$\alpha = 2 + y; \ \lambda = -\frac{1 + 2y}{2 + y}; \ \mu = \frac{y}{2 + y}.$$

If the strip be very long in comparison with the linear

dimension of cross-section, y may be considered zero, and we have

$$\alpha = 2$$
; $\lambda = -\frac{1}{2}$; $\mu = 0$.

Therefore

$$k = 2\frac{l}{l_o} \left\{ \mathbf{I} - \frac{\mathbf{I}}{2} \frac{l}{l_o} \right\} = \mathbf{I} - \left(\mathbf{I} - \frac{l}{l_o} \right)^2$$

- 3. Solid Cylinder.—For this form of grain there are two cases to be considered: (a) When the diameter of cross-section of the cylinder is the least dimension. (b) When the length of the cylinder is the least dimension. That is, a cylinder proper and a circular disk.
- (a) Cylinder Proper.—In accordance with the notation adopted, l_o will be the radius and m the length of the cylinder. We have by mensuration,

$$S_o = 2 \pi (l_o m + l_o^2); S' = 0; V_o = \pi l_o^2 m;$$

whence

$$a = -2\pi (4 l_o + m)$$
 and $b = 6\pi$.

Putting, as before, $\frac{2 l_o}{m} = x$, there results

$$\alpha = 2 + x; \ \lambda = -\frac{1 + 2x}{2 + x}; \ \mu = \frac{x}{2 + x}.$$

These are the same expressions for α , λ , μ as was found for a strip with square cross-section, as might readily be inferred.

If x be small in comparison with unity, that is, if the grains are long slender cylinders (thread like), like cordite, we have very approximately,

$$\alpha = 2; \lambda = -\frac{1}{2}; \mu = 0;$$

and, as before shown,

$$k = I - \left(I - \frac{l}{l_o}\right)^2$$

The fo	llowing	table	was	computed	bv	this	formula:
--------	---------	-------	-----	----------	----	------	----------

$\frac{l}{l_o}$.	k.	First Differences.
0.0	0.00	• • • •
0.2	0.36	0.36
0.4	0.64	0.28
0.6	0.84	0.20
0.8	0.96	0.12
,		0.04
1.0	1.00	I . 00

Comparing this table with that given for "strips," it will be seen that the burning of cordite is not so progressive as that of axite.

If the length of the solid cylindrical grain be the same as its diameter, then x = 1; and we have

$$\alpha = 3$$
; $\lambda = -1$; $\mu = \frac{1}{3}$

as for spherical and cubical grains.

(b) Circular Disk.—With this form of grain the thickness becomes the least dimension instead of the diameter. Let $2 l_o$ be the thickness of the disk and R its radius.

Then

$$S_o = 2 \pi R (2 l_o + R); S' = 2 \pi (R - l_o)^2; V_o = 2 \pi l_o R^2.$$

Whence

$$a = -4 \pi (2 R + l_0)$$
 and $b = 6 \pi$.

Therefore making $\frac{2l_o}{2R} = \frac{l_o}{R} = x$, we have, as has already

been found for square flat grains,

$$\alpha = 1 + 2x; \lambda = -\frac{x(2+x)}{1+2x}; \mu = \frac{x^2}{1+2x}$$

4. Cylinder with Axial Perforation.—Let R = radius of grain, r = radius of perforation, and m = its length. We then have

$$2 l_o = R - r,$$

and

$$R + r = 2 (R - l_o)$$
. $\therefore R^2 - r^2 = 4 l_o (R - l_o)$.

By the rules of mensuration, we find, after reduction,

$$S_o = 2 \pi m (R + r) + 2 \pi (R^2 - r^2) = 4 \pi (m + 2 l_o) (R - l_o).$$

 $S' = 4 \pi (m - 2 l_o) (R - l_o)$

$$V_o = 4 \pi l_o m (R - l_o)$$

Therefore

$$a = -16 \pi (R - l_0)$$
 and $b = 0$.

Making, as before, $x = \frac{2 l_0}{m}$ we have

$$\alpha = 1 + x; \lambda = -\frac{x}{1+x}; \mu = 0.$$

Therefore

$$k = (\mathbf{I} + x) \frac{l}{l_o} \left\{ \mathbf{I} - \frac{x}{\mathbf{I} + x} \frac{l}{l_o} \right\}$$

As an example of this form of grain, suppose the length to be three hundred times the thickness of web. Then

$$x = \frac{1}{300}$$
; $\alpha = \frac{301}{300}$; $\lambda = -\frac{1}{301}$; $\mu = 0$.

The expression for k is

$$k = \frac{301}{300} \frac{l}{l_o} \left\{ 1 - \frac{1}{301} \frac{l}{l_o} \right\} = \frac{l}{l_o} \left\{ \frac{301}{300} - \frac{1}{300} \frac{l}{l_o} \right\}$$

The following	table	was	computed	by	this	formula:
---------------	-------	-----	----------	----	------	----------

$\frac{l}{l_o}$.	<i>k</i> .	First Differences.
0.0	0.0000	
0.2	0.2005	0.2005
0.4	0.4008	0.2003
0.6	0.6008	0.2000
0.8	0.8005	0.1997
		0.1995
1.0	1.0000	1.0000

This form of grain is very progressive, much more so than any other form that has been proposed, and seems well adapted for guns of all calibers. The first differences show that for all practical purposes the emission of gas may be considered constant during the entire burning of the grain.

From (11) we have, when $\mu = 0$,

$$S' = (\mathbf{I} + 2\lambda) S_o = \frac{\mathbf{I} - x}{\mathbf{I} + x} S_o.$$

Therefore in this example, when $x = \frac{1}{300}$, we have

$$S' = \frac{299}{301} S_o,$$

and the burning surface during the entire combustion lies between its initial value S_o and its final value $\frac{299}{301}S_o$.

5. Multiperforated Grains.—These grains, which are used exclusively with the heavy artillery of the army and navy of the United States, are cylindrical in form and have seven equal longitudinal perforations, one of which coincides with the axis

of the grain, while the others are disposed symmetrically about the axis, their centres joined forming a regular hexagon. web thickness $(2 l_0)$ is the distance between any two adjacent circumferences; and therefore, if R is the radius of the grain and r the radius of each of the perforations, we have the relation

$$2 l_o = \frac{R - 3 r}{2}$$

From the geometry of the grain as defined above we have the following relations:

$$S_o = 2 \pi \left\{ R^2 - 7 r^2 + m (R + 7 r) \right\} . . . (13)$$

$$S' = S_o + 4 \pi l_o (3 m - 2(R + 7 r) - 9 l_o) . . (14)$$

$$S' = S_o + 4 \pi l_o (3 m - 2(R + 7 r) - 9 l_o)$$
 (14)

$$V_o = \pi \ m \left(R^2 - 7 \ r^2 \right) = \frac{m}{2} \left\{ S_o - 2 \ \pi \ m \left(R + 7 \ r \right) \right\} \tag{15}$$

$$V'_{o} = l_{o} S_{o} + 2 \pi l_{o}^{2} (3 m - 2 (R + 7 r) - 6 l_{o}) . \quad (16)$$

In these expressions S_o and V_o are the initial surface of combustion and volume, respectively, while S' and V'_{o} are the vanishing surface and volume burned, when l is about to become l_o and the grain to break up into slivers. If we substitute the values of S_o , S' and V'_o from the above equations in (6) and (7), they reduce to

$$a = 4 \pi (3 m - 2 (R + 7 r))$$

and

$$b = -36 \pi$$

These values of a and b, in equations (8), give

$$\alpha = \frac{R^2 - 7r^2 + m(R + 7r)}{R^2 - 7r^2 + (m - 2l_0)(R + 7r + 3l_0)}.$$
 (17)

$$\lambda = \frac{l_o \left\{ 3 \, m - 2 \, (R + 7 \, r) \right\}}{R^2 - 7 \, r^2 + m \, (R + 7 \, r)} \quad . \tag{18}$$

$$\mu = -\frac{6 l_o^2}{R^2 - 7 r^2 + m (R + 7 r)} \qquad . \qquad . \qquad . \qquad (19)$$

These values of α , λ , and μ satisfy the equation of condition

$$\alpha (1 + \lambda + \mu) = 1,$$

since when $l=l_o$ the volume V'_o has been consumed. When this occurs, the original form of the grain disappears and there remain twelve slender, three-cornered pieces with curved sides technically called "slivers." These of course must be treated differently. In the applications of these formulas given in Chapter V, the form characteristics of the slivers are assumed to be $\alpha=2$, $\lambda=\frac{1}{2}$ and $\mu=0$, with good results.

The form characteristics deduced in (17), (18), and (19), if substituted in (9), will give the fraction of volume V'_o burned. But what is required in practice is the fraction of the entire grain (or charge). This is found by employing V_o instead of V'_o . By this means we find

$$\alpha = 2 l_o \left\{ \frac{I}{m} + \frac{R + 7 r}{R^2 - 7 r^2} \right\}; \quad . \quad (20)$$

and this value of α will be used in all the applications. The expressions for λ and μ , being independent of the volume (see equations (8)), are those deduced above.

Substituting the form characteristics in (9) and making

$$l = l_o$$

we shall have the fraction of the entire grain burned when the web thickness is burned. Calling this fraction k' it will be found that

$$k' = \frac{2 l_o}{m} \left\{ 1 + \frac{(m - 2 l_o) (R + 7 r + 3 l_o)}{R^2 - 7 r^2} \right\} . \qquad (21)$$

This expression for k' would also be obtained by dividing (16) by (15).

It will be seen that for the same web thickness α and k' decrease as m increases, but within moderate limits, their limiting values, when m is infinite, being

$$\alpha = \frac{2 l_o (R + 7 r)}{R^2 - 7 r^2}$$

and

$$k' = \frac{2 l_o (R + 7 r + 3 l_o)}{R^2 - 7 r^2}$$

For the grains employed in the United States service, the ratio $\frac{R}{r}$ varies but little from 11. If we adopt this ratio, the expressions for the form characteristics α , λ , μ and k' become

$$\alpha = \frac{12}{19} + \frac{2 l_o}{m}$$

$$\lambda = \frac{2 (m - 6 l_o)}{19 l_o + 6 m}$$

$$\mu = -\frac{4 l_o}{19 l_o + 6 m}$$

$$k' = \frac{16}{19} + \frac{6 l_o}{19 m}$$

We also have $R = 5.5 l_o$ and $r = 0.5 l_o$.

If, in addition, we make $m = n l_o$ we have

$$\alpha = \frac{12}{19} + \frac{2}{n}$$

$$\lambda = \frac{2(n-6)}{6n+19}$$

$$\mu = -\frac{4}{6n+19}$$

$$k' = \frac{16}{19} + \frac{6}{19n}$$

It may be noted that the limiting values of these form characteristics, as the length of the grain is indefinitely increased, are,

$$\alpha = \frac{12}{19}$$
; $\lambda = \frac{1}{3}$; $\mu = 0$ and $k' = \frac{16}{19}$.

Also that λ is zero when n = 6 and becomes negative when the grain is still further shortened.

It will be seen that the percentage of slivers can never be greater than about 16.

For the grains in use n is approximately 26, which gives

$$\alpha = \frac{175}{247} = 0.70850$$

$$\lambda = \frac{8}{35} = 0.22857$$

$$\mu = -\frac{4}{175} = -0.022857$$

$$k' = \frac{211}{247} = 0.85425$$

There seems to be no valid reason why these, or other simple ratios, for R/r and m/l_o should not be adopted by powder manufacturers for all sizes of m.p. grains, making the diameter of the grain and perforations, and also its length, depend upon the web thickness adopted for a particular gun. For example, the web thickness adopted for the 14-inch gun is 0.1454 inch. Therefore the dimensions of the grains would be

Diameter =
$$5.5 \times 0.1454 = 0.7997$$
 in.
Diameter of perforation = $0.1454/2 = 0.0727$ in.
Length = $13 \times 0.1454 = 1.89$ in.

These dimensions are practically the same as those of the actual grains. From equation (26') of this chapter it will be seen that the initial surface of one pound of these grains would vary inversely as the web thickness.

For these grains, equations (13) to (16) reduce to

$$S_o = 525 \pi l_o^2$$

 $S' = 729 \pi l_o^2$
 $V_o = 741 \pi l_o^3$
 $V'_o = 633 \pi l_o^3$

The vanishing surface is therefore about 39 per cent. greater than the initial surface.

Captain Hamilton has shown conclusively that the m.p. grains now in use are much too short to secure a proper alignment in the powder chamber, and that this lack of alignment conduces to excessive pressure.*

If we make n = 200, that is, make the length of the grains 100 times the web thickness, we should have

$$\alpha = 0.64158$$
 $\lambda = 0.31829$
 $\mu = -0.00328$
 $k' = 0.84368$

This value of n would make the length of the grains for the 14-inch gun 14.52 inches; which would not only secure a good alignment of the grains in the containing bag, but would also give a much less initial surface of combustion to the charge and would thus reduce the maximum pressure.

The general expression for the surface of combustion of an m.p. grain with 7 perforations, in terms of the thickness of web burned, is by (1),

$$S = S_o + 4\pi (3m - 2(R + 7r)) l - 36\pi l^2$$

Differentiating twice, we have

$$\frac{ds}{dl} = 4\pi (3 m - 2 (R + 7 r)) - 72\pi l$$

$$\frac{d^2 s}{dl^2} = -72\pi$$

There is, therefore, a maximum value of S which occurs when

$$l = \frac{3 m - 2 (R + 7 r)}{18}$$

^{*} Journal U. S. Artillery, July-August, 1908, page 9.

and the maximum surface of combustion is

$$S_{max.} = S_o + \frac{\pi (3 m - 2 (R + 7 r))^2}{9}$$

From these formulas are easily deduced the following:

- 1. When 3m 2(R + 7r) = 0, S is a decreasing function of l during the entire burning of the web thickness.
- 2. When 3m 2(R + 7r) is equal to, or greater than, 18 l_0 the grain burns with an increasing surface.
- 3. When 3m 2(R + 7r) lies between 0 and 18 l_o the surface of combustion is at first increasing and then decreasing.

Expression for Weight of Charge Burned.—If we assume that the entire charge is ignited at the same instant, which is practically the case with an igniter at both ends of the cartridge, the combustion of the charge will be expressed by the same function that applies to a single grain. Therefore if y is the weight of the charge burned at any period of the combustion and $\tilde{\omega}$ the weight of the entire charge, we may assume the equality (since the weights are proportional to the volumes)

$$k = \frac{y}{\tilde{\omega}} = \alpha \frac{l}{l_o} \left\{ 1 + \lambda \frac{l}{l_o} + \mu \frac{l^2}{l_o^2} \right\} \quad . \quad (22)$$

In this equation α is always positive from its definition, viz: $\alpha = \frac{S_o \, l_o}{V_o}$. It varies in value from 3 (spheres and cubes) to less than unity (service multiperforated grains). The smaller α is, cateris paribus, the less will be the maximum pressure for a given charge. Of the other characteristics, λ and μ , either may be positive, negative, or zero, but not both at the same time.

Expressions for Initial Volume and Surface of Combustion of a Charge of Powder.—Let N be the number of grains in unit weight of powder, V' the volume of unit weight of water, and δ the specific gravity of the powder. Then, from the definition of specific gravity,

$$\delta = \frac{V'}{N V_o}, \qquad (23)$$

since we may assume that the weights are proportional to the volumes. The number of grains in unit weight of powder can be counted, and, with the carefully moulded grains now in use, V_o can be calculated with great accuracy. Thus (23) can be employed to determine the specific gravity of a powder when it is not given by the manufacturer, as is usually the case. For the large grains designed for seacoast guns the number of grains in 100 units should be counted, estimating the fraction of a grain in excess. For small-arms powder, if the specific gravity of the mass of which the grains are made is known, the number of grains in unit weight may be computed by the formula

$$N = \frac{V'}{\partial V_o} (24)$$

The units to be used in these and other formulas that will be deduced will be considered later.

Initial Surface of Unit Weight of Powder and of the Entire Charge.—Let S_1 be the initial surface of the grains of unit weight of powder. Then if S_o is the surface of one grain, we have, by $\binom{24}{2}$

$$S_1 = N S_o = \frac{V' S_o}{\delta V_o}. \qquad (25)$$

But by (8)

$$S_o = \frac{\alpha V_o}{l_o}$$

Therefore

$$S_{i} = \frac{\alpha V'}{\delta l_{o}} = \frac{\alpha N V_{o}}{l_{o}}, \qquad (26)$$

for one unit weight of powder; and for $\tilde{\omega}$ units weight,

$$S_{\tilde{\omega}} = \frac{\alpha \ \tilde{\omega} \ V'}{\delta \ l_o} = \frac{\alpha \ \tilde{\omega} \ N \ V_o}{l_o} \quad . \quad . \quad . \quad (26')$$

This simple formula was first published in the Journal U. S. Artillery for November-December, 1905. It shows that for two charges of equal weight and made up of grains of the same density and thickness of web, but of dissimilar forms, the entire surfaces of all the grains in the two charges are proportional to the corresponding values of α . It also shows that if the initial surfaces of two charges of equal weight but made up of grains of dissimilar forms, are to be the same, the web thicknesses must be inversely as the values of α . For example, if the two charges are made up, the one of cubes and the other of long slender cylinders (axite and cordite), the web thickness of the former must be one-half greater than the latter to obtain the same initial surface for each charge. These principles are important since the maximum pressure in a gun varies very nearly with the initial surface of the charge.

Volume of Entire Charge.—Let $V_{\hat{\omega}}$ be the volume of a charge of $\tilde{\omega}$ units weight supposed to be reduced to a single homogeneous grain. For a single grain of unit weight (23) gives

$$V_o = \frac{V'}{\delta}$$

and for $\tilde{\omega}$ units

$$V_{\hat{\omega}} = \frac{\tilde{\omega} V'}{\delta} \quad . \quad . \quad . \quad . \quad (27)$$

Gravimetric Density.—Gravimetric density is the density of a charge of powder when the spaces between the grains are considered. It is measured by the ratio of the weight of any given volume of the powder grains to the weight of the same volume of water. Since one pound of water fills 27.68 cubic inches we may say that the gravimetric density of a powder is the weight in pounds of 27.68 cubic inches of the powder not pressed together except by its own weight. Or, if we take a cubic foot as the unit and designate the gravimetric density by γ , the weight of a cubic foot of the powder grains by $\tilde{\omega}'$, and by w the weight of

a cubic foot of water, we shall have by definition,

$$\gamma = \frac{\tilde{\omega}'}{w} = \frac{\tilde{\omega}'}{1728/27.68} = \frac{\tilde{\omega}'}{62.427}.$$

It is evident that γ will vary not only with the density of the individual grains but also with the volume of the interstices between them; and this latter varies with the general form of the grains, or, in other words, with their ability to pack closely or the reverse. It is evident that the maximum value of γ is the weight of a cubic foot of solid powder, in which case the above ratio would be the specific gravity of the powder, designated by δ . The gravimetric density is therefore always less than the specific gravity. For modern powders gravimetric density is of very little importance.

Density of Loading.—Density of loading is defined to be the "ratio of the weight of charge to the weight of a volume of water just sufficient to fill the powder chamber." Let Δ be the density of loading and V_c the volume of the powder chamber. Since V' is the volume of unit weight of water it is evident that V_c/V' is the weight of a volume of water equal to the volume of the chamber. Hence by definition,

$$\Delta = \frac{\tilde{\omega} V'}{V_c} \qquad (28)$$

From (27) we have,

$$\tilde{\omega} = \frac{\delta V_{\hat{\omega}}}{V'}$$

and this substituted in (28), gives

$$\Delta = \frac{\delta V_{\hat{\omega}}}{V_c} \quad . \quad . \quad . \quad . \quad (29)$$

From this last equation the density of loading may be defined as the ratio of the volume of the powder grains supposed to be reduced to a single grain, to the volume of the chamber, multiplied by the density of the powder. If $V_{\hat{\omega}} = V_c$, that is, if the

chamber is filled by a single grain, then $\Delta = \delta$; and this is the superior limit of density of loading. The inferior limit is, of course, zero, namely, when $V_{\hat{\omega}} = 0$. If the density of loading is unity it follows from (28), that

$$\tilde{\omega} = \frac{V_c}{V'};$$

that is, the weight of charge equals the weight of water that would fill the chamber.

Reduced Length of Initial Air Space.—By initial air space is meant that portion of the volume of the chamber not occupied by the powder grains constituting the charge. The reduced length of the initial air space is the length of a cylinder whose cross-section is the same as that of the bore, and whose volume is equal to the initial air space. Denote this length by z_o and the area of cross-section of the bore by ω . Then as $V_c - V_{\hat{\omega}}$ is the volume of the air space we have

$$z_o = \frac{V_c - V_{\hat{\omega}}}{\omega}$$

Substituting for V_c and $V_{\hat{\omega}}$ their values from (28) and (27), we have

$$z_o = \frac{\tilde{\omega} \, V'}{\omega} \Big(\frac{\mathrm{I}}{\Delta} - \frac{\mathrm{I}}{\delta} \Big).$$

Put

$$a = \frac{1}{\Lambda} - \frac{1}{\delta} = \frac{\delta - \Delta}{\Lambda \delta}$$
.

Then.

$$z_o = \frac{a \,\tilde{\omega} \,V'}{\omega} \quad . \qquad . \qquad . \qquad . \qquad (30)$$

Working Formulas for English and French Units.—The English units used with formulas (23) to (30), inclusive, are the pound and inch. Therefore

V' = 26.78 cubic inches, nearly.

The French units employed with the same formulas are the kilogramme and decimetre. For these units we have

$$V' = 1$$
 cubic decimetre.

The two sets of formulas in working form are therefore:- .

English Units			French Units		
$\delta = \frac{27.68}{N V_o}$	(23')		$\delta = \frac{\mathrm{I}}{N V_o}$	(23")	
$N = \frac{27.68}{\delta V_o}$	(24')		$N = \frac{\mathbf{I}}{\partial V_o}$	(24")	
$S_{\tilde{\omega}} = \frac{27.68 \alpha}{\delta l_o} \frac{\alpha}{\delta}$	(26')		$S_{\hat{\omega}} = \frac{\alpha \ \tilde{\omega}}{\partial \ l_o}$	(26")	
$V_{\hat{\omega}} = \frac{27.68 \ \tilde{\omega}}{\delta}$	(27')		$V_{\hat{\omega}} = \frac{\tilde{\omega}}{\delta}$	(27")	
$\Delta = \frac{27.68 \ \tilde{\omega}}{V_c}$	(28')		$\Delta = \frac{\tilde{\omega}}{V_c}$	(28")	
$z_o = \frac{27.68 a \tilde{\omega}}{\omega}$	(30′)		$z_o = \frac{a \ \tilde{\omega}}{\omega}$	(30")	

EXAMPLES

1. Compute the number of grains in a pound of the powder used with the service magazine rifle. Also the initial surface.

The grains of this powder are pierced cylinders of the following dimensions:

$$R = 0.$$
".045
 $r = 0.015$ \therefore 2 $l_o = 0.$ ".03
 $m = \frac{1}{21}$ in.
 $\delta = 1.65$
 $\alpha = 1.63$
 $\tilde{\omega} = 1$

From (24') and (26'), we have

$$N = \frac{27.68}{4 \pi l_o m (R - l_o) \delta} = 62300$$

$$S_1 = \frac{27.68 \times 1.63}{1.65 \times 0.015} = 1823 \text{ in.}^2$$

It is officially stated that the number of grains per pound varies from 83,000 to 93,000. This discrepancy is partly due to shrinkage and partly to the breaking and chipping of the grains. Possibly also to the method of counting.

2. What is the entire initial surface of a charge of 70 lbs. of the m.p. powder designed for the 8-inch rifle? For this powder we have

$$R = o''.256$$
; $r = o.''.0255$; $m = 1''.029$; $\delta = 1.58$
 $l_o = \frac{R - 3r}{4} = o''.044875$; $\alpha = 0.72667$
 $\therefore S_{\tilde{\omega}} = \frac{27.68 \times 0.72667 \times 70}{1.58 \times 0.044875} = 19813 \text{ in.}^2$

3. Suppose the powder of example 2 to be made into cubes having the same thickness of web. What would be the initial surface of the charge?

For a cube $\alpha = 3$. Therefore

$$S_{\hat{\omega}} = \frac{3 \times 19813}{0.72667} = 81795 \text{ in.}^2$$

To make the initial surface of the latter charge the same as the former the web thickness would have to be

$$2 l_o = \frac{3 \times 0.08975}{0.72667} = 0''.37$$

4. The volume of the chamber of the 12-inch rifle is 17487 cubic inches. If the charge is 400 lbs. what is the density of loading?

Ans.: $\Delta = 0.633$.

CHAPTER IV

COMBUSTION AND WORK OF A CHARGE OF POWDER IN A GUN

It has been established by experiment that a grain of modern powder burns in concentric, parallel layers, and that the velocity of combustion under constant pressure is uniform. Let l_o be one-half the web thickness of a grain and τ the time of burning this thickness under the constant pressure of the atmosphere. We then have, since the web burns on both sides,

$$\frac{l_o}{\tau}$$
 = velocity of combustion = constant = v_c (say). (1)

In the bore of a gun, however, the pressure surrounding the grain is very far from being constant and greatly exceeds the atmospheric pressure. All writers on interior ballistics agree that the velocity of combustion may be regarded at each instant as proportional to some power of the pressure; but they differ widely among themselves as to what this power is. Sainte-Robert, Vieille, Gossot, and Liouville give reasons (based, however, upon experiments made with a small quantity of powder exploded in an eprouvette of a few cubic inches capacity) for adopting the exponent $\frac{2}{3}$. Centervall makes the exponent $\frac{9}{10}$ for "Nobel N K" powder. Sebert and Hugoniot, from observations of the recoil of a 10-cm. gun mounted on a free-recoil carriage, deduced a law of burning directly proportional to the pressure. This law is the most simple of all and allows an easy and complete integration of the equations entering into the problem.* But simple as is this law of Sebert and Hugoniot,

^{*} See Journal U. S. Artillery, vol. 7, pp. 62-82.

we prefer to make use of Sarrau's law of the square root of the pressure, because the resulting formulas are easily worked and give results which "agree very well with facts" as stated by Sarrau, and as has been repeatedly shown by the writer and others.

Sarrau's law of burning under a variable pressure p leads directly to the equation,

$$\frac{d\,l}{d\,t} = \frac{l_o}{\tau} \left(\frac{p}{p_o}\right)^{\frac{1}{2}} \quad . \quad . \quad . \quad . \quad (2)$$

in which p_o is the atmospheric pressure and l the thickness of layer burned in time t.

It will be assumed that the variable pressure p in the bore is measured by the energy of translation imparted to the projectile (which is many times the sum of all the other energies entering into the problem); and it will be taken for granted that all the other work done by the expansion of the powder gas may be accounted for by giving suitable values to the constants so as to satisfy the firing data by means of which they are determined. This procedure will be fully illustrated further on.

If p is the variable pressure per unit of surface upon the base of the projectile at any instant, ω the area of the base, and u the corresponding distance travelled by the projectile from its firing seat, we have from the principle of energy and work,

$$\omega p = \frac{w}{g} \frac{d^2 u}{d t^2}$$

in which w is the weight of the projectile.

But from mechanics and calculus,

$$\frac{d^2 u}{d t^2} = \frac{d v}{d t} = \frac{d v}{d u}. \quad \frac{d u}{d t} = v \frac{d v}{d u} = \frac{1}{2} \frac{d (v^2)}{d u},$$

in which v from now on represents velocity.

Therefore

Combining (2) and (3), we have

$$\frac{dl}{dt} = \frac{l_o}{\tau} \left(\frac{w}{2 g \omega \rho_o} \right)^{\frac{1}{2}} \left(\frac{d (v^2)}{d u} \right)^{\frac{1}{2}} \quad . \quad . \quad (4)$$

Since z_o is the reduced length of the initial air-space and u the distance travelled by the projectile from its firing seat, we may say very approximately, by the principle of the covolume, that u/z_o is the number of volumes of expansion of the gas during the travel u, and this whether the charge is all converted into gas or not. If we make $u/z_o = x$, and therefore $du/dx = z_o$, (3) and (4) become

$$\omega p = \frac{w}{2 g z_o} \cdot \frac{d (v^2)}{d x} \qquad . \qquad . \qquad . \qquad . \qquad (5)$$

and

$$\frac{d l}{d t} = \frac{l_o}{\tau} \left(\frac{w}{2 g \omega p_o z_o} \right)^{\frac{1}{2}} \left(\frac{d (v^2)}{d x} \right)^{\frac{1}{2}} \quad . \quad . \quad (6)$$

It will be seen that (6) connects the velocity of burning of the grain with the velocity of travel of the projectile in the bore. It will be better to make x the independent variable in the first member as well as the second.

We have from calculus,

$$\frac{dl}{dt} = \frac{dl}{dx} \cdot \frac{dx}{du} \cdot \frac{du}{dt} = \frac{v}{z_0} \cdot \frac{dl}{dx}$$

Therefore, substituting in (6),

$$\frac{d l}{d x} = \frac{l_o}{\tau} \left(\frac{w z_o}{2 \varrho \omega p_o} \right)^{\frac{1}{2}} \left(\frac{d (v^2)}{d x} \right)^{\frac{1}{2}} \frac{1}{v}$$

Integrating between the limits o and x, we have

$$\frac{l}{l_o} = \frac{1}{\tau} \left(\frac{w z_o}{2 g \omega p_o} \right)^{\frac{1}{2}} \int_o^x \left(\frac{d (v^2)}{d x} \right)^{\frac{1}{2}} \frac{1}{v} d x \qquad (7)$$

In order to perform the integration indicated, we must know the relation existing between v and x, that is between the velocity of the projectile in the bore at any instant and the corresponding number of volumes of expansion of the gas. We get this relation from (19) Chapter II, which is

$$v^2 = \frac{6 g f y}{w} \left\{ 1 - \frac{1}{(1+x)^{\frac{1}{3}}} \right\}$$
 (8)

From this equation we deduce by simple differentiation

$$\frac{d(v^2)}{dx} \cdot \frac{1}{v} = \frac{1}{\sqrt{3}(1+x)^{\frac{1}{2}}\sqrt{(1+x)^{\frac{1}{3}}-1}}$$

Substituting this in (7) and making

$$X_o = \int_0^x \frac{dx}{\sqrt{(1+x)^{\frac{1}{2}}}\sqrt{(1+x)^{\frac{1}{2}} - 1}}, \quad . \quad . \quad (9)$$

we have

$$\frac{l}{l_o} = \frac{\mathrm{I}}{\tau} \left(\frac{w \, z_o}{6 \, g \, \omega \, \dot{p}_o} \right)^{\frac{1}{2}} X_o \, . \qquad . \qquad . \qquad . \tag{10}$$

It will be observed that X_o is a function of a ratio and is independent of any unit, and may therefore be tabulated with x as the argument.

If we put

$$K = \frac{1}{\tau} \left(\frac{w z_o}{6 g \omega \rho_o} \right)^{\frac{1}{2}} \quad . \qquad . \qquad . \qquad . \qquad (11)$$

we have

$$\frac{l}{l_o} = K X_o \quad . \quad . \quad . \quad . \quad (12)$$

Substituting the value of l/l_o from (12) in (22), Chapter III, we have

$$k = \frac{y}{\tilde{\omega}} = \alpha K X_o (1 + \lambda K X_o + \mu (K X_o)^2)$$
 . (13)

an equation which gives the fraction of the charge burned at any instant in terms of the volumes of expansion of the gases generated. When the powder is all burned in the gun (if it be all burned before the projectile leaves the bore), we have $y = \tilde{\omega}$ and $l = l_o$.

If, therefore, we distinguish X_o by a dash when $l = l_o$, (12) becomes

$$K\overline{X_o} = \mathfrak{1}, \ldots (\mathfrak{1}_4)$$

and (13) reduces to

$$I = \alpha (I + \lambda + \mu)$$

a fundamental relation established in Chapter III.

Substituting the value of K from (14) in (13), we have, while the powder is burning, the relation

$$\frac{y}{\tilde{\omega}} = \alpha \frac{X_o}{\overline{X}_o} \left\{ \mathbf{1} + \lambda \frac{X_o}{\overline{X}_o} + \mu \left(\frac{X_o}{\overline{X}_o} \right)^2 \right\} . \qquad (15)$$

Expression for Velocity of Projectile while the Powder is Burning.—Substituting the value of y from (15) in (8) and making

$$X_1 = X_o \left(\mathbf{I} - \frac{\mathbf{I}}{(\mathbf{I} + x)^{\frac{1}{3}}} \right)$$
 . . . (16)

we have

$$v^{2} = 6 g \alpha f \frac{\tilde{\omega}}{w} \cdot \frac{X_{1}}{\overline{X}_{o}} \left\{ \mathbf{1} + \lambda \frac{X_{o}}{\overline{X}_{o}} + \mu \left(\frac{X_{o}}{\overline{X}_{o}} \right)^{2} \right\}. \quad (17)$$

This equation holds only while the powder is burning and ceases to be true when $X_o > \overline{X}_o$.

Velocity of Projectile when $y = \tilde{\omega}$. When $X_o = \overline{X}_o$ and, therefore, $X_1 = \overline{X}_1$, equation (17) reduces to

$$\overline{v}^{2} = 6 g f \frac{\tilde{\omega}}{w} \frac{\overline{X}_{1}}{\overline{X}_{2}} \alpha (\mathbf{I} + \lambda + \mu);$$

or, since

$$\alpha (\mathbf{1} + \lambda + \mu) = \mathbf{1},$$

it becomes

$$\overline{v}^2 = 6 g f \frac{\tilde{\omega}}{w} \frac{\overline{X}_1}{\overline{X}_o}$$

This equation is, of course, the same as (8) from which it is

derived as is evident from (16). Putting

$$\frac{\overline{X}_1}{\overline{X}_o} = \overline{X}_2,$$

or, generally,

$$\frac{X_1}{X_2} = X_2, \dots (18)$$

the expression for \overline{v}^2 becomes

$$\overline{v}^2 = 6 g f \frac{\tilde{\omega}}{w} \overline{X}_2 \qquad . \qquad . \qquad . \qquad . \qquad (19)$$

It should be remembered that all symbols employed in this work affected with a dash refer to the position of the projectile, either in the bore on in the bore prolonged, when the powder has all been burned, and therefore where $y = \tilde{\omega}$.

From (19), we have

$$6 g f = \frac{\overline{v^2}}{\overline{X_2}} \frac{w}{\tilde{\omega}} ; \dots \qquad (20)$$

and this substituted in (17), gives

$$v^{2} = \alpha \overline{v}^{2} \frac{X_{1}}{\overline{X}_{1}} \left\{ \mathbf{I} + \lambda \frac{X_{o}}{\overline{X}_{o}} + \mu \left(\frac{X_{o}}{\overline{X}_{o}} \right)^{2} \right\} \qquad (21)$$

For convenience, put

$$M=rac{\alpha \overline{v}^2}{\overline{X}_1}; \ N=rac{\lambda}{\overline{X}_0}; \ N'=rac{\mu}{\overline{X}_0^2}=rac{\mu}{\lambda^2}N^2.$$

Then, finally, while the powder is burning,

$$v^2 = M X_1 \{ I + N X_o + N' X_o^2 \}$$
 . . (22)

Velocity of Projectile after Powder is all Burned.—The velocity of the projectile after the powder is all burned is given by (8), substituting $\tilde{\omega}$ for y. Reducing by means of (16), (18), and (20), and denoting velocity after the powder is all burned by capital V, equation (8) becomes

$$V^2 = \bar{v}^2 \frac{X_2}{\overline{X}_2}$$
 (23)

The velocity then after the powder is all burned varies directly as the square root of X_2 . From (16) and (18), we have

$$X_2 = I - \frac{I}{(I+x)^{\frac{1}{3}}} \dots \dots (24)$$

and therefore the superior limit as x (or u) increases indefinitely is unity. On this supposition (23) becomes

$$V^2 = \frac{v^2}{\overline{X}_2} = V_1^2 \text{ (say)} \qquad (25)$$

We may regard V_1 then as the theoretical limiting velocity after an infinite travel. In terms of V_1 (23) becomes

$$V^2 = V_1^2 X_2 \dots (26)$$

Since from $(25)^{\overline{v}^2} = \overline{X}_2 V_1^2$, therefore

$$\frac{\bar{v}^2}{\bar{X}_1} = \frac{\bar{X}_2 V_1^2}{\bar{X}_1} = \frac{V_1^2}{\bar{X}_0} \text{ (by (18))}$$

and, therefore,

$$M = \frac{\alpha V_1^2}{\overline{X}_0} \quad . \quad . \quad . \quad (27)$$

Pressure on Base of Projectile while Powder is Burning.—Differentiating (17) with respect to the independent variable x and putting for simplicity

$$X_3 = \frac{dX_1}{dx}$$
; $X_4 = X_0 + \frac{X_1}{X_3} \frac{dX_0}{dx}$; $X_5 = X_0^2 + \frac{2X_0X_1}{X_3} \frac{dX_0}{dx}$,

we have

$$\frac{d(v^2)}{dx} = M X_3 \{ 1 + N X_4 + N' X_5 \}.$$

Therefore, from (5)

$$p = \frac{w}{2 g \omega z_0} M X_3 \{ 1 + N X_4 + N' X_5 \}.$$

Combining the constants outside the brackets into one multiplier by making

we have the following expression for the pressure per unit of surface, on the base of the projectile:—

$$p = M' X_3 \{ \mathbf{1} + N X_4 + N' X_5 \} . . . (29)$$

Pressure after the Powder is all Burned.—Differentiating (26) with reference to x and substituting the differential coefficient in (5) we have, employing capital P to express pressure in this case,

$$P = \frac{w V_1^2}{2 \varrho \omega z_0} \frac{d X_2}{d x}.$$

But from (24)

$$\frac{dX_2}{dx} = \frac{1}{3(1+x)^{\frac{4}{3}}};$$

whence, putting

$$\frac{w V_1^2}{6 g \omega z_o} = P' \qquad (30)$$

we have finally

$$P = \frac{P'}{(\mathbf{I} + x)^{\frac{2}{3}}}. \qquad . \qquad . \qquad . \qquad (31)$$

If we make x = 0 in (31), we have

$$P = P'_{\cdot} . \qquad . \qquad . \qquad . \qquad . \qquad . \qquad (32)$$

Therefore P' is the pressure per unit of surface at the origin supposing the powder to be all burned before the projectile moves from its seat.

Relation Between f and P'.—From (19) and (25) we get

$$f = \frac{wV_1^2}{6 g \tilde{\omega}} \qquad (33)$$

Combining this with (30) there results

$$P' = \frac{\tilde{\omega} f}{z_o \omega} \quad . \quad . \quad . \quad . \quad (34)$$

Since f is (at least theoretically) the pressure per unit of surface of the gases of one pound of powder at temperature of combustion, occupying unit volume, it follows from (34) that

P' is the pressure per unit of surface of the gases of $\tilde{\omega}$ pounds of powder (the entire charge), occupying a volume equal to the initial air space z_o ω , as has already been shown by equation (32). Equation (31) is, therefore, the equation of the pressure curve upon the supposition that the charge is all converted into gas before the projectile has moved from its seat. From equation (30'), Chapter III, we have,

$$z_o \omega = \frac{27.68}{1728} a \tilde{\omega}$$
 cubic feet.

Therefore, from (34)

$$P' = \frac{1728}{27.68} \frac{f}{a} : \dots$$
 (35)

Values of the X Functions.—These values may be most easily and simply expressed by means of auxiliary circular functions. Thus let

$$(1 + x)^{\frac{1}{6}} = \sec \phi$$
 (36)

Then, by trigonometry,

$$\sin^2 \phi = I - \frac{I}{(I+x)^{\frac{1}{3}}} = X_2.$$
 (from (22))

and

$$\tan \phi = \sqrt{(1+x)^{\frac{1}{3}} - 1}$$

Also

$$dx = 6\sec^6\phi\tan\phi\,d\phi$$

Substituting these values in the expression for X_o we have

$$X_o = 6 \int_o^{\phi} \sec^3 \phi \, d\phi \quad . \tag{37}$$

Integrating, we have

$$X_o = 3 \sec \phi \tan \phi + 3 \log_e (\sec \phi + \tan \phi) \qquad . \tag{38}$$

By substituting the values of $\sec \phi$ and $\tan \phi$ given above in (38), we get an expression for X_o in terms of x. But it is of no practical interest. The definite integral in (37) is a well-known function of ϕ and has been extensively used in exterior ballistics.

A table of this function computed for every minute of arc up to 87° was first published by Euler, and has recently been reprinted (1904) at the Government Printing Office and issued as "Supplement No. 2, to Artillery Circular M." By means of this table it is easy to compute X_o for any value of x. First compute ϕ by (36), and then take the definite integral corresponding to ϕ , which has been symbolized by (ϕ) , from the table just mentioned. We then have

$$X_o = 6 (\phi)$$
 (39)

Since

$$X_2 = \sin^2 \phi$$
. (40)

we have from (18)

$$X_1 = X_o \sin^2 \phi \qquad . \qquad . \qquad . \qquad . \qquad . \qquad (41)$$

By definition

$$X_3 = \frac{d X_1}{d x}$$

But from (41)

$$\frac{dX_1}{dx} = \frac{dX_0}{dx}\sin^2\phi + 2X_0\sin\phi\cos\phi\frac{d\phi}{dx}$$

From (9) and (36) we deduce

$$\frac{dX_o}{dx}\sin^2\phi = \sin\phi\cos^4\phi.$$

Also, from what precedes,

$$2 \sin \phi \cos \phi \frac{d \phi}{d x} = \frac{\sin \phi \cos \phi}{3 \sec^6 \phi \tan \phi} = \frac{\cos^8 \phi}{3}.$$

Therefore

$$X_3 = \sin\phi \cos^4\phi + \frac{1}{3}X_o\cos^8\phi.$$

Let

$$X = \frac{1}{1 + \frac{1}{3} X_o \cos^4 \phi \csc \phi} \quad \text{wher}$$

Then we have Lang!

$$X_3 = \frac{\sin\phi\cos^4\phi}{X} \quad . \quad . \quad . \quad . \quad (42)$$

From the foregoing equations we find

$$\frac{X_1}{X_3} \frac{d X_o}{d x} = X X_o$$

by means of which are easily deduced from the definitions of X_4 and X_5 the following simple equations:—

$$X_4 = X_o (1 + X) \dots (43)$$

and

$$X_5 = X_o^2 (\mathbf{1} + 2 X)$$
 . . . (44)

By means of equations (39), (40), (41), (42), (43) and (44), the table of the logarithms of the X-functions given at the end of the volume was computed.

Some Special Formulas. Dividing (21) by (15) and reducing by (25), we have, since $y/\tilde{\omega} = k$,

$$v^2 = k V^2 = k V_1^2 X_2 \dots (45)$$

That is, the velocity of the projectile at any travel before the charge is all burned is equal to what the velocity would have been at the same travel had all the charge been converted into gas before the projectile moved, multiplied into the square root of the fraction of charge burned.

For spherical, cubical, and certain other forms of grain, we have $\alpha = 3$, $\lambda = -1$ and $\mu = \frac{1}{3}$. Substituting these in (15), we have by obvious reductions,

$$k = \mathbf{I} - \left(\mathbf{I} - \frac{X_o}{\overline{X}_o}\right)^3 \quad . \quad . \quad . \quad (46)$$

and therefore

$$X_o = \overline{X}_o \{ \mathbf{1} - (\mathbf{1} - \mathbf{k})^{\frac{1}{3}} \}$$
 . . . (47)

For cordite and similar grains we have $\alpha = 2$, $\lambda = -\frac{1}{2}$ and $\mu = 0$. Substituting these in (15), gives

$$k = 1 - \left(1 - \frac{X_o}{\overline{X}_o}\right)^2 \qquad . \qquad . \qquad . \qquad (48)$$

and

$$X_o = \overline{X}_o \{ \mathbf{1} - (\mathbf{1} - k)^{\frac{1}{2}} \}$$
 . . . (49)

Equations (46) and (48) give the fraction of the charge consumed for any given travel of the projectile, and, conversely, (47) and (49) enable us to determine the travel of projectile for any given fraction of charge burned. For any other forms of grain the solution of a complete cubic equation is necessary to determine X_o when k is given. See equation (15).

Expressions for Maximum Pressure.—It is well known that the maximum pressure in a gun occurs when the projectile has moved but a comparatively short distance from its seat, or when u and x are relatively small. The position of maximum pressure is not fixed but varies with the resistance encountered. As a rule it will be found that the less the resistance to be overcome by the expanding gases the sooner will they exert their maximum pressure, and the less will the maximum pressure be. The differentiation of (29) gives an analytical expression for the maximum value of p; but it is too complicated to be of any practical use. A reference to the table of the X functions shows that X_3 is approximately a maximum when x = 0.64, while X_4 and X_5 increase indefinitely. When λ is negative it is evident that p is a maximum when x is less than 0.64; and when λ is positive, when x is greater than 0.64. Therefore there will be two cases depending upon whether the grains burn with an increasing or a decreasing surface. These will be considered separately.

(a) When the grains burn with a decreasing surface; or what is the same thing, when λ is negative. A function at, or near, its maximum changes its value slowly. Therefore a moderate

91

variation of the position of maximum pressure will have no practical effect upon its computed value. It has been found by trial in numerous cases that x = 0.45 gives the position of maximum pressure when λ is negative with great precision. For this value of x the table gives,

$$\log X_3 = 9.85640 - 10$$

 $\log X_4 = 0.48444.$
 $\log X_5 = 0.93587.$

Substituting these in (29) and designating the maximum pressure by p_m , we have approximately, when λ is negative,

$$p_m = [9.85640 - 10] M' \{ 1 - [0.48444] N + [0.93587] N' \}$$
 (50) or,
$$p_m = 0.71846 M' \{ 1 - 3.0510 N + 8.6273 N' \} .$$
 (50')

(b) When the grains burn with an increasing surface. When the grains burn with an increasing surface λ is generally positive, and it will not be far wrong to assume that the maximum pressure

$$\log X_3 = 9.86027 - 10.$$

 $\log X_4 = 0.60479$
 $\log X_5 = 1.17352$

Substituting these in (29), we have,

 $p_m = [9.86027 - 10] M' \{ 1 + [0.60479] N - [1.17352] N' \}$ (51) or,

occurs when x = 0.8. For this value of x the table gives:

 $p_m = 0.72489 M' \{ 1 + 4.0252 N - 14.911 N' \}$. (51')

Expressions for Computing τ and the Velocity of Combustion. From (11) and (14) we have

 $\tau = \left(\frac{w \ z_o}{6 g \omega p_o}\right)^{\frac{1}{2}} \overline{X}_o \ . \ . \ . \ (52)$

If v_c is the velocity of combustion under atmospheric pressure we shall have

$$v_c = \frac{l_o}{\tau} \; ;$$

and therefore

$$v_c = \left(\frac{6 g \omega p_o}{w z_o}\right)^{\frac{1}{2}} \frac{l_o}{\overline{X}_o} \qquad (53)$$

Let v'_c be the velocity of combustion at any instant under the varying pressure p. Then from (2) we have

$$v'_{c} = v_{c} \left(\frac{p}{p_{o}}\right)^{\frac{1}{2}} = \left(\frac{6 g \omega p}{w z_{o}}\right)^{\frac{1}{2}} \frac{l_{o}}{\overline{X}_{o}} \quad . \quad . \quad (53')$$

Working Formulas. English Units.—It is customary in our service, following the English practice, to express the volumes of the powder chamber and bore in cubic inches; the various pressures in pounds per square inch; the caliber, reduced length of initial air space, and travel of the projectile in the bore, in inches; while the velocity of the projectile is expressed in foot-seconds and its weight in pounds and ounces. These units are apt to cause confusion and error in the applications of ballistic formulas; and to avoid this as much as possible it will be well to reproduce the most important of the formulas deduced in the preceding pages with all the reductions made and the mathematical and physical constants introduced and combined into one numerical coefficient. The physical constants adopted for English units (foot-pound), are the following:

g = 32.16 f.s. (mean for the United States)

 $p_o = 14.6967$ lbs. per in.²

V' = 27.68 cubic inches.

f is taken in pounds per square inch. The formulas are renumbered for convenience.

$$a = \frac{\mathbf{I}}{\Delta} - \frac{\mathbf{I}}{\delta} = \frac{\delta - \Delta}{\Delta \delta} \qquad (55)$$

$$z_o = \frac{4 \times 27.68}{\pi} \cdot \frac{a \, \tilde{\omega}}{d^2} = [1.54708] \frac{c \, \tilde{\omega}}{d^2} \text{ (inches)}$$
 (56)

$$V_1^2 = 144 \times 6 g \frac{f \tilde{\omega}}{w} = [4.44383] \frac{f \tilde{\omega}}{w} \text{ (foot-seconds)}$$
 (58)

$$V_1^2 = \frac{v^2}{k X_2} = \frac{V^2}{X_2} = \frac{M \overline{X}_o}{\alpha} = \frac{\lambda M}{\alpha N} = \frac{3 M P'}{M'} = \frac{\overline{v}^2}{\overline{X}_2}$$
 (59)

$$M' = \frac{6}{27.68g} \cdot \frac{w M}{a \tilde{\omega}} = [7.82867 - 10] \frac{w M}{a \tilde{\omega}} (61)$$

$$P' = \frac{2}{27.68 g} \frac{w V_1^2}{a \, \tilde{\omega}} = [7.35155 - 10] \frac{w V_1^2}{a \, \tilde{\omega}} \quad . \quad . \quad . \quad (63)$$

$$f = \frac{1}{144 \times 6 g} \cdot \frac{w V_1^2}{\tilde{\omega}} = [5.55617 - 10] \frac{w V_1^2}{\tilde{\omega}} \quad . \tag{64}$$

$$\frac{aP'}{f} = \frac{1728}{27.68} , \qquad (65)$$

$$\tau = \left(\frac{2 \times 27.68}{9 \pi^2 g p_o}\right)^{\frac{1}{2}} \frac{\overline{X}_o \sqrt{a w \tilde{\omega}}}{d^2} = \left[8.56006 - 10\right] \frac{\overline{X}_o \sqrt{a w \tilde{\omega}}}{d^2} \quad (66)$$

$$v_c = \frac{l_o}{\tau} = \left[1.43994\right] \frac{l_o d^2}{\overline{X}_o \sqrt{a w} \tilde{\omega}} \text{ (inches per second)} \quad . \tag{67}$$

$$v_c' = v_c \left(\frac{p}{p_o}\right)^{\frac{1}{2}} = [9.41639 - 10] v_c \sqrt{p}$$
 . . . (68)

$$\overline{X}_{o} = \frac{\left[1.43994\right] \tau d^{2}}{\sqrt{a w \tilde{\omega}}} = \frac{\left[1.43994\right] l_{o}}{v_{c} \sqrt{a w \tilde{\omega}}} d^{2} (69)$$

$$k = \frac{y}{\tilde{\omega}} = \frac{v^2}{V_1^2 X_2} \quad . \tag{70}$$

It must be remembered that v and p refer to the period when the powder is burning and V and P to the period after the powder is all burned.

FRENCH UNITS

In metric units we shall take V_c in cubic decimetres, p in kilogrammes per square centimetre, d in centimetres and z_o , u and v in metres. Also g=9.80896 m.s. With these units our formulas become,

$$z_o = \frac{40}{\pi} \cdot \frac{a \, \tilde{\omega}}{d^2} = [1.10491] \frac{a \, \tilde{\omega}}{d^2} . \qquad (72)$$

$$V_1^2 = [2.76977] \frac{f \tilde{\omega}}{w} (f \text{ in kilos. per cm.}^2)$$
 . . . (73)

$$M' = [7.70735 - 10] \frac{w M}{a \tilde{\omega}}$$
 (kilos. per cm.²) . . (74)

$$P' = [7.23023 - 10] \frac{w V_1^2}{a \tilde{\omega}} \text{ (kilos. per cm.}^2)$$
 . . . (75)

$$f = [7.23023 - 10] \frac{w V_1^2}{\tilde{\omega}}$$
 (kilos. per cm.²) . . (76)

$$v_c = [0.63128] \frac{l_o d^2}{\overline{X}_o \sqrt{a w \tilde{\omega}}}$$
 (cm. per. sec.) . . . (77)

$$\overline{X}_{o} = [0.63128] \frac{l_{o} d^{2}}{v_{c} \sqrt{a w \tilde{\omega}}} = [0.63128] \frac{\tau d^{2}}{\sqrt{a w \tilde{\omega}}}. \quad (78)$$

$$\tau = [9.36872 - 10] \frac{\overline{X}_o \sqrt{a w \tilde{\omega}}}{d^2} (79)$$

Characteristics of a Powder.—The quantities f, τ , α , λ and μ were called by Sarrau the characteristics of the powder because they determine its physical qualities. Of these quantities f depends principally upon the composition of the powder, and, with the same gun, for service charges, is practically constant for all powders having the same temperature of combustion. The value of τ depends generally upon the density and least

dimension of the grain. The factors α , λ and μ called "form characteristics" depend upon the form of the grain, and for the carefully moulded powders now employed their values may be determined with great precision. They are constant so long as the grain in burning retains its original form.

Expressions for M, M', N and N' in Terms of the Characteristics of the Powder.—When f and v_c are known from experimental firings or otherwise, for any gun and powder, the quantities V_1^2 and \overline{X}_o can be determined either from (58) and (69), or (73) and (78). Substituting these in the proper expressions for M, M', N and N' they become

For English Units

$$M = [3.00389] \frac{\alpha f v_c}{d^2 l_o} \left(\frac{a \tilde{\omega}^3}{w} \right)^{\frac{1}{2}} (80)$$

$$M' = \left[0.83256\right] \frac{\alpha f v_c}{d^2 l_o} \left(\frac{w \tilde{\omega}}{a}\right)^{\frac{1}{2}} \qquad (81)$$

$$N = [8.56006 - 10] \frac{\lambda v_c}{d^2 l_o} \sqrt{a w \tilde{\omega}} . . . (82)$$

For Metric Units

$$M = \left[2.48268\right] \frac{\alpha f v_c}{d^2 l_o} \left(\frac{a \, \omega^3}{w}\right)^{\frac{1}{2}} \quad . \quad . \quad . \quad (84)$$

$$N = [9.71291 - 10] \sqrt{a w \tilde{\omega}}$$
 . . . (86)

$$N' = \frac{\mu}{\lambda^2} N^2$$
 (87)

If we substitute the value of M' from (81) in (50) or (51), and reject the terms within the brackets, we have in effect Sarrau's

monomial formula for maximum pressure. But it is evident there can be no monomial formula for velocity or pressure unless λ and μ are approximately zero. Equations (80) to (87) are useful for determining the values of M and N (upon which all the other constants depend), when the charge varies or when there are variations in the weight of the projectile. In these formulas α , λ , μ and l_o are independent of $\tilde{\omega}$ and d and are strictly grain constants. v_c is a powder constant, varying only with the composition and density of the powder. f is approximately constant for full service charges of the same kind of powder, in guns of all calibers. For example the magazine rifle, caliber 0.3 inches, and the 16-inch B. L. R. give approximately the same value to f when computed by equation (64) or (65). This factor, however, varies with the charge in the same gun, for it is evident that its effective value as measured by projectile energy must decrease with the charge. Indeed if the charge be sufficiently reduced it is obvious that f becomes zero since we have omitted from our formulas all consideration of the force necessary to start the projectile. The law of variation is not known; but we will assume provisionally that f varies with the charge according to the law expressed by the equation

$$f = f_o \left(\frac{\tilde{\omega}}{\tilde{\omega}_o}\right)^n \qquad . \qquad . \qquad . \tag{88}$$

where $\tilde{\omega}_o$ is the service charge by means of which M and N were determined and f_o the corresponding value of f computed by (64) or (76). If the weight of the projectile also varies we will assume that f may be determined by the equation

$$f = f_o \left(\frac{\tilde{\omega}}{\tilde{\omega}_o}\right)^n \left(\frac{w}{w_o}\right)^{n'} \qquad . \qquad . \qquad . \tag{89}$$

The exponents n and n' must be determined from experimental data. If we make

$$K = \frac{f_o}{\tilde{\omega}^n \ w^{n'}} \quad . \qquad . \qquad . \qquad . \qquad . \qquad (90)$$

(89) becomes

$$f = K \tilde{\omega}^n w^{n'} \dots \dots (90')$$

Substituting this expression for f in (80) and (84) gives, for English units:

$$M = [3.00389] K \frac{\alpha v_c}{d^2 l_0} \left(\frac{a \tilde{\omega}^{3+2n}}{w^{1-2n'}} \right)^{\frac{1}{2}}. \quad . \quad (91)$$

and for metric units:

$$M = [2.48268] K \frac{\alpha v_c}{d^2 l_o} \left(\frac{a \tilde{\omega}^{3+2n}}{w^{1-2n'}} \right)^{\frac{1}{2}} . . . (91')$$

In the applications of these equations f_o must be computed by (64) or (76) and v_c by (67) or (77).

7

CHAPTER V

APPLICATIONS

THE principal formulas deduced in Chapter IV are here reproduced for convenience of reference. They are the following:

(a) Formulas which Apply Only While Powder is Burning.—

$$v^2 = M X_1 \{ 1 + N X_o + N' X_o^2 \}$$
 . . (1)

$$p = M' X_3 \{ 1 + N X_4 + N' X_5 \} . . . (2)$$

It will be observed that these formulas for velocity and pressure are identical in form, and that the constants within the brackets are common to both. Also that M' is a simple multiple of M. Moreover, from the manner of deriving p from v^2 , the velocity and pressure deduced from these formulas correspond at every point so that one can be easily and exactly computed from the other without the necessity of laying down velocity curves in order to obtain the pressures.

(b) Formulas which Apply Only After the Powder has All Been Burned.—

$$V^2 = V_1^2 X_2 = \frac{M \overline{X}_o X_2}{\alpha} = \frac{\lambda M}{\alpha N} X_2$$
 . (3)

$$P = \frac{P'}{(1+x)^{\frac{4}{3}}} = P'(1-X_2)^4 \quad . \quad . \quad . \quad (4)$$

(c) Formulas Which Apply at the Instant of Complete Combustion.—

$$\overline{v}^2 = M \overline{X}_1 \{ \mathbf{I} + N \overline{X}_o + N' \overline{X}_o^2 \}, \text{ (from (I))} : .$$
 (5)

and

$$\overline{v}^2 = \frac{M\overline{X}_1}{\alpha} \text{ (from (3))} \qquad (5')$$

Equations (5) and (5') give the same value to \overline{v}^2 , since the

former equation reduces to the latter at the point \overline{u} . But (1) and (3) are not tangent at the point \overline{u} unless the vanishing surface (S') of the grain is zero, as with cubes, spheres, solid cylinders, etc.

From (2) we have at the travel \overline{u}

$$\overline{p} = M' \overline{X}_3 \left\{ 1 + N \overline{X}_4 + N' \overline{X}_5 \right\} \quad . \quad . \quad (6)$$

and from (4)

$$\overline{p} = \frac{P'}{(1+\overline{x})^{\frac{4}{3}}} \quad . \quad . \quad . \quad . \quad (6')$$

Equations (6) and (6') give the same value to \overline{p} for all grains whose vanishing surface is zero, as may be thus shown:

Substituting for M' in (6) its value from (59), Chapter IV, and giving to N and N' their values in terms of \overline{X}_o , we have

$$\overline{p} = 3 P' \frac{\overline{X}_3}{\overline{X}_o} \alpha \left\{ \mathbf{I} + \frac{\overline{X}_4}{\overline{X}_o} \lambda + \frac{\overline{X}_5}{\overline{X}_o^2} \mu \right\}$$
But $\frac{\overline{X}_4}{\overline{X}_o} = \mathbf{I} + \overline{X}$ and $\frac{\overline{X}_5}{\overline{X}_o^2} = \mathbf{I} + 2 \overline{X}$.
$$\therefore \overline{p} = 3 P' \frac{\overline{X}_3}{\overline{X}_o} \alpha \left\{ \mathbf{I} + \lambda + \mu + (\lambda + 2 \mu) \overline{X} \right\}$$

For all grains whose vanishing surface is zero we have the relations (Equs. (10) and (12), Chapter III.)

$$\alpha (I + \lambda + \mu) = I$$

and

$$\alpha (\lambda + 2 \mu) = -1$$

$$\therefore \overline{p} = 3 P' \frac{\overline{X}_3}{\overline{X}_o} (\mathbf{I} - \overline{X}),$$

which readily reduces to (6'). Therefore for all forms of grain whose vanishing surface is zero the pressure curves (2) and (4) are tangent at \overline{u} . This is not true for grains for which S' > 0.

For these the pressure at travel \overline{u} given by (2) is greater than that given by (4), and this difference increases with S'.

Monomial Formulas for Velocity and Pressure While the Powder is Burning.—The expressions for velocity and pressure while the powder is burning (equations (1) and (2)) are generally trinomials because equations (9) and (22), Chapter III, are trinomials. And these are so because of the geometrical characteristics α , λ and μ . In order to have a monomial expression for velocity or pressure λ and μ must both be zero. But upon this supposition (22), Chapter III, would become

$$k = \frac{\alpha \, l}{l_o}$$

or, the fraction of grain burned would be directly proportional to the thickness of layer burned; which is impossible, since the grain burns on all sides. This same supposition would also make

$$l_o S_o = V_o$$

which is not true, at least for finite volumes.

It has been shown in Chapter III, that all grains which, under the parallel law of burning, retain their original form until wholly consumed and for which S'>0, have one or the other of the following expressions for α , namely, $\mathbf{1}+x$ or $\mathbf{2}+x$, x being the ratio of the thickness of web to the length (or breadth) of the grain. Only grains for which $\alpha=\mathbf{1}+x$ can give approximate monomial expressions for velocity and pressure, and this by making x so small that it may be omitted in comparison with unity, in which case α becomes practically unity and λ and μ zero. To this class belong thin, flat grains and long cylindrical grains with axial perforation.

When $\alpha = 1$ and λ and μ are zero, equations (1) and (2) become

$$v^2 = M X_1 \dots (7)$$



$$p = M'X_3 = [7.82867 - 10] M \frac{w}{a'\bar{\omega}} X_3$$
 (8)

Also, by equation (3), since $\alpha = 1$, we have, after the powder is all burned,

$$M = \frac{V_1^2}{\overline{X}_0} = \frac{V^2}{\overline{X}_0 X_2} \dots$$
 (9)

It will be seen that the pressures by the monomial formula are directly proportional to X_3 , which therefore gives, to the proper scale, the typical pressure curve. Its maximum value, as seen from the table of the X functions, occurs when x = 0.64, and its logarithm is 9.86390 - 10.* Applying this in equation (8) gives for the maximum pressure,

$$p_m = [7.69257 - 10] M \frac{w}{a \ \omega}$$
 . . . (10)

If the maximum pressure, assumed to be the crusher-gauge pressure, is known by experiment, we may compute M from the last equation. Thus we have

Substituting this in (7) and (8) we have while the powder is burning,

$$v = \begin{bmatrix} 1.15371 \end{bmatrix} \left(\frac{a \tilde{\omega} p_m}{w} \right)^{\frac{1}{2}} \sqrt{X_1} \quad . \quad . \quad (12)$$

and

$$p = [0.13610] p_m X_3 \dots (13)$$

Since by (12) the velocity is proportional to $\sqrt{X_1}$, this function represents the typical velocity curve while the powder is

^{*} The maximum value of X_3 occurs when x = 0.6336+. But the value of x given above is near enough for all practical purposes. It may be noted here that the curve of X_3 has a point of inflection when x = 1.3891.

burning. After the powder is all burned the monomial formulas

$$V = V_1 \sqrt{X_2} = [2.22191] \left(\frac{f \tilde{\omega} X_2}{w}\right)^{\frac{1}{2}} .$$
 (14)

and

$$P = \frac{P'}{(1+x)^{\frac{1}{3}}} = \frac{M' \, \overline{X}_o}{3a \, (1+x)^{\frac{1}{3}}} \, . \qquad . \qquad . \qquad (15)$$

are to be employed.

Example.—As an example of monomial formulas for velocity and pressure take the following data from "Notes on the Construction of Ordnance," No. 89, pages 43–47:

Gun: 8-inch B. L. R., Model 1888. $V_c = 3617$ in.³; $u_m = 205.25$ in.

Powder: Nitrocellulose composition, single-perforated grains of the following dimensions: length (m) 47.69 in.; outside diameter (2 R) 0.4455 in.; diameter of perforation (2 r) 0.1527 in.

From these dimensions we find by the formulas of Chapter III.,

$$2 l_o = \frac{1}{2} (0.4455 - 0.1527) = 0.1464 \text{ in.}$$

$$x = \frac{2 l_o}{m} = 0.0030525.$$

$$\alpha = 1 + x = 1.0030525.$$

$$\lambda = \frac{x}{1 + x} = 0.0030433.$$

$$\mu = 0.$$

We may, therefore, in this case, assume $\alpha = 1$ and $\lambda = 0$ without material error, and employ the monomial formulas (7) and (8), computing M either by (9) or (11), according as we take the observed muzzle velocity or crusher-gauge pressure for this purpose. If the crusher-gauge pressure (assumed to be p_m) is employed equations (12) and (13) may be used. If it is known that the powder is all burned at, or near, the muzzle (9) becomes

$$M = \frac{v_m^2}{X_1} \quad . \quad . \quad . \quad . \quad . \quad (9')$$

in which both symbols in the second member refer to the muzzle. If the charge is not all consumed at the muzzle and we know the value of v_c for the powder used, \overline{X}_o can be found by (69), Chapter IV., and then N can be computed by (9). Finally if v_c is not known equations (12) and (13) must be employed.

As an example one shot was fired with a charge of 78 lbs., and a projectile weighing 318 lbs. The observed muzzle velocity was 2040 f. s., and crusher-gauge pressure 30450 lbs. per in.², and it was known that the combustion of the charge was practically complete at the muzzle. From the given data we find (taking $\delta = 1.567$), $\Delta = 0.5969$, log a = 0.01584, and $z_o = 44.548$ in. Therefore,

$$x_m = \frac{u_m}{z_0} = \frac{205.25}{44.548} = 4.6073;$$

and from Table 1, for this value of x_m ,

 $\log X_o = 0.77147$

 $\log X_1 = 0.41207.$

 $\log X_2 = 9.64060 - 10.$

 $\log M = 2 \log 2040 - 0.41207 = 6.20719.$

Also by (61), Chapter IV, $\log M' = 4.63036$.

The equations for the velocity and pressure curves for this shot are, therefore,

$$v = [3.10359] \sqrt{X_1} \dots (16)$$

and

$$p = [4.63036] X_3 (17)$$

The first of these equations gives, of course, the observed muzzle velocity; and the second gives (by taking x = 0.64) a maximum pressure of 31208 lbs. per in.², exceeding the crusher-gauge pressure by 758 lbs.

If we determine the value of M by means of the crusher-gauge pressure we shall have by (11), $\log M = 6.19652$; and the equations for velocity and pressure now become

$$v = [3.09826] \sqrt{X_1}$$

and

$$p = [4.61969] X_3$$

This last equation gives the observed crusher-gauge pressure while the first makes the muzzle velocity 25 f. s. less than the observed. As muzzle velocities can be more accurately measured than maximum pressures, the first set of formulas are probably the more accurate and will be used in what follows in preference to the other set.

The expression for fraction of charge burned at any travel of projectile is found from (70), Chapter IV., and is for this example,

$$k = [3.02134 - 10] \frac{v^2}{X_2}$$
 . . . (18)

The travel of projectile is given by the equation

$$u = z_0 x = 44.548 x \text{ inches}$$
 . . . (19)

The following table computed by means of equations (16), (17), (18), and (19), is represented by the curves v and p in Fig. 1.

<i>x</i>	Travel u inches	Velocity v ft. secs.	Pressure p 1bs. per in. 2	Fraction of charge burned.	Pressure P lbs. per in. ²	Velocity V ft secs.
0.0	0.0	0.0	00	0.0	84084	0.0
0.2	8.910	379.7	26115	0.257	65939	749.3
0.4	17.819	600.2	30260	0.357	53687	1005.1
o .6	26.729	770.3	31194	0.430	4493 I	1175.0
o .8	35.638	910.1	30949	0.489	38402	1301.6
1.0	44.548	1029.2	30214	0.539	33369	1401.5
1.2	53.457	1133.1	29281	0.583	29387	1483.4
1.4	62.367	1225.2	28285	0.623	26167	1552.3
1.6	71.277	1308.0	27288	0.659	23519	1611.5
1.8	80.186	1383.2	26322	0.692	21306	1663.1
2.0	89.096	1452.0	25401	0.722	19434	1708.6
2.5	111.370	1602.2	23322	0.790	15823	1802.8
3.0	133.644	1729.3	21548	0.849	13242	1877.0
3.5	155.918	1839.1	20034	0.901	11318	1937.5
4.0	178.192	1936.0	18733	0.948	9834	1988.2
4.5	200.466	2022.5	17606	0.991	8661	2031.5
4.6073	205.250	2040.0	17384	1.000	8441	2040.0

The last two columns in the table represented by the curves V and P, Fig. 1, show the velocity and pressure upon the supposition that the powder was all converted into gas at the tempera-

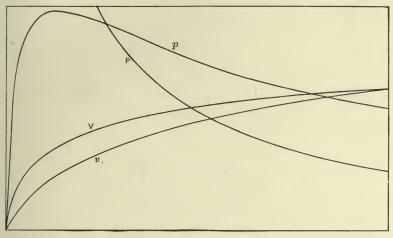


Fig 1.

ture of combustion before the projectile had moved. They were computed by the formulas

$$V^2 = V_1^2 X_2 = [6.97866] X_2 \dots (20)$$

and

$$P = \frac{[4.92471]}{(1+x)^{\frac{4}{3}}}. (21)$$

The force of the powder (f) and the velocity of combustion in free air (v_c) for this particular charge and brand of powder can now be computed by equations (64) and (67), Chapter IV. We find f = 1396.9 lbs. per in.² and $v_c = 0.13614$ in. per sec.

If we wish to compute velocities and pressures in this 8-inch gun when the charge varies K must be computed by (90) and M by (91), Chapter IV. Since the weight of projectile is constant n' is zero; and for an 8-inch gun we will assume that $n = \frac{2}{3}$,

—this assumption to be tested by experiment. With these values of n and n' we have

$$K = \frac{1396.9}{(78)^{\frac{2}{3}}} = 76.519$$

and therefore,

$$f = 76.519 \ \tilde{\omega}^{\frac{2}{3}}.$$
 . . . (22)

Substituting this value of K, and the gun and powder constants in (91), we have

$$M = [2.09974] a^{\frac{1}{6}} \tilde{\omega}^{\frac{13}{6}}. \qquad (23)$$

Also, from (61), Chapter IV,

$$M' = [2.43084] \frac{\tilde{\omega}^{\frac{7}{6}}}{a^{\frac{1}{2}}} \dots \dots (24)$$

Therefore from (7), (8), and (10),

$$v = [1.04987] a^{\frac{1}{4}} \tilde{\omega}^{\frac{18}{5}} \sqrt{X_1} \dots (25)$$

$$p = [2.43084] \frac{\tilde{\omega}^{5} X_{3}}{a^{\frac{1}{2}}} \dots \dots (26)$$

and

$$p_m = [2.29474] \frac{\tilde{\omega}^{\bar{b}}}{a^{\frac{1}{2}}} \quad . \quad . \quad . \quad (27)$$

which are the formulas for velocity and pressure for this gun and brand of powder in terms of the weight of charge.

As an example, what would be the maximum pressure with a charge of $79\frac{1}{2}$ lbs.? We first find $\Delta = 0.6084$ and then $\log a = 0.00238$. We then have by (27)

$$\log p_m = 2.29474 + \frac{7}{6} \log 79.5 - \frac{1}{2} \log a = 4.51065$$

$$\therefore p_m = 32408 \text{ lbs. per in.}^2$$

This agrees very closely with observation.

We have the means of testing the accuracy of these equations to a limited extent, since there were four shots fired with charges of 70, 78, 85, and 88 lbs. The following table gives the results of all the necessary preliminary calculations for the four shots

fired and also for two others "estimated from prolonged empirical curves." The data from the shot fired with a charge of 78 lbs. have been taken as the basis of the calculations. The gun constants will be found on page 102.

ũ	Λ	log a	log z ₀		Ат	Muzzle	
lbs.	Δ			x	$\log X_0$	$\log X_1$	$\log X_2$
60 70 78 85 88 95	0.4592 0.5357 0.5969 0.6505 0.6734 0.7370	0.18742 0.08940 0.01584 9.95382 9.92777 9.86767	1.70647 1.67540 1.64883 1.62414 1.61315 1.58629	4.0347 4.3339 4.6073 4.8769 5.0018 5.3210	0.74978 0.76150 0.77147 0.78066 0.78475 0.79467	0.36944 0.39260 0.41207 0.42987 0.43770 0.45663	9.61965-10 9.63110 9.64060 9.64919 9.65295 9.66196

The computed muzzle velocities and maximum pressures in the following table were obtained (with the exception of the first two muzzle velocities) by equations (25) and (27). The values of f were computed by (22) and \overline{X}_o by (69), Chapter IV.

ũ	$\log \overline{X_0}$	f	Muz	ZLE VELO F. S.	CITY,		MUM PRES	
lbs.	log A ₀	lbs. per inch²	Observed	Com- puted	0C.	Observed	Com- puted	0C.
60 70 78 85 88 95	0.74265 0.75819 0.77147 0.78382 0.78931 0.80274	1173 1300 1397 1479 1514 1593	1600 1839 2040 2200 2275 2454	1600 1844 2040 2205 2276 2441	0 - 5 0 - 5 - 1 +13	18000 24889 30450 35600 39301 47280	18859 25272 31206 37051 39756 46583	-859 -383 -756 -1451 -455 +697

For the first two shots the powder was all burned before the projectile had reached the muzzle, as is shown by the values of $\log \overline{X}_o$. For these the muzzle velocities were computed by equation (14).

It will be observed that the equations by means of which the muzzle velocities and maximum pressures given in this table were computed depend for their constants upon one measured velocity only, due to a charge of 78 lbs. The measured crusher-gauge pressure for this charge has not been made use of at all. The constant M upon which all the other constants depend might have been determined by equation (11) in which the muzzle velocity does not enter. But muzzle velocities can be more accurately measured than maximum pressures and are, therefore, better adapted to the determination of ballistic constants. The computed maximum pressures in the table are probably nearer the actual pressures on the base of the projectile than those given by the crusher gauge. The agreement between the computed and measured muzzle velocities is all that could be expected from any ballistic formulas.

To determine the travel of projectile when all the charge was burned we take \overline{x} by interpolation from the table of the X functions corresponding to the values of $\log \overline{X}_o$. We then have:

$$\overline{u} = \overline{x} z_0$$
.

For the travel of projectile when the pressure is a maximum. we have, calling this travel u',

$$u' = 0.64 z_0$$
.

The following table gives the values of u' and \overline{u} for all the shots:

ω̂ lbs.	u' inches	inches	$u_m - \overline{u}$ inches	k	Remarks
60 70 78 85 88 95	32.56 30.31 28.51 26.94 26.26 24.69	196.56 201.13 205.25 209.30 211.16 215.88	8.69 4.12 0.00 - 4.05 - 5.91 -10.63	1.0000 0.9926 0.9895 0.9813	

It will be observed that as the charge increases the sooner it exerts its maximum pressure. The last column gives the fraction of the charge burned at the muzzle and shows that approximately the entire charge for the series was consumed at the muzzle. k was computed by the formula

$$k = [6.17483 - 10] \frac{v^2}{\tilde{\omega}^{\frac{1}{8}} X_2}$$

In order to determine the velocity and pressure curves for any given charge we should compute M and M' by equations (23) and (24), and then employ (7) and (8) as has already been done for a charge of 78 lbs. For example, determine the velocity and pressure curves for a charge of 95 lbs. We have, from (23) and (24),

$$\log M = 2.09974 + \frac{1}{2}\log a + \frac{13}{6}\log \tilde{\omega} = 6.31864$$
$$\log M' = 2.43084 + \frac{7}{6}\log \tilde{\omega} - \frac{1}{2}\log a = 4.80435$$

Therefore

$$v = [3.15932] \sqrt{X_1}$$

and

$$\dot{p} = [4.80435] X_3$$

are the equations required.

Example.—Suppose the thickness of web of the grain we have been considering to be increased 10 per cent., all other conditions remaining the same. Deduce the velocity and pressure curves for a charge of 78 lbs. In this case it is evident that all the charge would not be burned in the gun and that in consequence both the maximum pressure and muzzle velocity would be diminished.

It will be seen from (69), Chapter IV, that, other things being equal, the value of \overline{X}_o varies directly with the web thickness. Therefore if this is increased by 10 per cent., or, what is the same thing, is multiplied by $\overline{1.1}$, \overline{X}_o will also be multiplied by $\overline{1.1}$; and from (60) and (61), Chapter IV, M and M' will be divided by 1.1. Therefore (16) and (17) will in this case become,

$$v = [3.08290] \sqrt{X_1},$$

and

$$p = [4.58897] X_3.$$

These equations give $v_m = 1945$ f. s., and $p_m = 28371$ lbs. This is a loss of 95 f. s. in muzzle velocity and a diminution of 2070 lbs. in maximum pressure. To determine the fraction of the charge burned at the muzzle, we have from (45), Chapter IV,

$$k = \frac{v^2}{V_1^2 X_2} \; ;$$

which gives, by employing the muzzle velocity just computed,

$$k = 0.909.$$

Therefore on account of the increased thickness of web, seven pounds of the charge remained unburned when the projectile left the gun.

We may next inquire what effect a decrease of 10 per cent. in web thickness would have upon the muzzle velocity and maximum pressure. In this case we must multiply the original value of \overline{X}_a by 0.9 and divide M and M' by the same fraction. We thus get

$$\log \overline{X}_o = 0.72571$$

 $\log M = 6.25295$
 $\log M' = 4.67612$

Therefore, from (9), the muzzle velocity in this case is found to be 2040 f. s.; and, by (10), the maximum pressure, 34675 lbs. That is, the muzzle velocity remains the same while the maximum pressure is increased by 4225 lbs. per in.2 These examples show that for the greatest efficiency (muzzle velocity and maximum pressure both considered), the web thickness for this form of grain should be such that the charge is all consumed at the muzzle. From the value of \overline{X}_a given above we find, by interpolation, that $\overline{x} = 3.4890$; and, therefore, $\overline{u} = 155.43$ inches. For this travel the above values of M and M' give $\overline{v} = 1936$ f. s., and $\overline{p} = 22294$ lbs. The muzzle pressure comes out 8433 lbs.

Suppose for a hypothetical 7-inch gun we assume the following data:

$$d = 0''.7$$

 $V_c = 4,000 \text{ c. i.}$
 $u_m = 40 \text{ calibers} = 280 \text{ inches.}$
 $\Delta = 0.6.$
 $\delta = 1.5776$
 $f = 1396.9 \text{ lbs.}$
 $v_c = 0.13614 \text{ in. per sec.}$
 $w = 205 \text{ lbs.}$

What muzzle velocity and maximum pressure would be obtained, supposing the charge to be all consumed at the muzzle; and what must be the thickness of web?

The weight of charge due to the given chamber capacity and density of loading is found to be 86.7 lbs. We next compute the following numbers by formulas given in Chapter IV:

$$\begin{array}{l} \log a = \text{0.01221} \\ \log z_o = \text{1.80713} \\ x_m = 4.3655 \\ \log \overline{X}_o = \log X_{om} = \text{0.76269} \\ \log \overline{X}_1 = \log X_{1m} = \text{0.39492} \\ \log V_1^2 = 7.21529 \text{ (By (58), Chapter IV),} \\ \log M = 6.45260 \text{ (By (9))} \end{array}$$

Then by (7) and (10) we find

Muzzle velocity = 2653 f. s.

and

Maximum pressure = 32112 lbs. per in.²

The muzzle pressure, by (8), is 18413 lbs.

The necessary thickness of web in order that the charge may all be consumed at the muzzle, is 0.158 inches. The other dimensions of the grains are immaterial.

If the volume of the chamber is taken at 3,000 c. i., all the

other data remaining the same, we should have the following results:

$$\tilde{\omega} = 65.03 \text{ lbs.}$$
 $M. V. = 2413 \text{ f. s.}$
 $p_m = 28868 \text{ lbs. per in.}^2$
 $M. P. = 14105 \text{ lbs. per in.}^2$
 $2 l_0 = 0.152 \text{ in.}$

If $V_c = 4500$ c. i., we have the following:

$$\tilde{\omega} = 97.54 \text{ lbs.}$$
 $M. V. = 2753 \text{ f. s.}$
 $p_m = 33574 \text{ lbs. per in.}^2$
 $M. P. = 24496 \text{ lbs. per in.}^2$
 $2 l_o = 0.160 \text{ in.}$

Binomial Formulas for Velocity and Pressure.—Binomial formulas pertain to grains for which μ is zero or so small that it may be neglected, while λ must be retained on account of its magnitude. To this class belong all unperforated, long, slender grains of whatever cross-section, such as strips, ribbons, cylinders, etc. The binomial expressions for velocity and pressure for these grains are

 $v^2 = M X_1 \{ 1 - N X_o \}$. . . (28)

and

$$p = M' X_3 \{ r - N X_4 \}$$
 . . . (29)

The second term within the brackets has the negative sign because λ is always negative for these forms of grain.

Methods for Determining the Constants M and N.—The constants M and N can be determined when the given experimental data are such that two independent equations can be formed involving M and N. These data may be either two measured velocities of the same shot at different positions in the bore; or a measured muzzle velocity and crusher-gauge pressure,—the latter taken as the maximum pressure. In

addition to these all the elements of loading, as well as the powder and gun constants, are supposed to be known.

First Case.—Let v_1 and v_2 be two measured velocities in the bore at the distances u_1 and u_2 from the origin, which is the base of the projectile in its firing position. From the gun and firing constants compute z_o by (56), Chapter IV, and then x_1 and x_2 corresponding to u_1 and u_2 by the equation

$$x = \frac{u}{z_o}$$

With these values of x_1 and x_2 as arguments, interpolate from the table of the X functions the corresponding values of $\log X_o$ and $\log X_1$, distinguishing them by accents. We then have the two independent equations

$$v_1^2 = M X_1' (I - N X'_o)$$

 $v_2^2 = M X_1'' (I - N X''_o)$

from which M and N may easily be determined. For simplicity let

$$b = \left(\frac{v_2}{v_1}\right)^2 \cdot \frac{X'_1}{X''_1}$$
 and $b' = \frac{X'_0}{X''_0}$.

We then have in a form well adapted to logarithmic computation

$$N = \frac{\mathbf{I} - b}{(\mathbf{I} - bb') X''_{o}} (30)$$

and

$$M = \frac{v_1^2}{X'_1(I - N X'_0)} = \frac{v_2^2}{X''_1(I - N X''_0)} \quad . \tag{31}$$

These equations are equally adapted to English or French units.

Second Case.—When the powder is not all burned in the gun let v_m be the observed muzzle velocity and p_m the crusher-gauge pressure. We then have the two independent equations

$$v_m^2 = M X_1 (I - N X_0)$$

and ((50'), Chapter IV),

$$p_m = [9.85640 - 10] M' (1 - [0.48444] N)$$
 . (32)

Substituting for M' its value in terms of M ((61), Chapter IV), and making, for English units,

$$c = [7.68507 - 10] \frac{w \, v_m^2}{a \, \tilde{\omega} \, p_m \, X_1}$$

we have

$$N = \frac{1 - c}{X_o - 3.051 c} = \frac{1 - c}{X_o \left(\frac{1 - [0.48444] c}{X_o}\right)}, \quad (33)$$

and then M from (31). The X functions in these last two formulas refer to v_m . Any measured velocity within the bore before the powder is all burned may be used instead of v_m . For French units the logarithmitic multiplier in the expression for c is [7.56404 - 10].

Second Method.—If the powder is all burned before the projectile reaches the muzzle, we have from (3)

$$M = \frac{\alpha V_m^2 N}{\lambda X_2} \quad . \quad . \quad . \quad (34)$$

where V_m is the muzzle velocity and X_2 corresponds to V_m . N must be determined either by a velocity v_1 measured in the bore before the powder is consumed, or by the crusher-gauge pressure assumed to be p_m . If by the former, we have from (28)

$$v_1^2 = M X_1' (1 - N X_0').$$

Substituting M from (34) in this equation and solving for N we have

$$N = \frac{1}{2 X'_o} \left\{ 1 - \left(1 - \frac{4 \lambda X_2}{\alpha X'_2} \left(\frac{v_1}{V_m} \right)^2 \right)^{\frac{1}{2}} \right\} . \quad (35)$$

In this equation X'_o and X'_2 correspond to the measured velocity v_1 . The travel of projectile to the point where all the powder is burned is found by the equation $\overline{X}_o = \lambda/N$ and a reference to the table of the X functions. In using these last

two formulas N must first be computed, and then M. Equation (35) is independent of the units employed.

If, as is usually the case, there is no interior measured velocity available recourse must be had to the crusher-gauge pressure p_m . In this case we have by means of (32) and (34), and (61), Chapter IV, for English units,

$$N = [9.21453 - 10] \left\{ 1 - \left(1 - [2.79937] \frac{4 \lambda a \tilde{\omega} X_2 p_m}{\alpha w V_m^2} \right)^{\frac{1}{2}} \right\} (36)$$

and then M by (34). For metric units the logarithmetic multiplier within the braces becomes [2.92040].

Application to Sir Andrew Noble's Experiments.—These very important experiments were made at the Elswick works, Newcastle-on-Tyne, with a six-inch gun. They are thus described by Sir Andrew *: "The energies which the new explosives are capable of developing, and the high pressures at which the resulting gases are discharged from the muzzle of the gun, render length of bore of increased importance. object of ascertaining with more precision the advantages to be gained by length, the firm to which I belong has experimented with a six-inch gun of 100 calibers in length. In the particular experiments to which I refer, the velocity and energy generated has not only been measured at the muzzle, but the velocity and pressure producing this velocity have been obtained for every point of the bore, consequently the loss of velocity and energy due to any particular shortening of the bore can at once be deduced.

"These results have been attained by measuring the velocities every round at sixteen points in the bore and at the muzzle.

^{*&}quot;Report (1894) on methods of measuring pressures in the bore of guns"; and "Researches on Explosives, Preliminary Note." An abstract of these papers is given in the "English Text-book of Gunnery," 1902; in *Nature* for May 24, 1900; and in Encyclopædia Britannica, 11th edition, article "Ballistics."

These data enable a velocity curve to be laid down, while from this curve the corresponding pressure curve can be calculated. The maximum chamber pressure obtained by these means is corroborated by simultaneous observations taken with crusher gauges, and the internal ballistics of various explosives have thus been completely determined."

The velocities at the sixteen points in the bore were determined by registering the times at which the projectile passed these points. The registering apparatus is thus described by Sir Andrew in the "Report," page 11: "The chronograph which I have designed for this purpose consists of a series of thin disks made to rotate at a very high and uniform velocity through a train of geared wheels. The speed with which the circumference of the disks travels is between 1200 and 1300 inches per second, and, since by means of a vernier we are able to divide the inch into thousandths, the instrument is capable of recording the millionth part of a second.

"The precise rate of the disk's rotation is ascertained from one of the intermediate shafts, which, by means of a relay, registers the revolutions of a subsidiary chronoscope, on which, also by a relay, a chronometer registers seconds. The subsidiary chronoscope can be read to about the $\frac{1}{5000}$ th part of a second.

"The registration of the passage of the shot across any of the fixed points in the bore is effected by the severance of the primary of an induction coil causing a spark from the secondary, which writes its record on prepared paper gummed to the periphery of the disk. The time is thus registered every round at sixteen points of the bore.

"I have ascertained by experiment that the mean instrumental error of this chronoscope, due chiefly to the deflection of the spark, amounts only to about three one-millionths of a second. Usually the pressures were deduced from the mean of three consecutive rounds fired under the same circumstances." The following table gives the recorded experimental data for the various kinds of smokeless powders employed at the Elswick firings, and which will be used in the following discussions.

	Weight	Density	Mean Crusher-		ED VELOC		
Kind of Powder	of Charge, lbs.	of Loading	Gauge Pressure, lbs.	16.6 ft.	21.6 ft.	34.1 ft.	46.6 ft.
Cordite, 0".4 Cordite, 0".35 Cordite, 0".3 Ballistite, 0".3	27.5 22.0 20.0 20.0	0.55 0.44 0.40 0.40	47040 30352 36960 33936	f. s. 2794 2444 2495 2416	f. s. 2940 2583 2632 2537	f. s. 3166 2798 2821 2713	f. s. 3284 2915 2914 2806

The cordite used in these experiments contained 37 per cent. of gun-cotton, 58 per cent. of nitro-glycerine, and 5 per cent. of a hydrocarbon known as vaseline. The ballistite was nearly exactly composed of 50 per cent. of dinitrocellulose (collodion cotton) and 50 per cent. of nitro-glycerine.

DISCUSSION OF THE DATA FOR CORDITE, 0".4 DIAMETER.— The form characteristics of cordite are $\alpha=2$, $\lambda=-\frac{1}{2}$ and $\mu=0$. The equations for velocity and pressure are therefore,

$$V^2 = M X_1 (1 - N X_0)$$

and

$$p = M' X_3 (I - N X_4)$$

Since the second terms within the parentheses, which contain λ , have been made negative, λ in subsequent calculations must be regarded as positive.

For the preliminary calculations we have the following data:

$$\tilde{\omega} = 27.5 \text{ lbs.}$$

 $w = 100 \text{ lbs.}$
 $\Delta = 0.55$
 $\delta = 1.56$

From these data we find by the proper formulas:

$$\log a = 0.07084$$
$$\log z_o = 0.42178 \therefore z_o = 2.6411 \text{ ft.}$$

To determine whether the powder was all burned in the gun, the following table is formed which explains itself:

u ft.	$x = \frac{u}{z_0}$	(observed) f. s.	$\log v^2$	$\log X_2$	$\log V_1 ^2$	\logV_1	V ₁
16.6 21.6 34.1 46.6	6.2853 8.1784 12.9112 17.6446	2940 3166	6.89245 6.93669 7.00102 7.03281	9.68496-10 9.71799 9.76657 9.79440	7.20749 7.21870 7.23445 7.23841	3.60374 3.60935 3.61722 3.61920	4015 4068 4142 4161

The increase of V_1 as shown in the last column, indicates that the powder was all burned in the gun and between u = 34.1 and u = 46.6 ft.

We will compute M and N by means of the measured muzzle velocity (V_m) and the mean crusher-gauge pressure (p_m) , as these data can always be obtained without sacrificing a gun.

For cordite equations (36) and (34) become

$$N = [9.21453 - 10] \left\{ 1 - \left(1 - [2.79973] \frac{a \, \tilde{\omega} \, X_2 \, p_m}{w \, V_m^2} \right)^{\frac{1}{2}} \right\} (36')$$

and

$$M = \frac{4 V_m^2 N}{X_2} (34')$$

The numbers to be used in these formulas are

$$\tilde{\omega} = 27.5 \text{ lbs.}$$
 $w = 100 \text{ lbs.}$
 $p_m = 47040 \text{ lbs. per in.}^2$
 $V_m = 3284 \text{ f. s.}$
 $\log X_2 = 9.79440 - 10 \text{ (at muzzle)}$
 $\log a = 0.07084$

Performing the operations indicated in (36') and (34') we have

$$\log N = 8.73599 - 10$$
$$\log M = 6.57646$$

Also, by (62), Chapter IV,

$$\log M' = 4.89496.$$

The equations for the velocity and pressure curves while the powder is burning are, therefore,

$$v^2 = [6.57646] X_1 \{ I - [8.73599 - IO] X_o \}$$
 (37)

and

$$p = [4.89496] X_3 \{ 1 - [8.73599 - 10] X_4 \} . (38)$$

After the powder is burned we have from (34'), dropping the subscript from V_m , and reducing,

$$V = [3.61920]\sqrt{X_2} \dots \dots (39)$$

Also from (31) and (63), Chapter IV,

$$P = \frac{[5.07979]}{(1+x)^{\frac{2}{3}}} (40)$$

To determine the travel of projectile to the point where the powder was all burned \overline{u} we have, for cordite,

$$\overline{X}_o = \frac{1}{2N}$$

Therefore

$$\log \overline{X}_0 = 0.96298;$$

and by interpolation from the table of the X functions,

$$\bar{x} = 16.018.$$

Whence

$$\bar{u} = \bar{x} z_0 = 42.30$$
 ft.

The velocity \overline{v} may be computed by either of equations (37) and (39), as they both give the same value to \overline{v} , namely,

$$\bar{v} = 3261 \text{ f. s.}$$

It will be seen that the increase of velocity from $\overline{u} = 42.26$ ft. to u = 46.6 ft., a travel of 4.34 ft., is only 23 f. s.

Since the vanishing surface of a grain of cordite is zero, (38) and (40) give the same value to \overline{p} . We find by either equation,

 $\overline{p} = 2745$ lbs. per in.²

The muzzle pressure by (40) is 2431 lbs. The distance travelled by the projectile at point of maximum pressure is

$$0.45 \times 2.64 = 1.19$$
 ft. = 2.38 calibers.

Equations (37) to (40) give all the information that was obtained by Noble's experiment with cordite, o".4. The only question that can arise is as to their accuracy in giving the velocity and pressure at every point of the bore. Equation (38) gives the observed maximum pressure and (37) the corresponding velocity. Equation (39) gives the observed muzzle velocity and (40) the corresponding pressure. These equations may be further tested by computing the velocities for 16.6, 21.6, and 34.1 feet travel and comparing with the measured velocities. The following table shows the results of this procedure. The differences between the measured and computed velocities are in all cases less than the probable error in measuring them, and are entirely negligible.

Travel of Projectile	Observed Velocity	Computed Velocity	OC.	Remarks
16.6 ft.	2794 f. s.	2781 f. s.	13	
21.6	2940	2942	-2	
34.1	3166	3172	-6	
46.6	3284	3284	0	

The limiting velocity, V_1 , is 4161 f. s.

It only remains to compute the characteristics v_c and f to solve completely the problem pertaining to this round. These

are found to be

 $v_c = 0.38$ inches per second,

and

 $f = 2266 \text{ lbs. per in.}^2$

This value of f would mean, if the problem under consideration were completely solved, that one pound of the gases of this powder, at temperature of combustion, confined in a volume of one cubic foot, would exert a pressure of 2,266 pounds per square inch. But the problem is very far from being solved rigorously. In the deduction of equation (18), Chapter II, which is the basis of all our formulas, there were neglected the following energies:

- 1. The heat lost by conduction to the walls of the gun.
- 2. The work expended on the charge, on the gun and carriage, and in giving rotation to the projectile.
- 3. The work expended in overcoming passive resistances, such as forcing, friction along the grooves, the resistance of the air, etc. In short the entire work of expansion was supposed to be employed in giving motion of translation to the projectile, and to be measured by the acceleration produced.

It may be seen, however, from a careful consideration of equation (18) and the use made of it in deducing the X functions that these functions are independent of the value of f; and that when this factor has been determined so as to satisfy completely such experiments as we have just been considering, these neglected energies are practically allowed for. Indeed, they are all contained implicitly in the factors M and N. Similar remarks apply to τ whose deduced value from (2), Chapter IV, depends upon the exponent of p_o/p , about which there is considerable uncertainty. But these characteristics are unnecessary for determining the equations of the velocity and pressure curves from such data as we have been considering. But they are of

use in deducing the circumstances of motion when the charge varies, as has been already shown.

We will now give a few illustrative examples which can be solved by this one round.

Example 1.—What thickness of layer was burned from the grains when the projectile had travelled 16.6 ft.?

Combining equations (12) and (14), Chapter IV, gives

$$l = \frac{l_o X_o}{\overline{X}_o}$$

That is, the thickness of layer burned from the surface of a grain of powder of whatever shape, in the bore of a gun, varies directly as the function X_o . In this example, applying the known values of l_o , \overline{X}_o and X_o (for u = 16.6), we find

$$l = 0.1443$$
 in. Ans.

Example 2.—What was the velocity of combustion of the grains at the point of maximum pressure?

We have (equation (53'), Chapter IV),

$$v'_c = v_c \left(\frac{p}{p_o}\right)^{\frac{1}{2}} = 0.3795 \left(\frac{47040}{14.6967}\right)^{\frac{1}{2}} = 21.47 \text{ in. per sec.}$$
 Ans.

Example 3.—What must be the diameter of the grains of this powder in order that the charge of 27.5 lbs. should all be burned when the projectile has travelled 16.6 ft.?

When the only variation in the charge and conditions of loading is in the thickness of web, equation (53), Chapter IV, shows that the X_o function of the distance travelled by the projectile when the powder is all burned is directly proportional to the web thickness. We therefore find, by employing known numbers,

$$2 l_0 = 0.2885$$
 in. Ans.

Cordite, o".35.—Preliminary calculations show that the cordite fired in this round was not quite all burned in the gun.

We will therefore compute N and M by (30) and (31) with the following data:

log
$$a = 0.21265$$

log $z_o = 0.46666$ $\therefore z_o = 2.929$ ft.
 $v_1 = 2583$ f. s. = velocity for 21.6 ft. travel.
 $v_2 = 2915$ f. s. = muzzle velocity.
log $X'_o = 0.84614$
log $X''_o = 0.96201$
log $X'_1 = 0.55164$.
log $X''_1 = 0.74763$

These numbers give

$$\log N = 8.73582 - 10$$

 $\log M = 6.48164$
 $\log M^{1} = 4.75514$

The results of further calculations for this round are:

$$p_m = 34093$$

 $\overline{u} = 46.96$ ft.
 $f = 2277$ lbs.
 $v_c = 0.315$ in. per sec.

The differences between the observed and computed velocities for u = 16.6 ft. and u = 34.1 ft., are, respectively, 12 and 6 f. s.

The limiting velocity for this round is 3731 f. s.

The mean crusher-gauge pressure was 30352 lbs., which is certainly erroneous. The force of the powder is practically the same as for cordite, 0".4; but this latter seems to be a quicker powder than cordite 0".35.

Cordite, o".3.—The cordite fired with this charge was all burned in the gun and we will, therefore, compute N and M by equations (35) and (34), with the following data:

log
$$a = 0.26928$$

log $z_0 = 0.48190$. . . $z_0 = 3.033$ ft.

00

$$v_1 = 2495$$
 f. s. $v_m = 2914$ f. s. $\log X_2 = 9.78256 - 10 \log X'_2 = 9.66596 - 10 \log X'_o = 0.79917$

By means of these numbers the various formulas give

$$\begin{array}{l} \log N = 8.80138 - 10 \\ \log M = 6.54986 \\ \log M' = 4.80822 \\ \log P' = 4.92766 \\ \log \frac{V_1^2}{z} = 7.14642 \\ \log \frac{\overline{X}_o}{x} = 0.89759 \\ \frac{\overline{u}}{x} = 31.3 \text{ ft.} \\ \frac{\overline{v}}{p} = 3331.2 \text{ lbs. per in.}^2 \\ f = 2521 \text{ lbs.} \\ v_c = 0.309 \text{ in. per sec.} \end{array}$$

The computed maximum pressure is 37287 lbs., which is but 327 lbs. in excess of the mean crusher-gauge pressure. The corresponding velocity is 877.2 f. s., and travel of projectile 1.33 ft. The differences between the observed and computed velocities for u = 21.6 ft., and u = 34.1 ft. are 1 and 5 f. s. respectively.

The limiting velocity is 3743 f. s.

This powder is apparently stronger than either the o".35 or o".4 cordite. These latter are evidently of the same composition, known as "Mark 1," while the former may have been the so-called "Cordite M. D.," which is said to have a slightly reduced rate of burning and to give higher velocities. Its composition is gun-cotton 65 per cent., nitro-glycerine 30 per cent., and mineral jelly 5 per cent.

Example.—Suppose the cordite, o".3, to be moulded into

cubes of the same web thickness. Determine the equations of velocity and pressure. We have $\alpha=3$, $\lambda=-1$, and $\mu=\frac{1}{3}$, while V_1 and \overline{X}_o remain the same as already found. We now have $M=\frac{3V_1^2}{\overline{X}_o}$, $N=-\frac{1}{\overline{X}_o}$, and $N'=\frac{1}{3\overline{X}_o^2}$: The equations are therefore,

$$v^2 = [6.72595] X_1 \{ 1 - [9.10241 - 10] X_o + [7.72770 - 10] X_o^2 \}$$
 and

$$p = [4.98431] X_3 \{ 1 - [9.10241 - 10] X_4 + [7.72770 - 10] X_5 \}$$

The maximum pressure computed by this last formula is 45726 lbs. The muzzle velocity is, of course, the same as before, as is also the velocity \overline{v} .

Application to the Hotchkiss 57 mm. Rapid-Firing Gun.—The data for the following discussion are taken from a paper by Mr. Laurence V. Benét, printed in the *Journal U. S. Artillery*, Vol. 1, No. 3. The gun experimented with was a standard pattern, all steel, 57 mm. Hotchkiss rapid-firing gun, and the experiments consisted in "cutting off successive lengths from the chase and observing the velocities of a series of rounds fired with each resulting travel of projectile." The data necessary for this discussion are the following:

GUN DATA.

Area of cross-section of bore, 0.2592 dm.² Equivalent diameter, 5.745 cm. Net volume of powder chamber, 0.887 dm.³

POWDER AND PROJECTILE.

"Two brands of the same type of smokeless powder were employed, both of which were manufactured at the Poudrerie Nationale de Sevran-Livry; they were designated as $B N_1$

and BN_{144} . These powders are in the form of thin strips, which are scored longitudinally on one side with a series of parallel and very narrow grooves. The chemical composition is unknown." The grains were of the following dimensions and densities:

	$B N_1$.	$B N_{144}$.
Length of strips,	76 mm.	85 mm.
Distance between scores,	1.4 mm.	1.6 mm.
Thickness of strips,	0.5 mm.	0.6 mm.
Specific gravity,	1.57	1.78
The elements of loading wer	e as follows:	
Weight of charge,	0.460 kilos.	0.400 kilos.

Weight of projectile, 2.720 kilos. 2.720 kilos. Density of loading, 0.519 0.451

The velocities were measured by means of two Boulengé-Breger chronographs on independent circuits; and the pressures were determined by means of a crusher gauge seated in the breech block of the gun. The mean pressures at the breech were for $B N_1$ powder, 2547 kilos. per cm.²; and for $B N_{144}$, 2543 kilos. per cm.²

From the firing records was obtained the following table giving the velocity of the projectile corresponding to each length of travel in the bore:

Velocity in Bore with	TRAVEL	OF SHELL	Velocity in Bore with	Remarks
B N ₁ .	Metres	Calibers	B N ₁₄₄	
Metres per Sec.			Metres per Sec.	
543.1	0.880	15.44	503.7	
574.4	1.051	18.44	534.9	
595.0	I.222	21.44	553.1	
612.6	1.393	24.44	565.0	
622.3	1.564	27.44	573.5	
636.5	1.792	31.44	591.0	
648.3	2.020	35.44	600.7	

We will first consider the powder $B N_1$, and compute by the

proper formulas—already many times referred to,—the values of a, z_o and x for the given charge and travels of projectile. Then take from the table of the X functions the logarithms of X_o , X_1 and X_2 . All these are given in the following table for convenient reference:

$\log a =$	0.11053.	$\log z_o =$	9.35962	 10.
0	00	0 0	7 007	

u Metres	x .	$\log X_0$	$\log X_1$	$\log X_2$	Remarks
0.880	3.8447	0.74183	0.35356	9.61173	
1.051	4.5918	0.77092	0.41100	9.64008	
1.222	5.3389	0.79521	0.45765	9.66244	
1.393	6.0860	0.81604	0.49670	9.68067	
1.564	6.8362	0.83425	0.53013	9.69590	
1.792	7.8293	0.85540	0.56819	9.71279	
2.020	8.8254	0.87382	0.60062	9.72681	

It is known that the powder was all burned in the gun, as might be also inferred from the thinness of web; and the first step is to determine the travel of projectile when this takes place, in other words the value of \overline{u} . On account of the "series of parallel and very narrow grooves" with which the strips were scored on one side, it is difficult to ascertain the form characteristics from geometrical considerations. Their determination will therefore be left until M and N are computed from the measured velocities. μ will be considered zero.

The expression for V_1 , the limiting velocity, is

$$V_1 = \frac{V}{\sqrt{X_2}},$$

where V is any velocity after the powder is all burned and X_2 a function of the corresponding travel of projectile. If then we compute V_1 by this formula for all the measured velocities, and find that it is approximately constant for a certain number of measured velocities nearest the muzzle, we shall have an indication of the travel of projectile when the powder

is all burned. The following table gives the values of V_1 so computed:

и	V _t .
0.880 m.	849 m. s.
1.051	869
1.222	878
1.393	885
1.564	883
1.792	886
2.020	888

An examination of this table shows that \overline{u} lies between 1.222 and 1.393, or that \overline{x} lies between the numbers 5.3389 and 6.0860 and rather nearer the former than the latter.

We will assume $\bar{x} = 5.6$ and $V_1 = 885.5$ m. s., which is a mean of the last four values. Since

$$\overline{v} = V_1 \sqrt{\overline{X}}$$

we find

$$\bar{v} = 605.1$$
 m. s.

N and M can now be computed by (30) and (31), which do not contain the form characteristics.

The data are: $v_1 = 543.1$ m. s., $v_2 = 605.1$ m. s., $\log X'_o = 0.74183$, $\log X''_o = 0.80284$, $\log X'_1 = 0.35356$, and $\log X''_1 = 0.47205$.

The results of the calculations are:

$$\begin{cases} N = 8.68636 - 10 \\ \log M = 5.25171 \\ \log M' = 3.62063 \end{cases} \text{ While powder is burning.}$$

$$V = \begin{bmatrix} 2.94719 \end{bmatrix} \checkmark X_2 \\ P = \frac{\begin{bmatrix} 3.78589 \\ (1+x)^{\frac{3}{8}} \end{bmatrix}} \text{ After powder is all burned.}$$

The following table shows the agreement between the observed and computed velocities:

Travel of	VELO	CITIES	O. – C.	Remarks	
Projectile	Observed	Computed			
o.880 m.	543.1 m. s.	543.1 m. s.	0.0		
1.051	574.4	572.8	1.6		
I.222	595.0	597·4	-2.4		
I.393	612.6	613.1	-0.5		
1.564	622.3	623.9	-1.6		
1.792	636.5	636.3	0.2		
2.020	648.3	646.5	1.8		

The greatest of these differences is less than one-half of one per cent. of the observed velocity and the others are practically nil. The maximum pressure computed by the formula

$$p_m = [9.85640 - 10] M' \{ 1 - [0.48444] N \}$$

is 2555 kilos. per cm.², differing by less than one-third of one per cent. of the mean crusher-gauge pressure. These results show that the assumed value of $\overline{x} = 5.6$ is practically correct.

Finally, we have,

$$f = 7883$$
 kg. per cm.²

and

$$v_c = 0.438$$
 cm. per sec.
= 0.172 in. " "

The form characteristics α and λ can be computed by the formulas

$$\alpha = \frac{M \overline{X}_o}{V_1^2}$$
 and $\lambda = N \overline{X}_o$

From these we find $\alpha = 1.4460$ and $\lambda = 0.3045$.

For the $B \dot{N}_{144}$ powder the equations for the velocity and pressure curves are found, by a process entirely similar to the above, to be,—while the powder is burning,—

$$v^2 = [5.24187] X_1 (1 - [8.75166 - 10] X_0)$$

and

$$p = [3.56310] X_3 (1 - [8.75166 - 10] X_4)$$

After the powder is burned the equations become

$$V = [2.92007] \sqrt{X_2}$$

and

$$P = \frac{[3.68425]}{(1+x)^{\frac{4}{5}}}$$

The following table shows the agreement between the observed and computed velocities:

Travel of Projectile m.	Observed Velocity m. s.	Computed Velocities m. s.	OC.	Remarks
0.880	503.7	503.9	-0.2	
1.051	534.9	532.2	2.7	
1.222	553.1	553.6	-0.5	
1.393	565.0	566.0	-1.0	
1.564	573.5	576.5	-3.0	
1.792	591.0	588.4	2.6	
2.020	600.7	598.5	2.2	

The value of f for B N_{144} comes out 8001.5 kilos. per cm.², and v_c is found to be 0.5268 cm. per sec. This powder is therefore slightly "stronger" than B N_1 and about 22 per cent. quicker; and this notwithstanding its greater density.

Application to the Magazine Rifle, Model of 1903.—The following data pertaining to this rifle were obtained partly from a descriptive pamphlet issued by the Ordnance Department, and partly through the courtesy of officers of the Ordnance Department on duty at the Springfield Armory and Frankford Arsenal, to whom the writer is under special obligations:

Caliber, 0.3 inches.

Volume of chamber, 0.252 cubic inches.

Total travel of bullet in bore, 22.073 inches.

Mean weight of powder charge, 44 grains.

Weight of bullet, 220 grains.

"The standard muzzle velocity of this ammunition is 2300

f. s., with an allowed mean variation of 15 f. s. on either side of the standard. The powder pressure in the chamber is about 49,000 pounds per square inch."

The powder used with this rifle is composed essentially of 70 per cent. nitrocellulose and 30 per cent. nitro-glycerine. "The grains are tubular, being formed by running the powder colloid through a die 0.09 inch in diameter, with a pin 0.03 inch in diameter; and the string thus made is cut 21 to the inch." There are considerable variations in the length and diameter of the grains "due to the fact that the string is not cut exactly perpendicular to its axis, and to irregularities in shrinking. There are 83,000 to 91,000 grains per pound. The specific gravity is about 1.65, and the gravimetric density is from 0.90 to 0.94."

On account of the tubular form of the grains the characteristic μ is zero, and therefore the equations for velocity and pressure are binomials. We have reliable measured interior velocities for this rifle, obtained at the Springfield Armory in the fall of 1903, by firing with a rifle the barrel of which was successively cut off one inch. Five shots (sometimes more) were fired for each length of barrel and the velocities were measured at a distance of 53 ft. from the muzzle, and reduced to muzzle velocity by well-known methods. (See Table A.)

It is known that the charge in the magazine rifle is all burned at, or very near, the muzzle. We may, therefore, take the two extreme reduced velocities of the series for v_1 and v_2 and thereby minimize the effects of errors in measuring the velocities. The firing data are then,

$$v_1 = 1274 \text{ f. s.};$$
 $v_2 = 2277.6 \text{ f. s.}$
 $u_1 = 3.073 \text{ in.};$ $u_2 = 20.073 \text{ in.}$

The weight of charge in these firings was 45.1 grains and weight of bullet 220 grains. The preliminary calculations give

$$\Delta = 0.7077$$

$$\log a = 9.90686 - 10$$

$$\log z_o = 0.30878 \qquad \therefore z_o = 2.036 \text{ in.}$$

$$x_1 = 1.5093, \qquad \log X'_o = 0.57969, \qquad \log X'_1 = 0.00146.$$

$$x_2 = 10.84125, \quad \log X''_o = 0.90504, \quad \log X''_1 = 0.65421.$$

These numbers and the velocities v_1 and v_2 , substituted in (30) and (31), give

$$\log N = 8.73379 - 10$$

and

$$\log M = 6.30896.$$

We also find

$$\log M' = 4.91902.$$

The formulas for velocity and pressure are, therefore,

$$v^2 = [6.30896] X_1 \{ 1 - [8.73379 - 10] X_o \}$$

 $p = [4.91902] X_3 \{ 1 - [8.73379 - 10] X_4 \}$

We have

$$\alpha = \frac{M \overline{X}_1}{\overline{v}^2}$$

and this substituted in (15), Chapter IV, gives

$$y = \tilde{\omega} \frac{M\overline{X}_2}{v_2} X_o \{ \mathbf{1} - [8.73379 - \mathbf{10}] X_o \}$$

Since in this case \overline{v} and \overline{X}_2 refer to the muzzle, we have for the powder burned, in grains,

$$y = [0.99736] X_o \{ 1 - [8.73379 - 10] X_o \}$$

or, in another form more convenient for computation,

$$y = [4.68840 - 10] \frac{v^2}{X_2}$$

Table A gives the measured and computed velocities for the travels of projectile in the first column, and also the weight of powder burned at each travel.

TABLE A

Travel of Projectile, inches	Mean Velocity 53 Feet from Muzzle, f. s.	Muzzle Velocity Deduced from Measured f. s.	Computed Velocity, f. s.	OC.	Powder Burned, grains
3.073	1253	1274	1274	0	29.99
4.073	1402	1426	1432	- 6	32.61
5.073	1531	1558	1555	3	34.63
6.073	1633	1662	1656	6	36.25
7.073	1742	1772	1740	32	37.59
8.073	1771	1802	1812	-10	38.71
9.073	1860	1894	1874	20	39.66
10.073	1909	1943	1929	14	40.48
11.073	1957	1992	1976	16	41.19
12.073	1989	2023	2018	5	41.81
13.073	2016	2052	2057	- 5	42.36
14.073	2050	2086	2091	- 5	42.83
15.073	2069	2105	2122	-17	43.25
16.073	2104	2140	2151	-11	43.63
17.073	2129	2165	2177	-12	43.95
18.073	2183	2219	2200	19	44.25
19.073	2163	2200	2222	-22	44.50
20.073	2201	2238	2242	- 4	44.73
21.073	2203	2240	2261	- I	44.93
22.073	2240	2278	2278	0	45.10

Table B, on page 134, supplements Table A by giving computed velocities and pressures from the origin of motion. The velocity curve in the diagram, Fig. 2, on page 135, shows at a glance the agreement between theory and observation.

It will be observed that the computed pressures depend entirely upon two measured velocities. Also that the maximum pressure occurs when x = 0.45, and agrees with the official statement. The muzzle pressure is about 6,000 lbs. per in.²

Powder Characteristics.—The form characteristics of these grains according to the given dimensions are

$$\alpha = 1.63$$
 and $\lambda = 0.3865$.

But these minute grains, of which there are 560 in the service charge, shrink irregularly and many of them doubtless are more or less abraded and perhaps broken, so that it is impossible to determine the mean values of α and λ geometrically with any

TABLE B

x	u, inches	Computed Velocity, f. s.	Computed Pressure, lbs. per inch ²	Powder Burned, grains	Pressure on Base of Projectile, pounds
0.000	0,000	0.000	0.00000	0.000	000
0.001	0.002	8.590	4501	1.081	318
0.01	0.020	47.854	13835	3.375	978
0.I	0.204	254.91	37008	10.139	2556
0.2	0.407	408.98	45117	13.842	3189
0.3	0.611	531.70	48456	16.475	3425
0.4	0.814	635.14	49640	18.554	3509
0.45	0.916	681.48	49769	19.454	3518
0.5	1.018	724.89	49695	20.283	00
0.6	1.222	804.25	49118	21.766	
0.8	1.629	939.77	47039	24.221	
1.0	2.036	1052.5	44492	26.205	
I.2	2.443	1148.8	41882	27.867	
1.4	2.850	1232.6	39369	29.291	
1.6	3.258	1306.4	37023	30.533	
2.0	4.072		32854		
2.5	5.090		28550		
3	6.108		25061		
4	8.144		19812		
4 5 6	10.180		16086		
	12.216		13319		
7 8	14.252		11188		
	16.288		9501		
9	18.324		8134		
10	20.360		7006		
II	22.396		6045		

certainty. They may, however, be deduced from the values of M and N. From these we find

$$\alpha = 1.7710$$
 $\lambda = 0.4353$

Finally we find

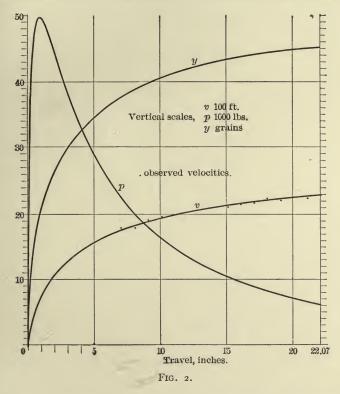
$$f = 1622.5$$
 lbs. per in.²

and

$$v_c = 0.28$$
 in. per second.

Formulas for Designing Guns for Cordite.—The caliber, of course, is given, and the weight of the projectile of desired length

and form of head can be computed by known methods.* grain characteristics and density of cordite and the values of f



and v_c are also known. The necessary formulas for this discussion, given in the order in which they will be used, are the following:-

$$\tilde{\omega} = [8.55783] \Delta V_c \quad . \quad . \quad . \quad (a)$$

$$a = \frac{1}{\Delta} - \frac{1}{\delta}$$
 (b)
$$z_o = [1.54708] \frac{d \tilde{\omega}}{d^2} \qquad . \qquad . \qquad . \qquad (c)$$

$$z_o = \left[1.54708\right] \frac{d \, \tilde{\omega}}{d^2} \qquad . \qquad . \qquad . \qquad . \qquad (c)$$

^{*} See the author's "Handbook" (Artillery Circular N), chapter xi.

$$\overline{X}_o = b \left\{ \mathbf{1} + \left(\mathbf{1} - \frac{[0.48444]}{b} \right)^{\frac{1}{2}} \right\}$$
 . . (e)

in which

$$b = \frac{[2.12890]f}{a p_m}$$

This equation is deduced from (32), eliminating M' and N.

$$M = \frac{2V_1^2}{\overline{X}_0}; \quad N = \frac{1}{2\overline{X}_0} \quad . \quad . \quad . \quad . \quad . \quad . \quad (f)$$

$$M' = [7.82867 - 10] M \frac{w}{a \tilde{\omega}} \dots \dots (g)$$

$$v^2 = M X_1 [I - N X_0]$$
 (h)

$$p = M' X_3 (I - N X_4) \dots (i)$$

$$p_m = [9.85640 - 10] M' (1 - [0.48444] N)$$
 . (j)

$$l_o = [8.56006 - 10] \frac{v_c \sqrt{a w \tilde{\omega}}}{d^2} \overline{X}_o$$
 (k)

Example.—Take the hypothetical 7-inch gun already considered on page 111, for which d = 7'' and w = 205 lbs. For cordite of o''.3 diameter we found f = 2521 lbs. per in.2, and $v_c = 0.309$ in. per sec. Also $\delta = 1.56$. The only assumptions necessary are the volume of the chamber (V_c) and the density of loading. And this last is not purely arbitrary, since considerations of safety to the gun and its efficiency restrict its value to narrow limits,—say from 0.4 to 0.6. This latter value is often exceeded, especially in our service; but it is believed that by choosing the proper shape and size of grain this can always be

avoided. As f and v_c are unusually large for cordite, we will take $\Delta = 0.4$; and for a first assumption will give the chamber a volume of 3,000 c. i.; which is less than the volume of the chamber of the 6-inch wire-wound gun. Finally we will take $p_m = 37,000$ lbs. per in.², leaving the muzzle velocity and travel in the bore for later consideration.

From the given data we find, by means of the above formulas, $\tilde{\omega}=43.352~$ lbs., $\log a=0.26928,~\log z_o=1.76317,~\log V_{1^2}=7.17066,~\log \overline{X}_o=0.90184,~\log M=6.56985,~\log M'=4.80398$ and $\log N=8.79713-10$.

The equations for velocity and pressure are, therefore,

$$v^2 = [6.56985] X_1 (1 - [8.79713 - 10] X_0)$$

and

$$p = [4.80398] X_3 (1 - [8.79713 - 10] X_4)$$

This last equation, which is the same as equation (j) when x = 0.45, makes $p_m = 37000$, thus verifying the calculations.

The muzzle velocity will, of course, depend upon where we place the muzzle, in other words upon the value adopted for u_m . If we regard 40 calibers as a suitable travel in the bore, we shall have

$$u_m = 40 \times 7'' = 280''$$

whence

$$x_m = u_m/z_o = 4.8305$$

For this value of x, Table I gives

$$\log X_o = 0.77912$$
$$\log X_1 = 0.42689$$

and these in the above velocity equation give

$$v_m = 2487$$
 f. s.

We find, from Table I, taking $\log \overline{X}_o$ as the argument,

$$\bar{x} = 10.613;$$

and by
$$(m)$$

$$\overline{u} = 615.19 \text{ in.}$$
Also by (k)

$$2 l_o = 0''.47$$
and by (l)

$$k_m = 0.9394.$$

That is 94 per cent. of the charge was burned at the assumed muzzle.

If the maximum pressure is increased to 38,000 lbs., the density of loading and volume of chamber remaining as before, the velocity for a travel of 280 inches will be increased to 2503 f. s., and the thickness of web, or diameter of the grain, will be reduced to 0".45. This slight diminution in the diameter of the grain increases the initial surface of combustion of the charge about $3\frac{1}{2}$ per cent., which fully accounts for the increased maximum pressure.

If we take $\Delta=0.5$, $V_c=3,000$ c. i. and $p_m=37,000$ lbs. per in.², there results $\tilde{\omega}=54.19$ lbs., $v_{280}=2570$ f. s., and $2\ l_o=0''.66$.

Trinomial Formulas for Velocity and Pressure.—Trinomial formulas occur when the grains of which the charge is composed are of such form and dimensions that the form characteristic μ cannot be regarded as zero. Spherical, cubical, and multiperforated cylindrical grains are of this kind. For the first two forms mentioned the second term is negative and the third positive; while for m.p. grains (those used in our service), the second term is positive and the third negative.

For spherical and cubical grains we may have, before the powder is all burned, the two independent equations,

$$v_1^2 = M X_1' \left(I - N X_o' + \frac{I}{3} N^2 X_o'^2 \right)$$
$$v_2^2 = M X_1'' \left(I - N X''_o + \frac{I}{3} N^2 X''^2_o \right)$$

Put for convenience,

$$\begin{split} a &= \left(\frac{v_2}{v_1}\right)^2 \frac{X'_1}{X_1''}, \ b &= \frac{X'_o}{X''_o}, \\ c &= \frac{3 \ (\mathbf{1} - a \ b)}{2 \ (\mathbf{1} - a \ b^2) \ X''_o} \ \text{and} \ d &= \frac{2 \ c \ (\mathbf{1} - a)}{(\mathbf{1} - a \ b) \ X''_o} \ . \end{split}$$

Then the quadratic equations give, using the sign applicable to this problem,

$$N = c - \sqrt{c^2 - d} = c \left(1 - \left(1 - \frac{d}{c^2} \right)^{\frac{1}{2}} \right)$$
 . (41)

The value of M may now be computed by either of the above expressions for v^2 . Or, if V_m is the muzzle velocity, that is, if the powder is all burned in the gun, M may be computed by the formula derived from (3),

If the powder is not all burned in the gun and our data are a muzzle velocity and the crusher-gauge pressure (assumed to be the maximum pressure), N may be computed by the following process: Compute the auxiliary quantities b, c, and d by the formulas:

$$b = [7.68507 - 10] \frac{w v_m^2}{a \tilde{\omega} p_m X_1}; c = \frac{3 \left(1 - \frac{b X_4}{X_o}\right)}{2 X_o \left(1 - \frac{b X_5}{X_o^2}\right)}$$
$$d = \frac{3 (1 - b)}{X_o^2 \left(1 - \frac{b X_5}{X_o^2}\right)}$$

Then

$$N = c \left\{ \mathbf{1} - \left(\mathbf{1} - \frac{d}{c^2} \right)^{\frac{1}{2}} \right\} \qquad (43)$$

The functions X_4 and X_5 pertain to the tabular value of x which gives the maximum pressure. If we take this to be 0.45

no material error will ensue. We therefore have

$$\log X_4 = 0.48444$$
$$\log X_5 = 0.93587$$

The function X_o pertains to the muzzle.

It should be remembered that equations (41), (42), and (43) are applicable to cubical and spherical grains only.

Application to Noble's Experiments with Ballistite.—The ballistite consisted of equal parts of dinitrocellulose and nitroglycerine and was in the form of cubes 0.3 of an inch on a side. The gun, powder, and firing data are as follows: d=6 inches, $\Delta=0.4$, $\delta=1.56$, $\tilde{\omega}=20$ lbs., and w=100 lbs. From these we find $\log a=0.26928$ and $\log z_o=0.48190$. $\therefore z_o=3.033$ ft. The following table, which explains itself, is formed for convenient reference:

u, ft.	$x=u/z_0$	Observed Velocity f. s.	$\log X_0$	$\log X_1$	$\log X_2$	Remarks
16.6	5.473	2416	0.79917	0.46513	9.66596-10	
21.6	7.121	2537	0.84069	0.54181	9.70112	
34.1	11.242	2713	0.91049	0.66339	9.75289	
46.6	15.363	2806	0.95685	0.73940	9.78256	

As the powder was not quite all burned in the gun the extreme measured velocities are available for determining N and M by means of (41). The data are $v_1 = 2416$ f. s., $v_2 = 2806$ f. s., $\log X'_o = 0.79917$, $\log X''_o = 0.95685$, $\log X'_1 = 0.46513$ and $\log X_1'' = 0.73940$. Substituting these in (41), we find

$$\log N = 9.03843 - 10$$

$$\log N' = 7.59974 - 10$$

$$\log M = 6.62918$$

$$\log M' = 4.88754$$

The equations for velocity and pressure are, therefore,

$$v^2 = [6.62918] X_1 \{ 1 - [9.03843 - 10] X_o + [7.59974 - 10] X_o^2 \}$$

 $p = [4.88754] X_3 \{ 1 - [9.03843 - 10] X_4 + [7.59974 - 10] X_b \}$

The equation for velocity will, of course, give the observed velocities for u=16.6 ft., and u=46.6 ft. It should also give the observed velocities, if our method is correct, for u=21.6 ft. and u=34.1 ft., and indeed for every point in the bore from the firing seat to the muzzle. The velocities computed for these intermediate values of u are 2536.5 f. s., and 2710 f. s., respectively, which differ so slightly from the measured velocities as to be negligible. The maximum pressure, which occurs in this case when x=0.4, is 39,163 lbs. per in.²; and the corresponding velocity 865.8 f. s., and travel of projectile 1.213 ft.

The distance travelled by a projectile to the point where the powder is all burned is determined by means of \overline{X}_o and the table of the X functions. We have for cubical grains for which λ is unity, $\overline{X}_o = 1/N$. Therefore for this example,

$$\log \overline{X}_o = 0.96157$$

Corresponding to this value of $\log \overline{X}_o$, we find, by interpolation from the table,

$$\overline{x} = 15.8649$$
, $\log \overline{X}_1 = 0.74694$, and $\log \overline{X}_2 = 9.78536 - 10$.

To compute \overline{u} , we have $\overline{u} = 15.8649 \times 3.033 = 48.12$ ft. The charge was therefore not all burned in the gun, though the fraction of the charge remaining unburned was exceedingly small, practically zero. To get an expression for the fraction of the charge burned for any travel of the projectile, we have, from (45), Chapter IV,

$$v^2 = k V_1^2 X_2$$

But from (3), for cubic grains,

$$V_1^2 = \frac{M}{3 N}$$
 : $\log V_1^2 = 7.11363$ (for this example).

Therefore for cubic grains,

$$k = \frac{3 N v^2}{M X_2},$$

which for this example reduces to

$$k = [2.88637 - 10] \frac{v^2}{X_2}$$

Applying different velocities and the corresponding values of $\log X_2$ in this formula it will be found that the charge was practically consumed long before the projectile reached the muzzle. Indeed nineteen-twentieths of the charge was burned when the projectile had travelled 16.6 ft.

We may also determine k in terms of the travel of projectile by means of the equation

$$k = I - \left(I - \frac{X_o}{\overline{X}_o}\right)^3$$
 (Eq. (46), Chapter IV).

Or, if we wish to know the distance travelled by the projectile when a given fraction of the charge is burned, we have

$$X_o = \overline{X}_o \left\{ \mathbf{I} - (\mathbf{I} - k)^{\frac{1}{3}} \right\}$$

As an example, suppose $k = \frac{1}{2}$. Then

$$X_o = \overline{X}_o \left(\mathbf{I} - \left(\frac{\mathbf{I}}{2} \right)^{\frac{1}{3}} \right) = \frac{\mathbf{I} - \left(\frac{\mathbf{I}}{2} \right)^{\frac{1}{3}}}{N}$$

Applying the value of N found for this round and completing the calculations it will be found that one-half the charge was burned when the projectile had travelled one foot.

The expressions for V and P for this round are

 $V = [3.55681] \sqrt{X}$

and

$$P = \frac{[4.89487]}{(1+x)^{\frac{4}{3}}}$$

The last two formulas give the velocity and pressure upon the supposition that all the powder was converted into gas at temperature of combustion before the projectile had started. The initial value of P is found by making x zero. Whence P' = 78,500 lbs. per in.²

Finally we find

j = 2338 lbs.

and

 $v_c = 0.266$ in. per sec.

The following table was computed by the formulas deduced for this round for comparison with the deductions from Sir Andrew Noble's velocity and pressure curves. Unfortunately these curves, as published, are drawn to so small a scale and are so mixed up with other curves that it is difficult to get the velocities and pressures from them with much precision.

Note:—The velocities and pressures in the second and third columns were computed by formulas slightly different from those deduced above. But the differences are so small as to be of no account in the discussion.

x	u ft.	Computed Velocities, f. s.	Computed Pressures, lbs. per inch ²	Pounds of Powder Burned	V, f. s.	P, lbs. per in. ²
0.000	0.000	0.0 12.449	0 4196	0.0 0.716	0.0 65.786	78500 78397
0.01	0.030	68.889	12672	2.206	207.40	77466
0.05	0.152	221.52	25254	4.683	457.78	73557
O.I	0.303	359.32	32007	6.357	637.37	69132
0.2	0.607	569.36	37680	8.464	875.21	61560
0.4	1.213	869.33	39439	10.967	1174.0	50122
0.6	1.820	1087.2	37567	12.549	1372.5	41948
0.8	2.427	1257.7	34846	13.686	1520.3	35852
1.0	3.033	1396.6	32050	14.557	1637.0	31153
2.0	6.066	1842.4	21268	17.043	1995.9	18143
3.0	9.100	2093.0	15013	18.225	2192.5	12363
4.0	12.133	2257.I	11164	18.892	2322.4	9181
5.0	15.166	2374.2	8645	19.299	2416.9	7200
5.473	16.6	2419.0	7742	19.435	2453.5	6507
7.121	21.6	2538.0	5500	17.742	2554.9	4809
11.242	34.I	2710.0	2893	19.979	2711.8	2782
15.363	46.6	2806.0	1890	19.999	2806.0	1890

The computed velocities in the third column of this table, corresponding to the travels of projectile in the second column,

agree very well with those deduced from Sir Andrew Noble's velocity curve, from the origin of motion to the muzzle, a distance of 46.6 ft. As the velocities are thus shown to be correct, the pressures in the fourth column are, from their manner of derivation as given in Chapter IV, necessarily correct also. That is, they correspond to the energy of translation of a hundred-pound projectile. In this respect they are more accurate than the pressures given by Sir Andrew's pressure curve which was derived from his velocity curve by graphic methods not sufficiently precise for the great accelerations encountered in ballistic problems.

The writer is indebted to Colonel Lissak, formerly Instructor of Ordnance and Gunnery at West Point, for the accompanying diagram (Fig. 3) of the velocity and pressure curves whose coordinates are given in this table. Many interesting facts may be gleaned from an examination of these curves, and the formulas by which their coördinates were computed.

The two velocity curves v and V are both zero at the origin but immediately separate, attaining their greatest distance apart when the projectile has moved but a short distance. They then approach each other very gradually and become tangent at the point where the powder is all burned,—practically at the muzzle. Both curves are tangent to the axis of ordinates at the origin and parallel to the axis of abscissas at infinity. The pressure curve p begins at the origin, attains its maximum when the projectile has traveled about 15 inches, changes direction of curvature when u is about six feet and meets the axis of abscissas at infinity. The pressure curve P is convex toward the axis of abscissas throughout its whole extent. It lies above the curve p from u = 0 to u = 30 inches (about), then passes below p and the two curves become tangent at the point where the powder is all consumed. Finally the areas under the curves p and P are equal.

Example 1.—Suppose the charge in the example under

consideration to be increased from 20 to 25 lbs. Deduce the equations for velocity and pressure.

In solving this example, we will compute the new constants M, M', N and N' by equations (80) to (83), Chapter IV; and as the charge is increased by 25 per cent., a new value of f must be found by (90) and (90'). For a six-inch gun we will take $n = \frac{1}{3}$, provisionally; and since the weight of the projectile remains the same, n' must be zero. We therefore have

$$K = 86.12,$$

and the new value of f is 2518 lbs.

The new values of a and z_o for $\tilde{\omega} = 25$ lbs., are

$$\log a = 0.13321$$

 $\log z_o = 0.44277$ $\therefore z_o = 2.772$ ft.

Applying these numbers in the equations above mentioned we find for a charge of 25 lbs.,

$$\log M = 6.73881$$

$$\log M' = 5.03633$$

$$\log N = 9.01885$$

$$\log N' = 7.56058$$

which give the equations required.

These constants give $p_m = 55676$ lbs., and a velocity of 2841 f. s., for a travel of 16.6 ft. That is, an increase of 5 lbs. in the charge increases the maximum pressure 16,800 lbs. per in.², and the velocity at 16.6 ft. travel, 425 f. s. Taking the reciprocal of N gives

$$\log \overline{X}_o = 0.98115$$

and from the table,

$$\overline{x} = 18.1425$$

and

$$\overline{u} = 18.1425 \times 2.772 = 50.29 \text{ ft.}$$

The limiting velocity and fraction of charge burned are given by the equations

$$V_1^2 = \frac{M \overline{X}_o}{3}$$
 and $k = \frac{v^2}{V_1^2 X_2}$

Therefore

$$\log V_{1^2} = 7.24284$$

and

$$y = k \ \tilde{\omega} = [4.15510 - 10] \frac{v_2}{X_2}$$

From this last formula we find when u = 16.6 ft., y = 24.18 lbs.

The pressure at this point is found to be 10480 lbs. per in.² It is interesting to compare these results with those found with a charge of 20 lbs.

In order to lessen the maximum pressure the grains must be increased in size and thus diminish the initial burning surface. Suppose we increase the size of the cubes from o''.3 to o''.5 on a side. Determine the equations of velocity and pressure for a charge of 25 lbs. An examination of equations (80) to (82), Chapter IV, will show that when the only change in the data is in the thickness of web the new values of M, M', and N will be found by multiplying the previously determined values of these constants by the ratio of the web thicknesses,—in this case by o.6. We therefore have for 25 lbs. of o''.5 cubes

$$\log M = 6.51696$$

 $\log M' = 4.81448$
 $\log N = 8.79700 - 10$
 $\log N' = 7.11688 - 10$

From these we get

$$p_m = 38437$$
 lbs. per in.²

and

$$v = 2571$$
 f. s., for $u = 16.6$ ft.

The measured velocity for this travel of projectile was 2416

f. s., with a charge of 20 lbs. of o".3 cubes. Therefore by increasing the weight of charge 5 lbs., and at the same time enlarging the grain from o".3 to o".5 on a side the velocity is increased 155 f. s.,—and this without increasing the maximum pressure, though the mean pressure is, of course, considerably increased.

The pressure for u=16.6 ft., with a charge of 20 lbs. of the smaller grains, was 7741 lbs.; and with a charge of 25 lbs. of the larger grains, the pressure for the same travel would be 11181 lbs. The powder actually burned during this travel of projectile is a little more in this latter case than in the former, and the space in which it has been confined during its expansion is less, both of which facts account for the greater work performed.

From equation (19'), Chapter III, it follows that for two equal charges made up of grains of the same form and differing only in their size, the entire initial surfaces of the two charges vary inversely as the thickness of web. Therefore the initial surface of the charge of o."5 grains is $\frac{3}{5}$ of the initial surface of the same charge of o."3 grains. This accounts for the two charges giving the same maximum pressure. It may be remarked that the same results would have been obtained if the grains had been spherical instead of cubical.

Application to Multiperforated Grains.—A peculiar difficulty arises in the application of any system of interior ballistic formulas to multiperforated grains from the fact that they do not retain their original form until completely consumed—as do all other forms of grain in use,—but each grain breaks up, when the web thickness proper is burned through, into twelve slender rods, or "slivers," which burn according to a different law; and thus two independent sets of formulas become necessary to represent what actually takes place in the gun. It was previously sought to overcome this difficulty by supposing the web thickness to be slightly increased so as to

satisfy the equation of condition

$$\alpha (1 + \lambda - \mu) = 1$$

and thus ignoring the slivers.* This method represents quite satisfactorily the actual circumstances of motion so long as the grains retain their original form, but not afterward. It assumes that the slivers are all burned with the fictitious web thickness; that is, when, in all our guns, the projectile has performed approximately half its travel in the bore; while it is certain that in most cases with our service powders they are not completely consumed when the projectile leaves the bore. It is necessary, therefore, to divide the entire combustion of the grain into two periods and to deduce formulas that shall represent the law of burning, as well as the circumstances of motion, for each period.

From equation (22), Chapter III, we have, for m.p. grains

$$k = \frac{y}{\tilde{\omega}} = \alpha \frac{l}{l_o} \left\{ \mathbf{I} + \lambda \frac{l}{l_o} - \mu \frac{l^2}{l_o^2} \right\}$$

which gives the fraction of the charge consumed when any thickness l of the web has been burned, and this without any reference to the law of burning. When $l = l_a$, that is, when the entire web thickness has been burned, this equation becomes

$$k' = \alpha (1 + \lambda - \mu)$$

in which k' is the fraction of the charge less the slivers. If we substitute for α , λ and μ their values for any of our m.p. grains, we shall find for this critical point,

$$k' = 0.85$$
 (about),

and therefore the slivers constitute approximately 15 per cent. of the charge. These slivers burn according to another law. We may regard them as slender cylinders whose form characteristics are very approximately

$$\alpha = 2$$
, $\lambda = -1$, $\mu = 0$.

^{*} See Journal U. S. Artillery, vol. 24, p. 196, and vol. 26, pp. 141 and 276.

We will now deduce formulas for each period of burning. Designate all symbols referring to the point where the grains are converted into slivers by an accent, and those relating to the muzzle—including M, M' and N—by a subscript m.

Equation (11), Chapter IV, becomes, by suitable reductions,

$$K = [8.56006 - \text{ro}] \frac{v_c \sqrt{a w \tilde{\omega}}}{d^2 l_o} \quad . \quad . \quad (44)$$

in which v_c is the velocity of combustion under atmospheric pressure and l_o one-half the web thickness. From (12), Chapter IV, we have

$$l/l_o = K X_o$$

which, when the web thickness is burned through, gives for this critical point,

$$KX'_o = 1$$

Therefore from (44)

$$X'_{o} = [1.43994] \frac{d^{2} l_{o}}{v_{c} \sqrt{a w \tilde{\omega}}}$$
 . . . (45)

which gives X'_o when v_c and l_o are known. Also

$$v_c = [1.43994] \frac{d^2 l_o}{X'_o \sqrt{a w \tilde{\omega}}}$$
 . . . (46)

and

$$l_o = [8.56006 - \text{IO}] \frac{v_c X'_o \sqrt{a w \tilde{\omega}}}{d^2}$$
 . (46')

While the slivers are burning we have from (49), Chapter IV,

$$X_o = \frac{K}{2N_{ss}} \quad . \quad . \quad . \quad . \quad (47)$$

in which X_o refers to any travel between u' and u_m and

$$K = I - (I - k)^{\frac{1}{2}}$$

If we know the values of X_o and k for any point, we can

determine N_m by the equation

$$N_m = \frac{K}{2X_o} \quad . \quad . \quad . \quad . \quad (48)$$

Therefore at the point of breaking up into slivers (48) becomes

$$N_m = \frac{K'}{2 X'_o} \text{ or } X'_o = \frac{K'}{2 N_m} . . . (49)$$

The fraction k' which enters into K' can be computed for a grain of given dimensions by (21), Chapter III; and $\log K'$ can be taken from Table II with k' as the argument. Therefore when N_m is known X'_o can be found from (49), and then x', taken from Table I with X'_o as the argument, locates the point where the grains become slivers by the equation

In order to determine N_m it is necessary to assume a value for k_m , or the fraction of the entire charge burned at the muzzle, and check this assumed value by the given maximum (crushergauge) pressure. By (45), Chapter IV, we have,

$$V_{1}^{2} = \frac{v_{m}^{2}}{k_{m} X_{2m}} \qquad (51)$$

by means of which V_1 can be determined from muzzle data. The constants M, N and N' to be used in the velocity and pressure formulas from u = 0 to u' are given by the formulas

$$M = \frac{\alpha V_1^2}{X'_o}, \quad N = \frac{\lambda}{X'_o} \text{ and } N' = \frac{\mu N^2}{\lambda^2} \quad .$$
 (52)

Finally the value of M_m for the travel from u' to u_m is given by the formulas, deduced from (3),

$$M_m = \frac{\alpha}{\lambda} N_m V_{12} = 4 N_m V_{12}$$
 . . . (53)

As an example of this method we will take the mean crushergauge pressure and muzzle velocity of five shots fired March 14, 1905, with the 6-inch Brown wire gun, by the Board of Ordnance at Sandy Hook. The gun had been previously fired twenty-six times with charges varying in weight from 32 lbs. to 69 lbs., and at this time was very little eroded. The gun data are as follows:

 $V_c = 3120$ c. i. d = 6 inches $u_m = 252.5$ inches (total travel in bore).

The firing charge for these five shots was 70 lbs. of nitrocellulose powder, with 8 ounces of black rifle powder at each end of the cartridge for a primer. As it is impossible to isolate the action of each kind of powder, we will consider the charge in its entirety and take $\tilde{\omega}=71$ lbs.* The projectiles varied slightly in weight from 100 lbs. (about one-quarter of one per cent.); but no material error will result if we make w=100 lbs. The mean muzzle velocity (v_m) was 3330.4 f. s., and the mean crusher-gauge pressure (p_m) was 42497 lbs. per in.² The charges were made up of m.p. grains designed for an 8-inch rifle, and of the following dimensions: R=0''.256; r=0''.0255; m=1''.029. And, therefore, $l_0=0''.044875$; $\alpha=0.72667$; $\lambda=0.19590$; $\mu=0.02378$; k'=0.85174.

The granulation of this powder is 89 grains to the pound. The volume of a single grain computed by (15), Chapter III, is 0.197144 c. i.; whence by (23'), Chapter III, $\delta = 1.5776$. From these data are found by methods already fully illustrated,

$$\Delta = 0.6299$$
 $\log a = 9.97940 - 10$
 $\log z_0 = 1.82144$ $\therefore z_0 = 66.289$ in.
 $x_m = 3.8091$
 $\log X_{om} = 0.74029$

^{*} Gossot recommends to increase the weight of charge by one-third that of the igniter. But there is no practical difference in the results by the two methods.

We will assume $k_m = 0.973$. Therefore from Table II, $\log K_m = 9.92204 - 10$, and $\log K' = 9.78885 - 10$. From (48) we have

$$N_m = \frac{K_m}{2 X_{om}} = \frac{K'}{2 X'_o} (54)$$

whence

$$X'_{o} = \frac{K' X_{om}}{K_{m}} (55)$$

$$\log X'_{o} = 0.60710$$

We now find from the preceding formulas,

$$\log V_1^2 = 7.44669$$

 $\log M = 6.70093$
 $\log M' = 4.69894$
 $\log N = 8.68493 - 10$
 $\log N' = 7.16201 - 10$

Substituting these values of M', N and N' in equation (51), Chapter IV, gives $p_m = 42521$ lbs. per in.², differing insensibly from the mean crusher-gauge pressure. The assumed value of k_m is therefore correct. We now find from (54) and (53),

$$\log N_m = 8.88072 - 10$$
$$\log M_m = 6.92947$$

The value of x' taken from Table 1, by means of $\log X'_o$, is

$$x' = 1.757.$$

Therefore $u' = 1.757 \times 66.289 = 116.47$ inches.

The two sets of equations for velocity and pressure for a charge of 70 lbs., and primer of 1 pound, are:—

From
$$u = 0$$
 to $u' = 116.47$ inches:

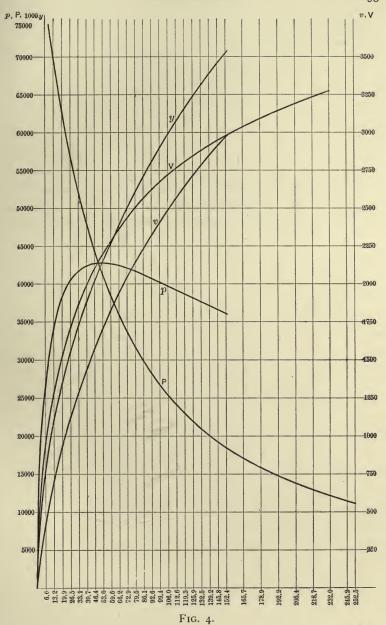
$$v^{2} = [6.70093] X_{1} \{ 1 + [8.68493 - 10] X_{o} - [7.16201 - 10] X_{o}^{2} \}$$

$$p = [4.69894] X_{3} \{ 1 + [8.68493 - 10] X_{4} - [7.16201 - 10] X_{5} \}$$

From u' = 116.47 in. to muzzle:—

$$v^2 = [6.92947] X_1 \{ 1 - [8.88072 - 10] X_o \}$$

 $p = [4.92748] X_3 \{ 1 - [8.88072 - 10] X_4 \}$



The X functions for the travel u' are

$$\log X'_{o} = 0.60710$$

$$\log X'_{1} = 0.06473$$

$$\log X'_{2} = 9.45763 - 10$$

$$\log X'_{3} = 9.79332 - 10$$

$$\log X'_{4} = 0.76498$$

$$\log X'_{5} = 1.48763$$

Both formulas for velocity give the same velocity for the travel u', namely v' = 2614 f. s. The pressure at this point by the first formula is 38431 lbs.; and by the second 20324 lbs. per in.² The discontinuity shown by the two curves P and p(see Fig. 4), at the travel u', where the grains break up into slivers is due to the sudden diminution of the surface of combustion of the grains at this point, whereby the rate of evolution of gas and heat suddenly falls and with it also the pressure. In this particular example the initial burning surface of each grain is 3.2 in.2, and goes on increasing until at the point of breaking up the vanishing surface is 4.2 in.2 It then suddenly falls to about 1.5 in.2, which is approximately the surface of the twelve slivers. Of course there is no such absolutely abrupt fall in the pressure as is indicated by the two pressure formulas. Neither can it be supposed that all the grains maintain their original form until the web thickness is completely burned. Nevertheless the two pressure formulas give very approximately the average pressure at or near this point. It might be possible to connect the two pressure curves by another curve of very steep descent; but this is hardly necessary.

The characteristics f and v_c are f = 1418 and $v_c = 0.134$. These characteristics, computed with the firing data of an 8-inch gun, were found on page 105 to be 1397 and 0.136, respectively.

The expression for y (powder burned) is by (45), Chapter IV,

$$y = [4.40457] \frac{v^2}{X_2}$$

For the travel u', this formula gives y' = 60.472 lbs. = 71 k'. At the muzzle, by the above formula, $y_m = 69.085$ lbs. = 71 k_m .

If it should be found in any case that the powder was all burned in the gun, it would be necessary to compute X_o by the formula

$$X'_{o} = K' \overline{X}_{o} \quad . \quad . \quad . \quad . \quad . \quad (56)$$

In this case we should assume a value for \overline{X}_o (or \overline{x}), and compute the maximum pressure for comparison with the crusher-gauge pressure, following the same steps as before.

We will consider a few additional problems illustrative of this method of treating m.p. grains.

Problem 1.—What must be the dimensions of the grains in the example just considered in order that the combustion of the entire charge may be completed at the muzzle? Also what would be the muzzle velocity and maximum pressure?

In solving this problem we must first consider the second period of combustion, namely, that of the slivers. It has already been shown that for a charge of 71 lbs., $\log V_1^2 = 7.44669$. We also have in this case, since $k_m = 1$,

$$N_m = \frac{1}{2 \, \overline{X}_o} \, \text{(by (48))}$$

 \overline{X}_o being the muzzle value of X_o ; and by (53),

$$M_m = 4 N_m V_1^2$$

We thus find for the second period of combustion,

$$v^2 = [7.00743] X_1 \{ 1 - [8.95868 - 10] X_o \}$$

 $p = [5.00544] X_3 \{ 1 - [8.95868 - 10] X_4 \}$

The muzzle velocity by the above equation is 3376 f. s., an increase of 46 f. s., due to the combustion of the entire charge in the gun. The muzzle pressure is 11433 lbs. per in.²

In order to deduce equations for velocity and pressure for the first period of combustion, it will be necessary to determine the value of k' from which to compute X'_o and $2 l_o$. Suppose we adopt grains for which R/r = 11 and $m/l_o = 30$.

By the method given in Chapter III, we find for grains having these ratios, $\alpha = \frac{199}{285}$, $\lambda = \frac{48}{199}$, $\mu = -\frac{4}{199}$ and $k' = \frac{81}{95}$. For this value of k' we find from Table II, $\log K' = 9.78966$; and since $K_m = 1$, we have from (55),

$$X'_{o} = K' X_{om}$$

which gives

$$\log X'_{o} = 0.52995.$$

By interpolation from Table 1, we find x' = 1.15217, and then $\log X'_1 = 9.88302 - 10$, $\log X'_3 = 9.83966 - 10$, $\log X'_4 = 0.67980$ and $\log X'_5 = 1.32095$. Next by equations (52), and equation (61), Chapter IV, we deduce the following equations for velocity and pressure, which apply from u = 0 to $u' = 1.1522 \times 66.289 = 76.38$ inches:

$$\begin{array}{l} v^2 = \left[6.76075\right] X_1 \left\{ 1 + \left[8.85245 - 10\right] X_o - \left[7.24333 - 10\right] X_o^2 \right\} \\ p = \left[4.75876\right] X_3 \left\{ 1 + \left[8.85245 - 10\right] X_4 - \left[7.24333 - 10\right] X_5 \right\} \end{array} B$$

Both sets of equations, A and B, give the same value to v', namely, 2319 f. s., while the pressures at u' by the two equations are, respectively, 51723 and 39560 lbs. per in.²,—a drop of more than 12000 lbs. when the grains break up into slivers. The maximum pressure (taking x = 0.8) is 52428 lbs. per in.²

The dimensions of the grains have yet to be determined. We have found for this powder $v_c = 0.134$ in. per sec. Substituting this and the value of $\log X'_o$, given above, in (46') gives,

$$l_o = 0.038$$
 in.

and then

$$r = l_o/2 = 0.019$$
 in.
 $R = 11 r = 0.209$ in.
 $m = 30 l_o = 1.14$ in.

A grain of these dimensions fulfils all the conditions of the problem. These calculations show in a striking manner the great effect which minute variations (scarcely measurable) in the dimensions of m.p. grains have upon the maximum pressure, increasing it in this case by 10,000 lbs. per in.²

The cause of this great increase in the maximum pressure is that the initial surface of combustion of the charge of the smaller grains is about 15 per cent. greater than that of the original grains, as is easily shown by equation (26'), Chapter III.

Problem 2.—What must be the dimensions of the grains of a charge of 71 lbs., in order that the burning of the web may be completed at the muzzle? Also determine the circumstances of motion.

To solve this problem we obviously have $u'=u_m$, and therefore $x'=x_m=3.8091$. As all the X functions relate to the muzzle only, we may drop the accents. We have from Table 1, $\log X_o = 0.74029$, $\log X_1 = 0.35048$, $\log X_2 = 9.61018$, $\log X_3 = 9.65311$, $\log X_4 = 0.91582$, and $\log X_5 = 1.78077$. Substituting the value of $\log X_o$ in (46), and making use of the known value of v_c , we find that for the new grains,

$$l_o = o''.06098$$

Therefore, as in Problem I, $r = \frac{l_o}{2} = o''.03049$
 $R = 5.5 \ l_o = o''.33539$
 $m = 30 \ l_o = 1''.8294$
 $k' = 0.85263 = k_m$

Since the limiting velocity V_1 is independent of the dimensions of the grains, we have as before, $\log V_1^2 = 7.44669$; and this, with the known values of α , λ and μ , substituted in equations (52), gives M, N and N'. We thus derive the following equations for velocity and pressure for a charge of 71 lbs. of these particular grains.

$$v^2 = [6.55041] X_1 \{ 1 + [8.64211 - 10] X_o - [6.82262 - 10] X_o^2 \}$$

 $p = [4.54842] X_3 \{ 1 + [8.64211 - 10] X_4 - [6.82262 - 10] X_5 \}$

From these formulas we get the following information:

Muzzle velocity, 3118 f. s.

Maximum pressure, 29897 lbs. per in.2

Muzzle pressure, 21014 lbs. per in.2

Powder burned in gun, 60.5 lbs. = 71 k'.

The maximum pressure is quite moderate, owing to the thickness of web which gives an initial surface of combustion but 71 per cent. of that of the original grains. The pressure is well sustained to the muzzle, where it would be considered excessive for all except wire-wound guns.

If we suppose the length of the grains to be twelve times the web thickness we should have $\alpha = \frac{163}{228}$, $\lambda = \frac{36}{163}$, $\mu = -\frac{4}{163}$,

$$k' = \frac{65}{76}$$
, and $m = 0.916$ in. Then, as before,

$$\log M = 6.56065$$
$$\log M' = 4.55866$$

$$\log N = 8.60383 - 10$$

 $\log N' = 6.90929 - 10$

These constants give—

Muzzle velocity, 3124 f. s.

Maximum pressure, 30163 lbs. per in.2

Muzzle pressure, 20875 lbs. per in.2

Powder burned in gun, 60.72 lbs. = 71 k'

The initial surface of combustion of the shorter grain is about 2.4 per cent. greater than that of the longer grain, which fact is shown in the maximum pressures.

Problem 3.—Suppose the powder we have been considering to be moulded into cylinders with an axial perforation.

If the length of the grain is 50 inches (approximately the length of the cartridge), and the diameter of the axial perforation

one-twentieth of an inch, what must be the diameter of the grain and thickness of web in order that a charge of 71 lbs. may all be burned just as the shot leaves the muzzle? Also determine the equations for velocity and pressure.

We have already found the thickness of web satisfying the conditions of the problem to be o".12196. (See Problem 1.) Therefore, by means of the formulas pertaining to this form of grain given in Chapter III, we find the diameter of the grains to be o".294 and

$$\alpha = 1.0024392$$
 $\lambda = 0.0024333$
 $\mu = 0$

Since
$$\log \overline{X}_o = 0.74029$$
 and $\log V_1^2 = 7.44669$, we find $v^2 = [6.70746] X_1 \{ 1 - [6.64590 - 10] X_o \}$
 $p = [4.70547] X_3 \{ 1 - [6.64590 - 10] X_4 \}$

which are the equations required. The muzzle velocity and maximum pressure by these formulas are

$$v_m = 3376 \text{ f. s.}$$

 $p_m = 37040 \text{ lbs. per in.}^2$

This latter, on account of the smallness of N, occurs when x = 0.64. The muzzle pressure is 22750 lbs. per in.²

A comparison of these results with those deduced in Problem 1 shows the great superiority of the uniperforated grain over the multiperforated grain so far as maximum pressure is concerned. The muzzle velocity is the same in both cases since the same weight of powder was burned in the gun. But the maximum pressure given by the m.p. grains is more than 15,000 lbs. greater, and the muzzle pressure 11,000 lbs. less than with the u.p. grains. For these latter grains the pressure is remarkably well sustained from start to finish.

The monomial formulas for velocity and pressure for this example are easily found to be

$$v = [3.35320] \sqrt{X_1}$$

and

$$p = [4.70441] X_3$$

The first of these gives the same value for the muzzle velocity as the complete formula; while the second gives maximum and muzzle pressures differing about o.i per cent. of their former values.

During the test-firing of the 6-inch Brown wire-wound gun at Sandy Hook, shots were fired with charges varying from 32½ lbs. to 75 lbs., thus enabling us to determine whether our formulas have any predictive value. Unfortunately the object of the firing was simply to test the endurance of the gun and no special effort was made to give to the results any scientific value. Many of the recorded velocities and pressures are inconsistent with each other as when, more than once, an increase of charge gave a diminished velocity and pressure. Some of the recorded muzzle velocities are so manifestly wrong that they cannot be used in getting averages. They suggest that the chronograph velocities were not always reduced to the muzzle.

We will compute the new values of f due to a change in the weight of charge by (88), Chapter IV, taking $\tilde{\omega}_o = 71$ lbs., and $f_o = 1418$ lbs. per in.², and for a six-inch gun, $n = \frac{1}{3}$. We therefore have

$$f = [2.53451] \, \tilde{\omega}^{\frac{1}{3}}$$

To determine X'_{o} , we have

$$X'_{o} = [1.43994] \frac{d^{2} l_{o}}{v_{c} \sqrt{a w \, \tilde{\omega}}}$$

which, by substituting the known values of d, l_o , v_c and w, reduces to

$$X'_{o} = \frac{\left[1.52243\right]}{\sqrt{a\,\tilde{\omega}}}$$

 N_m is given by (54), which easily reduces to (since $\log K' = 9.78885 - 10$)

$$N_m = [7.96539 - 10] \sqrt{a \, \tilde{\omega}}$$
 . . . (a)

Next we have from equation (58), Chapter IV, substituting for f its value given above and for w its value, 100 lbs.,

$$V_{1^2} = [4.97834] \, \tilde{\omega}^{\frac{4}{3}}$$

and lastly from (53),

$$M_m = 4 N_m V_1^2 = [3.54579] a^{\frac{1}{2}} \tilde{\omega}^{\frac{1}{6}}$$
 . . . (b)

The following table computed by these formulas shows the agreement between the observed and computed velocities for a range of charges between 75 lbs. and 33½ lbs. The differences in the last column follow no apparent law and are unimportant.

ω̃ lbs.	x _m	$\log M_m$	$\log N_m$	Observed Velocity	Computed Velocity	OC.
75.0	3.9574	6.95290	8.87242-10	3455	3477	$ \begin{array}{rrr} -22 \\ -37 \\ -21 \\ -5 \\ 0 \\ -3 \\ 16 \\ -9 \\ -52 \\ 17 \end{array} $
74.5	3.9383	6.95008	8.87347	3422	3459	
73.5	3.9005	6.94436	8.87557	3402	3423	
72.5	3.8635	6.93849	8.87764	3380	3385	
71.0	3.8091	6.92947	8.88072	3330	3330	
69.0	3.7392	6.91693	8.88474	3254	3257	
68.0	3.7052	6.91047	8.88672	3236	3220	
59.0	3.4244	6.84548	8.90384	2879	2888	
49.625	3.1742	6.76170	8.92037	2484	2536	
33.25	2.8146	6.55588	8.94643	1913	1896	

The two sets of equations for velocity and pressure for the charge of 75 lbs. are:

From
$$u = 0$$
 to $u' = 117.43$ inches:—

$$v^2 = [6.72436] X_1 \{ 1 + [8.67663 - 10] X_o - [7.14541 - 10] X_o^2 \}$$

 $p = [4.73896] X_3 \{ 1 + [8.67663 - 10] X_4 - [7.14541 - 10] X_5 \}$

From u' = 117.43 in. to muzzle:

$$v^{2} = [6.95290] X_{1}] \{ 1 - [8.87242 - 10] X_{o} \}$$

$$p = [4.96750] X_{3} \{ 1 - [8.87242 - 10] X_{4} \}$$

By the first equation for pressure we find $p_m = 46509$ lbs. per in.² And by the second, muzzle pressure = 15375 lbs. per in.² Both expressions for velocity give v' = 2744 f. s. For a charge of 62 lbs., the two sets of equations are

From
$$u = 0$$
 to $u' = 114.54$ inches:—

$$v^2 = [6.63999] X_1 \{ 1 + [8.70249 - 10] X_o - [7.19713 - 10] X_o^2 \}$$

 $p = [4.60288] X_3 \{ 1 + [8.70249 - 10] X_4 - [7.19713 - 10] X_5 \}$

From u' to muzzle:

$$v^2 = [6.86853] X_1 \{ \mathbf{I} - [8.89828 - 10] X_o \}$$

 $p = [4.83142] X_3 \{ \mathbf{I} - [8.89828 - 10] X_4 \}$

These formulas give a muzzle velocity of 3,000 f. s., with a maximum pressure of 34,263 lbs., and a muzzle pressure of 11,784 lbs. per in.² It would seem as if these last results are all that could be desired for a 6-inch gun.

APPLICATION TO THE FOURTEEN-INCH RIFLE

The 14-inch rifle was designed by the Ordnance Department to give a "muzzle velocity of 2,150 f. s. to a projectile weighing 1,660 lbs., with a charge of nitrocellulose powder of about 312 lbs., and with a maximum pressure not to exceed 38,000 lbs. per square inch." The gun has a powder-chamber capacity of 13,526 cubic inches and a travel of projectile in the bore of 413.85 inches. The type gun has been fired to date 55 times with charges varying from 102½ to 328 lbs., producing muzzle velocities ranging from 901 to 2,252 f. s., and crusher-gauge pressures from 4,875 to 46,078 lbs. per in.², this latter with a charge of 326 lbs.

The powder employed was "International Smokeless powder, lot 1, 1906, for 12-inch gun." The grains were cylindrical multiperforated (7 perforations), of the following dimensions:

Outside diameter, 0.826 in.
Diameter of perforations, 0.0815 in.
Length, 1.883 in.
Thickness of web, 0.145375 in.
These dimensions give:

 $\alpha = 0.71584$ $\lambda = 0.20974$ $\mu = 0.02151$ k' = 0.85058 $\log K' = 9.78778-10$.

The granulation of the powder is 20.6 grains to the pound, which by (24'), Chapter III, makes the density (δ) 1.4291.

We will base our calculations on round No. 55, fired January 23, 1911, with a charge of 328 lbs. of nitrocellulose powder plus an "igniter" of 9 lbs. of rifle, or saluting, powder. This round affords the following data:

 $\tilde{\omega} = 337 \text{ lbs.}$ w = 1664 lbs. $v_m = 2252 \text{ f.s.}$ $p_m = 43640 \text{ lbs. per in.}^2$

The preliminary calculations give

 $\Delta = 0.68965$ $\log a = 9.87523$ -10 $\log z_o = 1.65768$ $x_m = 9.1025$ $\log X_{om} = 0.87855$ $\log X_{1m} = 0.60885$

By a few trials it will be found that the observed values of v_m and p_m are satisfied when $k_m = 0.953$ and therefore from Table II, $\log K_m = 9.89388$ -10. We also find $\log X'_o = 0.77245$, $\log V_1^2 = 6.99575$, x' = 4.6354 and u' = 210.75 inches.

The equations of the velocity and pressure curves are found to be

From u = 0 to u' = 210.75 in.:

$$v^2 = [6.07812] X_1 \{ 1 + [8.54923-10] X_o - [6.78775-10] X_o^2 \}$$

 $p = [4.72511] X_3 \{ 1 + [8.54923-10] X_4 - [6.78775-10] X_5 \}$

From u' = 210.75 in. to muzzle:

$$v^2 = [6.31211] X_1 \{ 1 - [8.71430-10] X_o \}$$

 $p = [4.95910] X_3 \{ 1 - [8.71430-10] X_4 \}$

Both of the velocity formulas give v' = 1921 f. s. The first formula for pressure gives p' = 27457 and the second 19,772 lbs. per in.² The muzzle pressure comes out 9,485 lbs. per in.²

This round makes the powder characteristics, by (64) and (67), Chapter IV,

$$f = 1759.7$$

 $v_c = 0.10214$

For computing the velocity and pressure constants when the charge varies, we will consider v_c constant and assume f to vary directly as the weight of charge. That is, we will compute f by the formula

$$f = \frac{1759.7\tilde{\omega}}{337} = [0.71781] \tilde{\omega} . . . (a)$$

Equation (69), Chapter IV, becomes, by substituting the values of d^2 , l_o and v_c ,

Also (58), Chapter IV, becomes, by employing the expression for f given above,

$$V_{1^{2}} = \frac{\left[5.16164\right]\tilde{\omega}^{2}}{\varpi} (c)$$

We then have

$$M = \frac{\alpha V_1^2}{X_o'} = [1.43200] \tilde{\omega} \left(\frac{a\tilde{\omega}}{w}\right)^{\frac{1}{2}} (d)$$

$$N = \frac{\lambda}{X'_{a}} = [5.73722 - 10] \sqrt{aw\bar{\omega}} (e)$$

and

$$N' = \frac{\mu}{\lambda^2} N^2 = [9.68929 - 10] N^2$$
 . . . (f)

By (49), we have

$$N_m = \frac{K'}{2X'_o}$$

Combining this with the expression for N we have

$$N_m = \frac{K'N}{2\lambda} = (0.16507) N$$
 . . . (g)

Finally we have from (53)

$$M_m = 4 N_m V_1^2 = \frac{2K'M}{\alpha} = (0.23399) M$$
 . (h)

The following table gives the computed muzzle velocities and maximum pressures for certain charges, computed by these formulas, together with the observed velocities and crushergauge pressures for comparison:

û lbs.	w lbs.	Observed Velocity, f. s.	Computed Velocity, f.s.	OC. f. s.	Observed Pressure, lbs. per in. ²	Computed Pressure, lbs. per in. ²	OC.
337 335 334 284 263 239	1664 1660 1660 1662 ½ 1660	2252 2238 2232 1857 1738 1567	2252 2240 2232 1871 1724 1556	0 - 2 0 - 14 14 11	43640 42811 42877 25530 21190 16795	43628 42944 42637 29142 24431 19704	12 -133 240 -3612 -3241 -2909

The greatest difference between the observed and computed muzzle velocities is considerably less than one per cent. and may be disregarded. The same is true of the differences of the observed and computed maximum pressures of the first three charges. Then, as the charges are greatly reduced, these differences are largely increased. This may be accounted for if the same kind of copper cylinders were employed for all the charges.

For a charge of 314 lbs. of service powder and an igniter of 9 lbs. of rifle powder, making $\tilde{\omega} = 323$ lbs., and density of loading 0.661, the equations for velocity and pressure are as follows:

From u = 0 to u' = 208.75 inches.

$$\begin{array}{l} v^2 = [6.05003] \, X_1 \, \left\{ \, \mathrm{i} \, + [8.55696 - \mathrm{io}] \, X_o \, - [6.80321 \, - \, \mathrm{io}] \, X_o^2 \, \right\} \\ p = [4.67944] \, X_3 \, \left\{ \, \mathrm{i} \, + [8.55696 - \mathrm{io}] \, X_4 \, - [6.80321 \, - \, \mathrm{io}] \, X_5 \, \right\} \end{array}$$

From u' = 208.75 inches to muzzle.

$$v^2 = [6.28402] X_1 \{ I - [8.72203-I0] X_o \}$$

 $p = [4.91343] X_3 \{ I - [8.72203-I0] X_4 \}$

These formulas give

$$f = 1686.6$$

Muzzle velocity = 2152 f. s.
Maximum pressure = 39351 lbs. per in.²

This muzzle velocity is that for which the gun was designed, but the maximum pressure is about $3\frac{1}{2}$ per cent. greater. The muzzle pressure comes out 8714 lbs. per in.²

Example.—Suppose the volume of the powder chamber to be increased (as is proposed by the Ordnance Department) to 15,000 cubic inches, by lengthening the chamber 6.65 inches, thereby reducing the travel of the projectile to 407.2 inches. If the density of loading remain 0.661, what would be the charge, the muzzle velocity, and maximum pressure?

Answers:

$$\tilde{\omega} = 349.2 + 9 = 358.2 \text{ lbs.}$$

 $p_m = 41683 \text{ lbs. per in.}^2$
 $M. V. = 2233.5 \text{ f. s.}$

With a charge of 337 lbs. of service powder and an igniter of 9 lbs. of black powder, we should get, with the lengthened chamber, a muzzle velocity of 2150 f. s., with a maximum pressure of about 38,400 lbs. per in.² These results are practically those sought for in designing the present 14-inch gun.

Example 2.—Suppose, instead of enlarging the powder chamber of the 14-inch gun, we lengthen the grains of powder, and employ the ratios R/r = 11 and $m/l_0 = 200$. These ratios give, as is shown in Chapter III,

$$a = \frac{1219}{1900} = 0.64158$$

$$\lambda = \frac{388}{1219} = 0.31829$$

$$\mu = -\frac{4}{1219} = -0.00328$$

$$k' = \frac{1603}{1900} = 0.84368$$

$$\log K' = 9.78150 - 10. \text{ (By Table II.)}$$

Employing these grains, what muzzle velocity and maximum pressure may be expected with a charge of 314 lbs. of service powder and an igniter of 9 lbs. of black powder, in the gun as it is now, where $V_c = 13526$ c. i., and $u_m = 413.85$ in.?

The preliminary calculations give:

$$\Delta = 0.661$$
 $\log a = 9.91017$
 $\log z_0 = 1.67419$
 $x_m = 8.7630$
 $\log X_{om} = 0.87273$
 $\log X_{1m} = 0.59873$

By equations (a) to (h), inclusive, we find, the web thickness remaining as before,

$$f = 1686.6$$

 $\log V_1^2 = 6.95993$
 $\log X'_o = 0.76472$ $\therefore x' = 4.4201$ and $u' = 208.75$ in.

The equations for velocity and pressure are

From
$$u = 0$$
 to $u' = 208.75$ in.:
 $v^2 = 6.00247 X_1 \{ 1 + [8.73812 - 10] X_o - [5.98664 - 10] X_o^2 \}$
 $p = 4.63188 X_3 \{ 1 + [8.73812 - 10] X_4 - [5.98664 - 10] X_5 \}$
From $u' = 208.75$ in. to muzzle:
 $v^2 = [6.27775] X_1 \{ 1 - [8.71576 - 10] X_o \}$
 $p = [4.90716] X_3 \{ 1 - [8.71576 - 10] X_4 \}$

From these equations we find,

Maximum pressure = 37851 lbs. per in.²
Muzzle velocity = 2146 f. s.
Muzzle pressure = 8789 lbs. per in.²

$$v' = 1820.3$$
 f. s.
 $p' = \frac{26299}{18266}$ lbs. per in.²

The dimensions of these grains are found from the ratios given above, and are as follows:

Diameter of perforations =
$$l_o = 0.0727$$
 in.
Diameter of grain = 11 $l_o = 0.8$ in.
Length of grain = 200 $l_o = 14.54$ in.

The following table gives the pressures (p') at different points of the bore for a charge of 314 lbs. of service powder plus an igniter of 9 lbs., making $\tilde{\omega} = 123$ lbs., and also the pressures (p'') of the same charge made up of the grains whose dimensions are given above.

It will be seen from this table, and the previous calculations, that increasing the length of the powder grains relieves the maximum pressure more than is accomplished by lengthening the powder chamber, for the same muzzle energy:

x	u Inches.	þ' lbs. per in.²	⊅" lbs. per in.²	p'-p''	Remarks.
0. I	4·72	24325	22388	1937	
0.2	9·45	31329	29138	2191	
0.3	14·17	35084	32888	2196	
0.4	18.89	37234	35131	2103	
0.5	23.61	38453	36483	1970	
0.6	28.34	39090	37275	1815	
0.7	33.06	39351	37691	1660.	Maximum pressure.
0.8	37.78	39351	37851	1500	
0.9	42.50	39181	37833	1348	
I.O .	47.23	38892	37690	1202	
I.5	70.84	36650	36065	585	
2.0	94.45	34150	34019	131	
2.5 3.0	118.07 141.68	31849 29820 28049	32060 30293	- 211 - 473	
3·5 4·0 4·4201	165.29 188.91 208.75	26500 25344	28726 2734I 26299	- 677 - 841 - 955	The web thickness is
5.0	236.13	16308	16295	+ 13	burned at this point.
6.0	283.36	13566	13585	- 19	
7.0	330.59	11451	11496	- 45	
8.0 8.763	377.82	9775 8714	9838 8789	- 63 - 75	Muzzle.

CHAPTER VI

ON THE RIFLING OF CANNON

Advantages of Rifling.—The greater efficiency of oblong over spherical projectiles is twofold. In the first place they have greater ballistic efficiency,—that is, for the same caliber, muzzle velocity and range, an oblong projectile has a higher average velocity during its flight than a spherical projectile. This gives to the former a flatter trajectory which increases the probability of hitting the target. Experimental firing has demonstrated that the mean deviation of the shots from a rifled gun at medium ranges, when all known and controllable causes of deviation have been eliminated, is only one-third that from a smooth bore. This advantage results both from the greater sectional density of the oblong projectile whereby it is enabled the better to overcome the resistance of the air, and also because this resistance is diminished by the more pointed head.

In the second place the penetration of oblong projectiles, other things being equal, is much greater than can be realized with spherical shot, while the bursting charge of oblong shells is as great or even greater than that of spherical shells on account of their greater length. These are very substantial advantages; but to secure them it is essential that the oblong projectile should keep point foremost in its flight, otherwise it would have neither range, accuracy nor penetration, but would waste its energy beating the air.

The only way to secure steadiness of flight to an oblong projectile is to keep its geometrical axis in the tangent to the trajectory it describes by giving it a high rotary velocity about this axis. This is accomplished by rifling, as it is called,—that is, by cutting spiral grooves in the surface of the bore into which a projecting copper band, securely encircling the projectile near its base, is forced as soon as motion of translation begins, thus giving to the projectile a rotary motion in addition to its translation as it moves down the bore. The rifling may be such that the grooves (or rifles) have a constant pitch, that is, make a constant angle with the axis of the bore; or, this angle may increase. In the first case the gun is said to be rifled with a constant twist, and in the second case with an increasing twist. In all cases the twist at any point of the bore is measured by the linear distance the projectile would advance while making one revolution supposing the twist at that point to remain constant. This linear distance is always expressed in calibers, and is therefore independent of the unit of length employed.

The Developed Groove. Uniform Twist.—The element of a groove of uniform twist developed upon a plane is evidently a right line A C making, with the longitudinal element of the surface of the bore A B, the constant angle B A C, whose tangent is B C/A B.* Suppose A B to be the longitudinal element passing through the beginning of the groove at A, which is near the base of the projectile when in its firing seat and directly in front of the rotating band. Make A B = n d, n being the number of calibers the projectile travels while making one revolution. Then B C will be equal to the circumference of the projectile; or, B C = π d. If we designate the angle of inclination of the groove, B A C by β , we shall have

$$\tan \beta = \frac{BC}{AB} = \frac{\pi d}{nd} = \frac{\pi}{n} \quad . \quad . \quad (1)$$

Increasing Twist.—With a uniform twist the maximum pressure produced on the lands (or sides of the grooves) occurs (as will be shown presently) at the point of maximum pressure

^{*} The simple diagrams required in this Chapter can easily be constructed by the reader.

on the base of the projectile, which point, as we know, is near the beginning of motion. From this point the pressure on the lands decreases to the muzzle where it is not generally more than one-fourth of its maximum value. It is considered by gun-designers a desideratum to have the pressure on the lands as uniform as possible, and to this end recourse is had to an increasing twist—that is, the angle which a groove makes with the axis of the bore, instead of being constant as with a uniform twist, increases from the beginning of rifling toward the muzzle. If this variable angle be represented by θ , we shall still have, as before,

$$\tan \theta = \frac{\pi}{n} \quad . \quad . \quad . \quad . \quad (1)$$

in which n is now a variable, decreasing in value as the distance from the origin of rifling increases. If, as before, we take the origin of rectangular coördinates at A, the beginning of rifling, and suppose A B to be a longitudinal element of the bore and B C the length of arc revolved through by a point on the surface of the projectile while it travels from A to B, then the developed groove, A C, is a curve convex toward A B, the axis of abscissas, for the reason that by definition $\tan \theta$ increases from A toward B. The two forms of increasing twist that have been most generally adopted are the parabolic and circular.

General Expression for Pressure on the Lands.—Before attempting to decide upon the best system of rifling it will be necessary to deduce an expression for the pressure upon the lands. Take a cross-section of the bore and suppose for simplicity that there are only two grooves opposite to each other, and let the prolongation of the bearing surface pass through the axis of the bore, as is practically realized in the latest systems. Let M represent the point of application of the bearing surface of the upper groove. We will take three coördinate axes: one axis (x) is coincident with the axis of the bore, while the others (y) and (x)0 are in the plane of the cross-section of the bore and

perpendicular to each other. Let the axis y be directed along OM, and z in a direction perpendicular to OM.

The pressure at M, whatever may be its direction, can be replaced by the three following mutually perpendicular components: The first, perpendicular to the axis of the bore and consequently acting along the radius through M; the second lying in the plane tangent to the surface of the bore (normal to the radius O(M)) and acting along the normal to the groove; the third lying in this same plane and tangent to the groove.

Axis of x. Axis of z.
Force
$$R$$
 . $-R \sin \theta$ $R \cos \theta$
Force fR . $-fR \cos \theta$ $-fR \sin \theta$

The positive direction of the axis of x is toward the muzzle; that of z in the direction of the force R, and θ is the angle which the groove makes with the axis of x. The full component for the upper groove is:

On axis of
$$x$$
 . . . $-R(\sin \theta + fR\cos \theta)$
On axis of z . . $R(\cos \theta - fR\sin \theta)$

For the lower groove the component along the axis of x has the same value and sign as the upper one; while the component along the axis of z has the same value but the opposite sign. Besides these forces the projectile is also subjected to the variable

pressure of the powder gases on its base acting along the axis of x in the positive direction, which force call P. If we replace the grooves by the forces enumerated above, we may consider the projectile a free body and apply to it Euler's equations. These equations are six in number; but, as is readily seen, they reduce to two in the problem under consideration, namely: an equation of translation along the axis of x, and of rotation about the same axis, or, what is the same thing, the axis of the projectile. The first equation is

$$M \frac{d^2 x}{d t^2} = P - 2 R (\sin \theta + f \cos \theta)$$
 . . . (2)

and the second

$$M k^2 \frac{d \omega}{d t} = 2 r R (\cos \theta - f \sin \theta) . . . (3)$$

in which r is the radius of the projectile, ω its angular velocity about its axis and k its radius of gyration.

The angular velocity ω of a projectile about its geometrical axis for an increasing twist, continually increases as it moves along the bore from zero to its muzzle value, which is $\pi v/n r$, v being the muzzle velocity of translation and r the radius of the projectile. Its magnitude at any instant is given by the equation

$$\omega = \frac{d\varphi}{dt}$$

where φ is the angle turned through from the beginning of motion expressed in radians. The angular acceleration is found by differentiating this equation with respect to the time, which gives

Angular acceleration =
$$\frac{d \omega}{d t} = \frac{d^2 \varphi}{d t^2}$$
 . . . (4)

If we now take x and y as the rectangular coördinates of the developed groove with the origin at the beginning of rifling and the axis of abscissas parallel to the axis of the bore, then x will

represent at any instant the distance travelled by the shot, and the corresponding value of y will be $r\varphi$, where r is the radius of the projectile. Substituting y for φ in (4), gives

Substituting this value of $d \omega/d t$ in (3), it becomes, putting $k^2/r^2 = \mu$,

$$M \mu \frac{d^2 y}{d t^2} = 2 R (\cos \theta - f \sin \theta) (6)$$

Before these equations can be used for determining 2R we must eliminate dt; and this we can do by means of the equation of the developed groove. Let

$$y = f(x)$$

be this equation. Then employing the usual notation we have

$$\frac{dy}{dx} = f'(x) = \tan \theta \quad . \quad . \quad . \quad . \quad . \quad . \quad (7)$$

and

$$\frac{d^2 y}{d x^2} = f^{\prime\prime}(x)$$

Also, since

$$\frac{dy}{dt} = \frac{dy}{dx} \cdot \frac{dx}{dt} = f'(x) \frac{dx}{dt},$$

we have, by differentiating,

$$\frac{d^2 y}{d t^2} = f''(x) \left(\frac{d x}{d t}\right)^2 + f'(x) \frac{d^2 x}{d t^2} = v^2 f''(x) + \frac{d^2 x}{d t^2} \tan \theta$$

Substituting this value of $\frac{d^2y}{dt^2}$, and also the value of $\frac{d^2x}{dt^2}$

from (2) in (6), and solving for 2R, we have

$$2R = \frac{\mu \sec \theta \left\{ P \tan \theta + M v^2 f''(x) \right\}}{1 - \tan \theta \left\{ f - \mu \left(f + \tan \theta \right) \right\}}. \quad . \quad (8)$$

In using this equation f''(x) and $tan \theta$ are obtained from the equation of the developed groove, μ and M from the projectile,

V and P from the equations for velocity and pressure deduced in Chapter IV, while f is determined by experiment. The resulting value of 2R will be sum of the rotation pressures on all the lands.

Uniform Twist.—For uniform twist f''(x) is zero and θ becomes constant and equal to β . Its value is given by (1). Making these substitutions the expression for 2R becomes for uniform twist

$$2 R = \frac{\mu P \tan \beta \sec \beta}{1 - f \tan \beta + \mu \tan \beta (f + \tan \beta)} . \qquad (9)$$

In the second member of (9), P, the pressure on the base of the projectile, is the only variable; and therefore 2R is directly proportional to this pressure, and is a maximum when P is a maximum. In equation (8) there are four variables in the second member, namely, P, v, θ and f''(x); and it is not obvious on simple inspection where the point of maximum rotation is located. It will be shown, however, by examples that for an increasing twist this point is much nearer the muzzle than when the twist is uniform.

Increasing Twist. Semi-Cubical Parabola.—To continue an increasing twist quite up to the muzzle must conduce to inaccuracy of flight, and especially so when the projectile has partially left the bore so that it has lost its centering. To remedy this the acceleration of rotation near the muzzle is made either zero or constant (preferably the former), in order to relieve the rotating band as much as possible from pressure and to reduce the torsional effect upon the gun, far removed from its support at the trunnions. In all of our sea-coast guns the final twist is made constant beginning at about $2\frac{1}{2}$ calibers from the muzzle. The developed groove for the increasing twist is a semi-cubical parabola whose general equation is

$$y + b = p(x + a)^{\frac{3}{2}}$$
 (10)

The axis of x is parallel to the axis of the bore and the origin is at the beginning of the rifling, just in front of the rotating band of the projectile when in its firing seat. The coördinates of the vertex of the parabola are -a and -b, and these with the parameter p are determined by the particular twist adopted for the beginning and ending of the increasing twist. Suppose the rifling to start with a twist of one turn in n_1 calibers, and that at $2\frac{1}{2}$ calibers from the muzzle where the rifling begins to be constant it has a twist of one turn in n_2 calibers $(n_1 > n_2)$. For these two points we have, by (1),

$$\tan \theta_1 = \frac{\pi}{n_1}$$
 and $\tan \theta_2 = \frac{\pi}{n_2}$

 θ_1 and θ_2 being the inclinations of the grooves with the axis of x at the points considered. Differentiating (10), we have,

$$\frac{dy}{dx} = \tan\theta = \frac{3}{2} p(x+a)^{\frac{1}{2}} \qquad . \qquad . \qquad . \qquad (11)$$

At the origin x = 0, which gives

$$\tan \theta_1 = \frac{3 \not D \sqrt{a}}{2} = \frac{\pi}{n_1}$$

At $2\frac{1}{2}$ calibers from the muzzle where the increasing twist ends, $x = u_2$, and we have at this point

$$\tan \theta_2 = \frac{3 p \sqrt{u_2 + a}}{2} = \frac{\pi}{n_2}$$

From these two last equations we find

and

$$p = \frac{2\pi}{3 n_1 \sqrt{a}} = \frac{2\pi}{3 n_2 \sqrt{u_2 + a}} (13)$$

Since at the origin x and y are both zero, we find from (10) and (13),

Thus all the constants in the equation of the developed groove are given in terms of n_1 , n_2 and n_2 . Lastly, differentiating (II) gives

$$f''(x) = \frac{3p}{4\sqrt{x+a}} (15)$$

If the vertex of the semi-cubical parabola is at the origin, or beginning of rifling, a and b are zero, and (10) becomes

$$y = p x^3$$
 (16)

In this case the twist is zero at the origin and increases to one turn in n_2 calibers near the muzzle. The values of $\tan \theta$, p and f''(x) for this particular form of rifling are deduced from (11), (13), and (15), by making a zero. This form of groove is that adopted by the navy for all their heavy guns of recent construction.

Common parabola.—The equation of the common parabola is

$$y + b = p(x + a)^2$$
 . . . (17)

where -a and -b are the coördinates of the vertex. The constants are determined as for the semi-cubical parabola, and are as follows:

$$a = \frac{u_2}{\left(\frac{n_1}{n_2}\right) - 1} \qquad . \qquad . \qquad . \qquad (18)$$

$$b = \frac{\pi a}{2 n_1}$$
. (19)

$$p = \frac{\pi}{2 a n_1} = \frac{\pi}{2 n_2(u_2 + a)}$$
 . . . (20)

$$f''(x) = 2 p$$
 (21)

Relative Width of Grooves and Lands.—In our service siege and sea-coast guns the number, N, of grooves (or lands) is given by the equation

$$N = 6 d$$

in which d is the diameter of the bore in inches, and is a whole number for each of these guns. If w_g is the width of a groove and w_l the width of a land, we have the relation

$$w_g + w_l = \frac{\pi d}{6 d} = \frac{\pi}{6} = 0.5236$$
 inches.

The best authorities lay down the rule that the width of a groove should be at least double that of a land. In our guns the lands are made 0.15 in. wide, and the grooves are therefore 0.5236 - 0.15 = 0.3736 in. wide.

Application to the 10-inch B. L. R. Model 1888.—This gun has 60 grooves which, beginning at 20.1 inches from the bottom of the bore with a twist of one turn in 50 calibers, increase to one in 25 at 20 inches from the muzzle, and from thence continue uniform. We therefore have $n_1 = 50$ and $n_2 = 25$. The bore is 22.925 ft. long, and therefore $u_2 = 19.583$ ft. The developed groove is a semi-cubical parabola whose equation is (10). The constants are

$$a = \frac{19.583}{3} = 6.528 \text{ ft.}$$

$$b = \frac{6.528 \pi}{75} = 0.27344 \text{ ft.}$$

$$p = \frac{2 \pi}{150\sqrt{6.528}} = 0.016395.$$

The equation of the developed groove (changing x to u to indicate travel of projectile) is therefore,

$$y + 0.27344 = 0.16395 (u + 6.528)^{\frac{3}{2}}$$
 (22)

in which y will be given in feet.

From (11), we have

$$\tan \theta = 0.024592 \sqrt{u + 6.528}$$

Making u zero in this last equation gives

$$\theta_1 = 3^{\circ} 35' 42''$$

which is the inclination at which the groove starts. At 20 inches from the muzzle where the twist becomes uniform (and which is therefore a point of discontinuity on the developed groove) we have $u_2 = 19.583$; and at this point

$$\theta_2 = 7^{\circ} 9' 45''$$

This value of θ is retained to the muzzle.

From (15), we have

$$f''(x) = \frac{0.012296}{\sqrt{u + 6.528}}$$

This function decreases from the origin to the point of discontinuity. From this point to the muzzle f''(x) is zero.

If we put

$$K = \frac{\mu \sec \theta}{1 - \tan \theta \{f - \mu (f + \tan \theta)\}}. \quad . \quad (23)$$

equation (8) becomes

$$2R = K \{ P \tan \theta + M v^2 f''(x) \}$$
 . (24)

Captain (now Sir Andrew) Noble, as the result of very careful experiments made by him with 12-cm. quick-firing guns, found f = 0.2, and this value will be adopted in what follows. We also have for cored shot $\mu = 0.5$, nearly. Substituting these values of f and μ in (23), it will be found that K increases very slowly as θ increases. The values of K for u = 0 and u = 19.583 are, respectively, 0.5032 and 0.5064. We might therefore take for K the arithmetical mean of these two values and write (24)

$$2R = 0.5048 \{ P \tan \theta + M v^2 f''(x) \}$$
 . (25)

without any material error. This formula may be employed for all our sea-coast guns.

If the 10-inch gun were rifled with the kind of groove given by (16), we should find

$$p = \frac{2\pi}{75\sqrt{19.583}} = 0.018931$$

$$\tan \theta = 0.028397 \sqrt{u}$$

$$f''(x) = \frac{0.014198}{\sqrt{u}}$$

In this form of rifling the initial inclination of the grooves is zero and increases to 7° 9' 45'' at 20 inches from the muzzle, where the twist becomes uniform. Between this point and the muzzle, f''(x) is zero.

Uniform Twist.—If we suppose the 10-inch gun to be rifled throughout with a uniform twist of one turn in 25 calibers, we have $\beta = 7^{\circ}$ 9' 45". Employing the values of μ and f already given, (9) reduces to

$$2R = 0.063624 P.$$
 . . . (26)

Working Expressions.—If the equation of the developed groove is (10), we have

$$_{2}R = K \tan \theta \left\{ P + \frac{M v^{2}}{2(u+a)} \right\}.$$
 (27)

and

$$\tan \theta = \frac{\pi}{25} \left(\frac{u+a}{u_2+a} \right)^{\frac{1}{2}}$$

If (16) is the equation of the developed groove, we have

$$2 R = K \tan \theta \left\{ P + \frac{M v^2}{2 u} \right\} \qquad . \qquad . \qquad (28)$$

and

$$\tan \theta = \frac{\pi}{25} \left(\frac{u}{u_2}\right)^{\frac{1}{2}}$$

Pressure on the Lands of the 10-Inch B. L. R.—The equations for velocity and pressure for this gun are the following:

$$v^2 = [6.20536] X_1 \{ I - [8.5938I - IO] X_o \}$$
 . (29)

- and

$$p = [4.72060] X_3 \{ I - [8.5938I - IO] X_4 \}$$
 . (30)

The gun and firing data were $V_c=7064$ c. i., $u_m=22.925$ ft., $\tilde{\omega}=250$ lbs. of brown cocoa powder, w=575 lbs., muzzle velocity 1975 f. s., maximum pressure on base of projectile, 33300 lbs. per in.², $\Delta=0.98$, and $z_o=3.461$ ft. It will be convenient to change (30) so that it will give the entire pressure (P) on the base of the projectile; and to avoid large numbers we will adopt the ton as the unit of weight. We then have

$$P = \frac{\pi d^2 p}{8960}$$

and (30) becomes

$$P = [3.26544] X_3 \{ I - [8.5938I - IO] X_4 \}. \quad (30')$$

Finally, the mass of the projectile expressed in tons is

$$M = \frac{575}{2240 \, g} = [7.90210 - 10]$$

We have now all the formulas and data necessary for computing the pressures on the lands of the 10-inch B. L. R., by means of (26), (27), and (28), for the three principal systems of rifling adopted in our service. These calculations are given in the table on page 183.

The last three columns of this table show that the maximum pressure on the lands is greater for uniform twist than for either form of increasing twist; but the difference between these maxima is not very great. Moreover, the maximum pressure for uniform twist occurs at the trunnions where its torsional effect upon the gun—so far as deranging the aim is concerned—is a

Pressures on lands required to produce rotation of shot in the 10-inch B. L. R. for different systems of rifling. Charge 250 lbs. Projectile 575 lbs. Muzzle velocity 1975 f. s. Maximum pressure on base of projectile 33300 lbs. per square inch.

	Travel of	Velocity of	Pressure on Base of	Pressu	JRE ON LANDS.	Tons
x '	Shot,	Shot,	Shot,	Uniform	Increasing	Increasing
	feet	f. s.	tons	Twist	Twist Eq. (28)	Twist, Eq. (27)
0.0	0.	0.0	0.0	0.0	0.0	0.0
.I	0.3461	227.7	841.1	53.5	12.0	27.0
.2	0.6922	366.7	1036.4	65.9	21.5	36.9
0.3	1.0382	478.1	1122.4	71.4	29.0	42.3
•4	1.3843	572.4	1158.3	73.7	35.3	46.1
•5	1.7304	654.7	1167.4	74.3	40.9	48.9
0.6	2.0765	727.8	1161.1	73.9	44.8	51.1
.7	2.4226	793.6	1145.6	72.9	48.5	52.9
.8	2.7686	853.4	1124.7	71.6	51.6	54.3
0.9	3.1147	908.2	1100.7	70.0	54.3	55.5
1.0	3.4608	958.7	1075.0	68.4	56.7	56.5
I.I	3.8069	1005.6	1048.5	66.7	58.7	57.3
1.2	4.1530	1049.3	1021.9	65.0	60.5	58.1
1.3	4.4991	1090.2	995.5	63.3	62.1	58.7
1.4	4.8452	1128.6	969.7	61.7	63.5	59.2
1.5	5.1912	1164.7	944.5	60.1	64.7	59.7
1.6	5.5372	1198.9	920.1	58.5	65.8	60.1
1.7	5.8833	1231.4	896.4	57.0	66.7	60.5
1.8	6.2294	1262.1	873.6	55.6	67.5	60.8
1.9	6.5755	1291.4	851.7	54.2	68.2	61.1
2.0	6.9216	1319.4	830.5	52.8	69.0	61.3
3.0	10.3824	1543.4	659.4	42.0	72.7	62.4
4.0	13.8432	1702.8	541.4	34.4	73.5	62.3
5.0	17.3040	1824.9	456.4	29.0	73.2	61.6
5.6586	19.5833	1891.6	412.4	26.2	72.6	61.0
6.0000	20.7648	1922.8	392.4	25.0	25.0	25.0
6.6242	22.9250	1975.0	359.9	22.9	22.9	22.9

minimum; while the position of the maximum pressure upon the lands for either form of increasing twist is well down the chase. It is difficult to see any superiority of an increasing twist over a uniform twist, especially in view of the fact demonstrated by Captain Noble's experiments, that the energy expended in giving rotation to the projectile with rifling having an increasing twist is nearly twice as great as with a uniform twist.

Application to the 14-inch Rifle.—This gun has 126 grooves and the same number of lands, in this respect differing from the rule followed with the other seacoast guns. The values of n_1 , n_2 , θ_1 and θ_2 are the same as those found for the 10-inch rifle. The rifling begins 7.05 inches from the base of the projectile when in its firing seat and becomes uniform 22.8 inches from the muzzle. Therefore

$$u_2 = 413.85 - (7.05 + 22.8) = 384$$
 inches.

From (12), (13), and (14), we now find

$$a = 128$$
 $p = 0.0037024$

$$p = 0.0037024$$

 $b = 5.36165$

Therefore the equation of the developed groove is

$$y + 5.36165 = 0.0037024 (u + 128)^{\frac{3}{2}}$$

From (11) and (15), we have, finally,

$$\tan \theta = 0.0055536\sqrt{u + 128}$$

$$f''(x) = \frac{0.0027768}{\sqrt{u + 128}}$$

$$M = \frac{1660}{2240 g} = 0.02304$$

and

$$P = \frac{\pi \, d^2 \, p}{8960} = 0.06872 \, p$$

in which P is the entire pressure on the base of the projectile in

tons while p is the pressure in pounds per square inch given by the equation on page 166, for a charge of 314 lbs.

Substituting these expressions for $\tan \theta$, f''(x), and M in (25) and reducing, we have the working expression

$$2R = [7.44769 - 10]\sqrt{u + 128} \left\{ P + [8.06145 - 10]\frac{v^2}{u + 128} \right\}$$

in which 2R is the normal pressure on all the lands in tons and u the travel of the projectile in inches. To determine the normal pressure on each land 2R must be divided by 126.

For a uniform twist 2 R is given by (26).

In the following table v and p were computed by the formulas on page 166 for a charge of 314 lbs., and P and 2 R by the formulas given above:

SEx per week	inches.	v f. s.	P tons.	2 R Increasing T	2 R Uniform T
0.I	4.72	198.7	1671.7	54.1	106.4
0.2	9.45	325.2	2153.0	71.1	137.0
0.3	14.17	429.1	2411.1	81.1	153.4
0.4	18.89	518.9	2558.8	87.7	162.8
0.5	23.61	598.5	2642.6	92.2	168.2
0.6	28.34	670.3	2686.3	95.3	171.3
0.7	33.06	735.8	2704.3	97.6	172.I
0.8	37.78	. 796.1	2704.3	99.2	172.1
0.9	42.50	852.1	2692.6	100.4	171.3
1.0	47.23	904.3	2672.8	101.2	170.1
1.5	70.84	1123.3	2518.7	102.5	160.3
2.0	94.45	1295.1	2346.9	101.8	149:3
2.5	118.07	1436.7	2188.7	100.5	139.3
3.0	141.68	1557.5	2049.3	99.1	130.4
3.5	165.29	1662.9	1927.6	97.8	122.7
4.0	188.91	1756.5	1821.1	96.5	115.9
5.0	236.13	1891.0	1120.7	66.0	71.3
6.0	283.36	1981.2	932.3	59.3	59.3
7.0	330.59	2053.7	787.0	53.6	50.1
8.0	377.82	2113.3	671.8	48.8	42.7
8.763	413.85 Muzzle	2152.0	598.8		

Influence of the Rifling for a Uniform Twist.—For a uniform twist we have

$$\omega = \frac{\pi \, v^*}{n \, r}$$

where n is constant, and r is the radius of the projectile. Differentiating with respect to t we have

$$\frac{d\omega}{dt} = \frac{\pi}{nr}\frac{dv}{dt} = \frac{\pi}{nr}\frac{d^2x}{dt^2}$$

Substituting this value of $d \omega/d t$ in (3), it becomes

$$\frac{\pi \mu M}{n} \frac{d^2 x}{d t^2} = 2 R \left(\cos \beta - f \sin \beta\right) . \qquad (31)$$

Eliminating 2 R between equations (2) and (31), gives

$$P = \frac{d^2 x}{d t^2} \left\{ M + \frac{\pi \mu M}{n} \frac{f + \tan \beta}{1 - f \tan \beta} \right\}. \qquad (32)$$

If the bore were smooth the equation of translation of the projectile would be

$$M\frac{d^2x}{dt^2}-P$$
;

from which it appears that the effect of the grooves upon the motion of the projectile for a constant twist is equivalent to increasing the mass of the projectile by the quantity

$$\frac{\pi \mu M}{n} \cdot \frac{f + \tan \beta}{1 - \tan \beta}$$

By making f = 0.2, n = 25, $\mu = 0.5$ and $\beta = 7^{\circ}$ 9' 45", the value of this supplemental term is found to be $0.021 \, M$. That is, the retarding effect of a constant twist of one turn in 25 calibers is equivalent to increasing the weight of the projectile 2 per cent.

^{*} Artillery Circular, N, p. 201.

TABLES

														PAGE
I.	X Functions						•	•	•	•		•	•	189
II.	K = 1 - (1 -	$k)^{\frac{1}{2}}$				•					•	•	•	211
III	Work of Fired	Gui	100	wder										212



TABLE I

D	69719	7718	984	878	791
	29885	6500	962	859	776
	17397	5608	940	840	762
	12287	1033	918	822	747
	9487	1008	898	806	733
log X ₅	9.30059 9.99778 0.29663 0.47060	0.68834 0.76552 0.83052 0.88660 0.89693	0.90701 0.91685 0.92647 0.93587 0.94505	0.95403 0.96281 0.97140 0.97980 0.98802	0.99608 1.00399 1.01175 1.01937 1.02684
D	34885 14973 9728 6171 4771	3886 3276 2829 522 509	497 475 475 453	444 434 425 416 408	400 392 386 378 371
log X ₄	9.66437	0.35965	0.46987	0.49360	0.51487
	0.01322	0.39851	0.47484	0.49804	0.51887
	0.16295	0.43127	0.47969	0.50238	0.52279
	0.25023	0.45956	0.48444	0.50663	0.52665
	0.31194	0.46478	0.48907	0.51079	0.53043
D	32763 12434 6305 3855 2553	1756 1229 860 136 126	117 109 101 92 84	78 71 63 52	46 41 36 31 26
log X ₃	9.23296 9.56059 9.68493 9.74798	9.81206 9.82962 9.84191 9.85051	8.95313 9.85430 9.85539 9.85640 9.85732	9.85816 9.85894 9.85965 9.86028 9.86085	9.86137 9.86183 9.86224 9.86260 9.86260
D	68769	6752	811	709	628
	28746	5568	789	693	615
	16304	4709	767	676	601
	11240	859	746	659	587
	8482	834	727	643	575
log X ₂	7.52000	8.85541	9.04263	9.08103	9.11483
	8.20769	8.92293	9.05074	9.08812	9.12111
	8.49515	9.87861	9.05863	9.09505	9.12726
	8.65819	9.02570	9.06630	9.10181	9.13327
	8.77059	9.03429	9.07376	9.10840	9.13914
D	103529 43569 24888 17273 13121	10511 8720 7420 1357 1320	1285 1252 1219 1188 1159	1131 1104 1079 1053 1030	1007 986 965 944
log X ₁	7.05911	9.08291	9.37619	9.43722	9.49119
	8.09440	9.18802	9.38904	9.44853	9.50126
	8.53009	9.27522	9.40156	9.45957	9.51112
	8.77897	9.34942	9.41375	9.47036	9.52077
	8.95170	9.36299	9.42563	9.48089	9.53021
D	34760	3759	475	422	371
	14823	3152	463	413	371
	8584	2711	452	403	364
	6033	498	440	394	356
	4639	486	431	387	350
log X ₀	9.53911	0.22750	0.33356	0.35617	0.37636
	9.88671	0.26509	0.33831	0.36039	0.38015
	0.03494	0.29661	0.34294	0.36452	0.38386
	0.12078	0.32372	0.34746	0.36855	0.38750
	0.18111	0.32870	0.35186	0.37249	0.39106
*	0.01 0.05 0.10 0.15 0.20	0.25 0.30 0.35 0.40 0.41	0.42 0.43 0.45 0.45	0.47 0.48 0.49 0.50 0.51	0.52 0.53 0.54 0.55 0.55

TABLE I-Continued

н	$\log X_0$	D	$\log X_1$	D	$\log X_2$	D	$\log X_3$	D	$\log X_4$	D	$\log X_5$	D
.57	0.39456	344	9.53946	906	9.14489	563	9.86317	22	0.53414	364	1.03417	720
0.58	0.39800	338	9.54852	888	9.15052	550	9.86339	81	0.53778	3.58	1.04137	707
.59	0.40138	331	9.55740	870	9.15602	539	9.86357	14	0.54136	352	1.04844	695
09.0	0.40469	325	9.56610	853	9.16141	528	9.86371	01	0.54488	346	1.05539	682
19.0	0.40794	320	9.57463	837	69991.6	518	9.86381	9	0.54834	340	1.06221	129
.62	0.41114	314	9.58300	821	9.17187	507	0.86287	+	0.55174	222	1.06802	9
.63	0.41428	309	9.59121	908	9.17694	407	9.86390	0	0.55500	220	1.07552	970
6.64	0.41737	304	9.59927	792	9.18191	487	9.86390	4-	0.55838	323	1.08201	640
.65	0.42041	299	61209.6	777	9.18678	477	9.86386	7	0.56161	318	1.08840	628
99.	0.42340	294	9.61496	763	9.19155	469	9.86379	6	0.56479	313	1.09468	619
.67	0.42634	290	9.62259	750	9.19624	461	9.86370	12	0.56792	300	1.10087	009
89.0	0.42924	285	60069.6	737	9.20085	453	9.86358	15	0.57101	304	1.10696	009
69.	0.43209	280	9.63746	725	9.20538	444	9.86343	18	0.57405	300	1.11296	591
.70	0.43489	276	9.64471	712	9.20982	436	9.86325	20	0.57705	295	1.11887	583
.71	0.43765	272	9.65183	102	9.21418	429	9.86305	23	. 0.58000	291	1.12470	574
.72	0.44037	268	9.65884	689	9.21847	421	9.86282	25	0.58291	287	1.13044	266
.73	0.44305	264	9.66573	829	9.22268	414	9.86257	27	0.58578	283	1.13610	557
-74	0.44569	260	9.67251	299	9.22682	407	9.86230	29	0.58861	279	1.14167	248
.75	0.44829	256	81629.6	929	9.23089	400	9.86201	31	0.59140	275	1.14715	542
92.0	0.45085	252	9.68574	949	9.23489	393	9.86170	33	0.59415	271	1.15257	535
.77	0.45337	249	9.69220	636	9.23882	387	9.86137	7,	0.50686	268	1.15702	527
.78	0.45586	246	9.69856	627	9.24269	381	9.86102	36	0.50054	264	1.16210	220
62.0	0.45832	243	9.70483	617	9.24650	375	9,86066	38	0.60218	261	1,16839	513
- 80	0.46075	239	9.71100	809	9.25025	369	9.86028	40	0.60479	257	1.17352	206
-81	0.46314	236	0.71708	200	0.25204	262	88088	2	0,000	200	11001	200

TABLE 1—Continued

D	494 488 184	476 476 470	464 453 448 443	438 433 427 423 419	414 410 406 401 397	393 385 385 378
$\log X_{\rm b}$	1.18358	1.19821	1.20767 1.21231 1.21690 1.22143 1.22591	1.23034 1.23472 1.23905 1.24332 1.24755	1.25174 1.25588 1.25998 1.26404 1.26805	1.27202 1.27595 1.27984 1.28369 1.28750
D	251 248	242	236 233 230 228 228	223 220 218 215 215	211 209 206 204 202	200 198 197 194 192
$\log X_4$	0.60990	0.61733	0.62214 0.62450 0.62683 0.62913 0.63141	0.63366 0.63589 0.63809 0.64027 0.64242	0.64455 0.64666 0.64875 0.65081 0.65285	0.65487 0.65687 0.65885 0.66082 0.66276
D	445	5 4 4 4 8 8 8 1 1 8 1	50 53 53 55	55 57 57 59 59	60 62 63 63	64 65 65 66 66
$\log X_3$	9.85946	9.85811 9.85764	9.85716 9.85666 9.85615 9.85562 9.85508	9.85453 9.85397 9.85341 9.85284 9.85226	9.85167 9.85107 9.85046 9.84984 9.84921	9.84858 9.84794 9.84729 9.84664 9.84598
D	357	341	331 326 322 317 313	308 304 300 296 292	288 284 280 276 273	270 266 263 260 257
log X2	9.25757 9.26114 9.26466	9.26812	9.27489 9.27820 9.28146 9.28468 9.28785	9.29098 9.29406 9.29710 9.30010 9.30306	9.30598 9.30886 9.31170 9.31450 9.31726	9.31999 9.32269 9.32535 9.32798 9.33058
D	590 582 573	565	550 543 535 528 521	514 507 501 494 488	482 476 471 465 459	454 448 443 438 433
log X ₁	9.72307 9.72897 9.73479	9.74652	9.75174 9.75724 9.76267 9.76802 9.77330	9.77851 9.78365 9.78872 9.79373 9.79867	9.80355 9.80837 9.81313 9.81784 9.82249	9.82708 9.83162 9.83610 9.84053 9.84491
D	233	224 221	219 216 213 211 208	206 203 201 198 196	194 193 190 188 186	184 182 181 178 176
$\log X_0$	0.46550 0.46783 0.47014	0.47241	0.47686 0.47905 0.48121 0.48334 0.48545	0.48753 0.48959 0.49162 0.49363 0.49561	0.49951 0.50144 0.50334 0.50522	0.50708 0.50892 0.51074 0.51255 0.51433
*	0.82	0.85	0.87 0.88 0.89 0.90 0.91	0.92 0.93 0.95 0.95 0.96	0.97 0.98 0.99 1.00 1.01	1.02 1.03 1.04 1.05 1.05

TABLE I-Continued

D	374	370	367	363	359	256	252	350	346	343	240	338	334	331	328	225	323	320	317	315	312	300	307	304	302
log X ₅	1.29128	1.29502	1.29872	1.30239	1.30602	130061	1.21217	1.31670	1.32020	1.32366	1 22700	1.33049	1.33387	1.33721	1.34052	1.24280	1.34705	1.35028	1.35348	1.35665	1.25080	1,36292	1.36601	1.36908	1.37212
D	190	189	187	185	183	182	180	178	176	175	173	172	171	169	168	166	165	163	162	160	150	158	157	156	154
$\log X_4$	0.66468	0.66658	0.66847	0.67034	0.67219	0 67400	0.67584	0.67764	0.67942	0.68118	0.68202	0.68466	0.68638	0.68800	0.68978	0.69146	0.69312	0.69477	0.69640	0.69802	0.69962	0.70121	0.70279	0.70436	0.70592
D	29-	89	89	69	69	70	- 70	20	71	71	7.2	72	72	73	73	7.3	74	74	74	75	7.5	75	75	75	92
$\log X_3$	9.84532	9.84465	9.84397	9.84329	9.84260	0.84101	9.84121	9.84051	9.83981	9.83910	0.82820	9.83767	9.83695	9.83623	9.83550	0.83477	9.83404	9.83330	9.83256	9.83182	9.83107	9.83032	9.82957	9.82882	9.82807
D.	253	250	247	244	241	230	236	233	230	228	225	223	220	218	215	213	211	208	206	204	202	200	861	195	193
$\log X_2$	9.33315	9.33568	9.33818	9.34065	9.34309	0.34550	9.34789	9.35025	9.35258	9.35488	9.35716	9.35941	9.36164	9.36384	9.36602	9.36817	9.37030	9.37241	9.37449	9.37655	9.37859	9.38061	9.38261	9.38459	9.38654
D	428	423	418	414	409	405	400	396	392	388	384	380	375	372	368	364	361	357	354	350	347	343	341	336	334
$\log X_1$	9.84924	9.85352	9.85775	9.86193	20998.6	9.87016	9.87421	9.87821	9.88217	60988.6	9.88997	9.89381	19268.6	9.90136	9.90508	9.90876	9.91240	10916.6	9.91958	9.92312	9.92662	9.93009	9.93352	9.93693	9.94029
D	175	173	171	170	891	991	165	163	191	1091	158	157	156	154	152	151	150	149	148	146	145	144	143	141	140
$\log X_0$	0.51609	0.51784	0.51957	0.52128	0.52298	0.52466	0.52632	0.52797	0.52960	0.53121	0.5328I	0.53439	0.53596	0.53752	0.53906	0.54058	0.54209		0.54508	0.54656	0.54802	0.54947	0.55091	0.55234	0.55375
*	1.07	1.08	60·I	1.10	1.11	1.12	1.13	I.14	1.15	91.1	1.17	81.I	61.1	1.20	1.21	1.22	1.23	1.24	1.25	1.26	1.27	1.28	6z.1	1.30	1.31

TABLE I-Continued

H	log X ₀	D	$\log X_1$	D	$\log X_2$	D	$\log X_3$	D	$\log X_4$	D	log X5	D
1.32	0.55515	130	0.04363	331	9.38847	192	0.82731	92-	0.70746	152	1.27514	300
1.33	0.55654	138	9.94694	328	9.39039	190	9.82655	9/	0.70899	152	1.37814	207
1.34	0.55792	137	9.95022	324	9.39229	188	9.82579	9/	0.71051	150	1.38111	295
1.35	0.55929	136	9.95346	322	9.39417	186	9.82503	77	0.71201	140	1.38406	203
1.36	0.56065	135	9.95668	319	9.39603	184	9.82426	22	0.71350	148	1.38699	290
1 2 7	0000	124	0.000	316	100000	182	0 83340	1	20717	1	08080 1	000
1.00	0.30200	134	106666	310	70765.6	101	9.02349	170	0.71490	14/	1.30909	007
1.30	0.50334	132	9.90303	313	9.39909	101	9.022/2	21	0.71045	140	1.39277	280
1.39	0.50400	131	9.90010	310	9.40150	6/1	9.62190	79	0.71791	145	1.39503	283
1.40	0.50597	130	9.96926	307	9.40329	177	9.82119	78	0.71936	144	1.39846	280
1.41	0.56727	129	9.97233	304	9.40506	175	9.82041	77	0.72080	143	1.40126	279
4.2	928920	202	1000	202	199010	11	190180	1	20000	-	1000	1
24.1	0.30030	120	7.57.65	300	9.40001	+/1	4.01904	1:	0.72223	241	1.40405	117
1.43	0.50904	171	9.976.39	299	9.40055	1/3	7001006		0.72305	140	1.40082	275
1.44	0.57111	127	9.98138	298	9.41028	170	9.81810	78	0.72505	139	1.40957	273
1.45	0.57238	125	9.98436	295	9.41198	169	9.81732	28	0.72644	139	1.41230	272
1.46	0.57363	124	9.98731	292	9.41367	891	9.81654	78	0.72783	138	1.41502	270
						,	,			***		
1.47	0.57487	124	9.99023	290	9.41535	991	9.81576	28	0.72921	137	1.41772	268
1.48	0.57611	123	9.99313	287	9.41701	164	9.81498	78	0.73058	136	1.42040	266
1.49	0.57734	122	00966.6	284	9.41865	163	9.81420	77	0.73194	134	1.42306	263
1.50	0.57856	121	9.99884	282	9.42028	191	9.81343	78	0.73328	134	1.42569	261
1.51	0.57977	120	9910000	280	9.42189	091	9.81265	78	0.73462	133	1.42830	260
						((((
1.52	0.58097	611	0.00446	278	9.42349	158	9.81187	78	0.73595	132	1.43090	258
1.53	0.58216	118	0.00724	. 275	9.42507	157	60118.6	78	0.73727	131	1.43348	256
1.54	0.58334	118	0.00999	273	9.42664	156	9.81031	78	0.73858	130	1.43604	254
1.55	0.58452	911	0.01272	271	9.42820	155	9.80953	28	0.73988	129	1.43858	252
1.56	0.48468	911	0.01543	260	0.42075	153	0.80875	78	0.74117	128	1.44110	25.1

TABLE I—Continued

TABLE I—Continued

x $\log X_0$ D $\log X_3$ D	1												
0.61340 98 0.07907 222 946567 1124 9.78842 77 0.77201 109 0.61348 97 0.08129 220 946691 122 9.78642 77 0.77310 109 0.61535 97 0.08349 218 946813 122 9.78647 77 0.77419 108 0.61632 96 0.08567 217 946935 121 9.78529 78 0.77534 107 0.61918 97 0.0857 21 947857 17 0.77847 107 0.62012 94 0.09244 211 947414 118 9.78259 77 0.77847 105 0.62106 93 0.09648 209 947649 116 9.78425 17 0.78657 104 0.62106 94 0.09648 209 947649 116 9.78442 77 0.78657 105 0.62106 94 0.09648 209 947649	*	$\log X_0$	D	$\log X_1$	D	$\log X_2$	D	$\log X_3$	D	$\log X_4$	D	$\log X_5$	D
0.61438 97 0.08129 220 9,46691 122 9,78762 78 0.77310 109 0.61535 97 0.08349 218 9,46813 122 9,78684 77 0.77419 108 0.61535 97 0.08784 216 9,46913 122 9,78529 78 0.77534 107 0.61828 95 0.08784 216 9,47556 129 9,78529 78 0.77741 107 0.61918 95 0.090214 213 9,47255 119 9,78451 77 0.77847 107 0.62202 9,09033 210 9,47549 116 9,78296 77 0.77847 105 0.62204 9,09038 209 9,47649 116 9,78296 77 0.78657 104 0.62205 92 0.10675 209 9,47849 116 9,77842 105 0.62267 91 0.10675 203 9,48108 113	1.82	0.61340	86	0.07907	222	9.46567	124	9.78840	-78	0.77201	100	1.50135	213
0.61535 97 0.08349 218 9.46813 122 9.78684 77 0.77419 108 0.61535 96 0.08567 217 9.46935 121 9.78697 78 0.77741 108 0.61728 95 0.09000 214 9.47176 119 9.78451 78 0.77741 106 0.62012 94 0.09214 213 9.47176 119 9.78451 77 0.77547 105 0.62106 93 0.09427 211 9.47144 118 9.78219 77 0.77847 105 0.62106 93 0.09648 209 9.47449 116 9.78142 77 0.78567 104 0.62106 93 0.09648 209 9.47649 116 9.78465 77 0.78567 104 0.62276 91 0.10470 205 9.47860 114 9.77988 77 0.78468 103 0.62366 91 0.	1.83	0.61438	97	0.08129	220	9.46691	122	9.78762	78	0.77310	100	1.50348	212
0.61632 96 0.08567 217 946935 121 9.78607 78 0.77527 107 0.61728 95 0.08784 216 9.47056 120 9.78451 78 0.77741 106 0.61918 94 0.09214 213 9477295 119 9.78473 77 0.77847 105 0.622012 94 0.09214 211 947414 118 9.78373 77 0.77847 105 0.622109 93 0.09648 209 947532 116 9.78429 77 0.77847 105 0.62294 92 0.09648 209 947649 116 9.78442 77 0.77855 105 0.62267 91 0.10675 207 947860 114 9.77948 77 0.78878 103 0.62267 91 0.10675 203 948221 112 9.77847 77 0.78878 103 0.62367 90 0.1087	1.84	0.61535	97	0.08349	218	9.46813	122	9.78684	77	0.77419	801	1.50560	210
0.61728 95 0.08784 216 9.47056 120 9.78529 78 0.77634 107 0.61928 95 0.09000 214 9.47056 119 9.78451 78 0.77741 106 0.61918 94 0.09214 213 9.47255 119 9.78373 77 0.77847 106 0.622106 93 0.09638 210 9.47541 118 9.78296 77 0.77847 105 0.622106 93 0.09648 209 9.47749 116 9.78296 77 0.78267 104 0.62236 92 0.10657 209 9.47765 115 9.78055 77 0.78265 103 0.62364 91 0.10675 203 9.48108 113 9.77865 77 0.78674 101 0.62567 91 0.10878 202 9.48221 112 9.77849 77 0.78674 101 0.62658 90	1.85	0.61632	96	0.08567	217	9.46935	121	6.78607	78	0.77527	107	1.50770	200
0.61823 95 0.09000 214 947176 119 9.78451 77 0.77741 106 0.61918 94 0.09214 213 947255 119 9.78373 77 0.77847 105 0.622106 94 0.09627 211 947414 118 9.78296 77 0.77847 105 0.622106 93 0.09638 210 9477649 116 9.78249 77 0.78265 104 0.62292 92 0.10657 209 947765 115 9.78065 77 0.78265 104 0.62367 91 0.10675 209 94794 114 9.7791 77 0.78265 103 0.62567 91 0.10675 203 948221 112 9.7756 77 0.78265 103 0.62658 90 0.11080 202 948221 112 9.7755 76 0.78674 101 0.62658 90 0.11281 <td>1.86</td> <td>0.61728</td> <td>95</td> <td>0.08784</td> <td>216</td> <td>9.47056</td> <td>120</td> <td>9.78529</td> <td>78</td> <td>0.77634</td> <td>107</td> <td>1.50979</td> <td>208</td>	1.86	0.61728	95	0.08784	216	9.47056	120	9.78529	78	0.77634	107	1.50979	208
0.61918 9.5 0.09214 214 9.47179 119 9.78373 77 0.77847 100 0.62012 9.4 0.09214 213 9.47295 119 9.78373 77 0.77847 105 0.62016 9.4 0.09427 211 9.47544 118 9.78296 77 0.77847 105 0.6219 9.3 0.09848 200 9.47649 116 9.78296 77 0.78265 105 0.62236 9.00247 200 9.47765 115 9.78065 77 0.78265 105 0.62367 9.1 0.02675 209 9.47860 114 9.77948 77 0.78471 104 0.62567 91 0.02675 203 9.48108 113 9.77833 77 0.78471 101 0.62658 90 0.10878 202 9.48221 112 9.77656 77 0.78674 101 0.62658 90 0.11281	0	00000	1	00000		71110	6	0 1	0	1	, ,	1	,
0.01918 94 0.09214 213 9.47295 119 9.78373 77 0.77847 105 0.02012 94 0.09427 211 9.47414 118 9.78296 77 0.77952 105 0.62199 93 0.09638 209 9.47649 116 9.7812 77 0.78057 104 0.62199 93 0.09648 209 9.47649 116 9.78219 77 0.78161 104 0.62282 92 0.10057 205 9.47649 114 9.78055 77 0.78161 104 0.62367 91 0.10675 203 9.48108 114 9.77918 77 0.78471 101 0.62567 91 0.10675 203 9.48108 113 9.77833 77 0.78471 101 0.62567 90 0.10878 202 9.48221 112 9.77855 77 0.78574 101 0.62658 90 0.1	1.07	0.01823	95	0.00000	214	9.47170	119	9.78451	2/	0.77741	100	1.51187	207
0.62202 94 0.09427 211 9.47414 118 9.78296 77 0.77952 105 0.62106 93 0.09648 210 9.47532 117 9.78219 77 0.78161 104 0.62199 93 0.09648 209 9.47549 116 9.78042 77 0.78265 104 0.62234 92 0.10264 206 9.47890 114 9.77988 77 0.78471 104 0.62256 91 0.10470 205 9.47894 114 9.77891 77 0.78471 101 0.62567 91 0.10470 205 9.47821 112 9.77831 77 0.78471 101 0.62567 90 0.11080 201 9.48333 111 9.77756 77 0.78471 101 0.62638 89 0.11281 199 9.48444 110 9.77602 77 0.78475 100 0.62638 89 0.	1.88	0.61918	94	0.09214	213	9.47295	611	9.78373	22	0.77847	105	1.51394	205
0.622106 93 0.09638 210 9.47532 117 9.78219 77 0.78057 104 0.62292 93 0.09848 209 9.47649 116 9.78142 77 0.78265 104 0.62384 92 0.10264 206 9.4786 115 9.78065 77 0.78265 103 0.62567 91 0.10675 203 9.47994 114 9.77988 77 0.78471 102 0.62567 90 0.10878 202 9.48108 113 9.77833 77 0.78471 101 0.62567 90 0.10878 202 9.48221 112 9.77567 77 0.78674 101 0.62568 90 0.11281 199 9.48221 112 9.77756 77 0.78674 101 0.62588 90 0.11281 199 9.48334 110 9.77625 76 0.78975 100 0.62927 88 0.	68.1	0.62012	94	0.09427	211	9.47414	811	9.78296	77	0.77952	105	1.51599	204
0.62292 93 0.09848 209 947649 116 9.78142 77 0.78265 103 0.62384 92 0.10057 206 9.4786 115 9.78065 77 0.78265 103 0.62364 92 0.10264 206 9.4786 114 9.77918 78 0.78471 103 0.62567 91 0.10675 203 9.48108 113 9.77833 77 0.78471 102 0.62567 90 0.10675 203 9.48108 113 9.77833 77 0.78471 102 0.62567 90 0.10878 202 9.48221 112 9.77833 77 0.78471 101 0.62678 90 0.10878 202 9.48444 110 9.77602 77 0.78475 100 0.62838 89 0.11480 198 9.48544 110 9.77602 77 0.78875 100 0.62927 88 0.114	1.90	0.62106	93	0.09638	210	9.47532	117	9.78219	77	0.78057	104	1.51803	203
0.62292 92 0.10057 207 9.47765 115 9.78065 77 0.78265 103 0.62384 92 0.10264 206 9.4786 114 9.77918 77 0.78368 103 0.62476 91 0.10470 205 9.47994 114 9.77911 78 0.78471 102 0.62567 91 0.10675 203 9.48108 113 9.77833 77 0.78471 102 0.625748 90 0.10878 202 9.48221 112 9.77562 77 0.78575 100 0.62838 89 0.11281 199 9.48444 110 9.77602 77 0.78775 100 0.62927 88 0.11480 198 9.48544 110 9.77502 77 0.78875 100 0.63191 174 0.12070 387 9.48653 216 9.77502 77 0.78875 197 0.63355 172	16.1	0.62199	93	0.09848	209	9.47649	911	9.78142	77	0.78161	104	1.52006	202
0.62292 92 0.10057 207 947765 115 9.78065 77 0.78265 103 0.62384 92 0.10247 206 9.47880 114 9.77938 77 0.78368 103 0.62267 91 0.10470 205 9.47994 114 9.77911 78 0.78371 101 0.62567 91 0.10470 205 9.48221 112 9.7781 78 0.78573 101 0.62588 90 0.11080 201 9.48333 111 9.77575 77 0.78674 101 0.62838 89 0.11281 199 9.48454 110 9.77679 77 0.78875 100 0.62927 88 0.11480 198 9.48654 109 9.77525 76 0.78975 100 0.63015 174 0.12070 387 9.48879 214 9.77525 76 0.79477 195 0.63355 172 0.					,								
0.62384 92 0.10264 206 9.47880 114 9.77988 77 0.78368 103 0.62476 91 0.10470 205 9.47894 114 9.77911 78 0.78471 102 0.62567 90 0.10675 203 9.48221 112 9.77833 77 0.78471 101 0.62587 90 0.10878 202 9.48221 112 9.77556 77 0.78574 101 0.62838 90 0.11281 199 9.48334 110 9.77652 76 0.78575 100 0.62927 88 0.11480 198 9.48544 110 9.77652 76 0.78975 100 0.63015 176 0.11678 392 9.48654 109 9.77525 76 0.78975 100 0.63191 174 0.12070 387 9.48879 214 9.77296 153 0.79677 195 0.63357 170 <t< td=""><td>1.92</td><td>0.62292</td><td>92</td><td>0.10057</td><td>207</td><td>9.47765</td><td>115</td><td>9.78065</td><td>77</td><td>0.78265</td><td>103</td><td>1.52208</td><td>20I</td></t<>	1.92	0.62292	92	0.10057	207	9.47765	115	9.78065	77	0.78265	103	1.52208	20I
0.62476 91 0.10470 205 9.47994 114 9.77911 78 0.78471 102 0.62567 91 0.10675 203 9.48108 113 9.77833 77 0.78573 101 0.62658 90 0.10675 202 9.48221 112 9.77556 77 0.78573 101 0.62838 90 0.11080 201 9.48444 110 9.77602 77 0.78775 100 0.62838 90 0.11678 392 9.48643 110 9.77602 77 0.78775 100 0.62927 88 0.11678 392 9.48663 216 9.77525 76 0.78875 100 0.63191 174 0.12070 387 9.48663 214 9.77526 153 0.79467 193 0.63365 172 0.12457 382 9.49093 211 9.77143 152 0.79467 193 0.633707 168	1.93	0.62384	92	0.10264	206	9.47880	114	9.77988	77	0.78368	103	1.52409	200
0.62567 91 0.10675 203 9.48108 113 9.77833 77 0.78573 101 0.62658 90 0.10878 202 9.48221 112 9.77556 77 0.78573 101 0.62838 89 0.11281 199 9.48444 110 9.77602 77 0.78775 100 0.62927 88 0.11480 198 9.48544 110 9.77602 77 0.78775 100 0.63015 176 0.11678 392 9.48654 109 9.77525 76 0.78775 100 0.63191 174 0.12070 387 9.48879 214 9.77526 153 0.79467 195 0.63357 172 0.12457 382 9.49093 211 9.77143 152 0.79467 193 0.63537 166 0.13231 374 9.49304 208 9.76991 152 0.79467 191 0.63537 166	1.94	0.62476	16	0.10470	205	9.47994	114	11627.6	78	0.78471	102	1.52609	199
0.62658 90 0.10878 202 9.48221 112 9.77756 77 0.78674 101 0.62838 90 0.11080 201 9.48344 110 9.77602 77 0.78775 100 0.62927 88 0.11281 199 9.48444 110 9.77602 77 0.78875 100 0.62927 88 0.11480 198 9.48654 109 9.77525 76 0.78975 100 0.63015 176 0.11678 392 9.48654 109 9.77525 76 0.78975 100 0.63191 174 0.12070 387 9.48879 214 9.77296 153 0.79467 195 0.63357 170 0.12839 378 9.49304 208 9.76991 152 0.79467 193 0.63537 166 0.13217 374 9.49512 203 9.76839 151 0.80040 187 0.640441 165	1.95	0.62567	16	0.10675	203	9.48108	113	9.77833	77	0.78573	101	1.52808	198
0.62748 90 0.11080 201 9.48333 111 9.77679 77 0.78775 100 0.62838 89 0.11281 199 9.48444 110 9.77602 77 0.78775 100 0.62927 88 0.11480 198 9.48654 109 9.77525 76 0.78775 100 0.63015 176 0.11678 392 9.48663 216 9.77449 153 0.79075 100 0.63191 174 0.12070 387 9.48879 214 9.77449 153 0.79075 197 0.63365 172 0.12457 382 9.49903 211 9.77143 152 0.79467 193 0.63537 170 0.12839 378 9.49304 208 9.76991 152 0.79851 189 0.633707 166 0.13591 369 9.49971 203 9.76687 151 0.80040 187 0.640441 165	1.96	0.62658	96	0.10878	202	9.48221	112	9.77756	77	0.78674	IOI	1.53006	197
0.62748 90 0.11080 201 9.48333 111 9.77679 77 0.78775 100 0.62838 89 0.11281 199 9.4844 110 9.77602 77 0.78875 100 0.63915 176 0.11480 198 9.48544 110 9.77525 76 0.78875 100 0.63915 176 0.11678 392 9.48663 216 9.77449 153 0.79075 100 0.63191 174 0.12070 387 9.48879 214 9.77449 153 0.79075 197 0.63365 172 0.12457 382 9.49987 214 9.77296 153 0.79467 193 0.63367 170 0.12839 378 9.49304 208 9.76839 152 0.79660 191 0.63875 166 0.13591 369 9.49717 203 9.76839 151 0.80040 187 0.640441 165													
0.62838 89 0.11281 199 9.48444 110 9.77602 77 0.78875 100 0.63927 88 0.11480 198 9.48554 109 9.77525 76 0.78875 100 0.63191 176 0.11678 392 9.48663 216 9.77449 153 0.79075 197 0.63191 174 0.12070 387 9.48879 214 9.77449 153 0.79075 197 0.63365 172 0.12457 382 9.49093 211 9.77143 152 0.79467 193 0.63377 168 0.13217 374 9.4934 208 9.76991 152 0.79660 191 0.63875 166 0.13217 369 9.49717 203 9.76687 151 0.80040 187 0.640441 165 0.13590 365 9.49920 200 9.76536 151 0.80040 187	1.97	0.62748	96	0.11080	201	9.48333	III	6.77679	22	0.78775	100	1.53203	961
0.62927 88 0.11480 198 9.48554 109 9.77525 76 0.78975 100 0.63015 176 0.11678 392 9.48663 216 9.77449 153 0.79975 197 0.63191 177 0.12670 387 9.48879 214 9.77449 153 0.79975 197 0.63365 172 0.12457 382 9.49093 211 9.77143 152 0.79467 193 0.63537 168 0.13237 374 9.49342 208 9.76991 152 0.79660 191 0.63875 166 0.13591 369 9.49717 203 9.76687 151 0.80040 187 0.640441 165 0.13590 365 9.49920 200 9.76536 151 0.80040 187	1.98	0.62838	89	0.11281	199	9.48444	011	9.77602	77	0.78875	100	1.53399	195
0.63015 176 0.11678 392 9.48663 216 9.77449 153 0.79075 197 0.63191 174 0.12070 387 9.48879 214 9.77296 153 0.79272 197 0.63365 172 0.12457 382 9.49993 211 9.77143 152 0.79467 193 0.63537 170 0.12839 378 9.49304 208 9.76991 152 0.79660 191 0.63707 168 0.13517 369 9.49512 205 9.76839 152 0.79851 189 0.64041 165 0.13591 365 9.49920 200 9.76536 151 0.80040 187	1.99	0.62927	88	0.11480	861	9.48554	109	9.77525	92	0.78975	100	1.53594	194
0.63365 172 0.12457 387 9.48879 214 9.77296 153 0.79272 195 0.63365 172 0.12457 382 9.49093 211 9.77143 152 0.79467 193 0.63767 170 0.12839 378 9.49304 208 9.76891 152 0.79660 191 0.63767 168 0.13217 374 9.49512 205 9.76839 152 0.79660 191 0.64844 165 0.13960 365 9.499717 203 9.76687 151 0.80040 187 0.64044 165 0.13960 365 9.49920 200 9.76536 151 0.80227 185	2.00	0.63015	921	0.11678	392	9.48663	216	9.77449	153	0.79075	197	1.53788	383
0.63365 172 0.12457 382 9.49093 211 9.77143 152 0.79467 193 0.63537 170 0.12839 378 9.49304 208 9.76991 152 0.79467 191 0.63707 168 0.13217 374 9.49512 205 9.76839 152 0.79650 191 0.63875 166 0.13591 369 9.49717 203 9.76687 151 0.80040 187 0.640441 165 0.13960 365 9.49920 200 9.76536 151 0.80227 185	2.02	0.63191	174	0.12070	387	9.48879	214	9.77296	153	0.79272	195	1.54171	379
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$			1	1	-								
0.63537 170 0.12839 378 9.49304 208 9.76991 152 0.79660 191 0.63707 168 0.13517 374 9.49512 205 9.76839 152 0.79851 189 0.63875 166 0.13591 369 9.49717 203 9.76687 151 0.80040 187 0.64041 165 0.13960 365 9.49920 200 9.76536 151 0.80227 185	2.04	0.03305	172	0.12457	382	9.49093	211	9.77143	152	0.79467	193	1.54550	370
0.63707 168 0.13217 374 9.49512 205 9.76839 152 0.79851 189 0.63875 166 0.13591 369 9.49717 203 9.76687 151 0.80040 187 0.64041 165 0.13960 365 9.49920 200 9.76536 151 0.80227 185	2.06	0.63537	170	0.12839	378	9.49304	208	16692.6	152	09962.0	161	1.54926	373
0.63875 166 0.13591 369 9.49717 203 9.76687 151 0.80040 187 0.64041 165 0.13960 365 9.49920 200 9.76536 151 0.80227 185	2.08	0.63707	891	0.13217	374	9.49512	205	9.76839	152	0.79851	189	1.55299	369
0.64041 165 0.13960 365 9.49920 200 9.76536 151 0.80227 185	2.10	0.63875	991	0.13591	369	9.49717	203	6.76687	151	0.80040	187	1.55668	365
	2.12	0.64041	165	0.13960	365	9.49920	200	9.76536	151	0.80227	185	1.56033	361
											_		

TABLE I—Continued

	ı																											
D	358	354	350	347	344	240	240	227	23I	328)	325	322	319	317	314		32.1	300	305	302	300	6	767	295	292	290	287
log X ₅	1.56394	1.56752	1.57106	1.57456	1.57803	1 28147	1.00.1	1.58824	1.50158	1.59489		1.59817	1.60142	1.60464	1.60783	00119.1	1 61414	1.01414	1.01/25	1.62033	1.62338	1.62640	1 60040	1.02940	1.63237	1.63532	1.63824	1.64114
D	184	182	180	178	177	1	172	172	170	168		167	991	164	163	161	160	0 1	150	157	156	154	i i	133	152	150	149	148
$\log X_4$	0.80412	0.80596	0.80778	0.80958	0.81136	0.81717	20100	0.81661	0.81833	0.82003		0.82171	0.82338	0.82504	0.82668	0.82831	0 82002	0.02999	0.03152	0.83310	0.83467	0.83623	2 0 0	///50.0	0.83930	0.84082	0.84232	0.84381
D	- 150	150	150	149	149	148	0 0 0	148	147	147		146	146	146	145	145	144	+++	144	143	143	142	140	747	141	141	140	140
log X ₃	9.76385	9.76235	9.76085	9.75935	9.75786	0 75627	0.75480	9.75341	9.75193	9.75046	(9.74899	9.74753	9.74607	9.74461	9.74316	17117	17477	17405/	9.73003	9.73740	9.73597	7246	2.13433	9.73313	9.73172	9.73031	9.72891
D	197	195	192	190	188	187	183	180	178	176	*******	174	172	171	891	991	164	401	102	101	159	157	u	661	154	151	150	148
log X ₂	9.50120	9.50317	9.50512	9.50704	9.50894	0.51082	0.51267	9.51450	9.51630	9.51808		9.51984	9.52158	9.52330	9.52501	9.52669	0 52825	000000	9.52999	9.53101	9.53322	9.53481	80900	7.33030	9.53793	9.53947	9.54098	9.54248
D	361	357	352	348	344	240	327	333	329	326		322	319	315	312	309	306	200	302	299	296	293	300	267	287	285	282	279
log X ₁	0.14325	0.14686	0.15043	0.15395	0.15743	0.16087	0.16427	0.16764	0.17097	0.17426		0.17752	0.18074	0.18393	0.18708	0.19020	0 10220	0 10625	0.19033	0.19937	0.20230	0.20532	0.20825	0.2002.0	0.21115	0.21402	0.21687	0.21969
D	163	162	091	158	157	Y L	154	152	151	149	(148	147	145	144	143	141	747	140	139	137	136	125	100-	134	133	132	131
log X ₀	0.64206	0.64369	0.64531	0.64691	0.64849	0.65006	0.65161	0.65315	0.65467	0.65618	,	0.65767	0.65915	0.66062	0.66207	0.66351	10.66404	164000	0.00033	0.00775	0.00014	0.67051	0.67187	101/00	0.07322	0.67456	0.67589	0.67721
*	2.14	2.16	2.18;	2.20	2.22	2.24	2.26	2.28	2.30	2.32		2.34	2.36	2.38	2.40	2.42	2.44	2 46	0.00	04.7	2.50	2.52	27.0	+0	2.50	2.58	2.60	2.62

TABLE I—Continued

D	285 283	28 <u>1</u> 279 277	274 273	270 269 267	264 262 260 258 258	252 252 251 618 608	597 588 578 568 559
log X ₅	1.64401	1.64969 1.65250 1.65529	1.65806	1.66353 1.66623 1.66892	1.67159 1.67423 1.67685 1.67945 1.68203	1.68459 1.68713 1.68965 1.69216 1.69834	1.70442 1.71039 1.71627 1.72205 1.72773
D	147 146	144 441 444 143	141 140	139 138 137	136 135 134 133	131 130 130 319 314	308 302 297 293 289
$\log X_4$	0.84529	0.84822 0.84966 0.85110	0.85253	0.85534 0.85673 0.85811	0.85948 0.86084 0.86219 0.86353 0.86486	0.86618 0.86749 0.86879 0.87328	0.87642 0.87950 0.88252 0.88549 0.88842
D	- 140 139	138	138	137 136 136	135 135 134 134 133	133 133 132 328 328	323 320 318 315 315
$\log X_3$	9.72751	9.72472 9.72333 9.72195	9.72057	9.71782 9.71645 9.71509	9.71373 9.71238 9.71103 9.70969 9.70835	9.70702 9.70569 9.70436 9.70304 9.66976	9.69650 9.69327 9.69007 9.68689 9.68374
D	147 146	144 143 141	140 138	136 135 134	133 131 130 129 128	126 125 124 304 297	290 284 272 266
$\log X_2$	9.54396	9.54689 9.54833 9.54833	9.55117	9.55395 9.55531 9.55666	9.55800 9.55933 9.56064 9.56194 9.56323	9.56451 9.56577 9.56702 9.56826 9.57130	9.57427 9.57717 9.58001 9.58279 9.58551
D	277	271 269 266	264	259 257 254	252 250 248 246 246	242 239 237 584 572	561 549 538 528 518
$\log X_1$	0.22248	0.22799 0.23070 0.23339	0.23605	0.24130 0.24389 0.24646	0.24900 0.25152 0.25402 0.25650 0.25896	0.26140 0.26382 0.26621 0.26858 0.27442	0.28014 0.28575 0.29124 0.29662 0.30190
D	130 128	127 126 125	125 124	122 121 121	119	115 114 113 280 275	270 265 261 256
$\log X_0$	0.67852	0.68110	0.68488	0.08737 0.68859 0.68980	0.69101 0.69220 0.69339 0.69457 0.69574	0.69690 0.69805 0.69919 0.70032 0.70312	0.70587 0.70857 0.71122 0.71383 0.71639
×	2.64	2.70	2.74	2.80	2.84 2.86 2.90 2.92	2.94 2.96 2.98 3.00 3.05	3.10 3.15 3.20 3.25 3.30

TABLE I—Continued

D	550	541	533	525	513	80%	502	496	489	482	474	467	460	454	449	443	438	431	426	421	415	410	406	401	396
log X ₅	1.73332	1.73882	1.74423	1.74956	1.75481	1.75004	1.76502	1.77004	1.77500	1.77989	1.78471	1.78945	1.79412	1.79872	1.80326	1.80775	1.81218	1.81656	1.82087	1.82513	1.82934	1.83349	1.83759	1.84165	1.84566
D	285	279	275	271	267	262	250	25.5	252	247	245	243	238	235	232	220	226	223	220	218	215	212	210	208	205
$\log X_4$	0.89131	0.89416	0.89695	0.89970	0.90241	0.00508	0.90771	0.91030	0.91285	0.91537	0.91784	0.92029	0.92272	0.92510	0.92745	0.92977	0.93206	0.93432	0.93655	0.93875	0.94093	0.94308	0.94520	0.94730	0.94938
D	-310	307	305	302	300	207	205	293	290	288	285	283	282	279	277	275	272	270	268	267	264	262	260	259	257
log X ₃	9.68062	9.67752	9.67445	9.67140	9.66838	0.66528	9.66241	9.65946	9.65653	9.65363	9.65075	9.64790	9.64507	9.64225	9.63946	9.63669	9.63394	9.63122	9.62852	9.62584	9.62317	9.62053	9.61791	9.61531	9.61272
D	260	255	250	244	240	225	231	226	22I	217	214	210	205	202	199	19.5	192	188	186	182	179	176	173	171	891
log X ₂	9.58817	9.59077	9.59332	9.59582	9.59826	990090	9.60301	9.60532	9.60758	62609.6	961196	9.61410	9.61620	9.61825	9.62027	9.62226	9.62421	9.62613	9.62801	9.62987	9.63169	9.63348	9.63524	9.63697	9.63868
D	500	499	489	481	473	464	456	440	441	433	427	420	413	406	400	394	388	382	376	371	366	360	355	351	346
log X ₁	0.30708	0.31217	0.31716	0.32205	0.32686	0.33150	0.33623	0.34079	0.34528	0.34969	0.35402	0.35829	0.36249	0.36662	0.37068	0.37468	0.37862	0.38250	0.38632	0.39008	0.39379	0.39745	0.40105	0.40460	0.40811
D	249	244	240	236	233	220	226	223	219	216	213	210	207	204	201	661	197	194	161	681	187	184	182	180	177
log X ₀ .	0.71891	0.72140	0.72384	0.72624	0.72860	0.73003	0.73322	0.73548	0.73771	0.73990	0.74206	0.74419	0.74629	0.74836	0.75040	0.75241	0.75440	0.75637	0.75831	0.76022	0.76211	0.76398	0.76582	0.76764	0.76944
H	3.35	3.40	3.45	3.50	3.55	3.60	3.65	3.70	3.75	3.80	3.85	3.90	3.95	4.00	4.05	4.10	4.15	4.20	4.25	4.30	4.35	4.40	4.45	4.50	4.55

TABLE I—Continued

D	391	82	28	14	02	366	19	22	24	20	47	43	39	336	33	50	56	23	320	17	14	II	20	304
	436						_				_					,						_		
log X ₅	85352	85740	86122	36500	86874	1.87244	87610	12628	88328	88682	89032	89379	89722	19006.1	90307	90730	91059	91385	80/16:	92028	92345	92659	92970	1.93277
lol																_	_	_	_		_			
D	202	198	961	193	161	189	188	185	183	182	180	177	176	174	173	171	166	167	991	164	162	191	160	158
X_4	0.95143	545	743	939	132	0.96323	512	200	885	890,	250	430	209	0.97783	7957	3130	30I	3470	0.98637	803	2965	129	290	0.99450
log	0.95	0.95	0.95	0.0	96.0	96.0	0.96	0.0	96.0	0.97	0.97	0.97	0.97	0.97	0.07	0.08	0.0	96.0	96.0	96.0	0.08	0.0	0.99	0.99
D	-255 253	250	249	247	246	244	242	240	238	237	235	234	232	23I	220	228	225	225	223	22 I	220	219	217	215
X_3	15	20	57	 80	19	15	71	29	 68	51	14	62	45	13	82	53	200	00	75	22	31	II	92	75
log A	9.61015	9.605	9.602	9.600	9.59761	9.595	9.592	9.590	9.587	9.585	9.58314	9.580	9.578	9.576	9.573	9.571	9.569	9.56700	9.564	9.562	9.560	9.558	9.555	9.55375
D	165	160	158	155	153	151	149	146	145	142	141	138	137	135	133	131	129	128	126	124	123	121	120	811
	9 I	4	4	7	7	0	Н	0	9	н	3	4	- 2	6	4	7	.∞	7	2	н	10		6	6
$\log X_2$	9.64036	3.6436	9.6452	3.6468	9.64837	9.6499	9.6514	9.6529	3.6543	9.65581	3.6572	9.6586	9.6600).6613	3.6627	9.6640	9.6653	9.66667	9.6679	9.6692	9.6704	9.6716	9.67289	9.6740
							_	_			5 962					_	_	273			_		257 9	
D	340	33	32	32	31	314	31	30	30	29	29	29	58	~~	28	27	27	27	56	26	26	26	25	25
X_1	0.41157	833	165	492	0.42815	134	448	759	005	5.44368	299	963	255	543	828	0110	388	0.46663	936	205	471	734	J.47994	251
log	0.41	0.41	0.42	0.42	0.42	0.43	0.43	0.43	0.44	0.44	0.44	0.44	0.45	0.45	0.45	0.46	0.46	0.46	0.46	0.47	0.47	0.47	0.47	0.48
D	175	171	691	167	991	164	162	160	158	157	155	153	152	150	149	147	146	145	143	142	140	139	137	136
X ₀	21	20,	41	01	77	43	100	69	62	87	44	66	52	40.	54	03	50	96	41	84	26	99	.05	245
log A	0.77121	0.774	0.776	0.778	0.779	0.78143	0.783	0.784	0.786	0.78787	0.789	0.790	0.792	0.794	0.795	0.797	962.0	0.79996	0.801	0.802	0.804	0.805	0.80705	308.0
*	4.60	202	.75	08.	4.85	06	95	00.	.05	01.	15	20	25	5.30	35	40	45	.50	5.55	9.	.65	5.70	.75	08.

TABLE I—Continued

D	302 299 297 586 575	555 538 528 828 828	519 503 497 490	481 474 468 461 454	447 440 435 431 425
$\log X_5$	1.93581	1.95640	1.98372	2.00892	2.03230
	1.93883	1.96205	1.98891	2.01373	2.03677
	1.94182	1.96760	1.99402	2.01847	2.04117
	1.94479	1.97306	1.99905	2.02315	2.04552
	1.95065	1.97844	2.00402	2.02776	2.04983
D	157	293	270	250	233
	155	288	265	246	230
	154	283	262	242	226
	304	279	258	239	223
	298	274	258	236	223
$\log X_4$	0.99608 0.99765 0.99920 1.00074 1.00378	1.00676 1.00969 1.01257 1.01540 1.01819	1.02093 1.02363 1.02628 1.02890 1.03148	1.03402 1.03652 1.03898 1.04140	1.04615 1.04848 1.05078 1.05304 1.05527
D	-214	408	385	364	345
	213	405	381	360	341
	212	399	376	356	338
	419	394	372	352	333
	415	390	368	349	330
log X ₃	9.55160	9.53687	9.51691	9.49809	9.48028
	9.54946	9.53279	9.51306	9.49445	9.47683
	9.54733	9.52874	9.50925	9.49085	9.47342
	9.54521	9.52475	9.50549	9.48729	9.47004
	9.54102	9.52081	9.50177	9.48377	9.46671
D	117	213	191	172	156
	116	209	186	168	153
	114	204	183	165	151
	224	200	179	165	147
	218	195	176	159	145
log X ₂	9.67527	9.68316	9.69337	9.70252	9.71078
	9.67644	9.68529	9.69528	9.70424	9.71234
	9.67760	9.68738	9.69714	9.70592	9.71387
	9.67874	9.6942	9.69897	9.70757	9.71538
	9.68098	9.69142	9.60076	9.70919	9.71685
D	252	466	423	387	356
	249	457	415	380	349
	247	447	408	373	345
	485	439	400	368	339
	475	430	394	361	339
$\log X_1$	0.48505	0.50213	0.52452	0.54492	0.56361
	0.48757	0.50679	0.52875	0.54879	0.56717
	0.49006	0.51136	0.53290	0.55259	0.57066
	0.49253	0.51583	0.53698	0.55632	0.57411
	0.49738	0.52022	0.54098	0.56000	0.57750
D	135 134 132 262 256	252 248 244 240 236	232 2224 2224 2224 218	215 211 208 208 206 203	199 196 194 189 189
log X ₀	0.80978	0.81897	0.83117	0.84241	0.85284
	0.81113	0.82149	0.83349	0.84456	0.85483
	0.81247	0.82397	0.83577	0.84667	0.85679
	0.81379	0.82641	0.83801	0.84875	0.85873
	0.81641	0.82881	0.84023	0.85081	0.86065
*	5.85 5.90 5.95 6.00 6.1	6.3 6.3 6.5 6.5 6.5	6.7 6.9 7.0 7.1	2.7.7.3.7.5.4.5.7.5	7.7 7.8 7.9 8.0 8.1

TABLE I—Continued

D	419	393	369	345	331
	413	388	366	345	327
	407	383	362	338	324
	403	379	357	338	322
	398	374	353	338	318
$\log X_{\delta}$	2.05408	2.07448	2.09365	2.11172	2.12882
	2.05827	2.07841	2.09734	2.11522	2.13213
	2.06240	2.08229	2.10100	2.11867	2.13540
	2.06647	2.08612	2.10462	2.12209	2.13864
	2.07050	2.08991	2.10819	2.12547	2.14186
D	217	204	193	182	172
	215	202	190	179	171
	212	199	187	178	169
	209	196	186	176	167
	209	195	186	176	165
log X4	1.05748	1.06808	1.07804	1.08744	1.09633
	1.05965	1.07012	1.07997	1.08926	1.09805
	1.06180	1.07214	1.08187	1.09105	1.09976
	1.06392	1.07413	1.08374	1.09283	1.10145
	1.06601	1.07609	1.08560	1.09459	1.10312
D	-328 324 321 321 318 318	312 309 306 303 303	297 292 292 289 287	284 282 279 277 275	272 270 268 268 265
log X ₃	9.46341	9.44736	9.43206	9.41746	9.40349
	9.46013	9.44424	9.42909	9.41462	9.40077
	9.45689	9.44115	9.42614	9.41180	9.39807
	9.45368	9.43809	9.42322	9.40901	9.39539
	9.45050	9.43506	9.42033	9.40624	9.39274
D	142	130	121	112	104
	140	128	119	111	102
	138	127	116	108	101
	135	124	115	106	100
	133	123	113	105	98
log X ₂	9.71830 9.71972 0.72112 9.72250 9.72385	9.72518 9.72648 9.72776 9.72903	9.73150 9.73271 9.73390 9.73506 9.73621	9.73734 9.73846 9.73957 9.74065	9.74276 9.74380 9.74482 9.74583 9.74683
D	329	305	282	267	251
	324	301	282	264	249
	319	298	278	261	246
	315	289	274	257	242
	310	289	271	257	239
log X ₁	0.58084	0.59681	0.61167	0.62557	0.63860
	0.58413	0.59986	0.61452	0.62824	0.64111
	0.58737	0.60287	0.61734	0.63088	0.64360
	0.59056	0.60585	0.62012	0.63349	0.64606
	0.59371	0.60878	0.62286	0.63606	0.64848
D	187	175	165	156	147
	184	173	163	154	146
	182	171	161	152	145
	179	169	159	151	143
	177	166	157	151	142
log X ₀	0.86254	0.87163	0.88017	0.88822	0.89584
	0.86441	0.87338	0.88182	0.88978	0.89731
	0.86625	0.87511	0.88345	0.89132	0.89877
	0.86807	0.87682	0.88506	0.89284	0.90022
	0.86986	0.87851	0.88665	0.89435	0.90165
*	2.8.8.8.8. 2.5.4.8.0.	8.8 8.9 9.0 9.1	9.60	9.7 9.8 9.9 10.0	10.2 10.3 10.4 10.5 10.6

TABLE I—Continued

D	314	308	305	303	300	297	294	291	288	286	283	280	277	275	272	27.0	1/2	500	202	204	262	260	257	200	000	433
log X ₅	2.14504	2.15129	2.15437	2.15742	2.16045	2.16345	2.16642	2.10936	2.17227	2.17515	2.17801	2.18084	2.18364	2.18641	2 18016	2.10910	2.19109	2.19400	2.19729	2.19996	2.20260	2.20522	2.20782	2.21030	200100	4.212.94
D	163	191	159	157	156	154	153	151	150	149	147	146	144	143	142	140	145	140	138	138	136	134	124	122	130	132
$\log X_4$	1.10477	1.10802	1.10963	1.11122	1.11279	1.11435	1.11589	1.11742	1.11893	1.12043	1.12192	1.12339	1.12485	1.12629	011011	1.12//2	1.12915	1.13057	1.13197	1.13335	1.13473	1.13600	T 12742	1 1 2 8 1 1	1/0011	1.14010
D	-262	257	255	253	251	249	248	245	244	241	240	237	235	234		233	232	230	227	226	225	222	2010	177	077	219
log X ₃	9.39011	9.38490	9.38233	9.37978	9.37725	9.37474	9.37225	9.36977	9.36732	9.36488	9.36247	9.36007	9.35770	9.35535	1	9.35301	9.35008	9.34836	9.34606	9.34379	0.24153	0 22028	7.000	9.33703	9.33404	9.33204
D	96	95	93	92	16	89	 	87	98	20	84	83	82	81	G	00	79	78	77	77	7	1.0	+ 7	+ 5	13	72
log X ₂	9.74781	9.74972	9.75067	9.75160	9.75252	9.75343	9.75432	9.75520	9.75607	9.75693	9.75778	9.75862	9.75945	9.76027	. 1	9.70108	9.76188	9.76267	9.76345	9.76422	001920	0.76574	4/20/16	9.70040	9.70/22	9.70795
D	237	232	229	526	224	222	219	217	215	212	210	208	205	204		202	200	861	195	194	103	261	2001	100	/01	185
$\log X_1$	0.65087	0.65558	0.65790	0.66019	0.66245	0.66469	16999.0	0.699.0	0.67127	0.67342	0.67554	0.67764	0.67972	0.68177	0	0.68381	0.68583	0.68783	0.68981	92169.0	0,000	0,093/0	200000	0.09752	0.09941	0.70128
D	140	137	135	135	134	132	131	130	129	127	126	125	124	123		122	120	611	119	811	111	7 1 1	611	611	114	113
log X ₀	0.90307	0.90586	0.90723	0.90858	0.00093	0.91127	0.91259	0.91390	0.91520	00100	0.01776	0.01002	0.92027	0.92151		0.92274	0.92396	0.92516	0.92635	0.92754	0.0000	2/0260	0.92909	0.93104	0.93219	0.93333
*	10.7	10.0	0.11	II.I	11.2	11.3	11.4	11.5	9.11	1	× 11	0.11	12.0	12.1		12.2	12.3	12.4	12.5	12.6	0	1.71	12.0	12.9	13.0	13.1

TABLE I—Continued

D	251	241	455	426	397
	249	239	449	422	392
	247	238	443	408	386
	245	470	436	406	382
	243	463	431	406	377
log X ₅	2.21547	2.22782	2.24433	2.26647	2.28711
	2.21798	2.23023	2.24888	2.27073	2.29108
	2.22047	2.23262	2.25337	2.27495	2.29500
	2.22294	2.23500	2.25780	2.27903	2.29886
	2.22539	2.23970	2.25780	2.28309	2.30268
D	130 130 129 128 128	125 124 124 244 241	238 233 227 227	221 218 214 213 209	206 204 201 199 197
log X ₄	1.14142	1.14786	1.15644	1.16797	1.17872
	1.14272	1.14911	1.15882	1.17018	1.18078
	1.14402	1.15035	1.16115	1.17236	1.18282
	1.14531	1.15159	1.16346	1.17450	1.18483
	1.14659	1.15403	1.16573	1.17663	1.18682
D	217	209	339	377	355
	215	208	395	371	351
	214	208	390	367	347
	213	410	385	363	343
	213	405	381	359	340
log X ₃	9.33045	9.31975	9.30535	9.28585	9.26748
	9.32828	9.31766	9.30136	9.28208	9.26393
	9.32613	9.31558	9.29741	9.27837	9.26042
	9.32399	9.31350	9.29351	9.27470	9.25695
	9.32186	9.30940	9.28966	9.27107	9.25352
D	71	68	125	115	104
	70	67	122	110	102
	70	66	121	109	100
	69	130	117	108	98
	68	128	116	105	97
$\log X_2$	9.76867 9.76938 9.77009 0.77079	9.77216 9.77284 9.77351 9.77417	9.77675 9.77800 9.77922 9.78043 9.78160	9.78276 9.78391 9.78501 9.78610	9.78823 9.78927 9.79029 9.79129 9.79227
D	183 182 180 178	175 174 172 340 334	328 323 318 318 308	304 298 293 289 285	281 277 273 269 265
log X ₁	0.70313	0.71213	0.72408	0.73997	0.75466
	0.70496	0.71388	0.72736	0.74301	0.75747
	0.70678	0.71562	0.73059	0.74599	0.76024
	0.70858	0.71734	0.73377	0.74892	0.76297
	0.71036	0.72074	0.73689	0.75181	0.76596
D	111	108	203	189	177
	111	107	201	187	175
	110	106	197	181	173
	109	210	195	181	170
	108	206	192	180	169
log X ₀	0.93446	0.93996	0.94733	0.95721	0.96643
	0.93558	0.94104	0.94936	0.95910	0.96820
	0.93669	0.94211	0.95137	0.96097	0.96995
	0.93779	0.94317	0.95334	0.96282	0.97168
	0.93888	0.94527	0.95529	0.96463	0.97338
*	13.2 13.3 13.4 13.5 13.6	13.7 13.8 13.9 14.0	14.4 14.6 14.8 15.0 15.2	15.4 15.6 15.8 16.0	16.4 16.6 16.8 17.0 17.2

TABLE I—Continued

H	log X ₀	D	log X ₁	D	log X ₂	D	log X ₃	a	log X4	D	log X ₅	D
17.4	0.97507	991	0.76831	262	9.79324	9.5	9.25012	-336	0.18879	103	2.30645	272
17.6	0.97673	165	0.77093	258	9.79419	94	9.24676	332	1.19072	192	2.31017	368
17.8	0.97838	163	0.77351	255	9.79513	92	9.24344	329	1.19264	190	2.31385	365
18.0	0.98001	091	909220	250	9.79605	16	9.24015	326	1.19454	186	2.31750	3.58
18.2	19186.0	159	0.77856	248	96962.6	68	9.23689	322	1.19640	185	2.32108	355
18.4	0.98320	157	0.78104	245	0.70785	84	0 22267	210	T TO825	182	2 22462	
18.6	0.98477	155	0.78349	242	9.79872	82	0.23048	316	1.20008	180	2.22814	247
18.8	0.98632	153	0.78591	238	9.79959	85	9.22732	313	1.20188	179	2.33161	347
0.61	0.98785	152	0.78829	235	9.80044	84	9.22419	310	1.20367	176	2.33504	339
19.2	0.98937	149	0.79064	232	9.80128	82	9.22109	306	1.20543	174	2.33843	334
19.4	0,99086	140	0.79296	230	0.80210	8	0.21802	207	71700 1	17.7	77177	,
19.6	0.99235	147	0.79526	228	9.80292	8	9.21499	301	1.20891	171	2.34510	328
8.61	0.99382	145	0.79754	225	9.80372	79	9.21198	298	1.21062	168	2.34838	326
20.0	0.99527	145	0.79979	222	9.80451	77	9.20900	295	1.21230	175	2.35164	326
20.2	0.99672	145	0.80201	220	9.80528	92	9.20605	288	1.21405	291	2.35490	323
20.4	0.99817	144	0.80421	219	0.80604	75	0.20217	205	1.21572	163	2.25812	218
20.6	1966600	143	0.80640	218	9.80679	7.5	9.20022	291	1.21735	191	2.36131	315
20.8	1.00104	143	0.80858	217	9.80754	74	9.19731	288	1.21896	159	2.36446	310
21.0	1.00247	142	0.81075	215	9.80828	73	9.19443	287	1.22055	159	2.36756	306
21.2	1.00389	138	0.81290	209	10608.6	71	9.19156	286	1.22214	159	2.37062	303
21.4	1.00527	127	0.81400	208	0.80072	7.1	0.18870	940	1 22272	u u F	2 2776	× × ×
21.6	1.00664	126	0.81707	300	0.81017	10	0.18504	010	2010	CO 1	2.37.303	200
21.8	1 00800	121	0,1010	907	0,01043	2,9	9.10394	5/2	1.22320	661	2.37003	293
0.14	200001	101	0.01913	200	9.01113	50	9.10321	500	1.22003	152	2.37958	291
0.22	1.0093/	131	0.02119	199	9.01162	80	9.18053	205	1.22835	149	2.38249	287
22.2	1.01008	131	0.82318	199	9.81250	89	9.17788	263	1.22984	145	2.38536	283
1												

TABLE I—Continued

and the same of th											
log X ₀	D	$\log X_1$	D	log X ₂	D	log X ₃	D	$\log X_4$	Q	log X ₅	D
1.01199	127	0.82517	193	9.81318	99	9.17525	262	1.23129	145	2.38819	282
1.01326	124	0.82710	161	9.81384	65	9.17263	264	1.23274	145	2.39101	278
1.01450	123	0.82901	186	9.81449	64	9.16999	264	1.23419	144	2.39379	276
1.01573	122	0.83087	185	9.81513	64	9.16735	261	1.23563	141	2.39655	274
1.01695	117	- 0.83272	180	9.81577	63	9.16474	256	1.23704	141	2.39929	273
1.01812	911	0.83452	178	9.81640	62	9.16218	252	1.23845	140	2.40202	271
1.01928	911	0.83630	177	9.81702	19	9.15966	248	1.23985	140	2.40473	270
1.02044	911	0.83807	176	9.81763	99	9.15718	245	1.24125	139	2.40743	267
1.02160	911	0.83983	921	9.81823	99	9.15473	244	1.24264	. 139	2.41010	265
1.02276	115	0.84159	174	9.81883	59	9.15229	239	1.24403	138	2.41275	262
1.02391	115	0.84333	174	9.81942	59	9.14990	235	1.24541	138	2.41537	262
1.02506	115	0.84507	173	9.82001	258	9.14755	229	1.24679	138	2.41799	259
1.02621	114	0.84680	171	9.82059	57	9.14526	231	1.24817	137	2.42058	257
1.02735	114	0.84851	170	9.82116	57	9.14295	228	1.24954	137	2.42315	255
	_						_	1	_		

SUPPLEMENT TO TABLE I

D.	59 45 38	33 30 27	23 23	20 20 18	18 17 17	15 16 14	15
X_4	1.46	2.88	3.78	4.48	5.06	5.58	6.03
	2.05	3.21	4.03	88.48	5.24	5.73	6.18
	2.50	3.51	4.26	88.88	5.41	5.89	6.32
Q .	1276 638 333	151 + 58 - 8	50 77 95	106	117	115 112 109	106
X_3	0.4841	0.7088	0.7299	0.7077	0.6742	0.6392	0.6056
	0.6117	0.7249	0.7249	0.6971	0.6625	0.6277	0.5950
	0.6755	0.7307	0.7172	0.6859	0.6508	0.6165	0.5846
D	277 247 224	203 186 171	158 147 137	128 120 113	101 101 96	91 86 83	78
X_2	6.0313	0.1061	0.1621	0.2063	0.2424	0.2728	0.2988
	0.0590	0.1264	0.1779	0.2191	0.2531	0.2819	0.3066
	0.0837	0.1450	0.1926	0.2311	0.2632	0.2905	0.3142
D	556 647 694	718 728 731	727 722 712	703 691 680	669 656 645	634 622 611	590
Xı	0.0339	0.2236	0.4413	0.6574	0.8648	1.0618	1.2485
	0.0895	0.2954	0.5140	0.7277	0.9317	1.1252	1.3085
	0.1542	0.3682	0.5862	0.7968	0.9973	1.1874	1.3675
D	4336	2191	1670	1344	1137	992	884
	3238	2028	1543	1265	1083	952	853
	2661	1828	1434	1197	1035	917	826
X_0	1.0838	2.1073	2.7220	3.1867	3.5673	3.8928	4.1789
	1.5174	2.3364	2.8890	3.3211	3.6810	3.9920	4.2673
	1.8412	2.5392	3.0433	3.4476	3.7893	4.0872	4.3526
н	0.1 0.2 0.3	0.5	0.7	I.0 I.1 I.2	1.3 1.5	1.6	1.9 2.0 2.1

SUPPLEMENT TO TABLE I—Continued

D	13 13 12	13 11 12	111	11 01 :	01 00 9	01 00 01	18 17 17
X ₄	6.45 6.58 6.71	6.83 6.96 7.07	7.19 7.30 7.41	7.52	7.84 7.94 8.04	8.13 8.23 8.32	8.42 8.60 8.77
D	98	889 83	80 78 75	73 71 69	67 64 63	61 59 57	110 104 99
X ₃	0.5746	0.5463	0.5205	0.4972	0.4759	0.4565	0.4388
	0.5648	0.5374	0.5125	0.4899	0.4692	0.4504	0.4278
	0.5554	0.5288	0.5047	0.4828	0.4628	0.4445	0.4174
Q	69 67 64	61 60 57	53 53 52	50 44 47	46 44 43	2444	72 69
X_1 D X_2 D X_3	0.3214	0.3414	0.3592	0.3752	0.3897	0.4030	0.4152
	0.3283	0.3475	0.3647	0.3802	0.3943	0.4072	0.4228
	0.3350	0.3535	0.3700	0.3850	0.3987	0.4112	0.4300
D	570	542	517	493	472	453	866
	560	533	509	486	466	448	845
	551	524	501	480	460	442	825
X1	I.4254	1.5935	1.7534	1.9061	2.0520	2.1918	2.3261
	I.4824	1.6477	1.8051	1.9554	2.0992	2.2371	2.4127
	I.5384	1.7010	1.8560	2.0040	2.1458	2.2819	2.4972
q	799	731	677	629	590	557	1043
	776	713	660	616	578	545	1009
	754	694	645	604	567	535	975
X ₀	4.4352	4.6681	4.8819	5.0801	5.2650	5.4385	5.6022
	4.5151	4.7412	4.9496	5.1430	5.3240	5.4942	5.7065
	4.5927	4.8125	5.0156	5.2046	5.3818	5.5487	5.8074
*	2004	20.00	8.0.0	3.5 3.3 5.3	8.88 4.80 5.6	23.8	0.2.4.

SUPPLEMENT TO TABLE I—Continued

D	. 17 16 16	15 15 14	15	14 13 13	13	12 11	111
X ₄	8.94 9.11 9.27	9.43 9.58 9.73	9.87 10.02 10.16	10.29 10.43 10.56	10.69 10.81 10.94	11.06 11.18 11.30	11.41
D		81 76 73	70 67 63	62 58 56	522	44 44 44	4 4 4 4
X ₃	0.4075 0.3982 0.3893	0.3809 0.3728 0.3652	0.3579 0.3509 0.3442	0.3379 0.3317 0.3259	0.3203 0.3148 0.3096	0.3046 0.2998 0.2951	0.2907 0.2863 0.2822
D	65 63 60	57 55 53	50 49 47	44 42	41 40 38	37 37 35	33
X_2	0.4369 0.4434 0.4497	0.4557 0.4614 0.4669	0.4722 0.4772 0.4821	0.4868 0.4914 0.4958	0.5000 0.5041 0.5081	0.5119 0.5156 0.5193	0.5228 0.5262 0.5295
	805 788 770	754 738 723	709 694 683	669 657 647	635 625 614	604 595 586	577 568 560
X_1 D	2.5797 2.6602 2.7390	2.8160 2.8914 2.9652	3.0375 3.1084 3.1778	3.2461 3.3130 3.3787	3.5069 3.5069 3.5694	3.6308 3.6912 3.7507	3.8093 3.8670 3.9238
D	944 917 890	866 844 821	801 781 763	748 730 713	701 686 673	660 645 637	625 613 603
X_0	5.9049 5.9993 6.0910	6.1800 6.2666 6.3510	6.4331 6.5132 6.5913	6.6676 6.7424 6.8154	6.8867 6.9568 7.0254	7.0927 7.1587 7.2232	7.2869 7.3494 7.4107
*	5.0	2.5.5 2.4.0	5.8 6.0 6.2	6.6 4.9 8.9	7.2	7.6 8.7 8.0	2.88.8.2

SUPPLEMENT TO TABLE I—Continued

1							1
Q	27	222 2	23 22 21	21 21 20	91 81	19 17 18	17 17 16
X4	11.75 11.86 12.13	12.38 12.63 12.87	13.10 13.33 13.55	13.76 13.97 14.18	14.38 14.57 14.76	14.94 15.13 15.30	15.48 15.65 15.82
D .	- 39 92 85	82.88	64 60 57	53 51 47	44 41	38 37 35	34 32 30
X_3	0.2781 0.2742 0.2650	0.2565 0.2485 0.2412	0.2343 0.2279 0.2219	0.2162 0.2109 0.2058	0.2011 0.1966 0.1923	0.1882 0.1844 0.1807	0.1772 0.1738 0.1706
D	31 75 71	62 59	56 53 51	44 45 45	42 40 39	37 36 35	33 33 31
X_2	0.5327 0.5358 0.5433	0.5504 0.5570 0.5632	0.5691 0.5747 0.5800	0.5851 0.5899 0.5945	0.5990 0.6032 0.6072	0.6111 0.6148 0.6184	0.6219 0.6252 0.6285
D	553 1347 1304	1263 1223 1189	1155 1125 1094	1068 1041 1016	995 972 952	931 913 895	878 860 845
X_1	3.9798 4.0351 4.1698	4.3002 4.4265 4.5488	4.6677 4.7832 4.8957	5.0051 5.1119 5.2160	5.3176 5.4171 5.5143	5.6095 5.7026 5.7939	5.8834 5.9712 6.0572
О	595 1442 1387	1339 1294 1250	1212 1173 1143	1110 1080 1052	1027 1002 980	954 938 916	898 878 862
X ₀	7.4710	7.8134 7.9473 8.0767	8.2017 8.3229 8.4402	8.5545 8.6655 8.7735	8.8787 8.9814 9.0816	9.1796 9.2750 9.3688	9.4604 9.5502 9.6380
×	8.8	10.0	11.5 12.0 12.5	13.0 13.5 14.0	14.5 15.0 15.5	16.0 16.5 17.0	17.5 18.0 18.5
	X_{0} D X_{1} D X_{2} D X_{3} D X_{4}	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	X_0 D X_1 D X_2 D X_3 D X_4 7.4710 595 3.9798 553 0.5358 75 0.2781 —39 11.75 7.5305 1142 4.0351 1347 0.5358 75 0.2742 92 11.86 7.8134 1387 4.1098 1304 0.5433 71 0.2650 85 12.13 7.9473 1294 4.4265 1223 0.5570 62 0.2485 73 12.63 8.0767 1250 4.4265 1189 0.5672 59 0.2412 69 12.87 8.2017 1212 4.6677 1185 0.5691 56 0.2485 73 12.63 8.3229 1173 4.7832 1125 0.5691 56 0.2279 60 13.33 8.4402 1143 4.7832 1125 0.5690 51 0.2279 60 13.35 8.6555 11080

SUPPLEMENT TO TABLE I—Continued

6.1417
6.2248
6.3064
6.3866
6.4655
6.5432
9619.9
6.6947
6.7688
6.8418
6.9137
6.9846
7.0544

TABLE II $K = I - (I - k)^{\frac{1}{2}}$

				- (-				
k	$\log K$	D	k	$\log K$	D .	k	$\log K$	D
0.60	9.56531	930	0.908	9.84304	206	0.979	9.93201	177
0.61	9.57461	922	0.910	9.84510	207	0.980	9.93378	181
0.62	9.58383	915	0.912	9.84717	209	0.981	9.93559	185
0.63	9.59298	908	0.914	9.84926	210	0.982	9.93744	189
0.64	9.60206	902	0.916	9.85136	212	0.983	9.93933	194
0.04	9.00200	902	0.920	9.03230		0,903	9.93933	194
0.65	9.61108	896	0.918	9.85348	213	0.984	9.94127	199
0.66	9.62004	890	0.920	9.85561	215	0.985	9.94326	205
0.67	9.62894	886	0.922	9.85776	217	0.986	9.94531	211
0.68	9.63780	882	0.924	9.85993	218	0.987	9.94742	219
0.69	9.64662	879	0.926	9.86211	221	0.988	9.94961	227
0.70	9.65541	875	0.928	9.86432	222	0.989	9.95188	236
0.71	9.66416	872	0.930	9.86654	224	0.990	9.95424	122
0.72	9.67288	871	0.932	9.86878	226	0.9905	9.95546	125
0.73	9.68159	869	0.934	9.87104	229	0.9910	9.95671	128
0.74	9.69028	869	0.936	9.87333	231	0.9915	9.95799	132
	3.13.11		70-	77000	-0-		7.70177	-0-
0.75	9.69897	869	0.938	9.87564	234	0.9920	9.95931	135
0.76	9.70766	869	0.940	9.87798	236	0.9925	9.96066	139
0.77	9.71635	871	0.942	9.88034	239	0.9930	9.96205	144
0.78	9.72506	873	0.944	9.88273	242	0.9935	9.96349	150
0.79	9.73379	877	0.946	9.88515	245	0.9940	9.96499	154
0.80	9.74256	880	0.948	9.88760	248	0.9945	9.96653	162
0.81	9.75136	886	0.950	9.89008	252	0.9950	9.96815	169
0.82	9.76022	893	0.952	9.89260	256	0.9955	9.96984	178
0.83	9.76915	900	0.954	9.89516	260	0.9960	9.97162	189
0.84	9.77815	454	0.956	9.89776	264	0.9965	9.97351	203
•),,,	101		, ,,,		// //	, ,,,,,,	
0.845	9.78269	456	0.958	9.90040	269	0.9970	9.97554	218
0.850	9.78725	459	0.960	9.90309	274	0.9975	9.97772	241
0.855	9.79184	462	0.962	9.90583	280	0.9980	9.98013	272
0.860	9.79646	465	0.964	9.90863	285	0.9985	9.98285	319
0.865	9.80111	469	0.966	9.91148	292	0.9990	9.98604	414
0.870	9.80580	473	0.968	9.91440	300	0.9995	9.99018	982
0.875	9.81053	478	0.970	9.91740	152	1.0000	0.00000	
0.880	9.81530	482	0.971	9.91892	155			
0.885	9.82012	488	0.972	9.92047	157			
0.890	9.82500	492	0.973	9.92204	160			
0.895	9.82992	500	0.974	9.92364	161			
0.900	9.83492	201	0.975	9.92525	165			
0.902	9.83693	202	0.976	9.92690	167			
0.904	9.83895	204	0.977	9.92857	170	(
0.906	9.84099	205	0.978	9.93027	174			
3.900	74-79	0		, , ,				
-								

TABLE III.—Giving the total work that dry gunpowder of the W. A. standard is capable of performing in the bore of a gun, in foot-tons per lb. of powder burned.¹

Number of volumes of expansion.	Corresponding density of products of combustion.	Total work per 1b. burned in foot-tons.	Difference.	Number of volumes of expansion.	Corresponding density of products of combustion.	Total work per 1b. burned in foot-tons.	Difference.
1.00 1.01 1.02 1.03 1.04 1.05 1.06 1.07 1.08 1.09 1.10 1.11 1.12 1.13 1.14 1.15 1.16 1.17 1.18 1.19 1.20 1.21 1.22 1.23 1.24 1.25 1.26 1.27 1.28 1.29 1.30 1.32 1.34 1.36 1.38 1.40 1.42 1.44 1.46 1.48 1.50 1.52 1.54	1.000 .990 .980 .971 .962 .952 .943 .935 .926 .917 .909 .901 .893 .885 .877 .870 .862 .855 .847 .840 .833 .826 .820 .813 .806 .800 .794 .787 .769 .758 .746 .735 .746 .735 .746 .746 .746 .746 .746 .746 .746 .746			1.56 1.58 1.60 1.62 1.64 1.66 1.68 1.70 1.72 1.74 1.76 1.78 1.80 1.82 1.84 1.88 1.90 1.92 1.94 1.96 1.98 2.00 2.05 2.10 2.15 2.20 2.25 2.30 2.45 2.55 2.40 2.45 2.55 2.60 2.65 2.70 2.75 2.80 2.85 2.90 2.85	.641 .633 .625 .617 .610 .602 .595 .588 .562 .555 .549 .543 .537 .532 .526 .521 .515 .505 .505 .488 .476 .465 .444 .444 .435 .425 .417 .408 .400 .384 .377 .370 .363 .357 .351 .345 .339 .333	34.500 35.301 36.086 36.855 37.608 38.346 39.069 39.778 40.474 41.156 41.827 42.486 43.133 43.769 44.394 45.009 45.614 46.209 46.795 47.372 47.940 48.499 49.050 50.383 51.673 52.922 54.132 55.304 56.439 57.539 58.605 59.639 60.642 61.616 62.563 63.486 64.385 65.262 66.119 66.955 67.771 68.568 69.347	.819 .801 .785 .769 .753 .738 .723 .709 .696 .682 .671 .636 .625 .615 .605 .595 .586 .577 .568 .577 .568 .559 .551 I .333 I .290 I .249 I .210 I .172 I .135 I .100 I .066 I .034 I .003 .947 .923 .899 .877 .836 .816 .797 .779
1 1			1	1		, , , ,	

¹ From Noble and Abel's "Researches on Fired Gunpowder."

1 processors and the second							
Number of volumes of expansion.	Corresponding density of products of combustion.	Total work per lb. burned in foot-tons.	Difference.	Number of volumes of expansion.	Corresponding density of products of combustion.	Total work per lb. burned in foot-tons.	Difference.
3.05 3.10 3.15 3.20 3.25 3.30 3.45 3.45 3.50 3.65 3.65 3.70 3.85 3.80 3.85 3.90 4.10 4.20 4.30 4.40 4.50 4.60 4.70 4.80 4.90 5.00 5.10 5.20 5.30 5.40 5.50 6.00 6.10 6.20 6.30 6.40 6.50 6.60	. 328 . 322 . 317 . 312 . 308 . 303 . 298 . 294 . 290 . 286 . 282 . 278 . 274 . 270 . 266 . 263 . 260 . 256 . 253 . 250 . 244 . 238 . 232 . 227 . 213 . 208 . 204 . 200 . 196 . 188 . 185 . 182 . 178 . 175 . 172 . 169 . 161 . 159 . 154 . 151	70.109 70.854 71.584 71.584 72.301 73.002 73.690 74.365 75.027 75.677 76.315 76.940 77.553 78.156 78.749 79.332 79.905 80.469 81.024 81.570 82.107 83.157 84.176 85.166 86.128 87.064 87.975 88.861 89.724 90.565 91.385 92.186 92.968 93.732 94.479 95.210 95.925 96.525 97.310 97.981 98.638 99.282 99.915 100.535 101.145 101.744 102.333	. 762 . 745 . 731 . 716 . 701 . 688 . 675 . 662 . 650 . 638 . 625 . 613 . 593 . 583 . 573 . 564 . 555 . 546 . 555 . 546 . 537 I . 050 I . 019 . 990 . 936 . 911 . 886 . 841 . 820 . 801 . 782 . 764 . 747 . 731 . 715 . 700 . 685 . 671 . 657 . 657 . 657 . 644 . 633 . 621 . 609 . 589	7.10 7.20 7.30 7.40 7.50 7.60 7.70 7.80 7.90 8.00 8.10 8.20 8.30 8.40 8.50 8.60 8.70 8.80 9.00 9.10 9.20 9.30 9.40 9.50 9.60 9.70 9.80 9.90 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 30	. 141 . 139 . 137 . 135 . 133 . 131 . 130 . 128 . 126 . 125 . 123 . 122 . 120 . 119 . 117 . 116 . 115 . 114 . 110 . 109 . 108 . 105 . 104 . 103 . 105 . 104 . 103 . 107 . 071 . 066 . 062 . 059 . 055 . 052 . 052 . 050 . 047 . 045 . 043 . 042 . 040 . 033	105.125 105.655 106.176 106.688 107.192 107.688 108.177 108.659 109.133 .109.600 110.514 110.962 111.404 111.840 112.270 112.695 113.114 113.528 113.937 114.341 114.739 115.133 115.521 115.905 116.284 116.659 117.029 117.395 117.757 121.165 124.239 127.036 129.602 131.970 134.168 138.138 139.944 141.647 143.258 144.788 146.242 147.689 148.953 154.800	.539 .530 .531 .512 .504 .496 .489 .482 .474 .460 .454 .448 .442 .436 .436 .425 .419 .414 .409 .404 .398 .394 .388 .394 .379 .375 .376 .362 3.408 3.074 2.797 2.566 2.368 2.19
6.70 6.80 6.90 7.00	. 149 . 147 . 145 . 143	102.912 103.480 104.038 104.586	· 579 · 568 · 558 · 548	35 40 45 50	.028 .025 .022 .020	159.667 163.828 167.456 170.671	4.867 4.161 3.628 3.215



WORKS CONSULTED

HUTTON: "Mathematical Tracts," London, 1812.

RUMFORD: "Experiments to Determine the Force of Fired Gunpowder," London, 1797.

RODMAN: "Experiments on Metal and Cannon and Qualities of Cannon Powder," Boston, 1861.

NOBLE AND ABEL: "Researches on Explosives," London, 1874, 1879.

Noble: "On the Energy Absorbed by Friction in the Bores of Rifled Guns."
Reprinted as "Ordnance Construction Note," No. 60. "On Methods that have been Adopted for Measuring Pressures in the Bores of Guns," London, 1894. "Researches on Explosives." Preliminary Note, London, 1894.

Official: "English Text-book of Gunnery." Editions of 1897 and 1902.

SARRAU: "Recherches sur les effets de la poudre dans les Armes," and "Formules pratiques des vitesses et des pressions dans les Armes." A translation of these memoirs into English by Lieutenants Meigs and Ingersoll is given in Vol. X of the Proceedings U. S. Naval Institute. "Recherches théoriques sur le chargement des bouches à feu." Translation by Lieutenant Howard, O. D., in "Ordnance Construction Note," No. 42. "War Powders and Interior Ballistics." A translation by Lieutenant Charles B. Wheeler, O. D., as "Notes on the Construction of Ordnance," No. 67. Washington, 1895.

Gossot and Liouville: "The Ballistic Effects of Smokeless Powders in Guns." Translated by Major Charles B. Wheeler, O. D. "Notes on the Construction of Ordnance," No. 88. Washington, 1906.

Dunn: "Interior Ballistics." Part I. "Notes on the Construction of Ordnance," No. 89. Washington, 1906. Souich: "Poudres de Guerre. Balistique Intérieur," Paris, 1882.

BAILLS: "Traité de Balistique Rationnelle," Paris, 1883.

Longridge: "Internal Ballistics," London, 1889.

Meigs and Ingersoll: "Interior Ballistics," Annapolis, 1887.

PASHKIEVITSCH: "Interior Ballistics." Translated from the Russian by Captain Tasker H. Bliss, A. D. C. Washington, 1892.

Bergman: "Lärobok i Artilleriteknik." Del I. Krutlära. Stockholm, 1908. A part of this work was translated for the author by Colonel Lundeen, Coast Artillery Corps, U. S. Army.

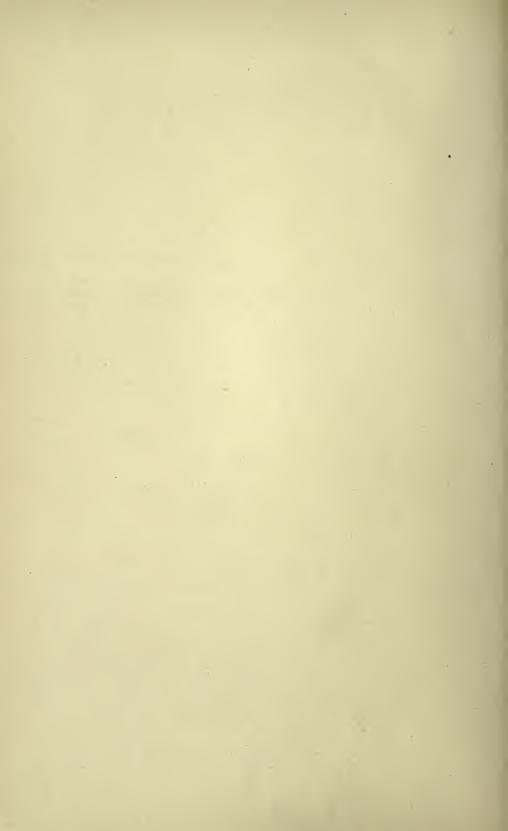
GLENNON: "Velocities and Pressures in Guns," Annapolis, 1889.

CROZIER: "On the Rifling of Guns." Ordnance Construction. Note No. 49. A. W.: "Des Armes de guerre Modernes et de leurs Munitions." Revue Militaire Belge, Vol. II, 1888.

McCulloch: "Mechanical Theory of Heat," New York, 1876.

Peabody: "Thermodynamics of the Steam Engine," New York, 1889.
Rontgen: "The Principles of Thermodynamics." Translated from the German by Professor A. Jay Du Boise. New York, 1889.

LISSAK: "Ordnance and Gunnery," New York, 1907. Encyclopædia Britannica, eleventh edition, 1911.



Absolute temperature, definition of, 17; of fired gunpowder, 40, 47.

Adiabatic expansion, definition of, 26.

Air space, initial, definition of, 76; expressions for reduced length of, 76, 77, 92, 94.

Angular acceleration, 174.

Applications of velocity and pressure formulas: To magazine rifle, 130 to 134; to Hotchkiss 57 mm. rapid-firing gun, 125 to 130; to 6-inch English gun, 115 to 125 and 140 to 147; to 6-inch Brown wire gun, 150 to 162; to 8-inch rifle, 102 to 110; to 10-inch rifle, 179; to 14-inch rifle, 162 to 169 and 184; to hypothetical 7-inch gun, 111, 136.

Artillery circulars M and N, references to, 88, 135, 186.

Atmospheric pressure, value of, 21.

Axite, form-characteristics of, 61.

Ballistic pendulum, 2, 3.

Ballistite, 61, 140.

Binomial formulas for velocity and pressure, 112.

Bliss, Captain Tasker H., 53.

Board of Ordnance, reference to, 151.

Boyle, Robert, 15.

B N powders, form-characteristics of, 61; computation of by velocity formula, 129.

CAVALLI, reference to, 8.

Centervall, law of combustion, 79.

Chamber, reduced length of, 31; alignment of grains in, 71; effect of varying volume of, 111, 112, 166.

Characteristic equation of gaseous state, 17.

Characteristics of a powder, 94.

Charge of powder, behavior of when ignited in a gun, 12, 13; in a close vessel, 12, 35; initial surface of, 73, 77.

Chase, excessive pressure in, 55, 158.

Chevreul, reference to, 8.

Chronograph, Noble's, 116; Boulengé-Breger, 126.

Coefficient of expansion of a perfect gas, 16.

Combustion of a grain of powder, 11; under constant pressure, 55, 79; under variable pressure, 79, 80.

Composition: of gunpowder, 1; of cordite, 11, 117, 124; magazine rifle powder, 131; ballistite, 140.

Constants, physical, adopted, 92, 94.

Cordite, composition of, 11, 117, 124; form-characteristics of, 63.

Cube, form-characteristics of, 61.

Cylindrical grains: solid, 63; with axial perforation, 65; with seven perforations (m.p. grains), 66 to 72.

D'ARCY's method of experimenting, 4.

Density: of powder, 11; of a gas, 21; of loading, 37, 75, 77.

Dulong and Petit, law of, 21, 23.

ELSWICK works, mention of, 115.

Encyclopædia Britannica, eleventh edition, reference to, 115.

Energies neglected in deducing equation for velocity, 121.

Energy of translation of projectile, 32, 51, 52, 53, 80, 144.

English Text-Book of Gunnery, reference to, 115.

Euler, mention of, 88; equations of, 174.

Examples: of expansion of gases, 28; of the formulas of Chapter III, 77; relating to 8-inch rifle, 109; to 6-inch gun, 122, 124, 144, 155; to 14-inch rifle, 166.

Expansion, work of: isothermal, 25; adiabatic, 26; in the bore of a gun, 30, 47.

FACTOR of effect, 49, 52, 54.

Force of the powder, 33, 36.

Formulas: Characteristic equation of gaseous state, 17. For specific heat under constant volume, 22. For work: of an isothermal expansion, 25; adiabatic expansion, 26, 27, 32; of gases of fired gunpowder, 49. For temperature: of an adiabatic expansion of a perfect gas, 26, 27; of gases of fired gunpowder, 47. For pressure: isothermal, 15, 17; adiabatic, 27; gases of fired gunpowder in close vessels, 6, 36, 39; in guns, 45. For pressure in guns with smokeless powders: While powder is burning, 85, 86, 101, 106,

112, 140, 152; after powder is all burned, 86, 102. Maximum pressure, 91, 101, 106. Initial pressure when powder is all burned before projectile moves, 86, 87, 93, 94, 99. For velocity of projectile in guns with smokeless powders: while powder is burning, 83, 84, 89, 100, 101, 103, 112; after powder is all burned, 84, 85, 102. For limiting velocity, 85, 93, 94, 98, 101, 127, 141, 146, 150, 164. For computing f, 32, 36, 84, 93, 94, 96, 97, 106, 164. For v_c , 91, 92, 93, 94. For k and k', 58, 59, 61, 62, 63, 65, 68, 69, 72, 89, 90, 109, 136, 141, 148. For M, M', N and N', 84, 85, 93, 94, 95, 97, 101, 102, 106, 112, 113, 114, 115, 118, 136, 139, 150, 155, 161, 165. For y, 39, 83, 93, 132, 146. For α , λ , μ , 58, 59, 60, 61, 62, 63, 64, 65, 67, 68, 69, 84, 129, 134. For P', 86, 87, 93, 94, 98. For the X functions, 87, 88, 89, For \overline{X}_0 or X_0' , 84, 93, 94, 119, 136, 149, 150, 152, 164. Working formulas, 77, 93, 94, 95, 97. For inclination of groove, 171, 172, 177. For pressure on lands, 175, 176, 180, 181, 185. For semi-cubical parabola, 176. Common parabola, 178.

Frankford arsenal, mentioned, 130.

Gas, perfect, 17.

Gay-Lussac, law of, 16; mentioned, 8.

Gossot, Colonel F., law of combustion, 79; igniter, 151.

Graham, mentioned, 8.

Grains of powder, combustion of under constant pressure, 55; vanishing-surface, 56; volume burned, 57; form-characteristics, 58; their relation to each other, 58, 59.

Granulation, 151, 163.

Groove, developed, 171; width of, 179.

Gun-cotton, 10, 11.

Gunpowder, 1, 2.

HAMILTON, Captain Alston, length of m.p. grains, 71.

Heat: mechanical equivalent of, 18; specific heats, 18, 19, 21, 22.

Hugoniot, law of combustion, 79.

Hutton, Dr. Charles, experiments with gunpowder, 3, 4.

Inflammation of a grain and charge of powder, 11, 12, 13. Isothermal expansion, 25.

JOURNAL U. S. Artillery, references to, 71, 79, 125, 148.

LANDS, width of, 179.

Lenk, General von, experiments with gun-cotton, 10.

Liouville, R., law of combustion, 79.

Lissak, Colonel O. M., ordnance and gunnery, 29; construction of velocity and pressure curves, 144.

Longridge, Atkinson, loss of energy in gun, 53.

MAGAZINE rifle, description of, 130.

Marriotte, law of, 15.

Maximum pressure in a gun, 90, 91

Maximum value of X_3 , 90, 101.

Mayevski, mention of, 8.

Monomial formulas, 100.

Muzzle velocities and pressures, computed, 107, 120, 123, 124, 129, 130, 133, 134, 143, 161, 169.

NATURE, reference to, 115.

Neumann, mentioned, 8.

Nobel, N. K. powder, law of combustion for, 79.

Noble and Abel, experiments with fired gunpowder in close vessels, and deductions therefrom, 33 to 54.

Noble, Sir Andrew, experiments with 6-inch gun, 115; coefficient of friction, 180.

Notation, 15, 17, 19, 23, 24, 31, 51, 56, 58, 60, 67, 72, 74, 75, 76, 79, 80, 81, 82, 83, 84, 85, 86, 91, 92, 108, 149, 171, 172, 174, 176.

Notes on the construction of ordnance, reference to, 102.

Ordnance Department, reference to, 130, 162, 166.

Otto, mentioned, 8.

PARALLELOPIPEDON, form-characteristics of, 60.

Pashkievitsch, Colonel, lost work in a gun, 53.

Piobert, mentioned, 8.

Point of inflection of X_3 , 101.

Powder grains. See Grains of powder.

Powder, smokeless. See Composition.

Pressure: of fired gunpowder in close vessels, 6, 7, 9, 35, 37; in guns, 41.

RADIUS of gyration of projectile, 174, 180, 186.

Retarding effect of uniform twist, 186.

INDEX 22I

Rifling of cannon, advantages of, 170.

Robins, Benjamin, experiments with fired gunpowder, 2.

Rodman, Captain T. J., experiments with fired gunpowder, 8; perforated grains, 9; cutter gauge, 9.

Rumford, Count, experiments with fired gunpowder, 4; comparison of results with those of Noble and Abel, 6.

SAINTE-ROBERT, Count de, law of combustion, 79.

Sandy Hook, mention of, 151, 160.

Sarrau, E., law of combustion, 80; monomial formula for pressure in a gun, 95. Schönbein of Basel, discoverer of gun-cotton, 10.

Sebert, law of combustion, 79.

Spherical grains, form-characteristics of, 59.

Springfield Armory, mentioned, 130, 131.

Tables, in text: of specific heats of certain gases, 22; of pressures in guns of fired gunpowder, 46; of velocities and pressures in guns, 104, 107, 129, 130, 133, 134, 143, 161, 165, 169, 183, 185; of pressure on lands, 183, 185.

Temperature of fired gunpowder, 45.

Trinomial formulas, 138.

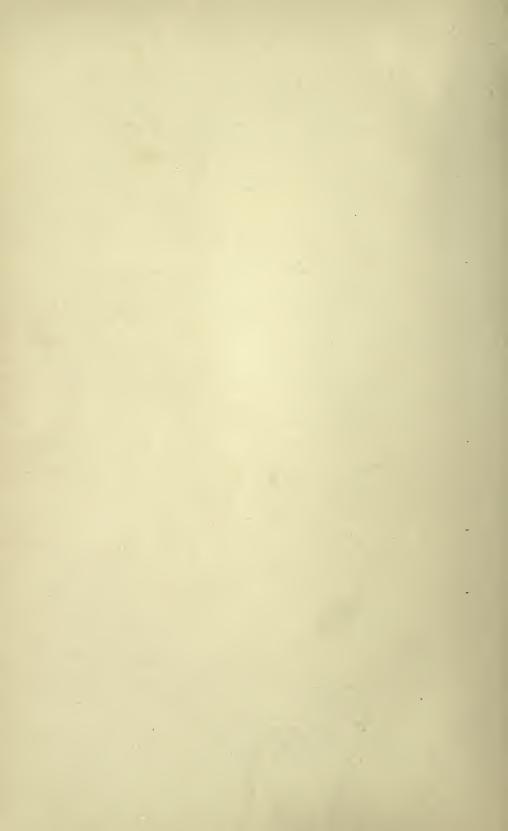
Twist, uniform, 171; increasing, 171.

VIEILLE, law of combustion, 79.

WEAVER, General E. M., Notes on explosives, referred to, 11.

Work of fired gunpowder, 47.

Working formulas, 77, 92, 181, 185.



SHORT-TITLE CATALOGUE

OF THE

PUBLICATIONS

OF

JOHN WILEY & SONS

NEW YORK

LONDON: CHAPMAN & HALL, LIMITED

ARRANGED UNDER SUBJECTS

Descriptive circulars sent on application. Books marked with an asterisk (*) are sold at net prices only. All books are bound in cloth unless otherwise stated.

AGRICULTURE—HORTICULTURE—FORESTRY.

Armsby's Principles of Animal Nutrition8vo,	\$4	00
* Bowman's Forest Physiography8vo,	5	
Budd and Hansen's American Horticultural Manual:		
Part I. Propagation, Culture, and Improvement	1	50
Part II. Systematic Pomology	1	50
Elliott's Engineering for Land Drainage	_	50
Practical Farm Drainage. (Second Edition, Rewritten.)12mo,	_	50
Graves's Forest Mensuration	_	00
* Principles of Handling WoodlandsLarge 12mo,	_	50
Green's Principles of American Forestry	_	50
Grotenfelt's Principles of Modern Dairy Practice. (Woll.)12mo,	_	00
* Herrick's Denatured or Industrial Alcohol		00
Holm's Milk in Denmark. (In Press.)	4	00
*Kemp and Waugh's Landscape Gardening. (New Edition, Rewritten.) 12mo.	-	=0
	_	50
* McKay and Larsen's Principles and Practice of Butter-making8vo,	_	50
Maynard's Landscape Gardening as Applied to Home Decoration12mo,	_	50
Sanderson's Insects Injurious to Staple Crops	1	50
Insect Pests of Farm, Garden, and Orchard. (In Press.)		
* Schwarz's Longleaf Pine in Virgin Forest12mo,		25
* Solotaroff's Field Book for Street-tree Mapping12mo,	0	75
In lots of one dozen	8	00
* Shade Trees in Towns and Cities8vo,	3	00
Stockbridge's Rocks and Soils	2	50
Winton's Microscopy of Vegetable Foods8vo,	7	50
Woll's Handbook for Farmers and Dairymen16mo,	1	50

ARCHITECTURE.

Atkinson's Orientation of Buildings or Planning for Sunlight. (In Press.)	
Baldwin's Steam Heating for Buildings12mo,	2 50
Berg's Buildings and Structures of American Railroads4to.	5 00

Birkmire's Architectural Iron and Steel		
Compound Riveted Girders as Applied in Buildings8vo, Planning and Construction of American Theatres8vo,		00
Planning and Construction of American Theatres		50
Skeleton Construction in Buildings8vo,		00
Briggs's Modern American School Buildings8vo,		00
Byrne's Inspection of Materials and Workmanship Employed in Construction.	٠,٠	00
16mo.	3	00
Carpenter's Heating and Ventilating of Buildings8vo,		00
* Corthell's Allowable Pressure on Deep Foundations	1	25
Eckel's Building Stones and Clays. (In Press.)		
Preitag's Architectural Engineering8vo,	3	50
Fire Prevention and Fire Protection. (In Press.)		
Fireproofing of Steel Buildings8vo,	2	50
Gerhard's Guide to Sanitary Inspections. (Fourth Edition, Entirely Re-		
vised and Enlarged.)		50
* Modern Baths and Bath Houses		00
Theatre Fires and Panics	_	50 50
* The Water Supply, Sewerage and Plumbing of Modern City Buildings,	T	90
8vo.	А	00
Johnson's Statics by Algebraic and Graphic Methods8vo,		00
Kellaway's How to Lay Out Suburban Home Grounds8vo,		00
Kidder's Architects' and Builders' Pocket-book16mo, mor.,		00
Merrill's Stones for Building and Decoration8vo,		00
Monckton's Stair-building4to,	4	00
Patton's Practical Treatise on Foundations8vo,	5	00
Peabody's Naval Architecture8vc,		50
Rice's Concrete-block Manufacture8vo,		00
Richey's Handbook for Superintendents of Construction 16mo, mor.		00
Building Foreman's Pocket Book and Ready Reference. 16mo, mor.	5	00
* Building Mechanics' Ready Reference Series: * Carpenters' and Woodworkers' Edition16mo. mor.	4	F O
* Cement Workers' and Plasterers' Edition16mo, mor.		50 50
* Plumbers', Steam-Fitters', and Tinners' Edition16mo, mor.		50
* Stone- and Brick-masons' Edition16mo, mor.	_	50
Sabin's House Painting		00
Siebert and Biggin's Modern Stone-cutting and Masonry8vo.		50
Snow's Principal Species of Wood8vo,	3	50
Wait's Engineering and Architectural Jurisprudence8vo,	6	00
Sheep,		50
Law of Contracts8vo,	3	00
Law of Operations Preliminary to Construction in Engineering and	-	
Architecture8vo,		00
Wilson's Air Conditioning	_	50
Wilson's Air Conditioning	1	50
Suggestions for Hospital Architecture, with Plans for a Small		
Hospital12mo,	1	25
	1	20
ARMY AND NAVY.		
Bernadou's Smokeless Powder Nitro-cellulose and the Theory of the Cellu		

Bernadou's Smokeless Powder, Nitro-cellulose, and the Theory of the Cellu-	
lose Molecule12mo,	2 50
Chase's Art of Pattern Making	2 50
Screw Propellers and Marine Propulsion	3 00
* Cloke's Enlisted Specialists' Examiner8vo,	2 00
*Gunner's Examiner8vo,	1 50
Craig's Azimuth4tu,	3 50
Crehore and Squier's Polarizing Photo-chronograph8vo.	3 00
* Davis's Elements of Law8vo,	2 50
* Treatise on the Military Law of United States8vo,	7 00
* Dudley's Military Law and the Procedure of Courts-martial Large 12mo	2 50
Durand's Resistance and Propulsion of Ships8vo.	5 00

* Dyer's Handbook of Light Artillery12mo,	\$3 00
Eissler's Modern High Explosives	4 00
* Fiebeger's Text-book on Field FortificationLarge 12mo,	2 00
Hamilton and Bond's The Gunner's Catechism18mo,	1 00
* Hoff's Elementary Naval Tactics8vo,	1 50
Ingalls's Handbook of Problems in Direct Fire8vo,	4 00
Interior Ballistics. (In Press.)	
* Lissak's Ordnance and Gunnery8vo,	6 00
* Ludlow's Logarithmic and Trigonometric Tables8vo,	1 00
* Lyons's Treatise on Electromagnetic Phenomena. Vols. I. and II8vo, each,	6 00
* Mahan's Permanent Fortifications. (Mercur)8vo, half mor.	7 50
* Mahan's Permanent Fortifications. (Mercur.)8vo, half mor. Manual for Courts-martial16mo, mor.	1 50
* Mercur's Attack of Fortified Places	2 00
* Elements of the Art of War8vo,	4 00
Nixon's Adjutants' Manual	1 00
Peabody's Naval Architecture8vo,	7 50
* Phelps's Practical Marine Surveying8vo,	2 50
Putnam's Nautical Charts 8vo	2 00
Rust's Ex-meridian Altitude, Azimuth and Star-Finding Tables8vo,	5 00
* Selkirk's Catechism of Manual of Guard Duty24mo,	50
Sharpe's Art of Subsisting Armies in War	1 50
* Taylor's Speed and Power of Ships. 2 vols. Text 8vo, plates oblong 4to,	7 50
* Tupes and Poole's Manual of Bayonet Exercises and Musketry Fencing.	1 00
	50
* Weaver's Military Explosives	
* Woodhull's Military Hygiene for Officers of the Line Large 12mo,	3 00
woodnun's winterly fryglene for Officers of the LineLarge 12mo,	1 50
ASSAYING.	
Dult IDAL t The LL	
Betts's Lead Refining by Electrolysis8vo,	4 00
*Butler's Handbook of Blowpipe Analysis	75
Fletcher's Practical Instructions in Quantitative Assaying with the Blowpipe.	
16mo, mor.	1 50
Furman and Pardoe's Manual of Practical Assaying8vo,	3 00
Lodge's Notes on Assaying and Metallurgical Laboratory Experiments8vo,	3 00
Low's Technical Methods of Ore Analysis8vo,	3 00
Miller's Cyanide Process	1 00
Manual of Assaving	1 00
Minet's Production of Aluminum and its Industrial Use. (Waldo.)12mo,	2 50
Ricketts and Miller's Notes on Assaying8vo.	3 00
Robine and Lenglen's Cyanide Industry. (Le Clerc.)8vc,	4 00
* Seamon's Manual for Assayers and ChemistsLarge 12mo,	2 50
Ulke's Modern Electrolytic Copper Refining8vo.	3 00
Wilson's Chlorination Process	1 50
Cyanide Processes	1 50
yamaa 11000303	1 00
ASTRONOMY.	
	0.50
Comstock's Field Astronomy for Engineers8vo,	2 50
Comstock's Field Astronomy for Engineers	3 50
Comstock's Field Astronomy for Engineers	3 50 3 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00
Comstock's Field Astronomy for Engineers. 8vo, Craig's Azimuth. 4to, Crandall's Text-book on Geodesy and Least Squares. 8vo, Doolittle's Treatise on Practical Astronomy. 8vo, Hayford's Text-book of Geodetic Astronomy. 8vo, Hosmer's Azimuth. 16mp, mor. * Text-book on Practical Astronomy. 8vo,	3 50 3 00 4 00 3 00 1 00 2 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00 2 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00 2 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00 2 00
Comstock's Field Astronomy for Engineers	3 50 3 00 4 00 3 00 1 00 2 00 2 50 3 00 5 00 2 00

Allen's Tables for Iron Analysis	\$3	00
Armsby's Principles of Animal Nutrition	4	
Arnold's Compendium of Chemistry. (Mandel.)	3	50
Association of State and National Food and Dairy Departments, Hartford	3	00.
Meeting, 1906		00
Jamestown Meeting, 1907 8vo, Austen's Notes for Chemical Students		50
Baskerville's Chemical Elements. (In Preparation.)	•	00
Bernadou's Smokeless Powder.—Nitro-cellulose, and Theory of the Cellulose		
Molecule	2	50
Molecule	1	25.
Laboratory Methods of Inorganic Chemistry. (Hall and Blanchard.)		
8vo,	3	00
* Bingham and White's Laboratory Manual of Inorganic Chemistry12mo.	1	00
* Blanchard's Synthetic Inorganic Chemistry12mo,	1	00.
Bottler's Varnish Making. (Sabin.) (In Press.) * Browning's Introduction to the Rarer Elements		
* Browning's Introduction to the Rarer Elements8vo,	1	50
* Butler's Handbook of Blowpipe Analysis		75
* Claassen's Beet-sugar Manufacture. (Hall and Rolle.)		00
Classen's Quantitative Chemical Analysis by Electrolysis. (Boltwood.).8vo, Cohn's Indicators and Test-papers		00
Cohn's Indicators and Test-papers12mo,		00·
Tests and Reagents	0	UU
* Danneel's Electrochemistry. (Merriam.)	1	25.
Dannerth's Methods of Textile Chemistry		00
Duham's Thermodynamics and Chemistry (Burgess)		00
Duhem's Thermodynamics and Chemistry. (Burgess.). 8vo, Effront's Enzymes and their Applications. (Prescott.). 8vo,		00
Eissler's Modern High Explosives8vo.	4	00.
* Fischer's Oedema		00
* Physiology of AlimentationLarge 12mo,		00-
Fletcher's Practical Instructions in Quantitative Assaying with the Blowpipe.		
16mo, mor.	1	$50 \cdot$
Fowler's Sewage Works Analyses12mo,	2	00.
Towler's bewage works maryses	-	
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)8vo,	5	00.
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)8vo,	5	$\begin{array}{c} 00 \cdot \\ 00 \cdot \end{array}$
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)8vo,	5	00.
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)8vo, Manual of Qualitative Chemical Analysis. Part I. Descriptive. (Wells.)8vc Quantitative Chemical Analysis. (Cohn.) 2 vols8vc, When Sold Separately, Vol. I, \$6. Vol. II, \$8. Fuertes's Water and Public Health	5	$\begin{array}{c} 00 \cdot \\ 00 \cdot \end{array}$
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5 0, 3 12 1 3	00. 00. 50. 50.
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5 12 1 1 3 2	00- 00- 50- 50- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5 0, 3 12 1 3	00. 00. 50. 50.
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5 , 3 12 1 3 2 1	00. 00. 50. 00. 00. 25.
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5 , 3 12 1 3 2 1	50 50 00 25
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5, 3, 12 1, 3, 2, 1 1, 2	00 50 50 00 00 25
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5, 3 12 1 3 2 1 1 2	00 50 50 00 00 25
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5 3 12 1 3 2 1 1 2 1 4	00 50 50 00 25 25 00 25
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5 5,3 12 1 3 2 1 1 2 1 4 5	00 00 50 00 00 25 25 00 00
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2	00- 00- 50- 00- 00- 25- 00- 00- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4	00- 00- 50- 50- 00- 25- 00- 25- 00- 00- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3	00- 00- 50- 50- 00- 25- 25- 00- 00- 00- 00- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4	00- 00- 50- 50- 00- 25- 00- 25- 00- 00- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3	00- 00- 50- 50- 00- 25- 25- 00- 00- 00- 00- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1	00- 00- 50- 50- 00- 00- 25- 00- 00- 00- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1	00- 00- 50- 50- 00- 00- 25- 00- 00- 00- 00- 00- 00-
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 2	00 00 50 50 00 00 25 25 00 00 00 00 00 00 50 50
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3	00 00 50 50 00 25 25 00 00 00 00 00 50 50
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	00- 00- 50- 50- 00- 25- 25- 00- 00- 00- 00- 50- 50- 00- 00- 00- 0
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3	00 00 50 50 00 25 25 00 00 00 00 00 50 50
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3 3 1	00 00 50 50 00 00 25 00 00 00 00 00 00 00 50 50 50
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3 3 1 3	00 00 50 50 00 00 25 25 00 00 00 00 00 50 50 00 25
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3 3 1 3	00 00 50 50 00 00 25 00 00 00 00 00 00 00 50 50 50
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3 3 1 3 3	00 00 50 50 00 25 25 00 00 00 00 00 50 50 00 25
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3 3 1 3 3	00 00 50 50 00 00 25 25 00 00 00 00 00 50 50 00 25
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 32 1 12 1 45 2 24 3 31 1 22 3 31 1 33	00- 00- 50- 50- 00- 25- 00- 25- 00- 00- 00- 00- 50- 50- 00- 00- 25- 00- 00- 00- 00- 00- 00- 00- 00- 00- 0
Fresenius's Manual of Qualitative Chemical Analysis. (Wells.)	5,3 12 1 3 2 1 1 2 1 4 5 2 4 3 1 1 2 2 3 3 1 7	00 00 50 50 00 25 25 00 00 00 00 00 50 50 00 25

		00
Low's Technical Method of Ore Analysis8vo,		00
Lowe's Paint for Steel Structures		00
* McKay and Larsen's Principles and Practice of Butter-making 8vo,	1	50
Maire's Modern Pigments and their Vehicles		00
Mandel's Handbook for Bio-chemical Laboratory	1	
Mandel's Handbook for Bio-chemical Laboratory	-	00
12mo,	0	60
Mason's Examination of Water. (Chemical and Bacteriological.)12mo,	1	25
Water-supply. (Considered Principally from a Sanitary Standpoint.)		
8vo,		00
* Mathewson's First Principles of Chemical Theory8vo,		00
Matthews's Laboratory Manual of Dyeing and Textile Chemistry8vo,		50
Textile Fibres. 2d Edition, Rewritten	4	00
* Meyer's Determination of Radicles in Carbon Compounds. (Tingle.) Third Edition	1	25
Miller's Cyanide Process		00
Miller's Cyanide Process. 12mo, Manual of Assaying. 12mo,		00
Minet's Production of Aluminum and its Industrial Use. (Waldo.)12mo,		50
* Mittelstaedt's Technical Calculations for Sugar Works. (Bourbakis.) 12mo,	1	50
Mixter's Elementary Text-book of Chemistry12mo,	1	50
Morgan's Elements of Physical Chemistry12mo,		00
* Physical Chemistry for Electrical Engineers12mo,		50
* Moore's Experiments in Organic Chemistry12mo,		50
* Outlines of Organic Chemistry		50
* Muir's History of Chemical Theories and Laws		50
Mulliken's General Method for the Identification of Pure Organic Compounds.	-	00
Vol. I. Compounds of Carbon with Hydrogen and Oxygen. Large 8vo,	5	00
Vol. II. Nitrogenous Compounds. (In Preparation.)		
Vol. III. The Commercial DyestuffsLarge 8vo,	5	00
* Nelson's Analysis of Drugs and Medicines		00
Ostwald's Conversations on Chemistry. Part One. (Ramsey.)12mo,		50
Part Two. (Turnbull.)12mo,		00
* Introduction to Chemistry. (Hall and Williams.)Large 12mo,		50
Owen and Standage's Dyeing and Cleaning of Textile Fabrics12mo,		00
* Palmer's Practical Test Book of Chemistry	1	25
Penfield's Tables of Minerals, Including the Use of Minerals and Statistics	-	20
of Domestic Production8vo.	1	00
of Domestic Production	5	00
Poole's Calorific Power of Fuels8vo,	3	00
Prescott and Winslow's Elements of Water Bacteriology, with Special Refer-		
ence to Sanitary Water Analysis12mo,	1	
* Reisig's Guide to Piece-Dyeing	25	00
	9	00
point		00
Rideal's Disinfection and the Preservation of Food8vo,		00
Riggs's Elementary Manual for the Chemical Laboratory8vo,	1	
Robine and Lenglen's Cyanide Industry. (Le Clerc.)8vo,	4	
Ruddiman's Incompatibilities in Prescriptions8vo,		
Whys in Pharmacy12mo,	1	
* Ruer's Elements of Metallography. (Mathewson.)	3	
Sabin's Industrial and Artistic Technology of Paint and Varnish 8vo,	3	00· 50·
Salkowski's Physiological and Pathological Chemistry. (Orndorff.)8vo, * Schimpf's Essentials of Volumetric AnalysisLarge 12mo,		50
Manual of Volumetric Analysis (Fifth Edition Rewritten). 8vo		00
Manual of Volumetric Analysis. (Fifth Edition, Rewritten)8vo, * Qualitative Chemical Analysis8vo,	1	
* Seamon's Manual for Assayers and ChemistsLarge 12mo,	2	50
Smith's Lecture Notes on Chemistry for Dental Students8vo,		50
Spencer's Handbook for Cane Sugar Manufacturers16mo, mor.		00
Handbook for Chemists of Beet-sugar Houses16mo, mor.	3	00
Stockbridge's Rocks and Soils8vo,	2	50
Stone's Practical Testing of Gas and Gas Meters8vo,		50
* Tillman's Descriptive General Chemistry8vo,	0	00

* Tillman's Elementary Lessons in Heat	\$1 50 3 00 4 00 5 00 1 50 3 00 4 00 5 00 2 00 3 00 1 50 1 50 1 50 1 25 3 50 1 50 7 50 3 00
CIVIL ENGINEEDING	
CIVIL ENGINEERING.	
BRIDGES AND ROOFS. HYDRAULICS. MATERIALS OF ENGIN ING. RAILWAY ENGINEERING.	EER-
* American Civil Engineers' Pocket Book. (Mansfield Merriman, Editor-	
in-chief.)16mo, mor.	5 00
Baker's Engineers' Surveying Instruments	3 00 25
tary Surveying	3 00
Vol. II. Higher Surveying8vo,	2 50
* Burr's Ancient and Modern Engineering and the Isthmian Canal8vo,	3 50
Comstock's Field Astronomy for Engineers	2 50 1 25
Crandall's Text-book on Geodesy and Least Squares	3 00
Davis's Elevation and Stadia Tables8vo,	1 00
Elliott's Engineering for Land Drainage	1 50
* Fiebeger's Treatise on Civil Engineering	5 00
Folwell's Sewerage. (Designing and Maintenance.)	3 00
Freitag's Architectural Engineering8vo,	3 50
French and Ives's Stereotomy8vo,	2 50
Gilbert, Wightman, and Saunders's Subways and Tunnels of New York.	
(In Press.) * Hauch and Rice's Tables of Quantities for Preliminary Estimates12mo,	1 25
Hayford's Text-book of Geodetic Astronomy8vo,	3 00
Hering's Ready Reference Tables (Conversion Factors.)16mo, mor.	2 50
Hosmer's Azimuth16mo, mor.	1 00
* Text-book on Practical Astronomy	$\begin{array}{cccc} 2 & 00 \\ 1 & 25 \end{array}$
* Ives's Adjustments of the Engineer's Transit and Level 16mo, bds.	25
Ives and Hilts's Problems in Surveying, Railroad Surveying and Geod-	
esy	1 50
* Johnson (J.B.) and Smith's Theory and Practice of Surveying. Large 12mo,	3 50
Johnson's (L. J.) Statics by Algebraic and Graphic Methods	$\frac{2}{3} \frac{00}{00}$
* Mahan's Descriptive Geometry8vo.	1 50
Merriman's Elements of Precise Surveying and Geodesy	2 50
Merriman and Brooks's Handbook for Surveyors	2 00
Orden's Sewer Construction	3 50
Sewer Design	2 00
Parsons's Disposal of Municipal Refuse	2 00
	7 50
6	

Reed's Topographical Drawing and Sketching4to, Riemer's Shaft-sinking under Difficult Conditions. (Corning and Peele.).8vo.	\$5 00	
Riemer's Shaft-sinking under Difficult Conditions. (Corning and Peele.).8vo.	3 00	
Siebert and Biggin's Modern Stone-cutting and Masonry8vo,	1 50	
Smith's Manual of Topographical Drawing. (McMillan.)8vo,	2 50	
Smith's Manual of Topographical Drawing. (McMillan.)8vo, Soper's Air and Ventilation of Subways12mo,	2 50	
* Tracy's Exercises in Surveying	1 00	
Tracy's Plane Surveying	3 00	
Venable's Garbage Crematories in America	2 00	
Methods and Devices for Bacterial Treatment of Sewage8vo,	3 00	
Wait's Engineering and Architectural Jurisprudence8vo,	6 00	
Sheep,	6 50	
Last of Contracts	3 00	
Law of Contracts	0 00	
Vaw of Operations Fremminary to Construction in Engineering and	5 00	
Architecture8vo,		
Sheep,	5 50	
Warren's Stereotomy—Problems in Stone-cutting8vo,	2 50	
* Waterbury's Vest-Pocket Hand-book of Mathematics for Engineers.		
* Enlarged Edition, Including Tables	1 00	
* Enlarged Edition, Including Tablesmor.	1 50	
Webb's Problems in the Use and Adjustment of Engineering Instruments.		
16mo, mor.	1 25	
Wilson's Topographic Surveying8vo,	3 50	
BRIDGES AND ROOFS.		
During the state of the state o	0.00	
Boller's Practical Treatise on the Construction of Iron Highway Bridges8vo,	2 00	
* Thames River BridgeOblong paper,	5 00	
Burr and Falk's Design and Construction of Metallic Bridges8vo,	5 00	
Influence Lines for Bridge and Roof Computations8vo,	3 00	
Du Bois's Mechanics of Engineering. Vol. IISma 4to,	10 00	
Foster's Treatise on Wooden Trestle Bridges4to,	5 00	
Fowler's Ordinary Foundations8vo,	3 50	
Greene's Arches in Wood Iron and Stone	2 50	
Bridge Trusses8vo,	2 50	
Roof Trusses8vo,	1 25	
Grimm's Secondary Stresses in Bridge Trusses8vo,	2 50	
	3 00	
Heller's Stresses in Structures and the Accompanying Deformations8vo, Howe's Design of Simple Roof-trusses in Wood and Steel8vo.	2 00	
Symmetrical Masonry Arches8vo,	2 50	
Treatise on Arches8vo,	4 00	
* Hudson's Deflections and Statically Indeterminate Stresses Small 4to,	3 50	
* Plate Girder Design	1 50	
* Jacoby's Structural Details, or Elements of Design in Heavy Framing, 8vo,	2 25	
Johnson, Bryan and Turneaure's Theory and Practice in the Designing of		
Modern Framed StructuresSmall 4to,	10 00	
* I-harry Device and Turnouse's Theory and Practice in the Designing of		
* Jonnson, Bryan and Tunisadire's Theory and Tractee in the Designing of Modern Framed Structures. New Edition. Part I	3 00	
* Part II New Edition	4 00	
Morrison and Incohy's Text-book on Roofs and Bridges		
Part I. Stresses in Simple Trusses	2 50	
Part I. Stresses in Simple Trusses	2 50	
Part II. Graphic Statics8vo,	2 50	
Part III. Bridge Design8vo,		
Part IV. Higher Structures8vo,	2 50	
Sondericker's Graphic Statics, with Applications to Trusses, Beams, and	0.00	
Arches8vo,	2 00	
Waddell's De Pontibus, Pocket-book for Bridge Engineers16mo, mor.	2 00	
* Specifications for Steel Bridges12mo,	50	
Waddell and Harrington's Bridge Engineering. (In Preparation.)		
William Parising		
TITLD DATIT TOO		
HYDRAULICS.		
HYDRAULICS.		
	3 00	
Rarnes's Ice Formation	3 00	
Barnes's Ice Formation		
Barnes's Ice Formation	2 00	
Barnes's Ice Formation		

Church's Diagrams of Mean Velocity of Water in Open Channels.		
Ublong 4to, paper,	\$1	50
Hydraulic Motors8vo,	2	00
Mechanics of Fluids (Being Part IV of Mechanics of Engineering) 8vo,		00
Coffin's Graphical Solution of Hydraulic Problems16mo, mor.		5 0
Flather's Dynamometers, and the Measurement of Power	3	00
Folwell's Water-supply Engineering8vo,	4	00
Frigall's Water-power	5	00
Fuertes's Water and Public Health	1	50
Water-filtration Works	2	50
Ganguillet and Kutter's General Formula for the Uniform Flow of Water in		
Rivers and Other Channels. (Hering and Trautwine.)8vo.	4	00
Hazen's Clean Water and How to Get ItLarge 12mo,	1	50
Filtration of Public Water-supplies8vo,	3	00
Hazelhurst's Towers and Tanks for Water-works8vo,	2	50
Herschel's 115 Experiments on the Carrying Capacity of Large, Riveted, Metal		
Conduits8vo.	2	00
Hoyt and Grover's River Discharge8vo,	2	00
Hubbard and Kiersted's Water-works Management and Maintenance.		
8vo,	4	00
*Lyndon's Development and Electrical Distribution of Water Power.	_	
> 8vo,	3	00
Mason's Water-supply. (Considered Principally from a Sanitary Stand-	_	
point.)8vo,	4	00
Merriman's Treatise on Hydraulics		00
* Molitor's Hydraulics of Rivers, Weirs and Sluices8vo,		00
* Morrison and Brodie's High Masonry Dam Design		50
* Richards's Laboratory Notes on Industrial Water Analysis8vo,		50
Schuyler's Reservoirs for Irrigation, Water-power, and Domestic Water-		00
supply. Second Edition, Revised and EnlargedLarge 8vo,	6	00
* Thomas and Watt's Improvement of Rivers4to,		00
Turneaure and Russell's Public Water-supplies		00
* Wegmann's Design and Construction of Dams. 6th Ed., enlarged4to,	_	00
Water-Supply of the City of New York from 1658 to 18954to,	10	
Whipple's Value of Pure WaterLarge 12mo,		00
Williams and Hazen's Hydraulic Tables		50
Wilson's Irrigation Engineering		00
wood's Turbines	4	50
MATERIALS OF ENGINEERING.		
Baker's Roads and Pavements8vo,	5	00
Treatise on Masonry Construction		00
Black's United States Public WorksOblong 4to,		00
Distance and Description Highway Perinsoning (In Description)	J	00
Blanchard and Drowne's Highway Engineering. (In Press.) Bleininger's Manufacture of Hydraulic Cement. (In Preparation.)		
Bottler's Varnish Making. (Sabin.) (In Press.) Burr's Elasticity and Resistance of the Materials of Engineering8vo,	-7	50
Burns 's Highway Construction		00
Byrne's Highway Construction	Э	UU
Inspection of the Materials and Workmanship Employed in Construction.	0	00
16mo,		00
Church's Mechanics of Engineering8vo,	0	00
Mechanics of Solids (Being Parts I, II, III of Mechanics of Engineer-		w .
ing8vo,	4	50
Du Bois's Mechanics of Engineering.	_	
Vol. I. Kinematics, Statics, KineticsSmall 4to, Vol. II. The Stresses in Framed Structures, Strength of Materials and	7	50
vol. 11. The Stresses in Framed Structures, Strength of Materials and		
Theory of FlexuresSmall 4to,	10	00
Eckel's Building Stones and Clays. (In Press.)		
* Cements, Limes, and Plasters		00
Fowler's Ordinary Foundations8vo,		50
Greene's Structural Mechanics8vo,	')	50
	4	
Holley's Analysis of Paint and Varnish Products. (In Press.)	_	
Holley's Analysis of Paint and Varnish Products. (In Press.) * Lead and Zinc Pigments	3	00

Johnson's (C. M.) Rapid Methods for the Chemical Analysis of Special Steels,	
Steel-making Alloys and GraphiteLarge 12mo,	\$3 00
Johnson's (J. B.) Materials of ConstructionLarge 8vo,	6 00
Keep's Cast Iron	2 50
Laure's Paints for Steel Structures 12mo	7 50
Lowe's Paints for Steel Structures	2 00
Maurer's Technical Mechanics8vo,	4 00
Merrill's Stones for Building and Decoration8vo,	5 00
Merriman's Mechanics of Materials8vo,	5 00
* Strength of Materials 12mo,	1 00
Metcalf's Steel. A Manual for Steel-users	2 00
Morrison's Highway Engineering8vo,	2 50
* Murdock's Strength of Materials	2 00
Rice's Concrete Block Manufacture8vo,	5 00 2 00
Richardson's Modern Asphalt Pavement8vo,	3 00
Richey's Building Foreman's Pocket Book and Ready Reference.16mo, mor.	5 00
* Cement Workers' and Plasterers' Edition (Building Mechanics' Ready	
Reference Series)16mo, mor.	1 50
Handbook for Superintendents of Construction16mo, mor.	4 00
*Stone and Brick Masons' Edition (Building Mechanics' Ready	
Reference Series)	1 50
* Ries's Clays: Their Occurrence, Properties, and Uses8vo,	5 00
* Ries and Leighton's History of the Clay-working Industry of the United	0 50
Sabin's Industrial and Artistic Technology of Paint and Varnish8vo,	2 50
* Smith's Strength of Material	$\frac{3}{1} \frac{00}{25}$
Snow's Principal Species of Wood	3 50
Spalding's Hydraulic Cement	2 00
Text-book on Roads and Pavements	2 00
* Taylor and Thompson's Extracts on Reinforced Concrete Design8vo,	2 00
Treatise on Concrete, Plain and Reinforced8vo,	5 00
Thurston's Materials of Engineering. In Three Parts8vo,	8 00
Part I. Non-metallic Materials of Engineering and Metallurgy8vo,	2 00
Part II. Iron and Steel	3 50
Part III. A Treatise on Brasses, Bronzes, and Other Alloys and their	0.70
Constituents	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
Tillson's Street Pavements and Paving Materials	4 00
Second Edition, Revised and Enlarged8vo,	3 50
Waterbury's Cement Laboratory Manual	1 00
Laboratory Manual for Testing Materials of Construction. (In Press.)	
Wood's (De V.) Treatise on the Resistance of Materials, and an Appendix on	
the Preservation of Timber8vo,	2 00
Wood's (M. P.) Rustless Coatings: Corrosion and Electrolysis of Iron and	
Steel8vo,	4 00
RAILWAY ENGINEERING.	
Andrews's Handbook for Street Railway Engineers3×5 inches, mor.	1 25
Berg's Buildings and Structures of American Railroads4to,	5 00
Brooks's Handbook of Street Railroad Location16mo, mor.	1 50
* Burt's Railway Station Service	2 00
Butts's Civil Engineer's Field-book	2 50
Crandall's Railway and Other Earthwork Tables	1 50 2 00
* Crockett's Methods for Earthwork Computations	1 50
Dredge's History of the Pennsylvania Railroad. (1879)	5 00
Fisher's Table of Cubic Yards	25
Godwin's Railroad Engineers' Field-book and Explorers' Guide. 16mo, mor.	2 50
Hudson's Tables for Calculating the Cubic Contents of Excavations and Em-	
bankments8vo,	1 00
Ives and Hilts's Problems in Surveying, Railroad Surveying and Geodesy	
Molitor and Beard's Manual for Resident Engineers	1 50
	1 00

Nagle's Field Manual for Railroad Engineers16mo, mor.	\$3 00
* Orrock's Railroad Structures and Estimates	3 00 3 00
Paymond's Pailroad Field Coometry 16mg mor	2 00
Elements of Railroad Engineering	3 50
Railroad Engineer's Field Book. (In Preparation.)	2 00
Roberts' Track Formulæ and Tables	3 00
Railroad Spiral	1 50
Taylor's Prismoidal Formulæ and Earthwork8vo,	1 50
Webb's Economics of Railroad Construction	2 50 5 00
Wellington's Economic Theory of the Location of RailwaysLarge 12mo.	5 00
Wellington's Economic Theory of the Location of RailwaysLarge 12mo, Wilson's Elements of Railroad-Track and Construction	2 00
DRAWING	
Barr and Wood's Kinematics of Machinery8vo,	2 50
* Bartlett's Mechanical Drawing 8vo, * " Abridged Ed 8vo,	3 00
* " Abridged Ed8vo,	1 50
* Bartlett and Johnson's Engineering Descriptive Geometry	1 50
Blessing and Darling's Descriptive Geometry. (In Press.) Elements of Drawing. (In Press.)	
Coolidge's Manual of Drawing	1 00
Coolidge and Freeman's Elements of General Drafting for Mechanical Engi-	0 50
neers	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
Emch's Introduction to Projective Geometry and its Application8vo,	2 50
Hill's Text-book on Shades and Shadows, and Perspective8vo,	2 00
Jamison's Advanced Mechanical Drawing	2 00 2 50
Elements of Mechanical Drawing	2 50
Part I. Kinematics of Machinery8vo,	1 50
Part II. Form, Strength, and Proportions of Parts8vo,	3 00
* Kimball and Barr's Machine Design	3 00
Kinematics; or, Practical Mechanism	5 00
Mechanical Drawing4to,	4 00
Velocity Diagrams	1 50
McLeod's Descriptive Geometry	1 50 1 50
Industrial Drawing. (Thompson.)8vo,	3 50
Moyer's Descriptive Geometry8vo,	2 00
Reed's Topographical Drawing and Sketching	$\frac{5}{2} \frac{00}{00}$
Text-book of Mechanical Drawing and Elementary Machine Design8vo,	3 00
Robinson's Principles of Mechanism8vo,	3 00
Schwamb and Merrill's Elements of Mechanism	3 00
Smith (A. W.) and Marx's Machine Design	$\frac{3}{2} \frac{00}{50}$
* Titsworth's Elements of Mechanical DrawingOblong 8vo.	1 25
Tracy and North's Descriptive Geometry. (In Press.) Warren's Elements of Descriptive Geometry, Shadows, and Perspective. 8vo,	
Elements of Machine Construction and Drawing8vo,	3 50 7 50
Elements of Plane and Solid Free-hand Geometrical Drawing12mo,	1 00
General Problems of Shades and Shadows8vo,	3 00
Manual of Elementary Problems in the Linear Perspective of Forms and	1 00
Shadow	1 00 1 50
Plane Problems in Elementary Geometry	1 25
Weisbach's Kinematics and Power of Transmission. (Hermann and Klein.)	- 05
Wilson's (H. M.) Topographic Surveying	5 00 3 50
* Wilson's (V. T.) Descriptive Geometry	1 50
Free-hand Lettering8vo.	1 00
Free-hand Perspective	2 50
Large 8vo,	3 00

ELECTRICITY AND PHYSICS.

* Abegg's Theory of Electrolytic Dissociation. (von Ende.)12mo, Andrews's Hand-book for Street Railway Engineers 3×5 inches mor. Anthony and Ball's Lecture-notes on the Theory of Electrical Measure-	1	25
ments		00
Anthony and Brackett's Text-book of Physics. (Magie.) Large 12mo, Benjamin's History of Electricity 8vo,		00
Betts's Lead Refining and Electrolysis		00
Burgess and Le Chatelier's Measurement of High Temperatures. Third	-	00
Edition. (In Press.)		
Classen's Quantitative Chemical Analysis by Electrolysis. (Boltwood.).8vo,		00
* Collins's Manual of Wireless Telegraphy and Telephony		50
Crehore and Squier's Polarizing Photo-chronograph		00 25
Dawson's "Engineering" and Electric Traction Pocket-book16mo, mor.		00
Dolezalek's Theory of the Lead Accumulator (Storage Battery). (von Ende.)		
12mo,		50
Duhem's Thermodynamics and Chemistry. (Burgess.)		00
Flather's Dynamometers, and the Measurement of Power		00 50
Gilbert's De Magnete. (Mottelay)		50
* Hanchett's Alternating Currents		00
Hering's Ready Reference Tables (Conversion Factors)16mo, mor.		50.
* Hobart and Ellis's High-speed Dynamo Electric Machinery8vo,		00
Holman's Precision of Measurements	2	00 75
* Hutchinson's High-Efficiency Electrical Illuminants and Illumination.		10
Large 12mo,	2	50
Jones's Electric Ignition. (In Press.)		
Karapetoff's Experimental Electrical Engineering:		=0
* Vol. I		50 50
Kinzbrunner's Testing of Continuous-current Machines		00
Landauer's Spectrum Analysis. (Tingle.)8vo,		00
Löb's Electrochemistry of Organic Compounds. (Lorenz.)8vo,		00
* Lyndon's Development and Electrical Distribution of Water Power8vo,		00
* Lyons's Treatise on Electromagnetic Phenomena. Vols, I. and II. 8vo, each, * Michie's Elements of Wave Motion Relating to Sound and Light8vo,		00
* Morgan's Physical Chemistry for Electrical Engineers12mo,		50
* Norris's Introduction to the Study of Electrical Engineering8vo,		50
Norris and Dennison's Course of Problems on the Electrical Characteristics of		
Circuits and Machines. (In Press.)	10	F0
* Parshall and Hobart's Electric Machine Design4to, half mor, Reagan's Locomotives: Simple, Compound, and Electric. New Edition.	12	อน
Large 12mo,	3	50
* Rosenberg's Electrical Engineering. (Haldane Gee-Kinzbrunner.)8vo,		00
Ryan, Norris, and Hoxie's Electrical Machinery. Vol. I8vo,		50
Schapper's Laboratory Guide for Students in Physical Chemistry12mo, * Tillman's Elementary Lessons in Heat8vo,		00 50
* Timbie's Elements of Electricity		00
Tory and Pitcher's Manual of Laboratory PhysicsLarge 12mo,		00
Ulke's Modern Electrolytic Copper Refining8vo,		00
* Waters's Commercial Dynamo Design8vo,	2	00
LAW.		
* Brennan's Hand-book of Useful Legal Information for Business Men.		
	5	
* Davis's Elements of Law		50
* Treatise on the Military Law of United States	7 2	
* Dudley's Military Law and the Procedure of Courts-martial. Large 12mo, Manual for Courts-martial		50 50
Wait's Engineering and Architectural Jurisprudence8vo,		00
Sheep,	6	
Wait's Law of Contracts8vo,	3	00

Wait's Law of Operations Preliminary to Construction in Engineering and		
Architecture8vo,		
Sheep,	5	50
MA MITTEN A TIME		
MATHEMATICS.		
"Dalanda Ellindia Dunations Syro	1	50
Baker's Elliptic Functions		00
* Buchanan's Plane and Spherical Trigonometry		00
* Duchanan's Flane and Spherical Higohometry		00
Byerly's Harmonic Functions		00
* Coffin's Vector Analysis		50
Compton's Manual of Logarithmic Computations12mo,		50
* Dickson's College AlgebraLarge 12mo,		50
* Introduction to the Theory of Algebraic Equations Large 12mo,		25
Emch's Introduction to the Theory of Algebraic Equations		50
Eights' Business of a Complex Veriable		00
Fiske's Functions of a Complex Variable		50
Elements of Geometry		75
* Rational Geometry		50
Synthetic Projective Geometry		00
* Hancock's Lectures on the Theory of Elliptic Functions8vo,		00
Hyde's Grassmann's Space Analysis		00
* Johnson's (J. B.) Three-place Logarithmic Tables: Vest-pocket size, paper,	-	15
* 100 copies,	5	00
* Mounted on heavy cardboard, 8×10 inches,	U	25
* 10 copies,	9	00
Johnson's (W. W.) Abridged Editions of Differential and Integral Calculus.	2	00
Large 12mo, 1 vol.	2	50
Curve Tracing in Cartesian Co-ordinates12mo,		00
Differential Equations		00
Elementary Treatise on Differential CalculusLarge 12mo,		5 0
Elementary Treatise on the Integral CalculusLarge 12mo,		50
* Theoretical Mechanics		00
* Theoretical Mechanics		50
Treatise on Differential CalculusLarge 12mo,		00
Treatise on the Integral CalculusLarge 12mo,		00
Treatise on Ordinary and Partial Differential EquationsLarge 12mo,		50
* Karapetoff's Engineering Applications of Higher Mathematics. Large 12mo,		75
Koch's Practical Mathematics. (In Press.)	U	10
Laplace's Philosophical Essay on Probabilities. (Truscott and Emory.). 12mo,	2	00
* Le Messurier's Key to Professor W. W. Johnson's Differential Equations.	2	00
Small 8vo,	1	75
* Ludlow's Logarithmic and Trigonometric Tables		00
* Ludlow and Bass's Elements of Trigonometry and Logarithmic and Other	-	00
Tables	3	00
* Trigonometry and Tables published separately		00
Macfarlane's Vector Analysis and Quaternions		00
McMahon's Hyperbolic Functions		00
Manning's Irrational Numbers and their Representation by Sequences and	-	00
Series	1	25
Mathematical Monographs. Edited by Mansfield Merriman and Robert		20
S. Woodward Octavo each	1	00
No. 1. History of Modern Mathematics, by David Eugene Smith. No. 2. Synthetic Projective Geometry, by George Bruce Halsted. No. 3. Determinants, by Laenas Gifford Weld. No. 4. Hyper-	-	00
No. 2. Synthetic Projective Geometry, by George Bruce Halsted		
No. 3. Determinants, by Laenas Gifford Weld No. 4 Hyper-		
bolic Functions, by James McMahon, No. 5, Harmonic Func-		
bolic Functions, by James McMahon. No. 5. Harmonic Functions, by William E. Byerly. No. 6. Grassmann's Space Analysis, by Edward W. Hyde. No. 7. Probability and Theory of Errors,		
by Edward W. Hyde. No. 7. Probability and Theory of Errors		
by Robert S. Woodward, No. 8. Vector Analysis and Chaternions		
by Alexander Macfarlane, No. 9. Differential Equations by		
by Alexander Macfarlane. No. 9. Differential Equations, by William Woolsey Johnson. No. 10. The Solution of Equations,		
by Mansfield Merriman. No. 11. Functions of a Complex Variable,		
by Thomas S. Fiske.		
Maurer's Technical Mechanics8vo,	4	00
Merriman's Method of Least Squares	2	00
	1	

* Moritz's Elements of Plane Trigonometry8vo, Rice and Johnson's Differential and Integral Calculus. 2 vols. in one.	\$2	00
Large 12mo,		50
Elementary Treatise on the Differential Calculus Large 12mo,		00
Smith's History of Modern Mathematics8vo, * Veblen and Lennes's Introduction to the Real Infinitesimal Analysis of One	1	00
Variable	9	00
* Waterbury's Vest Pocket Hand-book of Mathematics for Engineers.	4	00
	1	00
* Enlarged Edition, Including Tables		50
Weld's Determinants8vo,	1	00
Wood's Elements of Co-ordinate Geometry8vo,	2	00
Woodward's Probability and Theory of Errors8vo,	1	00
MECHANICAL ENGINEERING.		
MATERIALS OF ENGINEERING, STEAM-ENGINES AND BOILE	RS	
Bacon's Forge Practice	1	50
Baldwin's Steam Heating for Buildings		50
Barr and Wood's Kinematics of Machinery8vo,		50
* Bartlett's Mechanical Drawing8vo,		00
* " " Abridged Ed		50
* Bartlett and Johnson's Engineering Descriptive Geometry8vo,		50
* Burr's Ancient and Modern Engineering and the Isthmian Canal 8vo,	3	50
Carpenter's Heating and Ventilating Buildings8vo,	4	00
* Carpenter and Diederichs's Experimental Engineering8vo,		00
* Clerk's The Gas, Petrol and Oil Engine		00
Compton's First Lessons in Metal Working		50
Compton and De Groodt's Speed Lathe12mo,		50
Coolidge's Manual of Drawing	1	00
gineersOblong 4to,	9	50
Cromwell's Treatise on Belts and Pulleys		50
Treatise on Toothed Gearing		50
Dingey's Machinery Pattern Making		00
Durley's Kinematics of Machines		00
Flanders's Gear-cutting Machinery Large 12mo,		00
Flather's Dynamometers and the Measurement of Power12mo,	3	00
Rope Driving12mo,	2	00
Gill's Gas and Fuel Analysis for Engineers		25
Goss's Locomotive Sparks		00
* Greene's Pumping Machinery8vo,		00
Hering's Ready Reference Tables (Conversion Factors)16mo, mor. * Hobart and Ellis's High Speed Dynamo Electric Machinery8vo,		50 00
Hutton's Cas Engine		00
Hutton's Gas Engine		00
Elements of Mechanical Drawing		50
Jones's Gas Engine	4	00
Machine Design:		
Part I. Kinematics of Machinery8vo,		50
Part II. Form, Strength, and Proportions of Parts8vo,		00
* Kaup's Machine Shop PracticeLarge 12mo,		25
* Kent's Mechanical Engineer's Pocket-Book16mo, mor.		00
Kerr's Power and Power Transmission	2	00
* Kimball and Barr's Machine Design	3	00
mission	2	50
*Lanza's Dynamics of Machinery	_	50
* Lanza's Dynamics of Machinery		00
* Levin's Gas Engine8vo,		00
* Lorenz's Modern Refrigerating Machinery. (Pope, Haven, and Dean)8vo,	4	00
MacCord's Kinematics; or, Practical Mechanism8vo,		00
Mechanical Drawing		00
Velocity Diagrams8vo,	1	50
19		

MacFarland's Standard Reduction Factors for Gases	\$1	50 50
Mehrtens's Gas Engine Theory and DesignLarge 12mo.		50
Miller Berry and Riley's Problems in Thermodynamics and Heat Engineer-	0	75
iny. Svo, paper, Oberg's Handbook of Small Tools. Large 12mo, * Parshall and Hobart's Electric Machine Design. Small 4to, half leather,	2	50
* Parshall and Hobart's Electric Machine Design. Small 4to, half leather, * Peele's Compressed Air Plant. Second Edition, Revised and Enlarged. 8vo,	12 3	50 50
Perkins's General Thermodynamics. (In Press.)		
Poole's Calorific Power of Fueis	3	00
Randall's Treatise on Heat. (In Press.) * Reid's Mechanical Drawing. (Elementary and Advanced.)8vo,		
* Reid's Mechanical Drawing. (Elementary and Advanced.)		00
Richards's Compressed Air	1	50
Robinson's Principles of Mechanism		00
Smith (A. W.) and Marx's Machine Design	3	00
Smith's (O.) Press-working of Metals	3	00
Preston.)		00
Stone's Practical Testing of Gas and Gas Meters	3	50
12mo,		00
Treatise on Friction and Lost Work in Machinery and Mill Work8vo, * Tillson's Complete Automobile Instructor		00 50
* Titsworth's Elements of Mechanical DrawingOblong 8vo,	1	25
Warren's Elements of Machine Construction and Drawing 8vo, * Waterbury's Vest Pocket Hand-book of Mathematics for Engineers.	7	50
27×53 inches mor		00
* Enlarged Edition, Including Tables	1	50
Klein.)	5	00
		00
Machinery of Transmission and Governors. (Hermann—Klein.)8vo, Wood's Turbines8vo,	$\frac{5}{2}$	50
Wood's Turbines		
MATERIALS OF ENGINEERING.	2	50
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering8vo.	7	
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering8vo, Church's Mechanics of Engineering8vo, Mechanics of Solids (Being Parts I, II, III of Mechanics of Engineering).	7 6	50 50 00
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering 8vo, Church's Mechanics of Engineering 8vo, Mechanics of Solids (Being Parts I, II, III of Mechanics of Engineering). * Greene's Structural Mechanics 8vo	7 6 4	50
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering 8vo, Church's Mechanics of Engineering	7 6 4 2	50 50 00 50 50
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering 8vo, Church's Mechanics of Engineering	7 6 4 2	50 00 50 50 00
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering 8vo, Church's Mechanics of Engineering 8vo, Mechanics of Solids (Being Parts I, II, III of Mechanics of Engineering). * Greene's Structural Mechanics 8vo, Holley's Analysis of Paint and Varnish Products. (In Press.) * Lead and Zinc Pigments Large 12mo, Johnson's (C. M.) Rapid Methods for the Chemical Analysis of Special Steels, Steel-Making Alloys and Graphite	7 6 4 2 3	50 50 00 50 50 00
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3	50 50 50 50 50 00
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 6 2	50 00 50 50 50 00 00 00 50
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering. 8vo, Church's Mechanics of Engineering. 8vo, Mechanics of Solids (Being Parts I, II, III of Mechanics of Engineering). Svo, *Greene's Structural Mechanics. 8vo, Holley's Analysis of Paint and Varnish Products. (In Press.) * Lead and Zinc Pigments. Large 12mo, Johnson's (C. M.) Rapid Methods for the Chemical Analysis of Special Steels, Steel-Making Alloys and Graphite Large 12mo, Johnson's (J. B.) Materials of Construction. 8vo, Keep's Cast Iron. 8vo, King's Elements of the Mechanics of Materials and of Power of Transmission. 8vo, Larga's Applied Mechanics	7 6 4 2 3 6 2 2	50 50 50 50 50 00
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 6 2 7	50 50 50 50 50 50 50 50 50 50
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	2 7 6 4 2 3 3 6 2 2 7 1 2 4	50 50 00 50 50 50 00 00 50 50 50 00 00 0
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 3 6 2 2 7 1 2 4 5	50 50 00 50 50 50 00 00 50 50 50 00 00 0
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 3 6 2 2 7 1 2 4 5 1	50 00 50 50 50 50 50 00 00 00 00 00 00 0
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 3 6 2 2 7 1 2 4 5 1	50 50 00 50 50 50 00 00 50 50 50 00 00 0
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 3 6 2 2 7 1 2 4 5 1 2 2 3 1	50 50 50 50 50 50 50 50 50 50 60 60 60 60 60 60 60 60 60 60 60 60 60
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 3 6 2 2 7 1 2 4 5 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	50 50 50 50 50 50 50 50 50 50 60 60 60 60 60 60 60 60 60 60 60 60 60
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 3 6 6 2 2 7 1 2 4 5 1 1 1 8 8 2	50 50 00 50 50 50 00 00 50 50
MATERIALS OF ENGINEERING. Burr's Elasticity and Resistance of the Materials of Engineering	7 6 4 2 3 3 6 6 2 2 7 1 2 4 5 1 1 1 8 8 2	50 50 50 50 50 50 50 50 50 50 00 00 00 0

waterbury's Laboratory Manual for Testing Materials of Construction.		
Wood's (De V.) Elements of Analytical Mechanics8vo,	00	00
Treatise on the Decistores of Materials and an Appendix on the	33	00
Treatise on 'the Resistance of Materials and an Appendix on the Preservation of Timber8vo,	0	00
Wood's (M. P.) Rustless Coatings: Corrosion and Electrolysis of Iron and	2	00
Steel8vo,	-	00
51661	4	00
CAPAN DWGINDG AND DOLLEDG		
STEAM-ENGINES AND BOILERS.		
Berry's Temperature-entropy Diagram. Third Edition Revised and En-	1	
larged	2	50
Carnot's Reflections on the Notive Power of Heat. (Thurston) 12mo		50
Chase's Art of Pattern Making.		50
Carnot's Reflections on the Motive Power of Heat. (Thurston.) 12mo, Chase's Art of Pattern Making		00
Dawson's "Engineering" and Electric Traction Pocket-book 16mo, mor.		00
* Gebhardt's Steam Power Plant Engineering8vo,		00
Goss's Locomotive Performance8vo.		00
Hemenway's Indicator Practice and Steam-engine Economy12mo,		00
Hirshfeld and Barnard's Heat Power Engineering. (In Press.)		
Hutton's Heat and Heat-engines	5	00
Mechanical Engineering of Power Plants8vo,	5	00
Kent's Steam Boiler Economy8vo,	4	00
Kneass's Practice and Theory of the Injector8vo,	1	50
MacCord's Slide-valves8vo,	2	00
Meyer's Modern Locomotive Construction4to,	10	00
Miller, Berry, and Riley's Problems in Thermodynamics8vo, paper,		75
Moyer's Steam Turbine8vo,	4	00
Peabody's Manual of the Steam-engine Indicator	1	50
Tables of the Properties of Steam and Other Vapors and Temperature-		
Entropy Table		00
Thermodynamics of the Steam-engine and Other Heat-engines 8vo,		00
* Thermodynamics of the Steam Turbine8vo,		00
Valve-gears for Steam-engines8vo,		50
Peabody and Miller's Steam-boilers8vo,	4	90
Pupin's Thermodynamics of Reversible Cycles in Gases and Saturated Vapors.		~~
(Osterberg.)	T	25
Reagan's Locomotives: Simple, Compound, and Electric. New Edition.		
Large 12mo,		MO
Singlair's Locamative Engine Running and Management 12mg		50
Sinclair's Locomotive Engine Running and Management	2	00
Smart's Handbook of Engineering Laboratory Practice	2 2	00 50
Smart's Handbook of Engineering Laboratory Practice	2 2 3	00 50 00
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1	00 50 00 00
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1 2	00 50 00 00 50
Smart's Handbook of Engineering Laboratory Practice	$\begin{array}{c} 2 \\ 2 \\ 3 \\ 1 \\ 2 \\ \cdot 3 \end{array}$	00 50 00 00 50 00
Smart's Handbook of Engineering Laboratory Practice	$\begin{array}{c} 2 \\ 2 \\ 3 \\ 1 \\ 2 \\ \cdot 3 \end{array}$	00 50 00 00 50
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{c} 2 \\ 2 \\ 3 \\ 1 \\ 2 \\ \cdot 3 \\ 4 \end{array} $	00 50 00 00 50 00
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{c} 2 \\ 2 \\ 3 \\ 1 \\ 2 \\ \cdot 3 \\ 4 \end{array} $	00 50 00 00 50 00 00
Smart's Handbook of Engineering Laboratory Practice	2 3 1 2 ·3 4 5 1	00 50 00 00 50 00 00
Smart's Handbook of Engineering Laboratory Practice	2 3 1 2 ·3 4 5	00 50 00 00 50 00 00 00 50
Smart's Handbook of Engineering Laboratory Practice	2 3 1 2 ·3 4 5 1 5 10	00 50 00 00 50 00 00 00 50
Smart's Handbook of Engineering Laboratory Practice	2 3 1 2 ·3 4 5 1 5 10 6	00 50 00 00 50 00 00 00 50 00
Smart's Handbook of Engineering Laboratory Practice	2 3 1 2 ·3 4 5 1 5 10 6	00 50 00 00 50 00 00 00 50 00 00
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	00 50 00 00 50 00 00 00 50 00 00 00 00
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	00 50 00 00 50 00 00 50 00 00 00 00 00
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	00 50 00 00 50 00 00 50 00 00 00 00 00 0
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	00 50 00 00 50 00 00 50 00 00 00 00 00
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	00 50 00 00 50 00 00 50 00 00 00 00 00 0
Smart's Handbook of Engineering Laboratory Practice	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	00 50 00 00 50 00 00 50 00 00 00 00 00 0
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1 2 3 4 5 1 5 1 0 6 6 4 5 4	00 50 00 00 50 00 00 00 00 00 00 00 00 0
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1 2 3 4 5 1 5 1 0 6 6 4 5 4	00 50 00 00 50 00 00 00 00 00 00 00 00
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1 2 · 3 4 5 1 5 10 6 6 4 5 5 4 6 3	00 50 00 00 50 00 00 00 00 00 00 00 00 0
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1 2 · 3 4 5 1 5 10 6 6 4 5 5 4 6 3	00 50 00 00 50 00 00 00 00 00 00 00 00
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1 2 3 4 5 1 5 1 6 6 4 6 3 1	00 50 00 00 50 00 00 00 00 00 00 00 00 0
Smart's Handbook of Engineering Laboratory Practice	2 2 3 1 2 3 4 5 1 5 1 6 6 6 4 5 4 6 7 4 7 4 7 4 7 7 7 7 7 7 7 7 7 7 7 7	00 50 00 00 50 00 00 00 00 00 00 00 00 0

Dana's Text-book of Elementary Mechanics for Colleges and Schools .12mo, Du Bois's Elementary Principles of Mechanics:	\$1.5	(I)
Vol. I. Kinematics8vo,	3 5	0
Vol II Statics8vo,	4 0	
Mechanics of Engineering. Vol. I	7 5	0
Vcl. IISmall 4to,		
* Greene's Structural Mechanics	$\frac{2}{1} \frac{5}{2}$	
Hartmann's Elementary Mechanics for Engineering Students12mo, James's Kinematics of a Point and the Rational Mechanics of a Particle.	1 2	O -
James's Kinematics of a Point and the Rational Mechanics of a Particle. Large 12mo.	2 0	O:
* Johnson's (W. W.) Theoretical Mechanics12mo,	3 0	
* King's Elements of the Mechanics of Materials and of Power of Trans-	0 0	•
mission	2 5	0
Tanga's Applied Mechanics	7 5	0
* Martin's Text Book on Mechanics, Vol. 1, Statics	1 2	5
* Vol. II. Kinematics and Kinetics	1 5	
* Vol. III. Mechanics of Materials12mo,	1 5	
Maurer's Technical Mechanics8vo,	4 0	
* Merriman's Elements of Mechanics	1 0	
Mechanics of Materials	5 0	
* Michie's Elements of Analytical Mechanism	3 0	
Sanborn's Mechanics ProblemsLarge 12mo,	1 5	
Schwamb and Merrill's Elements of Mechanism8vo,	3 0	
Wood's Elements of Analytical Mechanics 8vo,	3 0	0.
Principles of Elementary Mechanics12mo,	1 2	5
MEDICAL.		
* Abderhalden's Physiological Chemistry in Thirty Lectures. (Hall and		
Defren.)	5 0	
von Behring's Suppression of Tuberculosis. (Bolduan.)12mo, * Bolduan's Immune Sera	1 0	
* Bolduan's Immune Sera	1 5	
Bordet's Studies in Immunity. (Gay.)	6 0	
Davenport's Statistical Methods with Special Reference to Biological Varia-	3 0	U.
tions16mo, mor.	1 5	O:
Ehrlich's Collected Studies on Immunity. (Bolduan.)	6 0	
Ehrlich's Collected Studies on Immunity. (Bolduan.)8vo, * Fischer's Nephritis	2 5	0
* Oedema8vo,	2 0	
* Physiology of Alimentation	2 0	
* de Fursac's Manual of Psychiatry. (Rosanoff and Collins.)Large 12mo,	2 5	
* Hammarsten's Text-book on Physiological Chemistry. (Mandel.)8vo,	$\frac{4}{1} \frac{0}{2}$	-
Jackson's Directions for Laboratory Work in Physiological Chemistry8vo, Lassar-Cohn's Praxis of Urinary Analysis. (Lorenz.)	$\frac{1}{1} \frac{2}{0}$	
Mandel's Hand-book for the Bio-Chemical Laboratory12mo,	1 5	
* Nelson's Analysis of Drugs and Medicines 12mo	3 0	
* Pauli's Physical Chemistry in the Service of Medicine. (Fischer.)12mo,	1 2	5
* Pozzi-Escot's Toxins and Venoms and their Antibodies. (Cohn.) 12mo,	1 0	
* Pauli's Physical Chemistry in the Service of Medicine. (Fischer.)12mo, * Pozzi-Escot's Toxins and Venoms and their Antibodies. (Cohn.)12mo, Rostoski's Serum Diagnosis. (Bolduan.)	1 0	~
Ruddiman's Incompatibilities in Prescriptions8vo,	2 0	
Whys in Pharmacy	$\begin{array}{cccc} 1 & 0 \\ 2 & 5 \end{array}$	
Salkowski's Physiological and Pathological Chemistry. (Orndorff.)8vo, * Satterlee's Outlines of Human Embryology	$\begin{array}{c} 2 & 3 \\ 1 & 2 \end{array}$	
Smith's Lecture Notes on Chemistry for Dental Students8vo,	$\frac{1}{2} \frac{2}{5}$	
* Whipple's Tyhpoid Fever	3 0	
* Whipple's Tyhpoid FeverLarge 12mo, * Woodhull's Military Hygiene for Officers of the LineLarge 12mo,		0,
* Personal Hygiene	1 0	0
Worcester and Atkinson's Small Hospitals Establishment and Maintenance,		
and Suggestions for Hospital Architecture, with Plans for a Small		-
Hospital12mo,	1 2	5
METALLURGY.		
Betts's Lead Refining by Electrolysis8vo,	4 0	0
Bolland's Encyclopedia of Founding and Dictionary of Foundry Terms used	2 0	0
in the Practice of Moulding	3 0	U
16		

fron Founder	\$2 50
* Borchers's Metallurgy. (Hall and Hayward.)8vo,	2 50
Borchers's Metallurgy. (Hall and Hayward.)	3 00
Burgess and Le Chatelier's Measurement of High Temperatures. Third	
Edition. (In Press.) Douglas's Untechnical Addresses on Technical Subjects	1 00
Goesel's Minerals and Metals: A Reference Book16mo, mor.	1 00-
	3 00
* Iles's Lead-smelting	2 50
Steel-making Alloys and GraphiteLarge 12mo,	0 00
	3 00
Keep's Cast Iron8vo,	2 50
Metalf's Steel. A Manual for Steel-users	
minet's Production of Aluminum and its industrial Use. (Waldo.) 12mo,	2 50
Palmer's Foundry Practice. (In Press.)	0.00
* Price and Meade's Technical Analysis of Brass	2 00
* Rier's Elements of Metallography. (Mathewson.)	3 00
Smith's Materials of Machines	1 00
Tate and Stone's Foundry Practice	2 00
Indiston's Materials of Engineering. In Time Parts	8 00
Part I. Non-metallic Materials of Engineering, see Civil Engineering,	
page 9.	0 40
Part II. Iron and Steel8vo,	3 50~
Part III. A Treatise on Brasses, Bronzes, and Other Alloys and their	0 10
Constituents8vo,	2 50
Ulke's Modern Electrolytic Copper Refining8vo,	3 00
West's American Foundry Practice	2 50
Moulders' Text Book	2 50
MINERALOGY	
MINERALOGY.	
D 1 211 (01 1 1 171 1 1 (1 1 1 1 1 1 1 1 1 1 1 1 1	
Baskerville's Chemical Elements. (In Preparation.)	
* Browning's Introduction to the Rarer Elements	1 50
Brush's Manual of Determinative Mineralogy. (Penfield.)8vo,	4 00
Butler's Pocket Hand-book of Minerals	3 00.
Butler's Pocket Hand-book of Minerals	3 00· 1 00·
Butler's Pocket Hand-book of Minerals	3 00· 1 00 1 25
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00-
Butler's Pocket Hand-book of Minerals	3 00· 1 00 1 25
Butler's Pocket Hand-book of Minerals	3 00 1 00 1 25 5 00 1 00
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00- 1 00- 1 50-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25 5 00- 1 00- 1 50- 2 00-
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00- 1 00- 1 50- 2 00- 1 50-
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 4 00-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 4 00- 1 00-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 4 00-
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 12 50- 4 00- 1 00- 1 25-
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 4 00- 1 25- 3 00-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 1 50- 1 50- 4 00- 1 25- 3 00- 3 50-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 12 50- 1 00- 1 25- 3 00- 3 50- 1 25-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25- 5 00- 1 00- 1 50- 1 50- 1 50- 1 25- 3 00- 3 50- 1 25- 1 50- 1 25- 1 50-
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 4 00- 1 25- 3 00- 3 50- 1 25- 1 50- 5 00- 1 50- 1 50-
Butler's Pocket Hand-book of Minerals	3 00- 1 00 1 25- 5 00- 1 00- 1 50- 1 50- 1 50- 1 25- 3 00- 3 50- 1 25- 1 50- 1 25- 1 50-
Butler's Pocket Hand-book of Minerals	3 00-1 00-1 25-00-1 00-1 25-01 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 00-1 00-1 25-1 50-5 00-5 00-1
Butler's Pocket Hand-book of Minerals	3 00- 1 00- 1 25- 5 00- 1 00- 1 50- 2 00- 1 50- 12 50- 4 00- 1 25- 3 00- 3 50- 1 25- 1 50- 5 00- 1 50- 1 50-
Butler's Pocket Hand-book of Minerals	3 00-1 00-1 25-5 00-1 250-1 250-1 25-1 25-1 500-5 00-5 00-1
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 50-1 250-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-5 00-5 00-60-60-60-60-1 25-1 50-1 25-1 50-5 00-5 00-60-60-60-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 25-1 25-1 25-1 25-1 25-1 25-1 25
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 50-1 250-1 250-1 25-1 500-1 25-1 500-5 00-5 00-1 00-1 00-1 00-1 00-1
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 50-1 250-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-5 00-5 00-60-60-60-60-1 25-1 50-1 25-1 50-5 00-5 00-60-60-60-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 25-1 25-1 25-1 25-1 25-1 25-1 25
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 50-1 25-0 1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-1 25-1 50-5 00-5 00-1 00-1 25-1 50-5 00-5 00-1 00-1 00-1 00-1 00-1 0
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 50-1 250-1 250-1 25-1 500-1 25-1 500-5 00-5 00-1 00-1 00-1 00-1 00-1
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 50-1 250-1 250-1 25-1 500-5 00-5 00-5 00-5 00-5 00-5 00-5
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 25-0 1 25-0 1 50-5 00-5 00-5 00-1 00-1 00-1 1
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 25-1 50-2 00-1 25-1 50-5 00-5 00-5 00-1 00-2 50-1 00-2 50-1 00-2 50-1 00-2 50-1 00-2 50-1 00-2 50-1 00-2 50-1 00-1 00-2 50-1 00-1 00-1 00-1 00-1 00-1 00-1 00-1
Butler's Pocket Hand-book of Minerals	3 00-1 00 1 25-5 00-1 00-1 25-0 1 25-0 1 50-5 00-5 00-5 00-1 00-1 00-1 1

* Ries and Leighton's History of the Clay-working inqustry of the United	\$ 2 50
States	1 25 2 00 2 00
MINING.	
* Beard's Mine Gases and Explosions Large 12mo,	3 00
* Crane's Gold and Silver	5 00 4 00
* 8vo. mor.	5 00
* Ore Mining Methods8vo,	3 00
* Dana and Saunders's Rock Drilling	4 00 1 00
Eissler's Modern High Explosives8vo,	4 00
Goesel's Minerals and Metals: A Reference Book16mo, mor.	3 00
Ihlseng's Manual of Mining	$\begin{array}{cccc} 5 & 00 \\ 2 & 50 \end{array}$
* Peele's Compressed Air Plant8vo,	3 50
Riemer's Shaft Sinking Under Difficult Conditions. (Corning and Peele.)8vo,	3 00
* Peele's Compressed Air Plant	2 50
Treatise on Practical and Theoretical Mine Ventilation 12mo,	1 25
CANYMARY COURSE	
SANITARY SCIENCE.	
Association of State and National Food and Dairy Departments, Hartford	
Meeting, 1906	3 00
* Bashore's Outlines of Practical Sanitation	1 25
Sanitation of a Country House	1 00
Sanitation of Recreation Camps and Parks. 12mo, * Chapin's The Sources and Modes of Infection. Large 12mo,	1 00
Folwell's Sewerage. (Designing, Construction, and Maintenance.)8vo,	3 00
Water-supply Engineering	4 00 2 00
Fowler's Sewage Works Analyses	2 50
Water and Public Health12mo,	1 50
Gerhard's Guide to Sanitary Inspections	1 50 3 00
Sanitation of Public Buildings. 12mo, * The Water Supply, Sewerage, and Plumbing of Modern City Buildings.	1 50
	4 00
8vo, Hazen's Clean Water and How to Get It Large 12mo,	4 00
Filtration of Public Water-supplies	3 00
* Kinnicutt, Winslow and Pratt's Sewage Disposal	3 00
Control8vo,	7 50
Mason's Examination of Water. (Chemical and Bacteriological)12mo,	1 25
Water-supply. (Considered principally from a Sanitary Standpoint). 8vo,	4 00
* Mast's Light and the Behavior of OrganismsLarge 12mo,	2 50
* Merriman's Elements of Sanitary Engineering. 8vo, Ogden's Sewer Construction 8vo,	2 00 3 00
Sewer Design. 12mo,	2 00
Parsons's Disposal of Municipal Refuse	2 00
Prescott and Winslow's Elements of Water Bacteriology, with Special Reference to Sanitary Water Analysis	1 50
* Price's Handbook on Sanitation	1 50
Richards's Conservation by Sanitation8vo,	2 50
Cost of Food. A Study in Dietaries	1 00
Cost of Living as Modified by Sanitary Science	1 00
Cost of Shelter	1 00 1 50
and with the start of the puter	1 30

Richards and Woodman's Air, Water, and Food from a Sanitary Stand-		
point8vo,	\$2	00
* Richey's Plumbers', Steam-fitters', and Tinners' Edition (Building		
Mechanics' Ready Reference Series) 16mo, mor.		50
Rideal's Disinfection and the Preservation of Food8vo,	_	00
Soper's Air and Ventilation of Subways		50
Turneaure and Russell's Public Water-supplies		00
Venable's Garbage Crematories in America		00
Method and Devices for Bacterial Treatment of Sewage 8vo,	3	00
Ward and Whipple's Freshwater Biology. (In Press.)		F0
Whipple's Microscopy of Drinking-water8vo,		50
* Typhoid Fever		00
Value of Pure WaterLarge 12mo,	-	00
Winslow's Systematic Relationship of the CoccaceæLarge 12mo,	Z	50
TELOGRA I AMBOLIO		
MISCELLANEOUS.		
* Burt's Railway Station Service	9	00
* Chapin's How to Enamel		00
Emmons's Geological Guide-book of the Rocky Mountain Excursion of the	-	00
International Congress of GeologistsLarge 8vo,	1	50
Ferrel's Popular Treatise on the Winds8vo,	_	00
Fitzgerald's Boston Machinist	_	00
* Fritz, Autobiography of John8vo,	_	-
Gannett's Statistical Abstract of the World24mo,		75
Haines's American Railway Management	2	50
Hanausek's The Microscopy of Technical Products. (Winton)8vo.		00
Jacobs's Betterment Briefs. A Collection of Published Papers on Or-	Ū	
ganized Industrial Efficiency	3	50
Metcalfe's Cost of Manufactures, and the Administration of Workshops8vo.		00
Putnam's Nautical Charts	2	00
Ricketts's History of Rensselaer Polytechnic Institute 1824-1894.		
Large 12mo.	3	00
* Rotch and Palmer's Charts of the Atmosphere for Aeronauts and Aviators.		
Oblong 4to,	2	00
Rotherham's Emphasised New TestamentLarge 8vo,	2	00
Rust's Ex-Meridian Altitude, Azimuth and Star-finding Tables8vo	5	00
Standage's Decoration of Wood, Glass, Metal, etc	2	00
Thome's Structural and Physiological Botany. (Bennett)16mo,		25
Westermaier's Compendium of General Botany. (Schneider)8vo,	2	00
Winslow's Elements of Applied Microscopy12mo,	1	50
HEBREW AND CHALDEE TEXT-BOOKS.		
Gesenius's Hebrew and Chaldee Lexicon to the Old Testament Scriptures.		
(Tregelles.)		00
Green's Elementary Hebrew Grammar		25
Green's Exementary Repress Grammar	1	20







fort soz

UNIVERSITY OF CALIFORNIA LIBRARY BERKELEY

Return to desk from which borrowed.

This book is DUE on the last date stamped below.

9May52HW	Timo of
3 Nay 5 2 Lu 3 Jan' 5 5 Kwy	Hawan INTER-LIBRARY LBAN
DEC 16 1958 10 DET 1 2 1955 18 13 May'56PL 5	INTERLIBRARY LOAN FEB 2 7 1930 UNIV. OF .CALIF., BERK. 1261

258715 D_F820

